Beeswax

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Beeswax is one of the natural waxes that have been used as a support ingredient in cosmetic and pharmaceutical formulations. Although it has well-known healing properties, it remains a secondary and poorly valued product, particularly in South American countries' apiarian production. In Latin America, for example, the apiarian activity can be traced back to stingless bees in pre-Columbian times. Then, with the arrival of the Spaniards in the 16th century, honey bees (*Apis mellifera iberian* and *A. m. mellifera*) were introduced, which were joined afterward by other breeds such as *A. m. ligustica* and *A. m. scutellata*. Over the years, honey has been the main product from the apiarian farmers, being beeswax a secondary product, mainly used for the regular renewal procedure of the hives. Nowadays, beeswax's cosmetic and pharmaceutical use is found at the level of small laboratories and small businesses.

beeswax wax biomaterials bee natural product

1. Beeswax

The term "wax" refers to mixtures of compounds such as esters, long-chain hydrocarbons, and ketones, which form materials with high melting points and excellent water resistance. In addition, there are waxes of animal, vegetable, and microorganism origin ^[1]. Beeswax is a fat produced by bees to build their honeycombs. The bees secrete the wax in the form of small rounded scales in the four ventral glands in the lower part of the abdomen, and it is synthesized as a reduction of sugars of food origin. The beeswax-making process phases are: bees eat honey and sugars (6 carbons molecules), which are absorbed in the intestine. From there, they pass to the interior of their body, where they are transformed into small fragments (2 carbons). Then, in the wax glands, they recombine differently to form, on the one hand, the fatty acids and hydrocarbons (between 18 and 36 carbons), and on the other, the wax esters and alcohols (between 24 and 33 carbons). The mixture of these products is known as beeswax ^{[2][3][4][5]}.

2. Uses and applications

Beeswax has played an essential role in history and popular tradition for many years. Historically, beeswax has been used for candle manufacture; it also was used in letter envelopes seals, sculpture making, and sealing coffins, among other applications. Due to beeswax characteristics, properties and benefits, it is used in both handcrafted and industrial products^{[6][7][8][9][10]}. Industry uses beeswax as an insulating and hydrophobic component of numerous products. For example, it is used in electrical cables to isolate copper from moisture, in

electronic circuits, to protect leather, in the preparation of varnishes, inks, matches, and protective waxes for cuttings [11][12].

Beeswax goes into the composition of ointments and creams as a fat base and thickener. The major use in this field is depilating wax, a mixture of beeswax and resins. Beeswax has anti-inflammatory and healing properties and is thus widely used in cosmetic and pharmaceutical products ^{[13][14]}. It is used to cover sewing cords in shoe production, paperboards, and even in some cultures to produce dried meat. It is also used in shoe polishes and creams to protect cans from acidic attacks from fruit juices and other corrosive agents. It is used to make models for pieces in jewelry and sculpture modeling due to its malleability^{[3][15]}. Beeswax in some Asia and African countries is used to create batik fabrics and manufacture small metal ornaments through the molten wax method^[16]. Companies such as Stockmar and Filana use beeswax to make wax crayons. Additionally, candelilla wax has been proposed in beeswax crayon formulations^[17]. Stockmar also manufactures modeling beeswax.

3. Beeswax physical properties

Beeswax is an inert material with high plasticity at a relatively low temperature (around 32°C)^[12]. Its melting point is not constant since the composition varies slightly depending on its origin. Typical values are between (62 °C to 65 °C)^[18]. Its relative density at 15 °C is reported between 0.958 g/cm³ to 0.970 g/cm³, while its thermal conductivity is approximately 0.25 W/m.K^[19]. It is also known that the viscosity at 100 °C is less than 20 mPa. The boiling point is unknown and has a flash point at temperatures higher than 180 °C^[20].

4. Pharmaceutical properties

Sterols present in beeswax are therapeutically beneficial compounds effective in lowering cholesterol levels. The incorporation of sterols into different foods may be convenient^[21]. It is used for delicate skin care in cosmetology, especially when it is dry. It cleans the epidermis and softens and nourishes the dermis, thus preventing skin aging. Products that contain beeswax soften the skin. White wax typically enters the composition of nourishing, astringent, cleansing creams and skin masks. The therapeutic properties of beeswax were already known in antiquity. In his famous "Canon of medicine," Avicenna cites several medicine formulas whose composition includes beeswax. In addition, archeological evidence of beeswax ointments has been found from as early as the 16th century^[22].

Nowadays, beeswax continues to occupy a prominent place in medicine preparations^[14]. According to Pharmacopoeia, plasters, ointments and creams should be prepared in pharmacies with a beeswax base. In addition, the white wax is included in the composition of creams, astringents, cleaning, whitening, and facial masks. In the United States, chewing gum (combs wax) is attributed to have specific valuable properties, among others, to activate the secretion of saliva and gastric juice, eliminate dental stones, and reduce nicotine concentrations in smokers^[23]. Recently, beeswax has been used to encapsulate drugs and flavors^{[24][25]}.

5. Beeswax chemical composition

Beeswax comprises at least 284 different compounds, of which 21 major compounds represent 56% of the total composition of the wax. The average composition of beeswax is presented in Table 1 ^[2].

Components	Percentage (%)
Hydrocarbons	14
Monoesters	35
Diesters	14
Triesters	3
Hydroxy monoesters	4
Hydroxy polyester	8
Monoester acids	1
Polyester acids	2
Free fatty acids	12
Unidentified material	7

Table 1. Beeswax composition

6. Beeswax purification

As found in combs, beeswax is yellow and has a particular smell similar to honey. Its purification is carried out through several procedures reported in the literature. The purification procedure consists in melting beeswax in a water bath at a temperature higher than 60 °C. Then it is bleached through a variety of methods, among them:

exposition to the sun^[26], through diatomaceous earth and activated carbon^[27], or with sulfuric acid ^[23]. The molten beeswax is then poured on a vessel and partially submerged in temperate water while slowly mixing, and the impurities are scraped off the surface. The purified beeswax is white and translucent and has thin edges.

7. Beeswax formulation for creams and ointments

In general, to obtain a dermo-cosmetic cream, the components of each phase must be mixed separately at a temperature close to 60 °C, then incorporated one phase into the other under mixing, cool, and homogenize. However, the preparation is more straightforward for ointments as a single phase. The procedure consists basically in melting the beeswax at a temperature higher than 65 °C and adding the formulation components. In this sense, the cream or ointment components must be chosen according to the objective pursued with the application on the skin. Thus, beeswax can be used as a component in moisturizing creams for burns, stretch marks, wrinkles, cellulite, lip balms, and even sunscreen formulations^{[28][29][30][31]}.

This entry is from: Technological use of beeswax for obtaining organic products, non-toxic for the human being.

References

- 1. Flaherty, B. Characterisation of Waxes by Differential Scanning Calorimetry. J. Appl. Chem. Biotechnol. 1971, 21 (5), 144–148. https://doi.org/10.1002/jctb.5020210507.
- 2. Tulloch, A. P. Beeswax: Structure of the Esters and Their Component Hydroxy Acids and Diols. Chem. Phys. Lipids 1971, 6 (3), 235–265. https://doi.org/10.1016/0009-3084(71)90063-6.
- 3. Tulloch, A. P. Beeswax—Composition and Analysis. Bee World 1980, 61 (2), 47–62. https://doi.org/10.1080/0005772X.1980.11097776.
- Blomquist, G. J.; Chu, A. J.; Remaley, S. Biosynthesis of Wax in the Honeybee, Apis Mellifera L. Insect Biochem. 1980, 10 (3), 313–321. https://doi.org/10.1016/0020-1790(80)90026-8.
- 5. Bogdanov, S. Quality and Standards of Pollen and Beeswax. Apiacta, 2004, 38, 334–341.
- Gao, Y.; Lei, Y.; Wu, Y.; Liang, H.; Li, J.; Pei, Y.; Li, Y.; Li, B.; Luo, X.; Liu, S. Beeswax: A Potential Self-Emulsifying Agent for the Construction of Thermal-Sensitive Food W/O Emulsion. Food Chem. 2021, 349, 129203. https://doi.org/10.1016/j.foodchem.2021.129203.
- Giampieri, F.; Quiles, J. L.; Cianciosi, D.; Forbes-Hernández, T. Y.; Orantes-Bermejo, F. J.; Alvarez-Suarez, J. M.; Battino, M. Bee Products: An Emblematic Example of Underutilized Sources of Bioactive Compounds. J. Agric. Food Chem. 2022. https://doi.org/10.1021/acs.jafc.1c05822.

- Szulc, J.; Machnowski, W.; Kowalska, S.; Jachowicz, A.; Ruman, T.; Steglińska, A.; Gutarowska,
 B. Beeswax-Modified Textiles: Method of Preparation and Assessment of Antimicrobial Properties.
 Polymers (Basel). 2020, 12 (2), 344. https://doi.org/10.3390/polym12020344.
- Diyana, Z. N.; Jumaidin, R.; Selamat, M. Z.; Suan, M. S. M. Thermoplastic Starch/Beeswax Blend: Characterization on Thermal Mechanical and Moisture Absorption Properties. Int. J. Biol. Macromol. 2021, 190, 224–232. https://doi.org/10.1016/j.ijbiomac.2021.08.201.
- 10. Ahnert, P. Beeswax Alchemy: How to Make Your Own Soap, Candles, Balms, Creams, and Salves from the Hive; Quarry Books, 2015.
- 11. Kester, J. J.; Fennema, O. Resistance of Lipid Films to Water Vapor Transmission. J. Am. Oil Chem. Soc. 1989, 66 (8), 1139–1146. https://doi.org/10.1007/BF02670100.
- 12. Hepburn, H. R. Honeybees and Wax: An Experimental Natural History; Springer Science & Business Media, 2012.
- Al-Waili, N. S. Topical Application of Natural Honey, Beeswax and Olive Oil Mixture for Atopic Dermatitis or Psoriasis: Partially Controlled, Single-Blinded Study. Complement. Ther. Med. 2003, 11 (4), 226–234. https://doi.org/10.1016/S0965-2299(03)00120-1.
- Fratini, F.; Cilia, G.; Turchi, B.; Felicioli, A. Beeswax: A Minireview of Its Antimicrobial Activity and Its Application in Medicine. Asian Pac. J. Trop. Med. 2016, 9 (9), 839–843. https://doi.org/10.1016/j.apjtm.2016.07.003.
- 15. Mladenoska, I. The Potential Application of Novel Beeswax Edible Coatings Containing Coconut Oil in the Minimal Processing of Fruits. 2012, 1 (2), 26–34.
- 16. FAO. La Cera de Abeja un producto útil y valioso http://www.fao.org/docrep/008/y5110s/y5110s07.htm (accessed Oct 10, 2021).
- 17. Gaytan, M. Proceso Para La Elaboración de Crayones a Base de Cera de Candelilla. Patent PA03011106A, 2005.
- Gaillard, Y.; Mija, A.; Burr, A.; Darque-Ceretti, E.; Felder, E.; Sbirrazzuoli, N. Green Material Composites from Renewable Resources: Polymorphic Transitions and Phase Diagram of Beeswax/Rosin Resin. Thermochim. Acta 2011, 521 (1), 90–97. https://doi.org/10.1016/j.tca.2011.04.010.
- Morgan, J.; Townley, S.; Kemble, G.; Smith, R. Measurement of Physical and Mechanical Properties of Beeswax. Mater. Sci. Technol. 2002, 18 (4), 463–467. https://doi.org/10.1179/026708302225001714.
- 20. Buchwald, R.; Breed, M. D.; Greenberg, A. R. The Thermal Properties of Beeswaxes: Unexpected Findings. J. Exp. Biol. 2008, 211 (1), 121–127. https://doi.org/10.1242/jeb.007583.

- Mellema, M. Co-Crystals of Beeswax and Various Vegetable Waxes with Sterols Studied by X-Ray Diffraction and Differential Scanning Calorimetry. J. Am. Oil Chem. Soc. 2009, 86 (6), 499– 505. https://doi.org/10.1007/s11746-009-1385-4.
- Baeten, J.; Romanus, K.; Degryse, P.; De Clercq, W.; Poelman, H.; Verbeke, K.; Luypaerts, A.; Walton, M.; Jacobs, P.; De Vos, D.; Waelkens, M. Application of a Multi-Analytical Toolset to a 16th Century Ointment: Identification as Lead Plaster Mixed with Beeswax. Microchem. J. 2010, 95 (2), 227–234. https://doi.org/10.1016/j.microc.2009.12.005.
- Valega, O. Usos terapéuticos de la cera de abeja https://www.apiservices.biz/es/articulos/ordenar-por-popularidad/972-como-produce-cera-laabeja-y-como-producir-mas-y-mejor (accessed Oct 12, 2020).
- Ranjha, N. M.; Khan, H.; Naseem, S. Encapsulation and Characterization of Controlled Release Flurbiprofen Loaded Microspheres Using Beeswax as an Encapsulating Agent. J. Mater. Sci. Mater. Med. 2010, 21 (5), 1621–1630. https://doi.org/10.1007/s10856-010-4034-4.
- 25. Fabra, M. J.; Hambleton, A.; Talens, P.; Debeaufort, F.; Chiralt, A.; Voilley, A. Influence of Interactions on Water and Aroma Permeabilities of ι-Carrageenan–Oleic Acid–Beeswax Films Used for Flavour Encapsulation. Carbohydr. Polym. 2009, 76 (2), 325–332. https://doi.org/10.1016/j.carbpol.2008.10.024.
- 26. Midgley, B. Guía Completa de Escultura Modelado y Cerámica: Técnicas y Materiales; 1982.
- 27. Orantes, F. Procedimiento de Descontaminación de Cera de Abejas y Cera de Abejas Resultante. Patent P200900541, 2012.
- Salisbury, R.; Leuallen, E. E.; Chavkin, L. T. The Effect of Phase Volume Ratio on Emulsion Type*: I. Beeswax-Borax Ointments. J. Am. Pharm. Assoc. (Scientific ed.) 1954, 43 (2), 117–119. https://doi.org/10.1002/jps.3030430213.
- 29. Herrero, J. Ointment for Burns. Patent P9801496, 2000.
- 30. Remiro, E. Oily Formulation for Topical Use, Procedure for the Prepa-Ration Thereof and Applications. Patent P9802060, 2000.
- 31. Lombardero, R. Pomada Para Quemaduras. Patent P201030393.

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