

# Biological Processes for Dairy Wastewater Treatment

Subjects: Engineering, Chemical

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Dairy wastewater (DWW) has a high fat content along with a high chemical oxygen demand (COD), which makes it problematic to dispose of. Biological treatment processes have shown great success in the remediation of this water. However, these are not without their shortfalls. A variety of biological processes have been listed here as well as suggestions to improve their effectiveness. To improve the treatment efficiency, there are two main lines of thinking: First, to optimise the process through the manipulation of the physical parameters of the systems (e.g., temperature, pH, hydraulic retention time (HRT), agitation, etc.). Second, to add either pre-treatment or post-treatment processes to the main process to increase the effectiveness of the entire process overall.

Keywords: dairy ; wastewater ; biological ; optimization ; treatment

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## 1. Introduction

The disposal of wastewater is a topic of great importance as improper disposal may lead to a detrimental effect on the lives of people as well as the degradation of the environment. Many laws and regulations have been put in place to ensure that the wastewater being released from various industrial processes meets a minimum standard of safety <sup>[1]</sup>. Different industries release varying quantities of wastewater with different types of contamination. The release of dairy wastewater (DWW) into water systems without proper treatment causes an array of environmental issues due to its high organic content. The decomposition of DWW in a water source causes the dissolved oxygen (O<sub>2</sub>) inside the water body to drop as the microbes (especially algae) flourish in the nutrient-rich environment. This results in a hypoxic environment which is unsuitable for aquatic animal and plant life. The increased algal growth produces elevated levels of CO<sub>2</sub> which lowers the pH of the body of water, which in turn slows the growth of fish and shellfish. The addition of DWW into fresh water also provides a breeding ground for various insects, which can upset the balance of the surrounding ecosystem. The degradation of casein also causes the formation of a sludge which is harmful to aquatic organisms <sup>[2]</sup>. This impacts the health of aquatic life and the potential for fishing which has a negative economic impact. In addition to these environmental implications, the processing of dairy products and the release of DWW can lead to several social impacts. These stem from two main sources, namely, the discharge of untreated or partially treated wastewater into nearby water bodies and the odour produced by the plant and effluents. The contamination caused by the release of DWW can promote the spread of bacteria such as *Salmonella* and *E. coli*, which can cause illness in humans and animals which use the water sources. Once contracted, these illnesses can easily be spread throughout the community.

## 2. Treatment Options

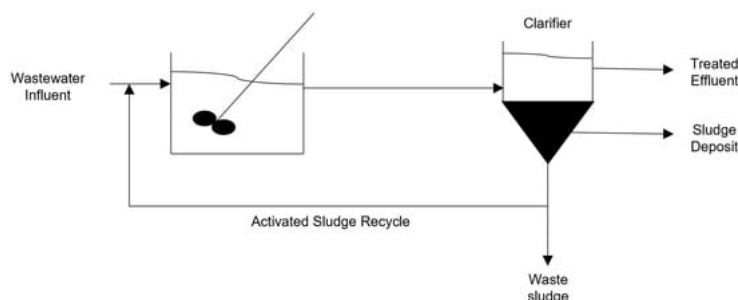
Biological treatments have been highly effective in reducing the organic load in various types of wastewaters. The nature of microbial systems allows for biological treatment methods to be versatile with respect to the type of wastewater being remediated. However, the physical conditions of the system, such as pH, temperature, and O<sub>2</sub> levels, need to be carefully controlled to avoid damaging the microbial consortia. It is also necessary to ensure that certain compounds do not exceed the microbes' tolerance, as an excess would kill them. One of the downsides to biological treatment methods is that there are certain substances (such as heavy metals) which cannot be processed as they would poison the microbial colonies. There are two main types of biological treatment methods: aerobic and anaerobic treatment. Anaerobic and aerobic treatments are commonly used in conjunction with one another as they each treat different components of the DWW. A combination of aerobic and anaerobic is a suitable solution for treating different components of DWW, replacing traditional one-phase biological systems <sup>[3]</sup>.

### 2.1. Aerobic Treatment Processes

Aerobic treatment occurs when the microbes are exposed to air and thus oxygen whilst anaerobic occurs without the presence of air. Aerobic treatment is effectively used to reduce the BOD<sub>5</sub> as well as the phosphorous and nitrogen content in dairy wastewater. Aerobic processes are additionally more cost-effective than anaerobic ones for the removal of fats from DWW <sup>[3]</sup>. Importantly, the removal of nitrogen is largely associated with the oxidation of ammonia into nitrates, which in turn reduces the odour of the DWW. Aerobic systems are associated with high biomass growth and are susceptible to clogging if solely used to treat DWW <sup>[4]</sup>.

## 2.2. Activated Sludge

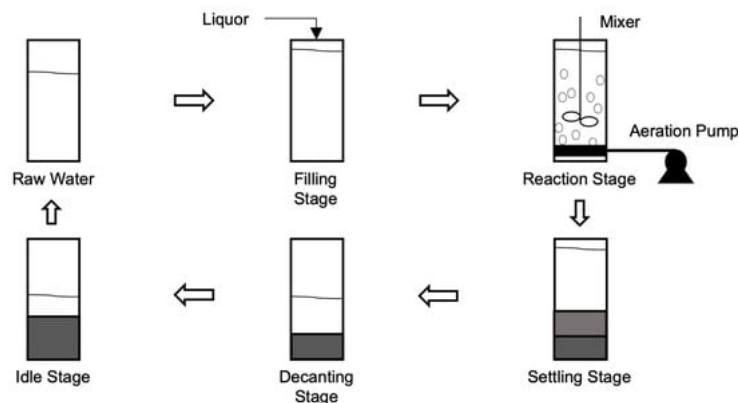
Activated sludge treatment methods are processes which introduce microbes into the wastewater stream to both treat the water and stimulate the growth of microbes. The microbial colonies are then separated from the bulk stream using a clarifier or filter, with a portion of the sludge being reintroduced to the reactor. A schematic representation of an activated sludge system can be seen in **Figure 1**. These methods are preferred for the removal of carbon, nitrates, and ammonia [5]. The constant recirculation of sludge allows for the microbes to acclimatize to the waste stream composition and increases the performance of the system, the longer it is operated for. These systems are also effective at breaking down complicated compounds such as lactose, FOG, and proteins [6].



**Figure 1.** Schematic diagram of an activated sludge process.

## 2.3. Sequencing Batch Reactors

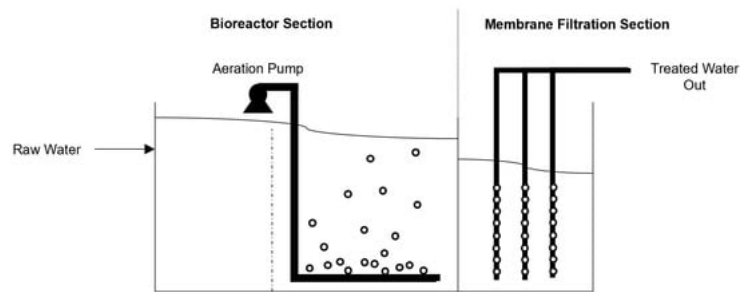
Sequencing batch reactors (SBRs) comprise a series of process steps within a single bioreactor. These steps occur in the sequence of filling, reacting, settling, decanting, and idling. The filling stage is the addition of the DWW and liquor, which contains microbes. The reaction stage can consist of cycles of aerobic and anaerobic phases. This can be achieved by alternating between periods of aerating and no aeration. For the settling stage, the aeration and mixing are stopped for the suspended solids to settle and be easily separated from the treated water. The decanting stage involves the removal of the supernatant fluid (treated wastewater). The idle stage is mostly needed for multi-reactor systems where there is a time delay between the filling stage in between units [3]. A schematic representation of a SBR system can be seen in **Figure 2**. SBRs have been observed to reduce COD by up to 95%, total solids by up to 60%, and up to 40% of the total nitrogen [7] from DWW. These systems do suffer from relatively low removal rates for phosphorous, nitrogen, and limited solid removal.



**Figure 2.** Schematic of an SBR treatment system.

## 2.4. Membrane Bioreactor

Membrane bioreactors (MBR) incorporate the principle of membrane filtration with a bioreactor. A schematic representation of an MBR can be seen in **Figure 3**. There are two main configurations of MBR, one where the membrane is submerged within the reactor and one where the membrane is located externally with a recycle back into the reactor. MBRs with submerged membranes are more popular as they are compact and have low energy usage [8]. One of the downsides of submerged membranes is that they are more prone to fouling than the external configuration. This is an issue that must be dealt with for most membrane systems. However, with an MBR, soluble microbial products add to the risk of fouling. This can be somewhat mitigated by treating the membranes with sodium metabisulfite to prevent microbial growth on the membrane [9].



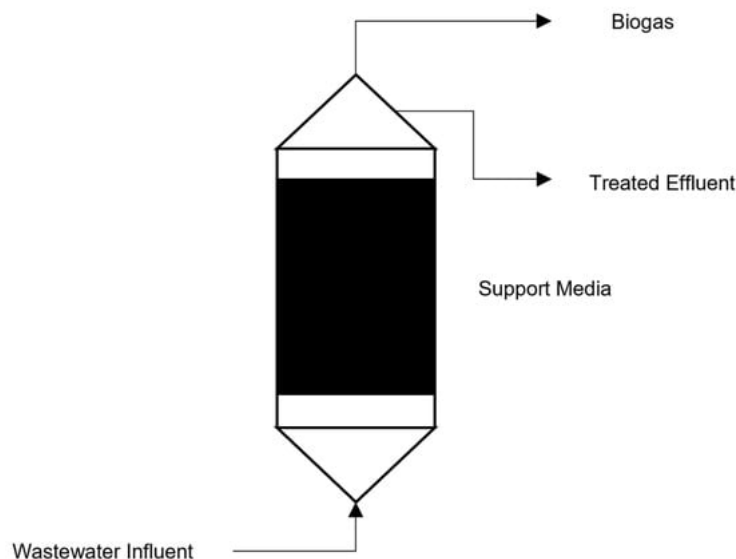
**Figure 3.** Schematic diagram of a membrane bioreactor.

## 2.5. Anaerobic Processes

Anaerobic treatment processes are primarily aimed at reducing the organic content of wastewater and are suited for the remediation of wastewater with high organic loads [4]. Anaerobic systems are also effective at decreasing the nitrites and nitrates by reducing them into nitrogen gas. These systems are, however, less effective at removing the FOGs, ammonia, and phosphorous [5] and may require further downstream treatment or a suitable pre-treatment method to meet discharge standards. FOGs and other complex polymeric compounds are difficult to break down using anaerobic consortia and usually require a hydrolysis or oxidation step to break them down into smaller, soluble molecules such as short-chain fatty acids and alcohols [10]. Hydrolysis is the process whereby a complex polymeric organic compound is broken down into smaller soluble compounds in the presence of water [11]. This process can be catalysed by the addition of acids or enzymes to the solution. This is relevant for anaerobic processes as the lipid content is usually a limiting factor for performance [12].

## 2.6. Anaerobic Filters

Anaerobic filters (AFs) are reactors that are packed with a porous support media with biomass in the packing void. AF reactors contain at least two filtration chambers within the system, with the filter media used to prevent biomass from passing through. A side effect of these filtration chambers is enhanced solid removal, which can be observed [2]. A schematic representation of an AF system can be seen in **Figure 4**.



**Figure 4.** Schematic diagram of an anaerobic filter.

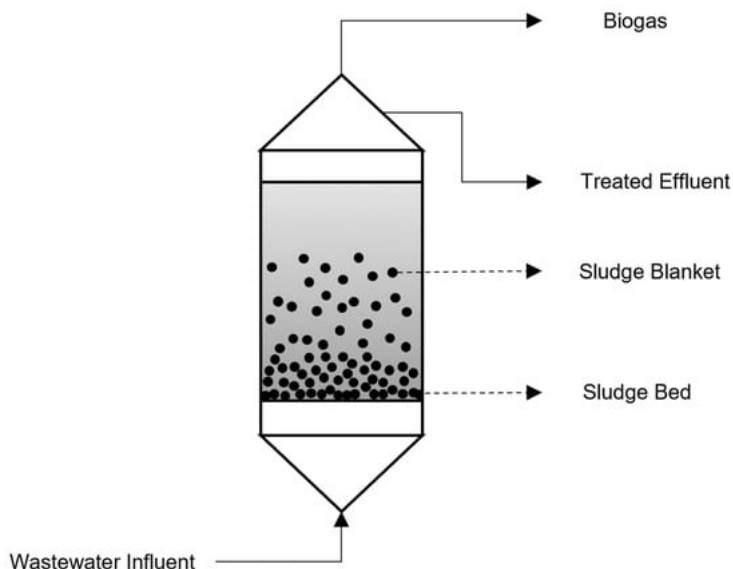
## 2.7. Anaerobic Packed Bed Reactors

Anaerobic packed bed reactors (APBRs) utilize sludge which has been immobilized, through which the wastewater is pumped either in an up-flow or downflow configuration. This reactor uses a packing material at the top and/or bottom to fix the biomass in place. This packing material is intended to support biofilm formation and not to act as a physical barrier for the solids contained within the DWW. This is a very similar configuration to that of an anaerobic filter; however, there are no filtration chambers present within the APBR configuration. An issue with this technology is that it can be prone to clogging due to the accumulation of biomass. This issue can be bypassed by allowing biomass to flow out of the reactor with the effluent stream, necessitating further downstream treatment.

## 2.8. Up-Flow Anaerobic Sludge Blanket Reactors

Up-flow anaerobic sludge blanket reactors (UASBRs) are some of the most used systems for the treatment of DWW. These types of reactors are fed wastewater through the bottom of a reactor column which is packed with anaerobic sludge granules that have been somewhat fixed in place. The treated effluent is then extracted from the top end of the reactor,

with the methane being siphoned off through a vent at the top of the reactor. A schematic representation of an UASBR can be seen in **Figure 5**.



**Figure 5.** Schematic diagram of an up-flow Anaerobic Sludge Blanket Reactor (UASBR).

## 2.9. Aerobic–Anaerobic Systems

Aerobic–anaerobic systems have been widely investigated for various wastewaters with high organic contents. The combination of these two technologies allows for a more complete remediation of DWW as each stage focuses on different pollutants within the wastewater. The aerobic stage effectively reduces the ammonia, phosphorous,  $\text{H}_2\text{S}$ , and  $\text{BOD}_5$  content, whereas the anaerobic stage reduces the COD and nitrate content of the water [3]. In theory, it should be possible to achieve removal yields from a two-stage system similar to the yields that can be achieved by the individual processes. This is due to the different pollutants that are being processed by each individual system. In a practical study of an anaerobic filter combined with an aerobic one, an SBR was able to achieve a COD removal of 90% from a high-strength dairy stream with an influent COD content of approximately 8000 g/L [13].

## 3. The Effects of Experimental Parameters on the Performance of Biological Treatment Systems

### 3.1. pH Adjustment

There are three main groups of microorganisms which are found in AD: acidogenic (acid producing), acetogenic (acetate-producing), and methanogenic (methane-producing). Each of these has a pH range which it prefers. Acidogens have an optimal pH range between 5.2 and 6.5; acetogens from 6.6 to 7.6; and methanogens from 7.5 to 8.5 [14]. The pH of the system should be carefully controlled because an imbalance, either too high or too low, can lead to an overproduction or accumulation of certain compounds, which can then result in digester failure [15]. It is important to select an appropriate pH based on the specific need of the system. However, the reactions which occur during the remediation of DWW produce volatile fatty acids (VFAs) from the breakdown of FOGs, which alter the pH as the process occurs.

### 3.2. Dissolved $\text{O}_2$ Levels

Aerobic digestion is enhanced by higher levels of dissolved  $\text{O}_2$ ; it therefore follows that anaerobic digestion will be enhanced by the removal of dissolved  $\text{O}_2$  from the wastewater. One straightforward approach to eliminate dissolved oxygen is to pass another gas through the wastewater. Some potential deoxygenating agents for anaerobic digestion are nitrogen or biogas. Biogas is a convenient option as an amount of  $\text{CO}_2$  and methane is produced by anaerobic digestion. However, the use of biogas poses an environmental issue due to methane being a greenhouse gas. Nitrogen gas is a suitable gas to use as it is non-reactive and does not pose any inherent environmental risks [15].

### 3.3. Hydraulic Retention Time

Hydraulic retention time (HRT) refers to the average time that a liquid substrate will spend within a reactor. HRT is largely dependent on the type of reactor being used and the organic load of the wastewater being treated. A small HRT can sometimes result in high biomass washout, whereas a long HRT can require large reactor volumes [15]. The growth of the various groups of microorganisms used in AD are favoured by different HRTs. Methanogens prefer a longer HRT period as opposed to acidogenic bacteria, which prefer low HRTs. The determination of optimal hydraulic retention time (HRT) is a complex decision that requires the consideration of both the process and the desired outcomes, and it should be evaluated on a case-by-case basis.

### 3.4. Aeration and Agitation

An important factor in aerobic treatments is the aeration which is introduced into the system. It is the driving force behind the reactions which aerobic digestion is utilized for, namely, the oxidation of ammonia and the breakdown of complex organic compounds. Therefore, an optimal aeration regime is essential for the performance of any treatment method used. At lower aeration levels, it was observed that there was a significant decrease in COD removal efficiency and an even more drastic decrease in the efficiency of ammonia removal [16]. This indicates that the available oxygen is being used to oxidise the carbon substrates in the wastewater as opposed to the ammonia. The most likely explanation is that ammonia requires a larger amount of  $O_2$  when compared to the organic components.

### 3.5. Temperature

The temperature plays a critical role in the reactions involved in wastewater remediation. Different temperature ranges can have both favourable and unfavourable effects on the performance of a bioreactor. High temperatures can cause the denaturation of proteins within the microbes, causing them to lose their enzymatic activity and die. Low temperatures can cause the microbes to become dormant, inactive, or die. It is, therefore, essential that the temperatures within a bioreactor are carefully controlled and are not allowed to go beyond the range within which the microbes are able to thrive. However, choosing an optimal temperature for which to run a system at depends on the specific microorganisms used within it. Most anaerobic digesters used for wastewater treatment typically operate at mesophilic conditions, between 35 °C and 37 °C. Temperature needs to be carefully controlled as large fluctuations can be detrimental for these processes [17].

## 4. Accessory Treatment Options

### 4.1. Hydrolysis

Anaerobic systems generally have difficulty in degrading FOGs and ammonia, as these are oxidation reactions which occur. High concentrations of FOGs can contribute to clogging within the reactor as they are poorly broken down through AD. An effective solution to these problems is a hydrolysis stage before the anaerobic digester. An up-flow ASB reactor (UASB) was observed to be able to effectively treat wastewaters with high FOG contents which had been hydrolysed before treatment. It was noted that when fed unhydrolyzed wastewater, the UASB showed unstable COD removal and a tendency to accumulate some of the fats within the sludge [18]. This suggests that the inclusion of a separate hydrolysis stage can mitigate the risk of clogging within the reactor.

### 4.2. Coagulation

Coagulation/Flocculation are currently the most commonly used processes for the removal of suspended and dissolved solids, colloids, and organic components in industrial wastewater [19]. There are two types of coagulants commonly used in wastewater treatment, namely, inorganic and organic coagulants. Inorganic coagulants are usually metal-based salts, usually containing aluminium or iron. The use of Alum ( $X \times Al(SO_4)_2 \times 12H_2O$ , where X is a metal ion such as potassium or sodium) as a coagulant in the treatment of DWW was observed to reduce the turbidity of the water by 95% and reduce COD by 68% [20]. This was further enhanced by the addition of polymeric coagulants, resulting in a reduction in COD by 85%. However, inorganic coagulants produce large amounts of metal-rich floc, which must be further treated before it is disposed of. They can also alter the pH of the water requiring pH control and corrosion-resistant equipment. When trying to integrate inorganic coagulants with biological systems, it is important to evaluate whether the microbial colonies will be able to function unhindered.

Organic coagulants can either be polymeric or natural coagulants. Polymeric coagulants produce longer polymer chains without any metals or hydroxides and produce smaller volumes of floc. They also do not impact the pH of the water. They also produce low-density floc, which does not always settle well. Natural coagulants are being investigated widely and there have been many studies which show their effectiveness in the treatment of wastewater. An extract made from the bark of *Guazuma ulmifolia* was used as a coagulant for DWW (3037 mg<sub>COD</sub>/L & 1283 mg<sub>BOD<sub>5</sub></sub>/L) and was observed to remove 95.8% of the turbidity, 76% of the COD and 81.2% of the BOD<sub>5</sub> [21]. Natural coagulants' low cost and eco-friendliness make them a suitable alternative to the more commonly used synthetic coagulants.

### 4.3. Membrane Filtration

Membrane filtration methods such as microfiltration are effective at significantly reducing the TSS within the wastewater. However, it has little effect on the TN, COD, and BOD<sub>5</sub> content and is commonly used as a pre-treatment step in a wastewater treatment process. Reverse osmosis (RO) is a viable option for dairy wastewater treatment and has been observed to reduce the TN and TOC by 94% and 84%, respectively [22]. Nanofiltration has been observed to be effective at reducing COD and TSS levels but does not remove all ions of interest, such as phosphates and nitrates, from the wastewater stream [23]. There have been some applications of nanofiltration and RO being used in conjunction with a bioreactor, which was observed to greatly improve the overall ion removal in addition to enhancing the reduction in COD and TSS. There are issues associated with RO and membrane filtration, which affect the long-term uses of this type of system. RO is an extremely energy-intensive process and is sometimes impractical to operate at the specifications required [24]. The replacement of this technology can be costly in the event of physical damage or fouling. In addition, this

process produces a highly concentrated retentate which needs to be disposed of. This is a major concern as the disposal of the concentrate that is formed is more problematic than that of the wastewater itself.

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