

Cr and Plants

Subjects: Environmental Sciences

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Definition

Extensive industrial activities resulted in an increase in chromium (Cr) contamination in the environment. The toxicity of Cr severely affects plant growth and development. Cr is also recognized as a human carcinogen that enters the human body via inhalation or by consuming Cr-contaminated food products.

1. Introduction

Heavy metal contamination is becoming a serious environmental issue worldwide for the past few decades due to their increased concentration beyond the permissible limit. Chromium (Cr) is a naturally occurring heavy metal and the 17th most abundant element in the earth's mantle ^[1]. Although Cr is required in trace amounts in plants and animals, at higher concentrations it serves as a major contaminant to the environment. Natural sources, as well as various anthropogenic activities, are responsible for the release of Cr in the soil, air, and water which have ultimately led to Cr pollution globally ^[2]. Regarding the consequences of the contamination of food lands and the drinking system, Cr easily enters into the food chain and affects the health of all life forms directly or indirectly ^{[3][4][5]}.

In plants, the toxic effects of Cr are also evident, showing symptoms such as delay in seed germination, damaged roots and reduction of root growth, reduced biomass, reduction in plant height, photosynthetic impairment, membrane damage, leaf chlorosis, necrosis, low grain yield and eventually plant death ^[6]. Cr is a fairly active metal and reacts easily with environmental oxygen. Different oxidation states of Cr have been reported ranging from 0 to +6. The trivalent Cr(III) and hexavalent Cr(VI) are the most stable forms of Cr in nature. Also, Cr(VI) shows higher toxicity than Cr(III) due to its higher solubility and mobility in the water system ^[1]. Both valence states of Cr i.e., Cr(III) and Cr(VI) are taken up by plants ^[7]. Cr(VI) is actively taken up into the plant cells by sulfate carriers ^[8]. On the other hand, Cr(III) enters passively by the cation exchange sites of the plant cell walls ^[9]. Moreover, the carboxylic acids present in the root exudates facilitate the Cr solubilization, and thus its uptake into the plants ^[10].

Phytoremediation is a rapidly growing field of research for heavy metal contaminated regions. There are various processes of phytoremediation such as phytovolatilization, phytoextraction, phytostabilization, and hyperaccumulation ^{[11][12]}. Numerous research studies have shown that many plant species are capable of effectively removing Cr from contaminated regions which could be useful for the phytoremediation process ^[13]. Cr hyperaccumulators, with their associated microflora, have been used around the industrial effluent sites to remove the excess toxic Cr, as well as organic matters. Plant-microbe interaction is also one of the efficient strategies for Cr detoxification due to its high efficiency, low cost, and eco-friendly nature ^{[14][15]}. At the molecular level, various pathways involved in Cr detoxification have been deciphered to provide tolerance in response to Cr toxicity. Cr stress activates Reactive Oxygen Species (ROS) signaling, antioxidant responses, defense proteins such as phytochelatins (PCs), metallothioneine (MTs), and glutathione-S-transferases (GSTs) followed by phytosequestration and compartmentalization thereby accelerating the bio-accumulating potential of the plants ^{[16][17][18]}. One approach could be developing transgenics by upregulating genes responsible for Cr uptake, transport, and sequestration to enhance the tolerance and accumulation rate of the plant.

Taking all into consideration, this review addresses Cr sources, effects, uptake, translocation, and subcellular distribution in plants. We also discussed Cr detoxification remedies in plants through phytoremediation and biotechnological approaches. Also, the molecular events underlying Cr toxicity and

the defense signal transduction have been discussed. Overall, the review summarizes the recent development of sustainable approaches for Cr detoxification in the environment.

2. Cr Occurrence and Sources

The distribution of Cr(III) and Cr(VI) containing compounds in the environment depends on the presence of oxidizing or reducing compounds, redox potential, the formation of Cr(III) complexes or insoluble Cr(III) salts, the kinetics of the redox reactions, pH, and the total chromium concentration [2]. In the environment, Cr(VI) occurs mostly as chromate ions (Cr_2O_4 and Cr_2O_7) or chromic acid (H_2CrO_4), whereas Cr(III) occurs in the form of oxides, hydroxides ($\text{Cr}(\text{OH})_n^{(3-n)+}$), and sulfates [2][9]. The chemical structure of various existing forms of Cr(VI) and Cr(III) in the environment has been given in Figure 1. Cr is also found in the air where it occurs in the form of aerosols. Cr(VI) has been reported to be 0.01–30% of the total Cr present in the air. Tobacco smokes result in a total Cr level of around 1000 ng/m³ in indoor air, i.e., 10–400 times higher than outdoor levels [2]. The average level of Cr in surface water, seawater, and rainwater is around 0.001–0.010 mg/L, 0.00004–0.0005 mg/L, and 0.0002–0.001 mg/L, respectively [2][5].

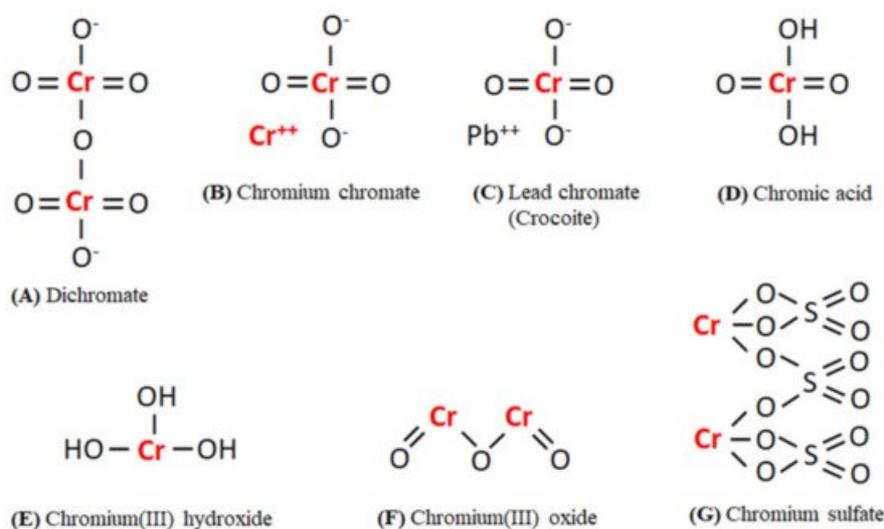


Figure 1. The chemical structures of existing forms of Cr(VI) (A–D) and Cr(III) (E–G) in the environment. Abbreviations: Cr, Chromium; O, Oxygen; H, Hydrogen; Pb, Lead; S, Sulphur.

Both natural, as well as anthropogenic sources, contribute to total Cr toxicity in the environment. Mineral leaching accounts for the natural origin of Cr in groundwater that is dominated by Cr(VI). However, above 70% of total Cr in the environment is due to the anthropogenic pollutants from effluent streams from paper and pulp mills, non-ferrous base metal smelters, leather tanning industries, refineries, releases from thermal generating stations, and urban stormwater runoff [2]. Among all sources, tanning industries play a vital role as they use $\text{Cr}_2(\text{SO}_4)_3$ as a tanning agent, of which 30–40% is unused and discharged into the environment via tannery effluent [19]. This inefficient usage of Cr brings about water defilement, carrying 500–1000 mg/L Cr from the current high-exhaust chrome tanning approaches and as high as 1500–3000 mg/L Cr from the conventional methodologies [20].

In nature, Cr(III) predominates in soil and occurs in small amounts in rocks. Worldwide, the average concentration of Cr in the soil is dependent on the bedrock and range between 10–100 mg/kg with an average concentration of 60 mg/kg [21][22]. The estimated acceptable level of Cr in the soil for environmental safety and human health is 64 mg/kg that is difficult to maintain because of the discharge of the anthropogenic pollutants [23]. Cr(III) is present in the form of chromite [$\text{FeCr}(\text{III})_2\text{O}_4$], bentorite $\text{Ca}_6(\text{Cr},\text{Al})_2(\text{SO}_4)_3$, and vauquelinite ($\text{CuPb}_2\text{CrO}_4\text{PO}_4\text{OH}$) in rocks from where it can be released into groundwater via weathering and erosion [1]. However, Cr(VI) seldom occurs naturally and is released from anthropogenic sources. The only natural source of Cr(VI) is an uncommon mineral, crocoite (PbCrO_4). High Cr(VI) concentration in California (0.006–0.036 mg/L in aquifer) [24], Italy (0.005–0.073 mg/L in groundwater) [25], Mexico (270–4120 mg/kg in rocks) [26], Zimbabwe (310–8600 mg/kg in soil) [27], Brazil (0.0025–0.11 mg/L in groundwater) [28], Indonesia (50–90 mg/kg in river sediment) and Japan (510–1420

mg/kg in river sediment) [29] were reported owing to the contact of water with ultramafic rocks and soil such as ophiolites, dunites, and serpentinite. Cr emitted into the air through sources such as Cr-based automotive catalytic convertors, tobacco smoke, and coal combustion also end up in soils [9].

Food is also a source of Cr exposure to the living population, containing a total Cr level spanning between <0.0005 to 1.3 mg/kg. However, fresh food is reported to be very low in Cr i.e., 0.02–0.05 µg/kg. As the food is recognized as a reducing medium, the total Cr found in it can be classed as Cr(III). Food items such as meat, seafood, fish, cereal products, black pepper, tea, cheese, some fruits, vegetables, and wheat germ have high Cr content (>0.1 mg/kg). Beer, spirits, and wine have total Cr concentrations of around 0.45, 0.135, and 0.30 mg/L respectively. Utensils, especially made up of stainless steel, can also add to total Cr concentrations in food [2].

3. The Effects of Cr on Plants

Oxidation state is the deciding factor of the toxicologic or physiologic effects of Cr. Higher concentrations of Cr(III) might be inhibitory to plant growth and development. On the contrary, Cr(VI) is highly toxic for plants and inhibits various morphological, physiological, and metabolic activities in plants, and may even lead to their complete damage [30][31].

3.1. Effects on Seed Germination

Seed germination in a Cr enriched environment depends on the plant's ability to withstand Cr toxicity. Excess Cr limits the seed germination rate of *Cucumis melo* L. (>10 mg/L Cr(III) in culture medium) [32], *Hibiscus esculentus* (>50 mg/kg Cr(VI) in soil) [6], *Triticum aestivum* (>25 mg/L Cr(VI) in the nutrient solution) [33], and *Echinochola colona* (2.5 mg/L Cr(VI) in the nutrient solution) [34]. Two freshwater plants, *Lemna minor*, and *Pistia stratiotes* when grown in nutrient solution containing 1, 5, or 10 mg/L of Cr(VI) resulted in retardation of the growth [35]. The seed germination of *Salvia sclarea* L. was inhibited when treated with different concentrations of Cr(VI) ranging from 1–10 mg/L [36].

3.2. Effects on Shoot Growth

Cr exposure affects the shoot length and biomass of the plant [37]. With the increase in Cr(VI) concentration, the shoot length of *Helianthus annuus* L. was found to be decreased [38]. In *Allium cepa*, Cr(III) concentration higher than 100 mg/L, reduced the shoot growth [39]. In *Citrus aurantium* L., an increase in Cr(III) concentration in the soil resulted in 39.3% and 90.4% reduced shoot length at 50 mg/kg and 200 mg/kg of Cr, respectively [40]. The small stem with a slow growth rate was observed in *Camellia sinensis* on exposure to 600 mg/kg Cr(III) [41]. It was reported that a lower Cr(VI) concentration (0.05 mg/L) increases the shoot length, while a higher Cr(VI) concentration (1 mg/L) retarded the shoot length and weight of *Myriophyllum spicatum* [30].

3.3. Effects on Root Growth

Being a primary organ for nutrient uptake, the roots are directly associated with Cr uptake and thus serve as a principal site of Cr toxicity in plants. In a greenhouse experiment, root growth of sour orange seedlings was analyzed under 50–200 mg/kg Cr(III) concentrations and the significant reduction in the root length was found at 200 mg/kg Cr(III) [40]. In *Pistia stratiotes*, Cr at low concentration (0.25 mg/L) promotes root length, laminal length, and breadth as compared to control but at higher concentration (2.5 mg/L), the root length was found to be decreased [42]. *C. sinensis* roots were severely affected on high Cr concentration (600 mg/kg) leading to a reduction in roots dry weight [41]. Cr(VI) at a concentration of 6, 12, 18, and 24 mg/kg caused a reduction in root length and root dry weight in *T. aestivum* [43]. Sundaramoorthy et al. [44] reported decreased root growth of field grown *Oryza sativa* L. treated with 200 mg/L Cr(VI) in distilled water. Also, the thin and brittle roots of *Pisum sativum* were observed under high Cr(VI) concentration (>1000 mg/L) [45].

3.4. Effects on Total Leaf Area

Plant takes up Cr via the root and it gets transported to the upper plant parts via various transporters. The leaf is one of the important organs of a plant that performs photosynthesis. The total leaf area is one of the determining factors for photosynthesis. The Cr(VI) toxicity affects the total leaf area and also showed a 50% reduction in leaf number per plant in *O. sativa* [44]. *Brassica oleracea* showed a decrease in leaf size, wilting, and chlorosis when grown in refined sand with complete nutritional media under 0.5 mM Cr(III) toxicity [46]. In *Lolium perenne* L., a noticeable wilting has been observed under 0.50mM Cr(VI) prepared in nutrient media [47]. Leaf chlorosis and leaf necrosis have been reported in *Saccharum officinarum* on Cr(VI) exposure of 40 mg/kg and 80 mg/kg, respectively [48]. *Phaseolus vulgaris* showed a reduction in leaf biomass under 0.01 mM Cr(III) treatment in nutrient solution [49]. The decreased number of leaves was reported in *Prosopis laevigata* under 3.4 mM Cr(VI) toxicity in nutrient media [50].

3.5. Effects on Grain Yield

The yield and productivity of crops are affected by Cr exposure as it exerts an adverse impact on biochemical and physiological processes. The reduction in yield of *Hordeum vulgare* and *Zea mays* was observed at a concentration of 100 or 300 mg/kg of Cr [51]. In *Daucus carota*, no harvestable yield was obtained on 270 or 810 kg/ha of Cr(VI) application [52]. In *O. sativa*, Cr(VI) application under 200 mg/L caused a decrease in grain weight and 80% loss [44].

3.6. Anatomical Changes

Cr induces structural and ultrastructural alterations in plant organs. In *Vigna radiata*, Cr exposure led to change in the epidermis, cortex, and stele in its stem [53]. The addition of 0.60 mM Cr(VI) in nutrient solution media caused less wax deposition and wide stomatal opening in the leaves of *Phyllanthus amarus* [54]. The number of palisade and spongy parenchyma cells were found to be decreased in fronds of *Pteris vittata* at 500 mg/kg of Cr(VI) and 1000 mg/kg of Cr(III) [55]. The roots of *Scirpus lacustris* L. showed an increase in pith and cortical tissue layer proportion under two different concentrations of Cr(IV) i.e., 4 and 8 mg/L [56]. *Mentha aquatica* roots showed structural changes such as damaged root cap, loss of root hairs, inhibition in lateral roots formation on exposure of Cr(VI) at a concentration of 20 or 40 mg/L [57].

3.7. Physiological Changes

Exposure of the elevated concentration of heavy metals leads to degradation of photosynthetic pigments that are responsible for deficiency in light-harvesting capacity. The 0.05 mg/L of Cr(VI) exposure resulted in a reduction in photosynthetic rate in *M. spicatum* [58]. Some species like *Citrus aurantium* [40], *Najas indica*, *Vallisneria spiralis*, and *Alternanthera sessilis* showed decrease in chlorophyll (Chl) content under Cr toxicity [30]. In *T. aestivum*, net photosynthetic rate (Pn) was reduced with a gradual increase in Cr exposure time. Less chlorophyll content was found in Cr(VI) treated wheat in comparison to control plants. Interestingly, it was due to the more reduction in Chl b than Chl a content [59]. However, in the case of *Chlorella pyrenoidosa*, both Chl a and b content were decreased under 0.1–50 mg/L of Cr(VI) [60]. The light-harvesting complex of the photosystem II was extensively affected in *T. aestivum* after treatment with 0.10 mM, 0.20 mM, and 0.30 mM of Cr(VI) [61].

In higher plants, Cr inhibits mitochondrial electron transport leading to higher ROS generation that causes oxidative stress, pigment, and chloroplast alterations [62]. Chromosomal impairment by Cr(VI) treatments (0.01 mM and 0.1 mM) has been reported in *Amaranthus viridis* plant tissues, where it regulated the activity of calmodulin that was further responsible for the activation of many key enzymes such as phospholipase and nicotinamide adenine dinucleotide kinase, involved in the chromosomal movement [63]. Furthermore, Cr has also been observed to affect the concentrations of free polyamines in *Avena sativa*, *Brassica napus*, and *H. vulgare* seedlings after treatment with 100 mg/L of Cr(III) for 1–14 days [62][64].

Wilting caused by Cr has been reported in various crops and plant species, but the effect of Cr on water relations of higher plants is less studied. Chatterjee and Chatterjee reported a decrease in water potential, transcription rate, and relative water content in *B. oleracea* leaves, grown in refined sand in a glass house with 0.5 mM of Cr(VI) toxicity [46]. In *Spinacea oleracea* leaves, the decrease in water potential and increase in diffusive resistance led to a decrease in physiological availability of water under 0.10 and 0.40 mM of Cr(VI) toxicity [65]. Essential nutrients uptake was reduced in *Spartina argentinensis* on Cr(III) exposure at 1500 mg/kg [66]. Longitudinal water movement was found to be reduced in beans due to a decrease in tracheary vessel diameter on high Cr(VI) level (0.096 mM) [67].

3.8. Effects on Nutrient Balance

The soil-plant transfer index of Cr(VI) is comparatively higher than Cr(III) due to its better adsorption and high solubility in the cells [68]. Cr(VI) modulates the intracellular concentration of essential nutrients viz phosphorus (P), calcium (Ca), manganese (Mn), magnesium (Mg), potassium (K), and iron (Fe) in plants [69]. It has also been reported that mycorrhizal fungi as well as organic acid such as citric acid increases the uptake of Cr(VI) in plants [70]. The field-grown *O. sativa* irrigated with different concentrations of Cr(VI), (50–500 mg/kg) showed a gradual decrease in uptake of macronutrients (K, P, and nitrogen (N)) and micronutrients manganese (Mn), zinc (Zn), copper (Cu) with increased Cr concentration [44][71]. In contrast, Mg concentration increased in nucleus and mitochondria with an increase in Cr(VI) concentration in *O. sativa*, showing a positive correlation [72]. Dube et al. [73] demonstrated that Cr(VI) exposure increased the accumulation of P, Mn, and Zn while decreased sulfur (S) and Cu content in *Citrullus vulgaris*. *Zea mays* roots showed a decrease in Cu absorption on Cr(VI) exposure [74]. In *B. oleracea*, a high concentration of Cr(VI) (0.5 mM) affected the Fe concentration and translocation of Zn, Cu, S, P, Mn from the roots to other plant parts [46].

3.9. Molecular Changes

The molecular changes after the Cr stress were explored through comparative transcriptome analyses in several plants to get insight of the underlying mechanism during Cr stress. Previous reports showed that the application of Cr modulates several biological processes to lessen the phytotoxicity of Cr stress. A recent microarray analysis on Cr(III) and Cr(VI) treated *O. sativa* seedlings showed that Cr stress induces the transcript levels of different antioxidant enzymes ascorbate peroxidase (APX), superoxide dismutase (SOD), peroxidase (POD), and glutathione peroxidase (GPX) coding genes in treated plants in comparison to control plants [75]. Likewise, Dubey et al. [17] observed the differential expression pattern of numerous genes (1138 up-regulated and 1610 down-regulated) in *O. sativa* roots exposed with 0.10 mM Cr(VI). Among all the up-regulated genes, most of the genes were related to secondary metabolites biosynthesis, transporters, and xenobiotics biodegradation. Interestingly, this study also showed the modulation of the sucrose degradation pathway which could be salvage machinery in response to Cr stress. On the other hand, Cr stress also stimulates the expression of microRNAs. Some Cr(VI)-responsive miRNAs have been identified in *Raphanus sativus* and *O. sativa* [76][77].

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Keywords

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