

# Various Uses of Lycopene

Subjects: Food Science & Technology | Chemistry, Medicinal | Oncology

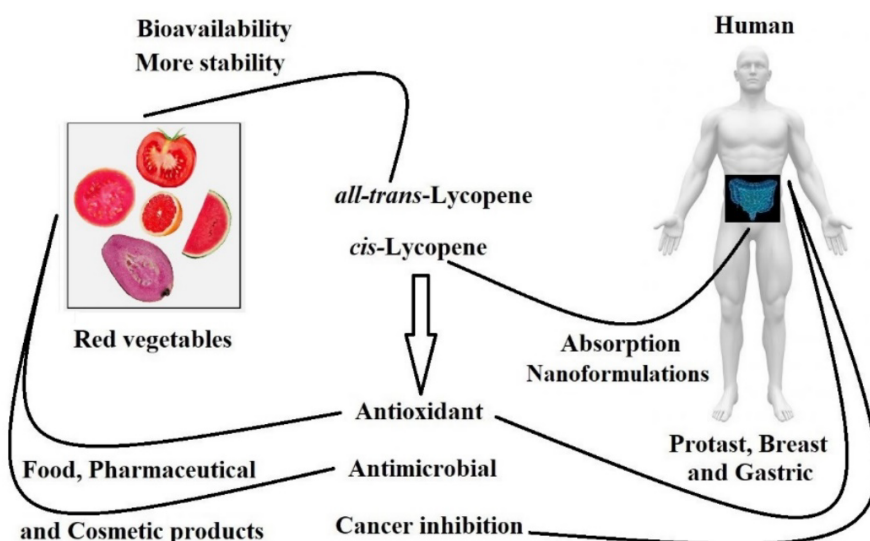
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Lycopene is a carotenoid abundantly found in red vegetables. This natural pigment displays an important role in human biological systems due to its excellent antioxidant and health-supporting functions, which show a protective effect against cardiovascular diseases, hypertension, cancers, and diabetes.

Keywords: lycopene ; carotenoids ; isomers ; antioxidants ; cancers ; food technology ; pharmaceuticals ; cosmetics ; biotechnology

## 1. Lycopene Bio-availability

Lycopene is a carotenoid abundantly found in red vegetables (**Figure 1**). This natural pigment displays an important role in human biological systems due to its excellent antioxidant and health-supporting functions, which show a protective effect against cardiovascular diseases, hypertension, cancers, and diabetes [1][2].



**Figure 1.** Representation of bio-availability and functions of lycopene in humans as well as products from foods, pharmaceutical and cosmetic products.

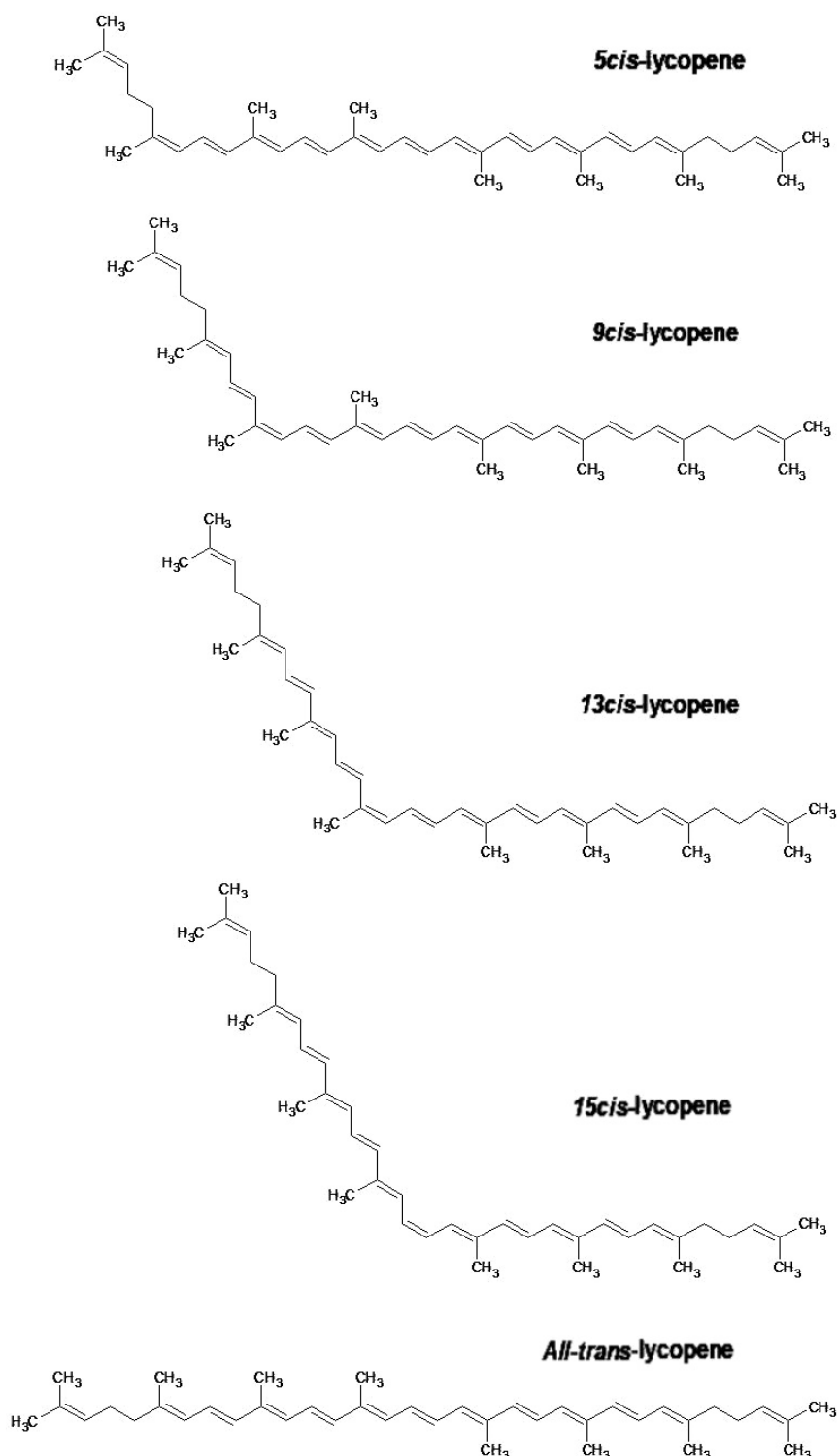
Furthermore, studies have shown the potential anticancer activity of lycopene, suggesting that its consumption may prevent prostate, esophagus, stomach, colorectal, pancreas, breast, and cervix cancers. However, the literature has not determined the cause–effect relationship and how the consumption of lycopene-rich food decreases cancer risk [3][4][5][6].

Lycopene is an organic molecule whose molecular formula is  $C_{40}H_{56}$  and molecular weight is  $536.85 \text{ g}\cdot\text{mol}^{-1}$ . It is insoluble in water but soluble in some organic solvents [7][8]. Its molecular structure contains 13 double bonds, 11 of which are conjugated and provide characteristics for lycopene's antioxidant activity and strong red color [7][9].

These double bonds are affected by the action of oxidants, and can be damaged by light, acid, and heat, which destroy or rearrange the structure of lycopene to different spatial *cis* configurations from *all-trans*-isomers [10]. These effects may deteriorate and/or lead to the loss of lycopene bioactivity [5][10]. The latter seems to depend on several factors, including lycopene content, the complex composition of food, and particle size consumed in the digestive process [4].

*All-trans*-lycopene (**Figure 2**) is interesting for industrial use in food and pharmaceuticals because of its high stability compared to other isomers of lycopene. *All-trans*-lycopene presents higher color intensity than *cis* isomers due to its low extinction coefficient. In nature, the *trans*-lycopene configuration has better stability compared with the *cis* isomer [3]. Nevertheless, the industry has also demonstrated interest in *cis*-lycopene structures (**Figure 2**) because they seem to

have better bio-availability when compared to the *all-trans* isomer and can also prevent breast and prostate cancer [11][12][13].



**Figure 2.** Representation of *cis* and *trans*-lycopene structure. Data deposited in computer by PubChem [14] and converted to 2D with ACD/Labs [15].

The nutritional efficacy and industrial applicability of lycopene is limited due to its insolubility in water. On the other hand, nano-emulsified or micro-encapsulated lycopene, as well as lycopene nanoparticles, have demonstrated great in vitro bio-accessibility throughout the liberation of lycopene content from the nanostructure [3][11][16][17]. Therefore, micro- or nano-encapsulated lycopene may both overcome and avoid the problems related to lycopene structure and can present opportunities for the use of this potent antioxidant.

## 2. Production and Extraction Process

Several processes for the production and extraction of carotenoids such as lycopene have been proposed (Table 1). The most used method for extraction is solvent extraction and supercritical fluid extraction—SFE (supercritical CO<sub>2</sub>) [3][18][19].

and for production the most used is biosynthesis in a reactor or flask fermentation using biological strains, such as bacteria and yeasts <sup>[1][8][20][21]</sup>. Regardless of the method used, the lycopene extracts obtained have a red color, and their color intensity depends on lycopene concentration in the extraction media <sup>[22]</sup>.

**Table 1.** Type of lycopene obtained from different extraction methods and biosynthesis from 2017 to 2021.

Technique	Solvent/Mobile Phase/Flow Rate	Device	Temperature/ Pressure/Time/ V/Hz/rpm	Lycopene Source/ Carotenoid Extracted	References
Extraction	Hexane	Soxhlet extractor	37 °C/6 h		
	* SC-CO <sub>2</sub> /1 mL·min <sup>-1</sup>	SFT 110 extractor	40 and 80 °C/30 and 50 MPa/30, 45, 60, 90, 120, 180, 240 min	Tomato peel and seed/ <i>cis</i> and <i>trans</i> -lycopene	[19]
	Hexane/Acetone/Ethanol (2:1:1)	Vortex	2 h		
	Washed in 0.1 M NaCl				
	OH pre-treatment		55 °C/1 min		
	Water/Ethanol (70%) (1:6 w/v) &				
	Water/Ethanol (1:6 w/v) & + OH solution	Thermal extraction	55 °C/15 min	Tomato peel and seed/ Lycopene and β-carotene	[22]
	OH application	Ohmic heating (OH) technology (6–11 V·cm <sup>-1</sup> ) pre-treatment	0–100 °C/30 min/ 60–280 V/ 25 kHz		
	* SC-CO <sub>2</sub> /1 L·min <sup>-1</sup>	SFT 110 extractor	60 °C and 40 MPa/ 30, 45, 60, 90, 120, 180, 240 min	Tomato peel and seed/ Highest <i>cis</i> -lycopene content	[23]
	Hexane	Soxhlet extractor/0.22 μm hydrophobic PTFE	12 h		
	Olive oil	Maceration (15 to 150 min)/Magnetic stirrer/Box–Behnken	40–80 °C/ 200–400 rpm	Tomato peel and seed/ Lycopene	[24]
	Methanol/Ethyl acetate/ Petroleum ether (1:1:1, v/v/v)				
	30% Methanolic potassium hydroxide		Room temperature /6 h	Tomato peel and seed from 10 varieties/ Lycopene, β-carotene, and lutein	[25]
	Saturated saline solution/ Diethyl ether/ Distilled water	Washed			
	Dry over anhydrous sodium sulfate	Rotary evaporator R-124	35 °C		
		PEF (pre-treatment)1,3; 5 kV·cm <sup>-1</sup> /0.012 kJ·kg <sup>-1</sup> , 0.160 kJ·kg <sup>-1</sup> , 0.475 kJ·kg <sup>-1</sup> /10 Hz/20 μs	20 ± 2 °C	Tomato peels/ <i>cis</i> and <i>all trans</i> -lycopene	[26]
	Acetone (1:40 w/v) &				
	Ethyl lactate (1:40 w/v) &	Extraction flask	25 °C/0–24 h /160 rpm		
		Thermal extraction	75 °C/1 or 2 h	Plum tomato peels/	
	Ethyl acetate	Ultrasounds	Approximately 0 °C/30 min	<i>cis</i> and <i>all trans</i> -lycopene	[27]

Technique	Solvent/Mobile Phase/Flow Rate	Device	Temperature/ Pressure/Time/ V/Hz/rpm	Lycopene Source/ Carotenoid Extracted	References
	Magnesium carbonate (20%) (sample/solution 1:1)	Orbital shaker	25 °C/2 h	Tomato peels/ Lycopene	[3]
	<i>n</i> -hexane/Acetone (3:1)	Ultrasonic	50 °C/30 min (10 times)		
		Centrifuge	10 °C/10 min		
		Reduced volume	40 °C/Low pressure		
Technique	Substrate	Device	Biological Strain	Carotenoid Produced	References
Biosynthesis	Isopropyl- $\beta$ -D-thiogalactoside (IPTG) as inducer	Shaking flask (37 °C and 200 rpm)	<i>E. coli</i>	Lycopene	[20]
	Glucose	Shake flask (30 °C, 300–600 rpm)	<i>S. cerevisiae</i>	Lycopene	[21]
	Glucose + Glycerol	Shake flask (37 °C, pH = 7.2, 48 h)	<i>Escherichia coli</i> R122	Lycopene	[1]
	Glucose				
	Oleic acid				
	Glucose	Bioreactor (37 °C, pH = 7.2, 48 h)	<i>Escherichia coli</i> FA03-PM	Lycopene and $\beta$ -carotene	[8]
	Glucose + oleic acid + yeast extract				
	Glucose + waste cooking oil + yeast extract				
	Lactic acid	Flask fermentation (120 h)	<i>B. trispora</i> NRRL 2895 (+) and N6 (–)		

\* SC-CO<sub>2</sub> = Supercritical Carbon Dioxide; (w/v) & = sample weight/solvent volume.

High temperatures (above 80 °C), light, oxygen, and exposure time may degrade lycopene, while the type of solvent can increase the isomerization from *all-trans*-lycopene to *cis*-lycopene. Acetone, for example, is one of the best solvents for extracting lycopene from fresh material once it provides better solubilization of the lipophilic intracellular content [26].

The use of electric processes to extract carotenoids from food has been successfully applied, with the advantage of promoting a selective extraction and improving carotenoid bio-availability [22]. Although the effects of electric processes on carotenoids are still unknown, applying low voltages could reduce the risk of damage to their structures [26]. These results have ushered in new studies about the factors that influence degradation, as well as “green” methods of lycopene recovery.

Supercritical CO<sub>2</sub> is a technique that significantly impacts lycopene extraction, as it is considered an environmentally friendly method when compared with those that use solvents and lower temperatures [1][18][19][26]. The bacteria and yeasts introduced into bioreactors and the species *Escherichia coli*, *Blakeslea trispora*, and *Saccharomyces cerevisiae* stood out for lycopene production [8][20][21]. These microorganisms consume glucose, lactic acid, and fatty acids as nutrients at a temperature of around 30 °C, producing  $\beta$ -carotene and lycopene as final products [1][21].

Therefore, biosynthesis methods followed by ultrasound or enzymatic lysis to damage the cell membrane before supercritical fluid extraction could be an alternative to obtaining lycopene using a clean methodology.

It is noteworthy that the extraction methods applied in obtaining lycopene from red guava are patented methodologies and, for this reason, these methods are not mentioned in **Table 1**.

### 3. Lycopene Bio-accessibility and Bio-availability—Novel Technologies

Carotenoids can offer numerous health benefits when consumed consistently (Table 2). Nevertheless, for this purpose, there must be a first release from the food matrix followed by carotenoid diffusion into oil droplets. The bile salts help to create micelles to assist the digestion of lipid forms, converting them in free fatty acids, mono- and diacylglycerides, lysophospholipids, and free cholesterol [28]. The harsh conditions throughout the absorption and assimilation process might conduct lycopene degradation due to exposure to pH changes, increased temperature, and oxidation [29]. Consequently, bio-availability and absorption are much lower than water-soluble molecules [30].

**Table 2.** Recent examples of methods used to encapsulate lycopene from 2017 to 2021.

Delivery System	Encapsulation Method	Results	References
$\beta$ -cyclodextrins	A mixture of methylene chloride solution of lycopene with ethanol at 37 °C.	Higher stability against oxidizing agents (AAPH and H <sub>2</sub> O <sub>2</sub> ).	[31]
$\beta$ -cyclodextrins	Lycopene inclusion complexes with $\beta$ -cyclodextrin were prepared by the precipitation method.	Increased thermal stability, photostability, and antioxidant activity.	[32]
Nanoliposomes	Sonication of lycopene, soybean phosphatidylcholine, cholesterol, and aqueous solution.	Neuronal protection against cerebral ischemia/reperfusion. Improved therapeutic efficacy and attenuated the cardiotoxicity of the chemotherapy drug doxorubicin.	[33]
Phospholipid nanoliposomes	Nanospheres of phospholipids with lycopene produced by evaporation and nanoliposomes produced by sonication with the presence of buffer and recovered by centrifugation.	Enhanced antioxidant activity. Prevented reactive oxygen species-induced kidney tissue damage.	[34]
Double-loaded liposomes	Lycopene, $\beta$ -cyclodextrins encapsulated with soy lecithin and cholesterol.	Prolonged-release. Improvement of lycopene solubility. Cardioprotective activity tested <i>in vivo</i> .	[35]
Oil-in-water nano-emulsions	Octenyl succinate anhydride-modified starch mixed with lycopene using high-pressure homogenization and medium-chain triglycerides as carrier oils.	Stable nano-emulsions system with potential application for functional foods.	[2]
Oil-in-water emulsions	Emulsion of water, pure whey isolate, citric acid, triglycerides, and lycopene created with pressure homogenizer.	Increased lycopene bio-accessibility. System critical for the delivery of lipophilic bioactive compounds in functional drinks.	[36]
Nanodispersions	Homogenization of lycopene dissolved in dichloromethane, aqueous phase, and Tween 20.	Small-size lycopene nanodispersions. Good stability for application in beverage products.	[37]
Feed emulsions	Homogenization of tomato powders, maltodextrin, and gum Arabic in aqueous solution and encapsulation made by spray-drying.	Increased lycopene stability.	[38]
Solid lipid nanoparticles (SLN)	Lycopene-loaded solid lipid nanoparticles using Precirol® ATO 5, Compritol® 888 ATO, and myristic acid by hot homogenization.	Stable after 2 months in an aqueous medium (4 °C).	[39]
Solid lipid nanoparticles (SLN)	Cold homogenization technique with glyceryl monostearate and lycopene.	Gel with a promising antioxidant therapy in periodontal defects.	[40]
Solid lipid nanoparticles (SLN)	Homogenization-evaporation technique of lycopene-loaded SLN with different ratios of biocompatible Compritol® 888 ATO and gelucire.	Particles showed <i>in vitro</i> anticancer activity.	[41]
Nanostructure lipid carriers (NLCs)	Ultrasonication of lycopene with Tween 80 and Poloxamer 188.	Enhanced oral bio-availability. Increased cytotoxicity against human breast tumor cells.	[42]

Delivery System	Encapsulation Method	Results	References
Nanostructure lipid carriers (NLCs)	Homogenization and ultrasonication method (aqueous phase with Tween 80, lecithin, and lycopene).	Increased lycopene aqueous solubility. Improved solubility masking tomato aftertaste. Increased homogeneity of fortified orange drink.	[43]
Nanostructure lipid carriers (NLCs)	Emulsion created with lycopene, a lipid mixture, Tween 80 followed by pressure homogenization.	Biphasic release pattern with fast release initially and a slower afterward.	[6]
Whey protein isolate nanoparticles	Lycopene loaded whey protein isolate nanoparticles.	Enhance the oral bio-availability of lycopene. Controlled release. Facilitated absorption through the lymphatic pathway.	[17]
Gelatin nanofibers	A mixture of gelatin from bovine skin and tomato extract is used in electrospinning.	Better retention of lycopene. Better antioxidant activity during 14-days storage.	[44]
Ionic gelation	Lycopene watermelon concentrate mixed with sodium alginate or pectin. Encapsulation by dipping in CaCl <sub>2</sub> and drying under vacuum.	More stable lycopene-rich beads. Good application as natural colorants/antioxidants in different types of food products.	[45]
Nano-encapsulation	CPCs (Chlorella pyrenoidosa cells) loaded with lycopene into a complex nutraceutical and exogenous.	Feasibility of lycopene encapsulation in the CPCs. Combined the activities of both materials. Novel nutraceuticals to reduce cellular oxidative stress.	[10]
Nano-emulsion	Lycopene from guava on nanoemulsifying system of natural oils.	Lycopene nano-emulsion with high stability. Significant inhibition of edema formation, suggesting a potential candidate for anti-inflammatory therapy.	[16]
Lipid-core Nanocapsules	Nano-encapsulation process mixed lycopene extract from guava with polycaprolactone polymer in acetone sorbitan monostearate.	The nanostructure was cytotoxic against cancer cells (human breast adenocarcinoma line MCF-7).	[12]
Nanoparticle	Polymer nanoparticle fucan-coated based on acetylated cashew gum and lycopene extract from guava.	Promising results for applicability in hydrophobic compounds carrying systems as lycopene with cytotoxic effect on the breast cancer cell.	[11]
Microencapsulation	Microencapsulation of lycopene from tomato peels by complex coacervation and freeze-drying.	The fine orange-yellow powder could be micro-encapsulated as stable lycopene applied to the food industry with properties against metabolic syndrome.	[3]

Strategies have been created to control problems related to the practical application of lycopene, which is strongly restricted due to its high sensitivity when exposed to light, oxygen, and heat, as well as contact with metal ions, besides processing conditions and low water solubility [11][46].

Encapsulation is one of the techniques that frequently uses oligosaccharides, such as cyclodextrins, to associate compounds in a hydrophobic core, while on the outside it forms a hydrophilic shell. The encapsulation protects lycopene from degradation and isomerization besides increasing its solubility in aqueous environments [46]. A report on the lycopene/ $\alpha$ - and  $\beta$ -CD complexes pointed that both could provide stable associations in water with profound differences in structure [47]. Maltodextrins were used during tomato processing to powders aiming to increase lycopene stability [48].

Respective delivery systems have been created to enhance lycopene bio-availability and absorption rates (**Figure 1**) in the gastrointestinal environment [37][49]. During digestion, carotenoids are assimilated with other lipids into mixed micelles containing bile salts and phospholipids, which perform as carriers to solubilize the carotenoids and transport them to the zone of maximum absorption in the intestine. Incorporating lycopene into the oil phase of emulsions is an alternative to protect it from oxidation and chemical degradation, providing better bio-availability and prolonging shelf-life [28]. Nano-emulsions have been reported to be suitable delivery systems with favorable results for the encapsulation of low-solubility compounds, such as lycopene [46]. Recent works have shown efficient delivery systems for lycopene: its incorporation in

oil-in-water emulsions for orange beverages [50], lycopene encapsulated in isolate-Xylo-oligosaccharide protein conjugates made by Maillard reaction [49], and oil-in-water emulsions with long- to short-chain triglycerides [37]. The literature also reports the formation of liposomes and nanoliposomes, which are spherical vesicles created with a concentric phospholipid bilayer of hydrophilic center. It was reported that lycopene tends to be entrapped in the hydrophobic bilayer, enhancing bio-accessibility when exposed to the gastrointestinal tract and with increased antioxidant capacity compared with the free form [51][52].

Oil-water nano-emulsions are nanoparticles dispersed in heterogeneous systems with an inner lipidic and an external aqueous phase stabilized by one or two surfactants. Unlike the nano-emulsions, lipid nanoparticles have an internal solid lipid phase since these nanoparticles are totally or mainly composed of solid lipids at room temperature. Such a solid matrix allows the controlled release of the encapsulated molecules and protects them from degradation while increasing the long-term stability of the system [53]. Lycopene-loaded SLNs demonstrated stability in an aqueous medium for two months, producing an applicable system for future in vivo trials in nutraceutical industries [39]. The encapsulation in SLN showed an improvement in lycopene oral delivery, and an ex vivo assessment determined that this carotenoid had better permeation besides causing more cytotoxicity against breast cancer cells [42]. Lycopene loaded into nanostructured lipid carriers (NLC) composed of Eumulgin SG, orange wax, and rice bran oil, employing high pressure in homogenization process, showed chemical stability and delayed degradation when put into cold storage [54].

Moreover, lycopene nano-emulsions have provided more thermal stability for lycopene and significantly inhibited edema formation. For this reason, these nanoparticles may be considered to be a potential candidate for anti-inflammatory therapy [16]. Lipid-core nanocapsules of lycopene, in turn, optimized stability for 7 months at 5 °C storage, and improved its toxicity against breast cancer cells. The nanocapsules also inhibited the production of intracellular peroxy radicals in human microglial cells and maintained the membrane integrity of erythrocytes, highlighting its potential to be employed in cancer treatment [42].

Lycopene encapsulated in polymeric nanoparticles showed high anti-tumor potential, with cytotoxicity against cancer cells at low concentrations and no toxicity against *Galleria mellonella*. Additionally, nanoparticles with sizes of  $162.10 \pm 3.21$  nm were efficient with a passive mechanism of permeability for targeting tumor tissues [11].

On the other hand, lycopene powder produced by complex coacervation and freeze-drying after microencapsulation had promising results as a biopolymeric composite with inhibitory effect potential on  $\alpha$ -amylase associated with metabolic syndrome, and demonstrated high antioxidant activity in formulations [3].

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