

Phytotoxic Ozone Concentration

Subjects: [Environmental Sciences](#)

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Tropospheric concentrations of phytotoxic ozone (O_3) have undergone a great increase from preindustrial 10–15 ppbv to a present-day concentration of 35–40 ppbv in large parts of the industrialised world due to increased emissions of O_3 precursors including NO_x , CO , CH_4 and volatile organic compounds.

[carbon uptake](#)

[manipulation experiment](#)

[stomatal ozone flux](#)

[interaction effect](#)

[tropospheric ozone](#)

1. Changes in O_3 Concentration

Concentrations of O_3 ($[O_3]$) have been increasing since the preindustrial era due to an increase of its precursors [\[1\]](#). As an important secondary phytotoxic air pollutant causing injury to plant tissue and a significant decrease of crop and timber yield, it deserves attention from farmers, scientists and the general public. Although the historical $[O_3]$ are unreliable because of a limited number of observations and disagreements in the retrospective modelling [\[2\]](#), it is considered that $[O_3]$ has increased from the baseline of 10–15 ppbv (parts per billion per volume, volume mixing ratio; [\[3\]](#)) to current concentrations of 35–40 ppbv in large parts of the industrialised world [\[4\]\[5\]](#). Plant species vary in their sensitivity to $[O_3]$, and it seems that genetically based detoxification processes [\[6\]](#) are significant and certainly sufficient to protect plants against any harmful effect of low pre-industrial $[O_3]$. The concept of “effective O_3 flux”, defined as a balance between stomatal O_3 flux and detoxifying capacity of the plant, was proposed [\[7\]](#). However, there is a huge variety of clones and cultivars (poplars, beans, etc.), which are sensitive even to low $[O_3]$, demonstrating a strong genetic basis for plant sensitivity to O_3 .

Ozone was discovered in 1839 by Christian Friedrich Schönbein during his experiments with the electrolysis of water. At the start of the modern era, $[O_3]$ was measured using classical procedures involving titration. In Europe, one of the first measurements was performed by Albert-Lévy in Paris. He showed $[O_3]$ to be 11 ± 2 ppbv over the period 1876–1910 [\[8\]](#). Even in high elevations, at Pic du Midi, France, 3000 m a.s.l., a concentration of only 10 ppbv was measured during 1874–1895 with a peak in spring and a minimum in winter [\[9\]](#). The oldest continuous measurements started at the Arkona-Zingst site (Germany) in 1956: they showed $[O_3]$ in the 1950s–1960s to be in the range of 15–20 ppbv [\[10\]](#).

The first harmful effects of O_3 were reported in the San Bernardino Mountains of Southern California, in *Pinus ponderosa* forest [\[11\]](#). During the 1970s, in inland valleys around Los Angeles, maximum annual $[O_3]$ reaching 300–

400 ppbv was common [12]. Air pollution led to an increase in the number of days with $[O_3] > 95$ ppbv from 114 in 1963 to 163 days in 1978 [13].

Elsewhere, an increase of 2–4 ppbv per decade was later reported [4], and an increase of 0.35 ppbv per year was seen in South Korea and Japan in 2000–2014 [14][15]. However, at highly polluted urban sites the increase was rapid, 2.6 ppbv per year in Beijing in 2005–2011 [16] and 2 ppbv per year in the Pearl River Delta region [17]. Recently, interannual and decadal changes are reported elsewhere: in the North China plains, there were increases of 3–5 ppbv (2001–2006), mostly attributed to a change of cloud cover and temperature [18] with only a low impact (1–2%) due to afforestation and increased VOC production [19].

In the southern hemisphere, with much less land area and industry, there is a trend of an increasing $[O_3]$ of 0.1 ppbv per year from 1990–2015 ranging from 0.04 at Baring head (New Zealand) to 0.21 at Arrival Heights (Antarctica). Overall there seems to be a concentration increase towards southern latitudes [20]. Similarly, an increase of 0.66 ppbv per decade has been observed in Chile at El Tololo mountain [21]. The increase is attributed to the poleward expansion of the Hadley Circulation, bringing the O_3 -rich air from the stratosphere [20].

Marked diurnal courses of $[O_3]$ have usually been found, particularly in large urban agglomerations [22]. O_3 is produced over the day, associated with high UV irradiance which drives the photochemistry, whilst at the same time, O_3 is being removed by wet and dry depositions on various surfaces and uptake by plants. Typically, NO_x is transported from urban areas at low elevations to rural forested areas where significant amounts of VOC are being produced as natural plant emissions. Thus the appropriate VOC/ NO_x ratio for O_3 production, ranging between 4 and 15, is achieved [23]. Such middle-range transport of NO_x is responsible for the enhanced production of O_3 in rural areas, often at high elevations, and may result in damage of vegetation. A globally averaged lifetime of tropospheric O_3 is approximately 23 days [24]. Therefore, O_3 could be transported even at long-range between continents [25]. However, its lifetime inside the boundary layer is much shorter because of the surface deposition and chemical reactions, such as reduction of O_3 to oxygen. These processes, as well as the spatio-temporal heterogeneity in $[O_3]$, are further modulated by the seasonal variability of microclimatic conditions (Figure 1).

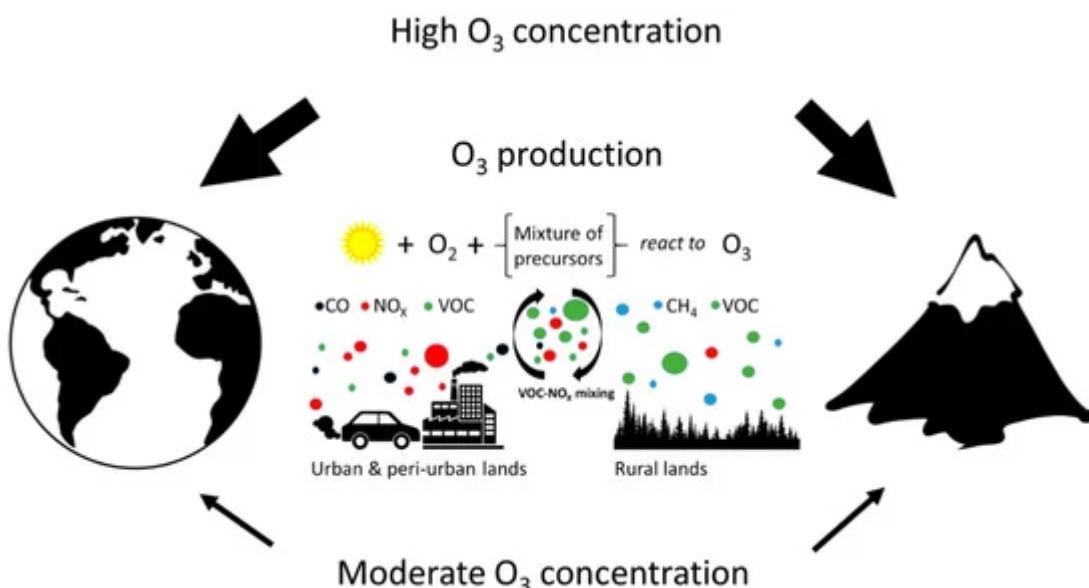


Figure 1. Scheme of tropospheric ozone (O_3) formation. Tropospheric ozone is formed in a complex series of photochemical reactions driven by ultraviolet (UV) solar radiation. NO_2 is photolyzed to form NO and an electronically excited oxygen atom, O, which reacts with molecular oxygen in the atmosphere (O_2) to form O_3 . However, O_3 may also regenerate NO_2 in the presence of NO, thus keeping a photo-stationary state. Therefore, net O_3 production occurs when O_3 precursors, such as carbon monoxide (CO), methane (CH_4) and volatile organic compounds (VOCs), are present in the atmosphere at appropriate concentrations. This chain of photochemical reactions is catalysed by hydroxide anion (OH^-), hydroperoxyl radical (HO_2), NO and NO_2 . Enhanced O_3 production thus occurs under high levels of UV radiation and when the concentration of precursors reaches critical levels. Transport of precursors and catalysts from urban and industrial lands (CO, NO_x and VOC) to rural conditions enhances mixing of polluted air plumes with clean rural air (enriched with CH_4 and VOC) and results in a VOC/ NO_x ratio conducive for O_3 formation. Therefore, latitudinal and elevational distribution differences in O_3 concentration are likely caused by the distribution of O_3 precursor sources associated with industrialization development and/or by an elevational increase in UV radiation.

At nightfall [O_3] rapidly decreases because of the oxidation of NO to NO_2 in the absence of production. Distinct seasonal behaviour patterns have been reported in industrialised and rural areas of Europe and the USA: (1) a broad spring-summer maximum of [O_3] in the industrialised parts, but (2) a minimum [O_3] in summer and autumn in remote regions [26]. Noticeably, spring [O_3] maximum is a northern hemispheric phenomenon, only found in northern and western parts of Europe. In the temperate zone of Central and Eastern Europe, the highest [O_3] are observed in summer months when temperatures and irradiances reach their highest values (reviewed in Monks [27]), while these are lowest in winter [28][29]. Moreover, substantially higher [O_3] are observed under clear skies than under cloudy skies, but not in winter [29].

In the Czech Republic, Central Europe, the annual maxima of [O_3] are being shifted towards the later parts of the year. The [O_3] peak has shifted from Day of Year (DOY) 120–170 at the beginning of the millennium towards DOY 160–175 over the following 20 years depending on the locality [30]. The shift is probably caused by the change of meteorological conditions towards warmer and dryer years with consequently more favourable conditions for O_3 formation [30]. However, contradicting results are found in the summer monsoon climate of Beijing: [O_3] maximum is in June, while the lowest values of [O_3] are in December [31]. Similarly, in the Yangtze River Delta, the maximum is found in July with a second maximum in September, followed by a minimum in November [32]. At 38 sites involved in the European Monitoring and Evaluation Programme (EMEP), there was a decrease of [O_3] reported in the 1990s, however later, around 2000, the [O_3] had increased; then, in the 2010s it decreased [33]. Interpretation of the trends and spatial patterns over several past decades has been challenging [34]; however in Europe, because of the successfully adopted measures to reduce O_3 precursors, O_3 surface concentration decreased by 2% from 2000 to 2014 [35].

In the Arctic, there is no clear trend in Barrow (USA, Alaska, 1981–2010) and Resolute (Canada), although there is an increasing trend in short-term periods [36]. In the southern hemisphere, the strongest increase in [O_3] is reported to be during the austral autumn (March-May) with an increase of 0.14 ppbv per year on average, while in other seasons the increase is only 0.07–0.12 ppbv per year [20]. The exception is South Africa with a sharp increase of 1

ppbv per year over the period 1992–2011 [37]. An overview of $[O_3]$ in different regions of the world, with model predictions for the future, is given by Archibald et al. [5].

Currently, $[O_3]$ and its changes are both measured and modelled, however modelling approaches based on state-of-the-art models may suffer from huge uncertainties [38], and some are unable to track accurately $[O_3]$ from the past.

2. Effect on Carbon Uptake from Leaf to Ecosystem Level

In the atmosphere, O_3 is known to react with double bonds between carbon atoms to produce aldehydes, ketones or higher oxidised molecules—that has been known since 1840 when O_3 was discovered. The mechanism is the same in plants, where, after penetrating through the stomatal apertures, ozone molecules oxidise the fatty acids of cell/organelle membranes; this leads to the formation of reactive oxygen species (ROS) causing damage of tissues (Figure 3). Such damage to photosynthetic membranes, despite the plant's increased defensive production of ROS scavenging enzymes ascorbate [39] and compounds with antioxidative capacity (carotenoids; [40][41]), inevitably leads to local necrotic cell death or early senescence [42]. Among others, Luwe and Heber [6] have shown that elevated $[O_3]$ increases concentrations of reduced and oxidised forms of ascorbate in the apoplast of leaves of different plant species. These transient increases are, however, often insufficient to protect leaf tissues. The yellowish mottling occurs particularly close to stomata and appears more often in older than young leaves [43]. Microscopic studies identified enlargement of intercellular space and chloroplast injuries, including thylakoid swellings and membrane disruption, as typical symptoms of O_3 impact [40]. Such reduced photosynthetically active leaf area leads to a reduced carbon uptake [44].

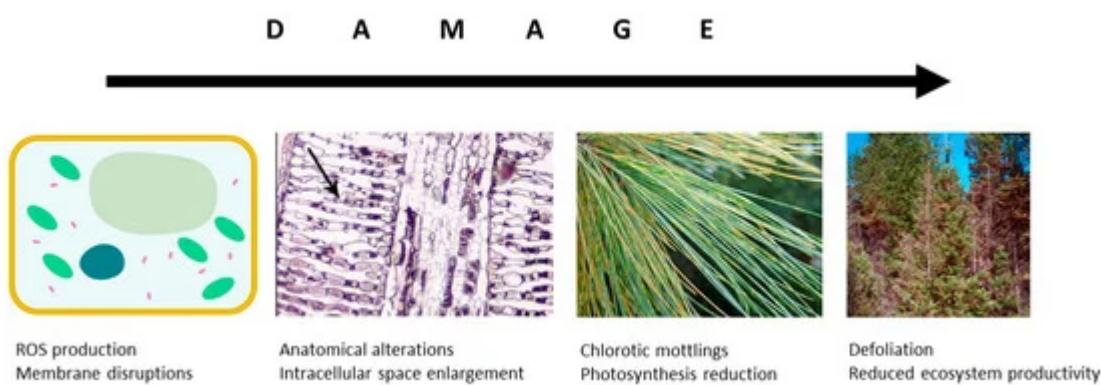


Figure 3. Damaging O_3 effects at the cellular and leaf levels influence the carbon allocation at tree and ecosystem level.

However, O_3 may affect carbon uptake at various physiological levels. Exposure to chronic $[O_3]$ closes stomatal pores leading thus to a reduced stomatal conductance to CO_2 diffusion and consequently to a reduced photosynthetic CO_2 assimilation [45]. Moreover, O_3 reduces photosynthetic CO_2 uptake via reduced Rubisco (ribulose-1,5-bisphosphate carboxylase/oxygenase) content [46][47]. These effects need to be related to growth and carbon economy at the ecosystem level. In a six-year free-air fumigation at a German forest, Matyssek et al. [48]

reported a 44% decline in stem productivity in *Fagus sylvatica* exposed to twice-ambient [O₃]. Reductions in biomass accumulation have been associated with a modified carbon allocation to plant organs. Based on the meta-analysis of temperate and boreal forests of the northern hemisphere, Wittig et al. [49] reported a significant decrease of the root-to-shoot ratio under elevated [O₃] indicating greater sensitivity of root biomass to [O₃]. O₃-induced reduction in root surface area per soil volume unit [50] can result in decreases of water and nutrition uptakes. Investigation of carbon pools revealed faster O₃-induced turnover of leaves/needles, reduction of canopy carbon pools and a substantial increase in carbon deposited to the forest floor [51].

Several metrics have been developed to assess the effect of O₃ on plants and to relate threshold [O₃] to relative yield loss. For example, the index AOT40 (accumulated dose of ozone over a threshold of 40 ppbv), which has to be interpreted with regional and meteorological aspect, has been established. This index is calculated over the sunlight hours and whole growing season, which is being prolonged towards a larger number of days in line with earlier phenological phase occurrence [52][53]. The highest and lowest AOT40 values are reported from Mediterranean regions (38,359 ppb h) and Northern Europe (5094 ppb h), respectively. In Continental Central Europe, AOT40 ranges between 13,636 and 23,515 ppb h, while it is 8207–13,751 ppb h in Atlantic Central Europe [35]. However, this AOT40 index takes into account only of O₃ exposure, but not the physiological properties enabling O₃ diffusion to plant tissues, which is directly responsible for the damage. Therefore, an alternative index based on stomatal O₃ uptake, POD_Y (phytotoxic O₃ dose above a flux threshold of Y nmol O₃ m⁻² s⁻¹) has also been advanced. The threshold is species-specific and depends on the detoxifying capacity of the plant (e.g., [54]). The value of Y ranges from 7 in *Alnus glutinosa* to 0–1 nmol O₃ m⁻² s⁻¹ in *Fagus sylvatica*. The minimum values of POD₀ were found in Northern Europe (14 mmol m⁻² year⁻¹), while maximum values of 29.7–32.1 mmol m⁻² year⁻¹ were observed in Mediterranean and Atlantic regions of Europe [35].

While POD_Y is mainly used in scientific and modelling studies, AOT40 still prevails in legislation (European Council Directive 2008/50/EC) and monitoring activities [55]. Protection of vegetation recommended by UNECE [56] sets an exposure-based critical level of AOT40 as 5000 ppbv h. Attitudes may change, and POD_Y is now being discussed as a potential integral part of new legislation in Europe [57]. While AOT40 decreased and POD₀ increase in Lithuanian forests over the period 2007–2014 [58], Klingberg et al. [59] reported a reduction of both indices in *Picea abies* at EMEP sites. More recently, Karlsson et al. [60] confirmed a reduction in AOT40 but did not find a change in POD₁ for the same tree species. Between 2000 and 2014, AOT40 decreased in most of the European countries (except rural northern areas of Iceland, Svalbard and Sweden), while POD₀ increased from 0.03 to 1.06 mmol O₃ m⁻² year⁻¹ across Europe [35]. However, in warm and dry years, AOT40 increased [61] and POD_Y decreased [62] when compared to wet seasons. To correctly determine the long-term trends in the development of these indices, continual time series over several decades are, therefore, very much needed.

Recently, a new flux-based index combining stomatal exposure and crown defoliation has been determined to define critical levels (CLef) for forest protection against O₃-induced visible injuries. Sicard et al. [63] recommended CLef to be less than 5 mmol m⁻² year⁻¹ POD₁ for broadleaved species and less than 12 mmol m⁻² year⁻¹ POD₁ for conifers. CLef representing ≥25% of crown defoliation is recommended to be maximal 17,000 and 19,000 ppbv

h of AOT40 for conifers and broadleaved species, respectively. It is obvious that those new indices are inevitably linked to POD_Y and AOT40 and only new limits are set.

As the injuries induced by O₃ deposition on cuticle are usually small [64], the negative effect of O₃ uptake is connected mainly to stomatal O₃ flux. The total flux of ozone to vegetation may be thought of as two components: stomatal flux (uptake through the stomatal pores) and non-stomatal flux (deposition to other surfaces in the canopy and also reaction with gaseous compounds in the canopy air-space). The ratio between stomatal and total O₃ flux depends on actual microclimatic conditions and differs in various ecosystems (Table 1). The highest seasonal maxima of total O₃ flux were recorded in *Quercus ilex* forest [62] followed by *Populus grandidentata* [65], *Larix decidua* and *Pinus halepensis* [66][67]. Daily mean values range from 0.8 nmol m⁻² s⁻¹ in *Pinus sylvestris* forest in Belgium [68] to 8.6 nmol m⁻² s⁻¹ in *Q. ilex* forest in Italy. See Table 1 for more details. Stomatal flux is determined by [O₃] and two resistances connected in series (leaf boundary layer resistance and the stomatal resistance). While boundary layer resistance depends on wind speed and heat flux, stomatal resistance is primarily influenced by irradiance and VPD [69]. Stomatal O₃ flux was found to be 37% of total O₃ flux in a northern mixed hardwood forest [65], but it was 21% in semi-arid regions of Israel [66], and only 15% in *Larix decidua*, Alps, Italy [67]. However, in subalpine coniferous forest dominated by Engelmann spruce (*Picea engelmannii*) and subalpine fir (*Abies lasiocarpa*) in southern Wyoming, USA, 59% of stomatal O₃ flux was found as an annual average [70]. Similarly, stomatal O₃ flux dominated in Czech *P. abies* mountainous forest under moderately cool and humid climate [71]. Juráň et al. [29] found that stomatal flux represents 53.5% of total O₃ flux during summer days with partly-cloudy conditions, but it decreases to 43.5% during sunny days. Moreover, a fraction of stomatal O₃ flux could be further modulated by forest age [71][72]. Comparison of modelled and measured fluxes could be found elsewhere [29]. See Table 1 for more details.

Table 1. Examples of fractions of stomatal O₃ fluxes to total O₃ fluxes from different forest ecosystems. Notes: EC—eddy-covariance; *—Total deposition flux in $\mu\text{g m}^{-2} \text{s}^{-1}$; **—value not specified.

| Forest Type | Species | Country | Total Deposition Flux (nmol m ⁻² s ⁻¹) | Stomatal Flux (% of Total) | Approach |
|-------------------------|--|----------------|---|----------------------------|-----------|
| Subalpine coniferous | <i>Picea engelmannii</i> and <i>Abies lasiocarpa</i> | Wyoming, USA | 0.5–0.6 * (summer max) | 59 | EC |
| Mountainous | <i>Picea abies</i> | Czech Republic | 7.09 (daily mean) | dominant ** | modelling |
| Mountainous | <i>Picea abies</i> | Czech Republic | 14 (summer max) 2 (winter max) | 43.5–53.5 | EC |
| Northern mixed hardwood | <i>Populus grandidentata</i> | Michigan, USA | 27.7 (seasonal max) | 37 | EC |
| Evergreen Mediterranean | <i>Quercus ilex</i> | Italy | 6.9–8.6 (daily average) | 34.4 | EC |

| Forest Type | Species | Country | Total Deposition Flux (nmol m ⁻² s ⁻¹) | Stomatal Flux (% of Total) | Approach |
|-------------|-------------------------|---------|---|----------------------------|-----------|
| | | | 51 (seasonal max) | | |
| Coniferous | <i>Pinus sylvestris</i> | Belgium | 0.8–5.8 (daily mean) | 26 | modelling |
| Coniferous | <i>Picea abies</i> | Denmark | 0.5 * (5-years mean) | 21 | modelling |
| Coniferous | <i>Pinus halepensis</i> | Israel | 5–10 (seasonal range) | 21 | EC |
| Alpine 3 | <i>Larix decidua</i> | Italy | 40 (summer daily max) | 15 | EC |

present. Chemical O₃ sink also contributes to non-stomatal flux involving the reactions of O₃ molecules with VOCs, NO and aerosols. Non-stomatal flux dominates in spring and summer because of the exponential increase of VOC concentration with increasing air temperature and solar radiation [73][74]. It is the period, when [O₃] is usually the highest. Contrary to that, non-stomatal fluxes are negligible over the winter in temperate forests due to low VOC emissions to the atmosphere [75]. They were significant even at a moorland site in Scotland, without the complications of complex forest canopies: the non-stomatal O₃ flux was up to 70% of the total flux [76]. These findings suggest that most of O₃ deposits on leaf cuticles and/or wet layer of the moss, below the sparse herbaceous canopy.

Here we summarise the effects of O₃ on NEP (net ecosystem productivity) and GPP (gross primary productivity) in several forest ecosystems estimated by eddy-covariance technique and modelling approaches (Table 2). The impacts of O₃ are very diverse. There is no effect in mature Belgian Scots pine forest on GPP measured over 15 years, although critical levels of AOT40 and POD₁ were exceeded in each year of measurement [77]. Similarly, in a poplar stand, Belgium, no effect on NEP was reported [78], even though stomatal O₃ flux amounted up to 59% of the total O₃ flux. On the other hand, a reduction of NEP was reported in Czech [71] and Swiss forests [79], particularly in Norway spruce and European beech stands. After 20 years of monitoring, the only mild effects of O₃ on GPP and photosynthesis were observed in a broad-leaf Harvard forest. These findings were attributed to the fact that 40% of photosynthesis occurs lower in the canopy, in shade, where stomatal conductance and [O₃] are lower [80]. So the canopy structure can also modulate the effect of O₃.

Generally, a strong correlation of GPP to AOT40 index was shown [81]. Among others, a tight linear decrease of whole-plant dry mass with increasing daylight AOT40 was found in Japanese larch (*Larix kaempferi*) and beech (*Fagus crenata*) seedlings [82]. Comparing to the preindustrial era, a reduction of 1–16% of GPP was reported for USA vegetation covers [83]. Similarly, Karlsson [84] reported a reduction of living biomass stock due to O₃ to be 2% in northern European countries but up to 32% in central European countries [85]. One of the highest, as much as 24.8%, reduction rates of NEP were for a Norway spruce forest [28] with the highest decrease in July, the warmest month of the year.

Wang et al. [86] modelled biomass carbon stock over 500 years involving a successional series of the temperate deciduous forests. No change of carbon stock was reported due to the change of forest species over the time period with the increasing dominance of isoprene-emitting species. Isoprene acts as a shielding agent preventing O₃ to enter the stomatal aperture—isoprene outside of the leaf reacts with O₃ [87]. It is clear, that O₃ was not an issue for half a millennium, however, it shows a possible direction of how natural ecosystems might possibly evolve and adapt in a O₃-rich world by a mechanism incorporating successional dynamics.

Table 2. Effects of O₃ on various carbon-related criterion. GPP—gross primary production, NEE—net ecosystem exchange, NEP—net ecosystem productivity.

| Type of Ecosystem | Dominant Plant | O ₃ Effect | Country | Criterion |
|--------------------------------|--------------------|--|----------------|-----------|
| Mature stand | Scots pine | neutral | Belgium | GPP |
| Plantation | mix of poplars | neutral | Belgium | NEE |
| Mature stand | Stone pine | neutral | Italy | GPP |
| Mixed hardwood/conifer forests | Red oak, Red maple | negligible | USA | GPP |
| USA vegetation | - | reduction 1–16% | USA | GPP |
| Young stand | Norway spruce | reduction | Czech Republic | NEP |
| Young stand | Norway spruce | reduction 24.8% | Czech Republic | NEP |
| Young stand | Ponderosa pine | reduction 12% | USA | GPP |
| Orchard | Orange orchard | reduction 19% | USA | GPP |
| Flux sites in Europe and USA | - | reduction 6–29% deciduous forest reduction 4–20% evergreen needle leaf forest | Europe, USA | biomass |

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