

Available Drinking Water Treatment Technologies

Subjects: Automation & Control Systems

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Water is one of the main sources of life's survival. It is mandatory to have good-quality water, especially for drinking. Many types of available filtration treatment can produce high-quality drinking water. In general, the treatment technologies for treating water depend on the type of raw intake water that comes from various water sources, such as surface water and groundwater. Membrane filtration is an advanced drinking water treatment that is widely used nowadays in water treatment processes, mainly for drinking water.

Keywords: filtration ; drinking water ; membrane fouling ; fouling prevention ; fouling prediction ; fouling control

1. Introduction

The quality of drinking water resources is being enthusiastically addressed around the world since it is essential to health and development issues. Due to uncontrolled industrial waste and low public awareness, water pollutants can be discharged either directly or indirectly to water resources such as lakes, ponds, rivers, seawater, and groundwater, which later become contaminated. The contaminated or poor quality of drinking water can cause various infectious diseases and negatively impact our overall health ^[1]. According to the World Health Organization (WHO), contaminated drinking water can cause serious diseases such as diarrhea, cholera, dysentery, hepatitis A, typhoid, and polio ^[2]. It is estimated that around 502,000 people die each year from diarrhea due to unsafe drinking water. The quality of water resources has been gradually depreciating due to industrialization and urbanization ^[3]. It has become a crucial problem due to the difficulty of meeting effluent quality standards with conventional treatment processes ^{[4][5][6]}. Good-quality drinking water helps people achieve maximum body health and well-being.

To obtain high-quality drinking water, a good and reliable water treatment process is desirable. Traditional drinking water treatment includes five common units such as coagulation, flocculation, sedimentation, filtration, and disinfection ^{[7][8][9]}. More than ten decades ago, the only treatment processes used in municipal and industrial water treatment were conventional filtration, such as clarification and granular media filtration, and chlorination methods. However, in the past twenty years, industrial water has shown high interest in the implementation of advanced water treatment technologies, particularly for water purification technologies such as membrane filtration, ultraviolet irradiation, the advanced oxidation process (AOP), ion exchange, and biological filtration for the removal of water contaminants in drinking water ^[10].

2. Available Drinking Water Treatment Technologies

In general, the treatment technologies for treating water depend on the type of raw intake water that comes from various water sources, such as surface water and groundwater.

2.1. Conventional Treatment

Conventional treatment is one of the popular approaches that has been used for water and wastewater treatment systems, where it involves several processes, including bar screening, grit removal, pre-oxidation, coagulation, flocculation, sedimentation, rapid/slow sand, granular active carbon filtration, and/or disinfection ^[11]. These processes can remove various solid sizes and organic matter from the liquid phase. It is also able to contribute to the reduction of microorganisms that cause concern for public health. There are several types of conventional filtration treatments, such as simple screen filters, slow and fast sand filters, diatom filters, and charcoal filters. The effect of filter media on the filtration process needs to be considered when designing the filtration unit. Additionally, the design of the backwash filter needs to be taken into account when high turbidity in effluent water increases head losses and requires long filtration operations ^[12] ^[13].

Many studies have been performed to investigate the effectiveness of conventional filtration in treating drinking water. The previous study of the removal of diclofenac from drinking water is reported by Rigobello et al. ^[14], where the conventional

sand filter is compared with granular activated carbon (GAC) filtration. The results showed that a sand filter could not effectively remove diclofenac, whereas a combination of a sand filter and GAC filtration could remove diclofenac with $\geq 99.7\%$ efficiency. A slow sand filter and charcoal filter have been used in the study by Murugan and Ram ^[15]. The application of a slow sand filter can help in the reduction of water turbidity and prevent fouling at the reactor tubes. The charcoal filter is used to help in the absorption of heavy metals that are present in the water. In this work, slow sand filters require periodic removal of the microbial layer, while charcoal must be replaced in the filter every month as there are no indications that the charcoal has reached its breakthrough.

Zheng et al. ^[16] investigate the use of a slow sand filter as a pre-treatment for the removal of organic foulants in secondary effluent. The investigation was conducted with different filtration rates and showed that the proposed pre-treatment can effectively control the fouling rate at low filtration rates with respect to biopolymer removal and cycle time. Another study on the effect of a flow configuration based on a slow sand filter was performed by Sabogal-Paz et al. ^[17], where a comparison study was performed for the household system between intermittent and continuous flows. The authors observe that the flow configuration of a slow sand filter cannot be applied as a single treatment because it is not able to remove the organic foulants effectively. The work proposed by Ahammed and Darva ^[18] investigates the effect of a modified slow sand filter by introducing a thin layer of iron oxide-coated sand. The performance of the proposed method is measured based on its capability to remove bacteria and turbidity. Results showed that the modified slow sand filter was able to increase the removal rate of bacteria, but there was no significant reduction in turbidity. Work by Mizuta et al. ^[19] presents bamboo powder charcoal and activated carbon filtration in the removal of nitrate and nitrogen from drinking water. The results showed that bamboo powder charcoal filtration was able to provide higher adsorption and less influence on temperature compared to activated carbon filtration. Bamboo charcoal filtration was studied by Zhang et al. ^[20] to remove microcystin-LR from drinking water. In this study, bamboo charcoal filtration was modified with chitosan, and the results indicate that the applied treatment was able to effectively remove the microcystin-LR, especially when the amount of bamboo charcoal was increased.

Based on previous studies of conventional treatment methods, it is clear that the method is incapable of producing satisfactory effluent quality. Most of the treatments require either modification or combination with other methods, which is costly due to frequent maintenance. Moreover, this treatment is considered economically unbeneficial for developing countries ^[21], where the treatments require a long operating period and a large footprint ^[22]. Due to the importance of having safe and healthy water, water utilities have started to consider alternative treatment technologies to traditional drinking water treatment.

2.2. Advanced Treatment

Here, several advanced treatments of water technologies, particularly for water purification technologies such as membrane filtration, ultraviolet irradiation, the advanced oxidation process, ion exchange, and biological filtration, are discussed. Recently, membrane filtration is increasingly being accepted and implemented in drinking water treatment plants ^[23]. Membrane technology is widely used in filtration systems, particularly for the removal of particulate matter in solid-liquid separation processes ^{[24][25]}. Moreover, the combination of membrane technology with a bioreactor is called a membrane bioreactor, and this technology has proven its high capacity for the removal of pollutants in water and wastewater treatment processes ^{[26][27]}. The main issue in membrane filtration is the fouling phenomenon, which, if not prevented, will affect the overall filtration performance in the long run.

Another advanced technology that is primarily used in drinking water is ultraviolet (UV) irradiation technology ^[28]. UV irradiation is used as a disinfection process and is commonly designed with a series of UV lamps so that the microorganisms in the water will be inactivated when exposed to UV light ^[29]. Although UV irradiation is a promising disinfection technology due to its compactness and low cost, it faces a challenge due to its reliance on electrical component sensitivity ^[30], which can result in high failure rates.

The advanced oxidation process (AOP) is another technology generally applied in water treatment. The AOP includes several processes that produce hydroxyl radicals for the oxidation of organic and inorganic water impurities ^[31]. Among the three main AOP processes are ozone, ozone with hydrogen peroxide addition, and UV irradiation with hydrogen peroxide addition. Each of the processes has its challenges and will not be discussed in detail here. To summarize, AOP can provide multiple uses in water treatment, such as color, oxidation of synthetic organic chemicals, taste and odor, and many more. However, the complexity of AOPs in terms of chemical reactions between processes makes it hard to achieve an optimum treatment system design ^[32]. The next advanced water treatment is ion exchange (IX) technology. This technology was previously limited to only softening water for use in water treatment plants. However, the limits are now also being set on several inorganic chemicals, making the IX a more interesting technology to explore in water treatment

applications. Lastly, biological filtration is another type of advanced treatment in water technology. The filtration is based on biological processes, which are different from the previously mentioned technologies that are based on physical and/or chemical processes. Works by Wang et al. [33] claim this biological filtration is the most effective process to produce biologically stable water. However, there are still unanswered issues regarding the proper design and implementation of biological filtration, particularly in terms of the size and type of filter media to be used. **Figure 1** summarizes the conventional and advanced filtration methods for drinking water treatment.

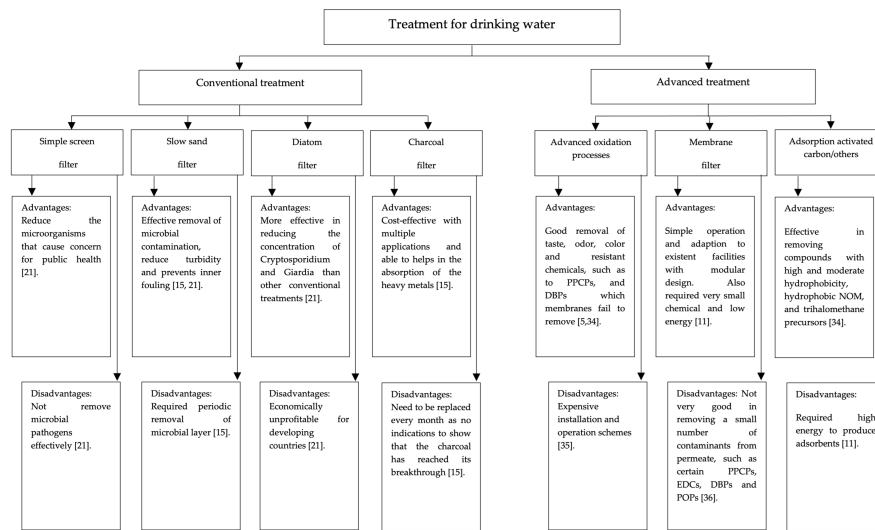


Figure 1. Available treatment for drinking water [5][11][15][21][34][35][36].

2.3. Hybrid Treatment

In general, most industrial drinking water treatments still involve conventional and advanced treatment processes [8]. **Figure 2** shows an example of industry-standard potable reuse water plants that involve conventional and advanced treatment processes [8]. In the primary treatment, the sedimentation of solid waste is performed. Water from secondary and tertiary treatment can be used for potable and non-potable reuse applications. The secondary treatment involves biological processes (e.g., the activated sludge process), and the tertiary treatment involves physical and/or chemical processes. For the disinfection process, chlorine is used to disinfect water to kill bacteria, parasites, and viruses in drinking water [37]. Alternatively, disinfectants such as chlorine dioxide, ozone, and ultraviolet radiation are also used. In advanced treatment, the integrated membrane system (IMS) and full advanced treatment (FAT) are implemented. The IMS uses a low-pressure membrane filtration process either microfiltration (MF) or ultrafiltration (UF). Meanwhile, FAT applies called either nanofiltration (NF) or reverse osmosis (RO), which are high-pressure membrane filtration processes. The application of IMS can provide high efficacy in the removal rate of particulate matter, microbial pathogens, and natural organic matter, whereas FAT is capable of removing magnificently organic–inorganic dissolved constituents such as salts and organic chemicals that are impossible to be removed by IMS. Ultraviolet and advanced oxidation processes act as post-treatment disinfection. In this stage, it will break down small neutral organic compounds that pass-through FAT. The final stage is known as degassing and lime dosing, which act as a water stabilizer and increase the pH and alkalinity of the water. The industry standard potable reuse water plant shown in **Figure 2** can meet the specification for drinking water quality, but there are several drawbacks, including a large footprint, high capital cost, and high energy consumption, which make it essential to discover another technology that can overcome the drawbacks [38].

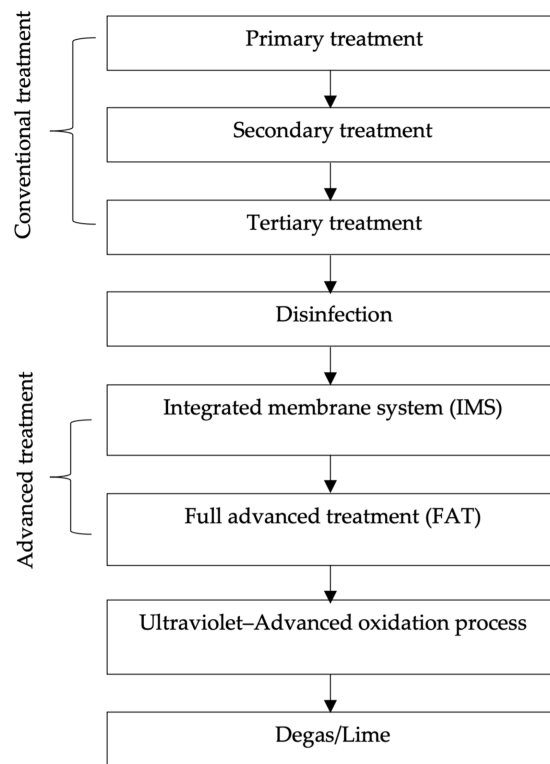


Figure 2. Industry-standard potable reuse plant.

The conventional design of the drinking water treatment process includes five common units, and four of them (coagulation, flocculation, sedimentation, and filtration) are the lines that remove suspended particles from surface water treatment plants. Filtration is the final step in the removal of suspended particles, and without it, the plants are considered untreatable. Therefore, proper control, design, and implementation of the filtration operation unit are crucial to improving the effluent quality and reducing the risk of waterborne diseases.

3. Membrane Filtration Technology

Membrane filtration is an advanced drinking water treatment that is widely used nowadays in water treatment processes, mainly for drinking water. Examples of types of membranes include microfiltration (MF), ultrafiltration (UF), nanofiltration (NF), reverse osmosis (RO), electrodialysis (ED), forward osmosis (FO), and membrane distillation (MD). Each method has its own specific range of membrane pore sizes, surface charge, and hydrophobicity that is produced from different materials ^[39]. **Table 1** shows the pore size ranges of various membrane filtration systems as compared to the size of common water contaminants.

Table 1. Contaminant with respective membrane filtration type.

Size (mm)	0.0001	0.001	0.01	0.1	1.0	10	100	1000
	Reverse osmosis							
	Nanofiltration		Ultrafiltration			Microfiltration		
Contaminant								Conventional
Metal Ions	✓	✓						
Aqueous Salts	✓	✓						
Humic Acids		✓	✓					
Viruses			✓					
Clays				✓	✓			
Assestor Fiber				✓	✓	✓		

Size (mm)	0.0001	0.001	0.01	0.1	1.0	10	100	1000
Bacteria				✓	✓	✓		
Cyst					✓	✓		
Algae					✓	✓	✓	
Sand							✓	✓

The application of membrane filtration technology to drinking water treatment on a large-scale ^[40] has received attention due to its advantages, including excellent effluent quality ^[41], simple process management ^[42], and strict solid-liquid separation with a small footprint requirement ^{[43][44]}. The technology is also easy to adapt to the existing treatment facilities ^[45], provides low energy consumption ^[11], and removes various contaminants ^[46]. The removal rate of contaminants depends on the characteristics of the membrane and the properties of the contaminant ^[36]. Aside from these benefits, the main disadvantage of this technology is the cost of the membrane itself, which can be reduced or eliminated if the membrane filtration process is handled properly. **Figure 3** shows the advantages and disadvantages of each membrane filtration treatment applied to drinking water treatment.

In general, membrane filtration can be classified into two categories: low-pressure membrane (10 to 30 psi) and high-pressure membrane (75 to 250 psi). The low-pressure membrane system includes MF and UF, while NF and RO are categorized as high-pressure membrane systems.

The low-pressure MF and UF membranes for the application of municipal surface water treatment have been studied and implemented since the 1980s. In these studies, the MF (nominal pore size of 0.2 mm) and UF (nominal pore size of 0.01 mm) have proven their high capabilities for the removal of particulate matter (turbidity) and microorganisms ^{[47][48]}. MF and UF membranes were proven to provide a barrier to microorganisms such as Giardia cysts and Cryptosporidium oocysts, while the UF was proven to be an absolute barrier to viruses due to its smaller pore size of 0.01 mm ^{[49][50]}. Previous studies ^{[51][52]} also demonstrated that low-pressure membranes were able to treat turbidity efficiently using pilot and full-scale plants. The low-pressure MF and UF membrane systems provide high performance for the removal of contaminants from surface water, and other advantages include a smaller footprint, low chemical usage, and more automation. However, the limitation of membrane technology, including MF and UF, is the high cost of membrane replacement and the lower effectiveness in removing dissolved organic matter in the treated water. The study of modified MF membrane technology is reported by Sinclair et al. ^[53], and it showed an improvement in reducing cost as they do not require any external driving force. Unfortunately, the modification resulted in an approximately 22% loss of membrane permeability.

Meanwhile, He et al. ^[54] published a study on improved UF technology in which they combined heterogeneous catalytic ozonation and a UF membrane filtration technique for the long-term degradation of bisphenol A (BPA) and humic acid (HA). Results have shown improvements in removal efficiency, reduction of membrane resistance, and mitigation of membrane fouling. Another study concerning UF was reported by Chew et al. ^[55], which compared and evaluated industrial-scale UF with conventional drinking water treatment systems. The study showed that UF systems can provide reliable filtrate quality even with the existence of fluctuation in the raw water quality. In addition, the UF system offers promising sustainability, with no coagulant required for high-quality filtrate and non-toxic sludge discharge.

High-pressure NF and RO membranes can provide an alternative method for removing organic and inorganic matter. The NF process is already known for its capabilities in the removal of total organic carbon (TOC) in surface water treatment ^[56]. This process has been implemented in several drinking water industries ^{[57][58][59]}. In an experiment conducted using pre-ozonation as a pre-treatment process for NF membranes proposed by Vatankhah et al. ^[60], it was found that pre-ozonation with a low specific ozone dose could effectively mitigate a significant portion of fouling. However, the removal performance of dissolved organic carbon (DOC) of the NF membrane did not show a substantial change, which may be due to the relatively low applied ozone dose. The RO process is applied for drinking water treatment, whether the source water comes from seawater, brackish water, or groundwater ^[21]. However, RO has a problem with the ability of suspended solids, colloidal material, and dissolved ions in raw water to foul the system ^[61]. A study conducted by Touati et al. ^[62] combined UF, NF, and RO processes for isotonic and drinking water treatment. Results showed that the UF process used as pre-treatment was able to eliminate natural organic matter (NOM), while the NF process was able to characterize the fouling mechanism. The overall performance's energy consumption is determined by salt rejection during the NF process.

Apart from RO, ED is another process that can be used to treat brackish water with high performance and energy efficiency ^{[63][64]}. The process involved the transfer of electrolytes or ions through a solution and membranes based on an applied electric field as the driving force ^[65]. Walha et al. ^[66] investigated the use of the NF, RO, and ED processes in

producing drinking water from a brackish water source. The results showed the treatment based on RO and ED processes is more efficient, as shown by the high rejection of inorganic matters present in the feed waters. The concentration of ions in the permeate flux can achieve World Health Organization (WHO) standards, and it is more economical than the NF process.

Forward osmosis (FO) and membrane distillation (MD) processes are driven by heat, which is different from the pressure-driven process usually used for potable water reuse [67]. FO processing operates at low or no hydraulic pressure, which may reduce irreversible fouling and achieve high rejection of contaminants [68]. However, Li et al. [69] reported that the water flux produced by the FO process was still inadequate compared to the RO process under a similar applied pressure. FO processes involve a permeable membrane and two solutions, known as feed and draw solutions. The feed and draw solution consists of different concentrations that produce the osmotic pressure gradient that acts as the driving force for water permeation across a semi-permeable membrane [70]. An experiment conducted by Tow et al. [71] studied the fouling propensity between RO, FO, and MD. The experiment was conducted using a single membrane module and showed that both FO and MD exhibit a significant advantage in fouling resistance but neither of them performed well with both organic and inorganic foulants.

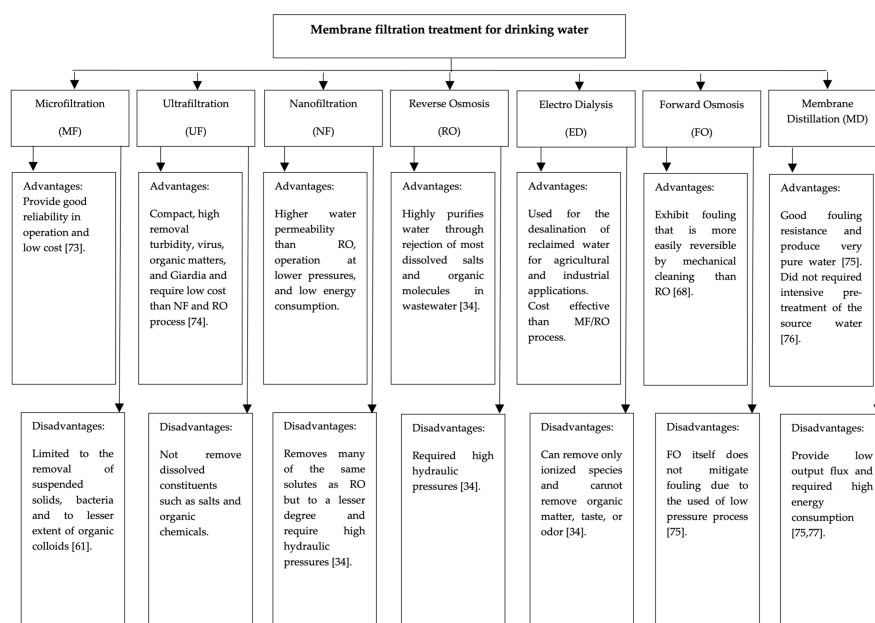


Figure 3. Available membrane filtration treatment for drinking water [34][61][68][72][73][74][75][76][77].

References

- Umar, M.; Kambai, J.; Mohammed, I.B.; Oko, J.O.; Obafemi, A.A.; Murtala, I.; Ajiya, K.G.; Yaya, A.A.; Abdulkarim, I.M.; Akafyi, D.E.; et al. Bacteriological Quality Assessment and Antibigram Profile of Bacteria Associated with Sachet Drinking Water Sold at Zaria, Northern Nigeria. *Int. J. Pathog. Res.* 2019, 2, 1–13.
- World Health Organization. *Drinking-Water, Sanitation and Hygiene in the Western Pacific Region*; World Health Organization: Geneva, Switzerland, 2018.
- Yoo, S.S.; Chu, K.H.; Choi, I.-H.; Mang, J.S.; Ko, K.B. Operating cost reduction of UF membrane filtration process for drinking water treatment attributed to chemical cleaning optimization. *J. Environ. Manag.* 2018, 206, 1126–1134.
- Saleem, M.; Alibardi, L.; Lavagnolo, M.C.; Cossu, R.; Spagni, A. Effect of filtration flux on the development and operation of a dynamic membrane for anaerobic wastewater treatment. *J. Environ. Manag.* 2016, 180, 459–465.
- Esplugas, S.; Bila, D.M.; Krause, L.G.T.; Dezotti, M. Ozonation and advanced oxidation technologies to remove endocrine disrupting chemicals (EDCs) and pharmaceuticals and personal care products (PPCPs) in water effluents. *J. Hazard. Mater.* 2007, 149, 631–642.
- Zhang, Y.; Gao, X.; Smith, K.; Inial, G.; Liu, S.; Conil, L.B.; Pan, B. Integrating water quality and operation into prediction of water production in drinking water treatment plants by genetic algorithm enhanced artificial neural network. *Water Res.* 2019, 164, 114888.
- Zhang, H.; Zheng, L.; Li, Z.; Pi, K.; Deng, Y. One-step Ferrate(VI) treatment as a core process for alternative drinking water treatment. *Chemosphere* 2019, 242, 125134.

8. Omar, I.A.; Aziz, S.Q. Research on Performance Evaluation and Improvement of Ifraz-2 Water Treatment Plant. *Recent Res. Adv. Biol.* 2021, 7, 9–22.
9. Abd Nasier, M.; Adel Abdulrazzaq, K. Conventional Water Treatment Plant, Principles, and Important Factors Influence on The Efficiency. *Des. Eng.* 2021, 8, 16009–16027.
10. Heydarifard, S.; Taneja, K.; Bhanjana, G.; Dilbaghi, N.; Nazhad, M.M.; Kim, K.-H.; Kumar, S. Modification of cellulose from paper for use as a high-quality biocide disinfectant filter for drinking water. *Carbohydr. Polym.* 2018, 181, 1086–1092.
11. Teodosiu, C.; Gilca, A.-F.; Barjoveanu, G.; Fiore, S. Emerging pollutants removal through advanced drinking water treatment: A review on processes and environmental performances assessment. *J. Clean. Prod.* 2018, 197, 1210–1221.
12. Lund, M.D.; Bording, T.; Andersen, T.R. Electrical resistivity tomography applied for monitoring backwash efficiency in drinking water filters. *Water Supply* 2022, 22, 6660–6671.
13. Jalil, A.T.; Emad, H.; Qurabiy, A.; Dilly, S.H.; Surendar, A.; Kadhim, M.M.; Aljeboree, A.M. CuO/ZrO₂ Nanocomposites: Facile Synthesis, Characterization and Photocatalytic Degradation of Tetracycline Antibiotic. *J. Nanostruct.* 2021, 11, 333–346.
14. Rigobello, E.S.; Dantas, A.D.B.; Di Bernardo, L.; Vieira, E.M. Removal of diclofenac by conventional drinking water treatment processes and granular activated carbon filtration. *Chemosphere* 2013, 92, 184–191.
15. Murugan, R.; Ram, C.G. Energy efficient drinking water purification system using TiO₂ solar reactor with traditional methods. *Mater. Today Proc.* 2018, 5, 415–421.
16. Zheng, X.; Ernst, M.; Jekel, M. Pilot-scale investigation on the removal of organic foulants in secondary effluent by slow sand filtration prior to ultrafiltration. *Water Res.* 2010, 44, 3203–3213.
17. Sabogal-Paz, L.P.; Campos, L.C.; Bogush, A.; Canales, M. Household slow sand filters in intermittent and continuous flows to treat water containing low mineral ion concentrations and Bisphenol A. *Sci. Total Environ.* 2020, 702, 135078.
18. Ahammed, M.M.; Davra, K. Performance evaluation of biosand filter modified with iron oxide-coated sand for household treatment of drinking water. *Desalination* 2011, 276, 287–293.
19. Mizuta, K. Removal of nitrate-nitrogen from drinking water using bamboo powder charcoal. *Bioresour. Technol.* 2004, 95, 255–257.
20. Zhang, H.; Zhu, G.; Jia, X.; Ding, Y.; Zhang, M.; Gao, Q.; Hu, C.; Xu, S. Removal of microcystin-LR from drinking water using a bamboo-based charcoal adsorbent modified with chitosan. *J. Environ. Sci.* 2011, 23, 1983–1988.
21. Omarova, A.; Tussupova, K.; Berndtsson, R.; Kalishev, M.; Sharapatova, K. Protozoan parasites in drinking water: A system approach for improved water, sanitation and hygiene in developing countries. *Int. J. Environ. Res. Public Health* 2018, 15, 495.
22. Shirazi, S.; Lin, C.-J.; Chen, D. Inorganic fouling of pressure-driven membrane processes—A critical review. *Desalination* 2010, 250, 236–248.
23. Xiang, H.; Min, X.; Tang, C.-J.; Sillanpää, M.; Zhao, F. Recent advances in membrane filtration for heavy metal removal from wastewater: A mini review. *J. Water Process. Eng.* 2022, 49, 103023.
24. Chadha, U.; Selvaraj, S.K.; Thanu, S.V.; Chalapadath, V.; Abraham, A.M.; Zaiyan, M.; Manikandan, M.; Paramasivam, V. A review of the function of using carbon nanomaterials in membrane filtration for contaminant removal from wastewater. *Mater. Res. Express* 2022, 9, 012003.
25. Zhang, W.; Xu, X.; Zhang, G.; Jin, S.; Dong, L.; Gu, P. Treatment of Membrane Cleaning Wastewater from Thermal Power Plant Using Membrane Bioreactor. *Membranes* 2022, 12, 755.
26. Deng, L.; Guo, W.; Ngo, H.H.; Zhang, X.; Chen, C.; Chen, Z.; Cheng, D.; Ni, S.-Q.; Wang, Q. Recent advances in attached growth membrane bioreactor systems for wastewater treatment. *Sci. Total Environ.* 2021, 808, 152123.
27. Oh, Y.; Sim, D.; Jeong, S.; Lee, J.; Son, H.; Bae, H.; Jeong, S. Multifunctional in-situ ferrate treatment and its removal mechanisms of membrane bioreactor residual pollutants. *Chem. Eng. J.* 2022, 446, 136956.
28. Buse, H.Y.; Hall, J.S.; Hunter, G.L.; Goodrich, J.A. Differences in UV-C LED Inactivation of *Legionella pneumophila* Serogroups in Drinking Water. *Microorganisms* 2022, 10, 352.
29. Sultan, T.; Ahmad, Z.; Hayat, K.; Chaudhry, I.A. Computational analysis of three lamp close conduit water disinfection UV reactor. *Int. J. Environ. Sci. Technol.* 2021, 19, 4393–4406.
30. Schmalwieser, A.W.; Hirschmann, G.; Eggers, J.; Sommer, R. A standardized method to measure the longitudinal UV transmittance of low-pressure-lamps in dependence of water temperature. *Water Supply* 2021, 22, 900–916.

31. Venkatesan, A.K.; Lee, C.-S.; Gobler, C.J. Hydroxyl-radical based advanced oxidation processes can increase perfluorinated alkyl substances beyond drinking water standards: Results from a pilot study. *Sci. Total Environ.* 2022, 847, 157577.
32. Giwa, A.; Yusuf, A.; Balogun, H.A.; Sambudi, N.S.; Bilad, M.R.; Adeyemi, I.; Chakraborty, S.; Curcio, S. Recent advances in advanced oxidation processes for removal of contaminants from water: A comprehensive review. *Process. Saf. Environ. Prot.* 2020, 146, 220–256.
33. Wang, J.Z.; Summers, R.S.; Miltner, R.J. Biofiltration performance: Part 1, relationship to biomass. *Am. Water Work. Assoc.* 2016, 87, 55–63.
34. Warsinger, D.M.; Chakraborty, S.; Tow, E.W.; Plumlee, M.H.; Bellona, C.; Loutatidou, S.; Karimi, L.; Mikelonis, A.M.; Achilli, A.; Ghassemi, A.; et al. A review of polymeric membranes and processes for potable water reuse. *Prog. Polym. Sci.* 2018, 81, 209–237.
35. Sillanpää, M.; Ncibi, M.C.; Matilainen, A. Advanced oxidation processes for the removal of natural organic matter from drinking water sources: A comprehensive review. *J. Environ. Manag.* 2018, 208, 56–76.
36. Snyder, S.; Adham, S.; Redding, A.M.; Cannon, F.S.; Decarolis, J.; Oppenheimer, J.; Wert, E.C.; Yoon, Y. Role of membranes and activated carbon in the removal of endocrine disruptors and pharmaceuticals. *Desalination* 2007, 202, 156–181.
37. Oh, K.S.; Leong, J.Y.C.; Poh, P.E.; Chong, M.N.; Von Lau, E. A review of greywater recycling related issues: Challenges and future prospects in Malaysia. *J. Clean. Prod.* 2018, 171, 17–29.
38. Holloway, R.W.; Miller-Robbie, L.; Patel, M.; Stokes, J.R.; Munakata-Marr, J.; Dadakis, J.; Cath, T.Y. Life-cycle assessment of two potable water reuse technologies: MF/RO/UV-AOP treatment and hybrid osmotic membrane bioreactors. *J. Membr. Sci.* 2016, 507, 165–178.
39. Gupta, V.; Ali, I. Chapter 5—Water Treatment by Membrane Filtration Techniques. *Environ. Water* 2013, 2013, 135–154.
40. Chew, C.M.; Aroua, M.; Hussain, M.; Ismail, W.W. Practical performance analysis of an industrial-scale ultrafiltration membrane water treatment plant. *J. Taiwan Inst. Chem. Eng.* 2015, 46, 132–139.
41. Song, J.; Zhang, Z.; Zhang, X. A comparative study of pre-ozonation and in-situ ozonation on mitigation of ceramic UF membrane fouling caused by alginate. *J. Membr. Sci.* 2017, 538, 50–57.
42. Zhang, X.; Wang, D.K.; Lopez, D.R.S.; da Costa, J.C.D. Fabrication of nanostructured TiO₂ hollow fiber photocatalytic membrane and application for wastewater treatment. *Chem. Eng. J.* 2014, 236, 314–322.
43. Kimura, K.; Hane, Y.; Watanabe, Y.; Amy, G.; Ohkuma, N. Irreversible membrane fouling during ultrafiltration of surface water. *Water Res.* 2004, 38, 3431–3441.
44. Kingsbury, B.F.; Li, K. A morphological study of ceramic hollow fibre membranes. *J. Membr. Sci.* 2009, 328, 134–140.
45. Le, N.L.; Nunes, S.P. Materials and membrane technologies for water and energy sustainability. *Sustain. Mater. Technol.* 2016, 7, 1–28.
46. Im, D.; Nakada, N.; Fukuma, Y.; Tanaka, H. Effects of the inclusion of biological activated carbon on membrane fouling in a combined process of ozonation, coagulation and ceramic membrane filtration for water reclamation. *Chemosphere* 2018, 220, 20–27.
47. Sandoval, A.D.O.; Brião, V.B.; Fernandes, V.M.C.; Hemkemeier, A.; Friedrich, M.T. Stormwater management by microfiltration and ultrafiltration treatment. *J. Water Process. Eng.* 2019, 30, 100453.
48. Bagga, A.; Chellam, S.; Clifford, D.A. Evaluation of iron chemical coagulation and electrocoagulation pretreatment for surface water microfiltration. *J. Membr. Sci.* 2008, 309, 82–93.
49. Hsu, B.-M.; Yeh, H.-H. Removal of Giardia and Cryptosporidium in drinking water treatment: A pilot-scale study. *Water Res.* 2003, 37, 1111–1117.
50. Bray, R.; Jankowska, K.; Kulbat, E.; Łuczkiewicz, A.; Sokołowska, A. Ultrafiltration process in disinfection and advanced treatment of tertiary treated wastewater. *Membranes* 2021, 11, 221.
51. Brehant, A.; Bonnelye, V.; Perez, M. Comparison of MF/UF pretreatment with conventional filtration prior to RO membranes for surface seawater desalination. *Desalination* 2002, 144, 353–360.
52. Shirakawa, D.; Shirasaki, N.; Matsushita, T.; Matsui, Y.; Yamashita, R.; Matsumura, T.; Koriki, S. Evaluation of reduction efficiencies of pepper mild mottle virus and human enteric viruses in full-scale drinking water treatment plants employing coagulation-sedimentation-rapid sand filtration or coagulation-microfiltration. *Water Res.* 2022, 213, 118160.
53. Sinclair, T.; Robles, D.; Raza, B.; Hengel, S.V.D.; Rutjes, S.; Husman, A.D.R.; de Grooth, J.; de Vos, W.; Roesink, H.D. Virus reduction through microfiltration membranes modified with a cationic polymer for drinking water applications. *Colloids Surf. A* 2018, 551, 33–41.

54. He, Z.; Xia, D.; Huang, Y.; Tan, X.; He, C.; Hu, L.; He, H.; Zeng, J.; Xu, W.; Shu, D. 3D MnO₂ hollow microspheres ozon e-catalysis coupled with flat-plate membrane filtration for continuous removal of organic pollutants: Efficient heterogeneous catalytic system and membrane fouling control. *J. Hazard. Mater.* 2018, 344, 1198–1208.
55. Chew, C.M.; Aroua, M.K.; Hussain, M.A.; Ismail, W.M.Z.W. Evaluation of ultrafiltration and conventional water treatment systems for sustainable development: An industrial scale case study. *J. Clean. Prod.* 2016, 112, 3152–3163.
56. Sengur-Tasdemir, R.; Urper-Bayram, G.M.; Turken, T.; Ates-Genceli, E.; Tarabara, V.V.; Koyuncu, I. Hollow fiber nanofiltration membranes for surface water treatment: Performance evaluation at the pilot scale. *J. Water Process. Eng.* 2021, 42, 102100.
57. Al-Amoudi, A.S. Factors affecting natural organic matter (NOM) and scaling fouling in NF membranes: A review. *Desalination* 2010, 259, 1–10.
58. Weber, R.; Chmiel, H.; Mavrov, V. Characteristics and application of new ceramic nanofiltration membranes. *Desalination* 2003, 157, 113–125.
59. Mohsen, M.S.; Jaber, J.O.; Afonso, M.D. Desalination of brackish water by nanofiltration and reverse osmosis. *Desalination* 2003, 157, 167.
60. Vatankhah, H.; Murray, C.C.; Brannum, J.W.; Vanneste, J.; Bellona, C. Effect of pre-ozonation on nanofiltration membrane fouling during water reuse applications. *Sep. Purif. Technol.* 2018, 205, 203–211.
61. del Pino, M.P.; Durham, B. Wastewater Reuse Through Dual Membrane Processes: Opportunities for sustainable water resources. *Desalination* 1999, 124, 271–277.
62. Touati, K.; Gzara, L.; Mahfoudhi, S.; Bourezgui, S.; Hafiane, A.; Elfil, H. Treatment of coastal well water using ultrafiltration-nanofiltration-reverse osmosis to produce isotonic solutions and drinking water: Fouling behavior and energy efficiency. *J. Clean. Prod.* 2018, 200, 1053–1064.
63. Ortiz, J.; Sotoca, J.; Expósito, E.; Gallud, F.; García-García, V.; Montiel, V.; Aldaz, A. Brackish water desalination by electrodialysis: Batch recirculation operation modeling. *J. Membr. Sci.* 2005, 252, 65–75.
64. Liu, R.; Wang, Y.; Wu, G.; Luo, J.; Wang, S. Development of a selective electrodialysis for nutrient recovery and desalination during secondary effluent treatment. *Chem. Eng. J.* 2017, 322, 224–233.
65. Sirivedhin, T.; McCue, J.; Dallbauman, L. Reclaiming produced water for beneficial use: Salt removal by electrodialysis. *J. Membr. Sci.* 2004, 243, 335–343.
66. Walha, K.; Ben Amar, R.; Firdaous, L.; Quéméneur, F.; Jaouen, P. Brackish groundwater treatment by nanofiltration, reverse osmosis and electrodialysis in Tunisia: Performance and cost comparison. *Desalination* 2007, 207, 95–106.
67. Bogler, A.; Lin, S.; Bar-Zeev, E. Biofouling of membrane distillation, forward osmosis and pressure retarded osmosis: Principles, impacts and future directions. *J. Membr. Sci.* 2017, 542, 378–398.
68. Lee, S.; Boo, C.; Elimelech, M.; Hong, S. Comparison of fouling behavior in forward osmosis (FO) and reverse osmosis (RO). *J. Membr. Sci.* 2010, 365, 34–39.
69. Li, D.; Yan, Y.; Wang, H. Recent advances in polymer and polymer composite membranes for reverse and forward osmosis processes. *Prog. Polym. Sci.* 2016, 61, 104–155.
70. Suwaileh, W.A.; Johnson, D.J.; Sarp, S.; Hilal, N. Advances in forward osmosis membranes: Altering the sub-layer structure via recent fabrication and chemical modification approaches. *Desalination* 2018, 436, 176–201.
71. Tow, E.W.; Warsinger, D.M.; Truworthy, A.M.; Swaminathan, J.; Thiel, G.P.; Zubair, S.M.; Myerson, A.S.; V, J.H.L. Comparison of fouling propensity between reverse osmosis, forward osmosis, and membrane distillation. *J. Membr. Sci.* 2018, 556, 352–364.
72. Hamid, K.I.A.; Sanciolo, P.; Gray, S.; Duke, M.; Muthukumaran, S. Comparison of the effects of ozone, biological activated carbon (BAC) filtration and combined ozone-BAC pre-treatments on the microfiltration of secondary effluent. *Sep. Purif. Technol.* 2019, 215, 308–316.
73. Gao, W.; Liang, H.; Ma, J.; Han, M.; Chen, Z.-L.; Han, Z.-S.; Li, G.-B. Membrane fouling control in ultrafiltration technology for drinking water production: A review. *Desalination* 2011, 272, 1–8.
74. Tow, E.W.; V, J.H.L. Unpacking compaction: Effect of hydraulic pressure on alginate fouling. *J. Membr. Sci.* 2017, 544, 221–233.
75. Saffarini, R.B.; Summers, E.K.; Arafat, H.A.; V, J.H.L. Economic evaluation of stand-alone solar powered membrane distillation systems. *Desalination* 2012, 299, 55–62.
76. Gil, J.D.; Roca, L.; Ruiz-Aguirre, A.; Zaragoza, G.; Berenguel, M. Optimal operation of a Solar Membrane Distillation pilot plant via Nonlinear Model Predictive Control. *Comput. Chem. Eng.* 2017, 109, 151–165.

77. Ullah, R.; Khraisheh, M.; Esteves, R.J.; Jr, J.T.M.; AlGhouti, M.; Gad-El-Hak, M.; Tafreshi, H.V. Energy efficiency of direct contact membrane distillation. *Desalination* 2018, 433, 56–67.
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