Ocotillol-Type Triterpenoids

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Ginseng is one of the most widely consumed herbs in the world and plays an important role in counteracting fatigue and alleviating stress. The main active substances of ginseng are its ginsenosides. Ocotillol-type triterpenoid is a remarkably effective ginsenoside from Vietnamese ginseng that has received attention because of its potential antibacterial, anticancer and anti-inflammatory properties, among others. The semisynthesis, modification and biological activities of ocotillol-type compounds have been extensively studied in recent years.

Keywords: biological activity ; modification ; semisynthetic ; ocotillol-type triterpenoids

1. Introduction

From the 1940s to the middle of 2019, approximately 33.5% of approved drugs were either natural products or directly derived from them ^[1]. The development of new drug entities based on natural products as sources of novel structures is still an area of active research. Ginseng, including Asian ginseng (*Panax vietnamensis* HA et GRUSHV.) and American ginseng (*Panax quinquefolium* L.), is one of the most widely consumed herbs in the world and plays an important role in counteracting fatigue and alleviating stress ^{[2][3]}. Ginseng contains a variety of active ingredients, but its main active substances are attributed to its ginsenosides. The ginsenosides with the highest content in Vietnamese ginseng are protopanaxadiol, protopanaxatriol, oleanolic acid and 20,24-epoxydammarane (ocotillol) (Figure 1) ^{[4][5]}.

Among the active components of ginseng, ocotillol-type compounds have received increasing attention because of their antibacterial, anticancer and anti-inflammatory properties ^{[G][Z]}. Their different pharmacologic effects and potential molecular mechanisms have been gradually elucidated. Compared with the structure of dammarane ginsenosides (including the protopanaxadiol and protopanaxatriol types), ocotillol-type saponins are tetracyclic triterpenoid saponins containing a furan ring linked with aglycones.

Ocotillol-type saponins were first isolated from *Fozrqwieria splendens* Eliselm. in 1965 by Warnhoff et al. They were also found in *Panax quinquefolium* L, *Panax vietnamensis* HA et GRUSHV, and *Panax japonicus* var, to name a few ^{[8][9][10][11]} ^{[12][13][14][15][16][17]}. However, because of the low content of ocotillol-type saponins in natural products, there were few studies on ocotillol-type derivatives in previous years ^{[18][19]} Fueled by the growing use of semisynthetic methods for the preparation of ocotillol-type derivatives, increased research of ocotillol-type derivatives has been recently observed. In 2016, Liu et al. published a review that focused on the discovery, semisynthesis, biological activities and metabolism of ocotillol-type saponins ^[6]. However, the structure of most derivatives and its structure–activity relationship (SAR) were not mentioned in the article. Compared with the previous review, this review summarized the semisynthesis, modification and pharmacological activities of ocotillol-type derivatives. All the structures of ocotillol-type derivatives and their SARs in antibacterial, anti-inflammatory and tumor multidrug resistance reversal were summarized. This review provides useful information for the development of ocotillol-type derivatives and gives a direction for further inspiration to enrich its structures with good pharmacological activities.



Figure 1. Ginsenosides are found in the highest abundance in Vietnamese ginseng.

2. Semisynthesis of Ocotillol-Type Compounds

Ocotillol-type sapogenins are less abundant in natural sources. Vietnamese ginseng contains higher amounts of ginseng saponins compared with other *Panax* genus species. The content of ocotillol-type saponins in *Panax Vietnamese* ginsengs is only 5.6%, while in *Panax quinquefolius*, it is less than 0.01% ^[20]. Additionally, 1 kg of fresh rhizome low-quality Vietnamese ginseng is about \$1000 in 2019. These factors may have led to the slow development of ocotillol-type ginsenosides in previous years.

In 2005, 20(*S*)-protopanaxadiol (20(*S*)-PPD) was used as a raw material to obtain **4** and **5** by a semisynthetic method $\frac{[21]}{2}$. Yang et al. optimized and improved the synthetic process and achieved the industrial production of **4** and **5** $\frac{[22]}{2}$.

Ocotillol-type sapogenins have been made using similar synthetic methods. 20(*S*)-PPD was used as the raw starting material and reacted with acetic anhydride, and then acetylated 20(*S*)-PPD was oxidized by *m*-CPBA. The molar ratio of the acetylated 20(*S*)-PPD to *m*-CPBA at -3 °C is approximately 1:4, 3 h. The ocotillol-type epimers (**4**, **5**) were obtained by the hydrolysis of the oxidation products. The synthetic route is shown in Figure 2A ^[23].

After further research by Meng et al., the synthesis mechanism of ocotillol-type epimers was proposed as follows (Figure 2B). 20(S)-PPD or 20(R)-PPD is oxidized by *m*-CPBA to generate the 24,25-epoxy intermediates, and then an intramolecular ring-opening loop reaction is carried out according to Baldwin's rule, and finally cyclization by a 5-exo-tet method forms a tetrahydrofuran ring [24][25][26].

Further research proved that the epimerization of C-24 could also result in remarkable differences in both the molecular conformation and the crystal packing arrangements. These remarkable differences may lead to diversity in both polarity and activity of the ocotillol-type epimers. The 24(*S*)-epimer (**5**) had two intramolecular hydrogen bonds, while the 24(*R*)-epimer (**4**) had one intramolecular hydrogen bond (Figure 3A,B) ^{[27][28]}. Crystal stacking showed that both the 20(*S*),24(*R*)-ocotillol and 20(*S*),24(*S*)-ocotillol generated an H-bonded tube, the 24(*R*)-epimer (**4**) generated a left-handed chiral channel, while the 24(*S*)-epimer (**5**) extended into the two-dimensional network with right-handed and left-handed chiral channels (Figure 3C–E) ^[29]. Additionally, the 24(*R*)-epimer (**4**) had weaker molecular polarity compared with the 24(*S*)-epimer (**5**). These differences in hydrogen bonding may contribute to the differences in the observed biological activity and molecular polarity of the 24(*R*)-epimer (**4**) compared with the 24(*S*)-epimer (**5**).

Ocotillol-type ginsenosides are rarely found in nature. Less than 20 naturally occurring ocotillol-type ginsenosides have been characterized and reported. The use of chemical methods to synthesize new ocotillol-type ginsenosides is a promising approach to generate structural diversity. Atopkina et al. reported the synthesis of ocotillol-type ginsenosides by coupling the acceptor **4** with α -acetobromoglucose and orthoester donors in the presence of mercury salts (Figure 2C) ^[30] ^[31]. In 2016, Shen et al. used a gold-catalyzed glycosylation scheme to synthesize ocotillol-type ginsenosides under neutral conditions (Figure 2C) ^[32]. Many ocotillol-type ginsenosides can be synthesized by this method, and further investigations of ocotillol-type ginsenosides should be pursued.



Figure 2. (A) Semisynthetic route for the preparation of ocotillol-type epimers. (B) Synthetic mechanism of ocotillol-type epimers. (C) Synthesis of ocotillol-type ginsenosides. Ac = acetyl; Bn = benzyl; Glc = β -d-glucopyranosyl; Bz = benzoyl. Reagents and conditions: (a) (CH₃CO)₂O, DMAP, pyridine, r.t.; (b) *m*-CPBA, CH₂Cl₂, r.t.; (c) NaOH, CH₃OH, H₂O, 65 °C; (d) alcohol, mercury cyanide, nitromethane, 90 °C, 1 h; (e) alcohol, *α*-acetobromoglucose, r.t.; (f) KOH/CH₃OH, THF, r.t.; (g) Ph₃PAuNTf₂, CH₂Cl₂, r.t.; (h) H₂, Pd(OH)₂/C, CH₃OH, r.t.



Figure 3. (**A**) The Oak Ridge Thermal Ellipsoid Plot Program for Crystal Structure Illustrations (ORTEP) figure of 20(S), 24(R)-ocotillol-type saponin (**4**). (**B**) The ORTEP figure of 20(S), 24(S)-ocotillol-type saponin (**5**). Thermal ellipsoids shown at 30% probability. (**C**) and (**D**) view of the H-bonded 1D left-handed chiral channel in 20(S), 24(R)-ocotillol-type saponin (**4**). (**E**) The 2D net with right-handed and left-handed chiral channels in 20(S), 24(S)-ocotillol-type saponin (**5**).

3. Pharmacological Activities and Chemistry

3.1. Antibacterial Effects

Evidence has shown that ginseng has antibacterial properties, and its extract may be effective for treating bacterial infections in the future ^[33]. Compound **5** had strong antibacterial activities against *Staphylococcus aureus* (*S. aureus*) and *Bacillus subtilis* (*B. subtilis*) with minimum inhibitory concentration (MIC) values of 8 μ g/mL ^[34]. Further research showed that **5** also had strong synergistic inhibition against community-associated methicillin-resistant *S. aureus* (MRSA; strain USA300), as **5** reduced the MIC of kanamycin (KAN) against MRSA USA300 from 1 μ g/mL to 0.125 μ g/mL giving a fractional inhibitory concentration index (FICI) of 0.14.

The furan ring, C-3 and C-12 are possible to explore in terms of chemical diversity as a modification of the furan ring, C-3, and C-12 significantly changed the antibacterial activity of ocotillol-type derivatives. Aromatic-substituted ocotillol-type derivatives **6–17** were synthesized by an esterification reaction, and their in vitro activity against *Escherichia coli (E. coli)*, *B. subtilis*, *S. aureus*, *Pseudomonas aeruginosa (P. aeruginosa)* and *Acinetobacter baumannii (A. baumannii)* was determined (Figure 4) ^[35]. Compounds **6** and **7** exhibited excellent antibacterial activities with MIC values of 1 µg/mL against *S. aureus* and *B. subtilis*, while compounds **9**, **10**, **12** and **16** exhibited moderate antibacterial activities against *S. aureus*. Further research showed that **6** and **7** displayed good antibacterial activities against MRSA USA300 with MIC values of 4 µg/mL. Additionally, **6** and **7** combined with KAN and chloramphenicol had strong synergistic inhibition against MRSA USA300 and reduced the MICs of KAN against MRSA USA300 from 1 µg/mL to 0.0156 and 0.0625 µg/mL (FICI = 0.078 and 0.020, respectively).



Figure 4. Synthesis of ocotillol-type derivatives 6–61. Reagents and conditions: (a) anhydrous CH₂Cl₂, anhydride or acids or Boc-amino acid, 1-ethyl-3(3-dimethylpropylamine) carbodiimide (EDCI), 4-dimethylaminopyridine (DMAP), r.t.; (b) trifluoroacetic acid (TFA), CH₂Cl₂, r.t.; (c) anhydrous pyridine, Ac₂O, DMAP, r.t.; (d) anhydrous pyridine, anhydride or acid chloride, DMAP, ref.; (e) CH₃OH, KOH, ref.

Bi et al. synthesized aliphatic ocotillol-type derivatives **18–33** (Figure 4) ^{[36][37][38]}. Compounds **18**, **20–23**, **25** and **30** showed good antibacterial activities against *S. aureus* and *B. subtilis*. Further screening results showed that **5**, **18** and **19** had similar antibacterial activities against MRSA USA300 with MIC values of 8 μ g/mL. Most ocotillol-type derivatives with an amino group at C-3 displayed excellent antibacterial activities, while those with a carboxylic group at C-3 showed moderate activities. A synergistic effect was observed for compound **19** as it reduced the MIC of KAN against MRSA USA300 from 1 μ g/mL to 0.25 μ g/mL with a FICI of 0.28.

A series of ocotillol-type derivatives **34–55** with an amino group was also synthesized (Figure 4) ^{[39][40]}. The antibacterial activity results showed that most of the ocotillol-type derivatives with an amino group had moderate to good inhibitory activities against Gram-positive bacteria but had no effect on Gram-negative bacteria. Compounds **38**, **40** and **51** had good inhibitory activities against MRSA USA 300 with MICs \leq 4 µg/mL, while **51** had the same antibacterial activity as KAN. A synergistic effect was observed for **39** when it was combined with KAN as shown by the significant enhancement of the MIC from 4 µg/mL to 0.25 µg/mL (FICI < 0.0088) against MRSA USA300.

A series of derivatives **57–61** were synthesized and screened. Among them, compound **58** had the best antibacterial activity against MRSA USA300 with a MIC of 8 µg/mL, and **60** had a moderate inhibitory effect against both Gram-positive and Gram-negative bacteria (Figure 4). Additionally, **58** combined with KAN had strong synergistic inhibition against MRSA USA300 with a FICI of 0.008.

The synthetic approaches to prepare compounds **6–56B** are only slightly different. Compound **6** was synthesized by the treatment of compound **4**, DMAP and phthalic anhydride in dry dichloromethane over 6 h to obtain **6** with 73% yield at room temperature. 1-ethyl-3(3-dimethylpropylamine) carbodiimide (EDCI) is an excellent dehydrating agent that can accelerate the esterification reaction. Compound **20** was synthesized by the treatment of **4**, DMAP, *N*-Boc-isonipecotic acid and EDCI in dry dichloromethane over 3 h to obtain the intermediate with 80% yield at room temperature. The use of EDCI can increase the speed and yield of the esterification reaction. It is noteworthy that the hydroxyl group at the C-12 does not easily react with anhydride or acid because of steric hindrance and the formation of hydrogens bond. After the addition of **56A**, DMAP and phthalic anhydride to anhydrous pyridine at 120 °C for 25 h, the yield of the intermediate is only 50%.

Ocotillol ketone derivatives **62–69** were synthesized by Zhou et al. (Figure 5). Compound **4** (0.21 mmol) in dry dichloromethane (8 mL) was added to pyridinium chlorochromate (0.40 mmol), and the mixture was stirred at room temperature for 3 h to obtain compound **62** with 66% yield. While compound **64** was synthesized by combining **4** (0.33 mmol) and pyridinium chlorochromate (1.00 mmol) in dry dichloromethane (8 mL), the reaction takes about 8 h to obtain intermediate with 76% yield at room temperature. Compound **65** had excellent antibacterial activities against *S. aureus* with a MIC of 16 μ g/mL, while compounds **67** and **69** had moderate inhibitory effects against *S. aureus*.



62, 64, 66, 68 24(R) 63, 65, 67, 69 24(S)

Figure 5. Synthesis of ocotillol-type derivatives **62–69**. Reagents and conditions: (a) anhydrous pyridine, Ac₂O, DMAP, r.t.; (b) anhydrous CH_2Cl_2 , pyridinium chlorochromate (PCC), r.t.; (c) CH_3OH , KOH, ref.; (d) anhydrous pyridine, NH₂OH·HCl, 80 °C.

Ocotillol-type derivatives with a nitric oxide (NO) donor **70–91** were synthesized, their NO release ability and the antibacterial abilities of some derivatives were studied (Figure 6) ^[41]. Compounds **70–91** showed similar NO-releasing capability at 100 μ M. Compounds **71**, **75**, **77**, **83**, **84**, **86**, and **91** showed better NO-releasing capability at 500 μ M as after 30 min of reaction, they all released more than 0.2 M NO. Compounds **83** and **86** demonstrated good activities against Gram-positive bacteria (MIC = 16 μ g/mL against *B. subtilis 168* and *S. aureus*). Compound **86** displayed broad-spectrum activity against Gram-positive and Gram-negative bacteria. Compound **86** used with chloramphenicol also showed good synergistic effects with a FICI = 0.03 against MRSA USA300.

A series of ocotillol-type lactone derivatives **92–108** was designed by Zhang et al. (Figure 7) ^[42]. Compounds **96–98**, **100** and **102** demonstrated good activities against *S. aureus* and *B. subtilis* with MIC values of 1 to 8 μg/mL. Compounds **96**, **100**, **101**, **102**, **105** and **107** showed good activities against MRSA USA 300. Compounds **96** and **102** also exhibited

bactericidal activities with minimum bactericidal concentration (MBC) values of 4 and 8 µg/mL. Additionally, **102** reduced the MICs of KAN and chloramphenicol against MRSA USA300 from 1 and 4 µg/mL to 0.125 and 1 µg/mL (FICI = 0.141 and 0.375), respectively. Zhang et al. also analyzed the antibacterial effect of the ocotillol-type lactone derivative **102** by scanning electron microscopy, a cytoplasmic β -galactosidase leakage assay and UV-visible analysis ^[42]. The results showed that **102** might exert its antibacterial effect by damaging bacterial cell membranes and disrupting the function of DNA. The precise mechanism of its DNA antibacterial action is currently under investigation.



Figure 6. Synthesis of ocotillol-type derivatives **70–91.** Reagents and conditions: (a) NaNO₂, HOAc, r.t., 1 h; (b) SOCl₂, pyridine, CH₂Cl₂, r.t., 8 h; (c) HOR₁OH, K₂CO₃, KI, CH₃CN, r.t., 3 h; (d) 1 M NaOH, ethyl chlorocarbonate, -5 °C; (e) Acetic anhydride, nitrosonitric acid, CH₂Cl₂, 0 °C; (f) cholamine, ethyl alcohol, r.t.; (g) K₂CO₃, 1,2-dibromoethane, THF, r.t.; (h) AgNO₃, CH₃CN, 70 °C, protection from light; (i) succinic anhydride, DMAP, CHCl₃, 42 °C, 6 h; (j) **A3-6**, **A10**, **A12**, DMAP, EDCl, 25 °C, CH₂Cl₂, 6 h; (k) Bromoacetic acid, 5-bromopentanoic acid or 6-bromohexanoic acid, Et₃N, DMAP, EDCl, dry CHCl₃, r.t.; (l) AgNO₃, CH₃CN, 70 °C, protection from light; (m) CrO₃, CH₃COOH, H₂O, r.t., 3 h; (n) NaOH, H₂O, CH₃OH, reflux, 6 h.



Figure 7. Synthesis of ocotillol-type derivatives **92–108**. Reagents and conditions: (a) Jones reagent, acetone, r.t.; (b) (1) KOH, MeOH, H₂O, 60 °C; (2) 50% H₂SO₄; (c) 1) corresponding acid, anhydride or Boc-amino acid, DMAP, EDCI, CH₂Cl₂, r.t.; (2) CH₂Cl₂, TFA, r.t.; (d) excess of PCC, CH₂Cl₂, r.t.; (e) 1 M of PCC, CH₂Cl₂, r.t.; (f) NaOH, THF, R₁NH₂, r.t.

At present, the antibacterial target of ocotillol-type derivatives is still not clear. Bi et al. synthesized the ocotillol-type probe **109A**, which had a MIC of 8 µg/mL against *B. subtilis*. An epifluorescent microscopy study showed that **109A** was mainly distributed on the bacterial cell membrane rather than within the nucleoid (Figure 8). On this basis, Bi et al. synthesized the ocotillol-type probe **109B**, which had a MIC of 1 µg/mL against MRSA 18–19 (Hospital-acquired methicillin-resistant *Staphylococcus aureus*, collected in Chengdu, China from 2018) ^[43]. The antibacterial mechanism of **109B** against MRSA 18–19 is currently underway. The number of ocotillol-type probes is small, which limits the discovery of their antibacterial target. In 2017, 28-hydroxy protopanaxadiol was synthesized as a novel probe template ^[44]. The synthesis of new ocotillol-type probes. Additionally, functional probes that target the cell membrane are needed. Further research of ocotillol-type probes will promote the discovery of the target protein and provide a reference for the development of more effective drugs.



Figure 8. Structure of the ocotillol-type derivative **109A-109D** and the epifluorescent microscopy images of *B. subtilis* strain BS125 (**top**), strain 168 with compound **109A** treatment (**middle**), and strain BS3 (**bottom**). Scale bar: 4 μm. Reagents and conditions: (a) DMAP, EDCI, *N*-Boc-*N*'-Fmoc-I-Lysine, CH₂Cl₂, r.t.; (b) TFA, CH₂Cl₂, r.t.

Based on the present research of the ocotillol-type derivatives, a preliminary SAR of their antibacterial activities is summarized in Figure 9. The 24(S)-configuration is preferred, while substitution at the 3-OH changes the conformation to render the 24(R)-compound bioactive. A hydrogen donor at C-3 and C-12 are preferred to maintain the activity against Gram-positive bacteria. Decreased activity was observed when the functional groups at C-3 and C-12 were a ketone. When R₂ is an ester, mild activity against Gram-negative bacteria was observed.



Figure 9. Structure-activity relationship (SAR) of the antibacterial activity of ocotillol-type derivatives.

3.2. Anti-Inflammatory Activities of Ocotillol-Type Derivatives

lipopolysaccharide-stimulated RAW 264.7 cells can release the inflammatory mediator NO, prostaglandin E₂ (PGE₂), tumor necrosis factor (TNF- α), interleukin-6 (IL-6) and anti-inflammatory mediator interleukin-10 (IL-10). The antiinflammatory activity of 20(S)-ocotillol (4, 5) and 20(R)-ocotillol (150, 151) was evaluated in RAW 264.7 cells. The results showed that both 20(S)-ocotillol and 20(R)-ocotillol inhibited the release of the inflammatory cytokines NO and interleukin-6 (IL-6). However, the 20(S)-epimers mainly inhibited the release of PGE₂ and primarily increased the release of the antiinflammatory mediator IL-10. The 20(R)-epimers inhibited the release of the inflammatory cytokine TNF- α [45][46]. Oral ocotillol-type ginsenosides such as 109C (Figure 8) are metabolized to ocotillol-type sapogenin in the gut by gut microbiota [47]. Ocotillol-type sapogenin showed the highest inhibitory effect. In vitro studies demonstrated that 20(R), 24(R)-ocotillol might ameliorate colitis by inhibiting the expression of the proinflammatory cytokines TNF- α , interleukin-1β, IL-6, interleukin-17 (IL-17), interleukin-23 (IL-23) and interferon-y (IFN-y). Additionally, 20(R),24(R)-ocotillol strongly ameliorated Trinitro-benzene-sulfonic acid-induced iNOS and cyclooxygenase-2 (COX-2) expression, as well as activation of their transcription factors NF-KB and MAPKs in mice [47][48]. In 2019, Wang et al. found that **109D** (Figure 8) could attenuate lipopolysaccharide (LPS)-induced acute lung injury. A further mechanistic study indicated that 109D reversed the LPS-induced increases of mRNA expression and protein levels of macrophage inflammatory protein-2 (MIP-2) and intercellular adhesion molecule-1 (ICAM-1) [49]. Compound 109D also possessed neuroprotective activity by inhibiting the TLR4-mediated transforming growth factor-β-activated kinase-1(TAK1)/ nuclear factor kappa-B kinase 2 (IKK) /NF-kB, MAPKs, and Akt signaling pathways to exert anti-neuroinflammatory effects on LPS-activated microglia. In vivo experiments demonstrated that 109D significantly inhibited microglial activation and proinflammatory factor expression in the mouse cortex and hippocampus after the LPS injection [50].

Ocotillol-type derivatives with NO-inhibitory activity were further studied (Figure 10) $\frac{51|(52)|(53)|(54)|}{112}$. Derivatives 6, 46, 110, 112, 113, 121, 132 and 136 showed significant NO-inhibitory activities, while 115, 116 and 119 had no obvious NO-inhibitory activities. Derivatives 46 and 136 exhibited the most potent NO-inhibitory activities and were even comparable to a steroid drug. Additionally, 46 and 136 significantly decreased LPS-induced TNF-α and IL-6 synthesis and iNOS and COX-2 expression via the NF-κB pathway.



Figure 10. Synthesis of ocotillol-type derivatives **110–143**. Reagents and conditions: (a) PCC, CH₂Cl₂, r.t.; (b) Hydroxylamine hydrochloride, pyridine, 80 °C; (c) NaCNBH₃, TiCl₃, AcONH₄, MeOH, r.t.; (d) *n*-Hexanoic acid, EDCl, DMAP, r.t.; (e) NaBH₄, *i*-PrOH, r.t.; (f) NaBH₄, MeOH, r.t.; (g) Ac₂O, CH₂Cl₂, r.t.; (h) trifluoromethanesulfonic acid tertbutyldimethylsilyl ester (TBS-OTF), lutidine, r.t.; (i) KOH, MeOH, THF, r.t.; (j) Boc-amino acid, EDCl, DMAP, CH₂Cl₂, r.t.; (k) TFA, CH₂Cl₂, r.t.; (l) O-benzotriazole-N,N,N',N'-tetraMethyl-uroniuM-hexafluorophosphate (HBTU), NEt₃, DMF, r.t.; (m) TFA, CH₂Cl₂, r.t.

Wang et al. synthesized a series of ocotillol-type derivatives (Figure 11) ^{[55][56][57]}. Compound **144** had a protective effect on the lung function of experimental model mice with hormone-resistant asthma caused by non-typeable *Hemophilus influenzae* and improved their hormone resistance. Compounds **58** and **145–149** had inhibitory effects on the IL-6 expression and promoting effects on the IL-10 expression in the serum of rats induced by chronic obstructive pulmonary disease (COPD) caused by cigarette smoking.



Figure 11. Synthesis of ocotillol-type derivatives **144–149**. Reagents and conditions: (a) Boc-amino acid, DMAP, EDCI, CH₂Cl₂, r.t.; (b) TFA, CH₂Cl₂, r.t.; (c) aliphatic acid, DCC, EDCI, CH₂Cl₂, r.t.

Based on the present research of ocotillol-type derivatives, a preliminary SAR of their anti-inflammatory effects is summarized, as shown in Figure 12. The 24(R)-configuration is preferred for the anti-inflammatory activity. An oxime at C-3 is preferred for good inhibitory activity of LPS-induced NO synthesis. Boc-amino groups seem to be preferred to inhibit

the activity of LPS-induced NO synthesis than amino groups at C-3. A hydrogen donor at C-12 is preferred to inhibit LPSinduced NO synthesis. A fatty acid or amino acid group at C-3 has inhibitory effects on the expression of IL-6 and promotes the expression of IL-10 in the serum of a rat model of COPD induced by cigarettes.



Figure 12. SAR of ocotillol-type derivatives with anti-inflammatory activity.

3.3. Anticancer Effects of Ocotillol-Type Derivatives

The antitumor effect of ocotillol-type derivatives is mainly concentrated in the ocotillol monomer or in substances directly extracted from plants; thus, there are few reports on its structural modification ^{[58][59]}. **172** (Figure 13A) showed effective antitumor-promoting activity on a mouse hepatic tumor and mouse skin ^{[60][61]}. A series of ocotillol-type derivatives had been studied for their cytotoxic activity against HeLa, A549 and MCF-7 cancer cells. Pharmacological experiments on HeLa cells showed that ocotillol-type derivatives had cytotoxicity. Among them, compounds **5**, **152** and **173** (Figure 13A) possessed good activities with IC₅₀ values of 11.53 ± 0.49 μ M, 4.58 ± 0.66 μ M and 19.84 ± 1.10 μ M toward HeLa cells, respectively ^[62]. Compounds **162**, **163**, **167** and **166** showed reduced cell viabilities toward HeLa cells at 48.59%, 47.39%, 52.82% and 59.02% at 100 μ g/mL, respectively ^[63].

Pharmacological results indicated that ocotillol-type derivatives had anticancer potential, and the configurations at C-20 or C-24 and the number of glycosyl units at C-3 could have an important influence on the cytotoxicity in vitro. There are only a small number of studies on ocotillol-type derivatives with anticancer activity, and thus, there is an opportunity to increase the number of ocotillol-type derivatives with anticancer activity.

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