

# Molybdenum Disulfide

Subjects: [Chemistry](#), [Applied](#)

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Molybdenum disulfide ( $\text{MoS}_2$ ) is one of the compounds discussed nowadays due to its outstanding properties that allowed its usage in different applications. Its band gap and its distinctive structure make it a promising material to substitute graphene and other semiconductor devices. It has different applications in electronics especially sensors like optical sensors, biosensors, electrochemical biosensors that play an important role in the detection of various diseases' like cancer and Alzheimer. It has a wide range of energy applications in batteries, solar cells, microwave, and Terahertz applications. It is a promising material on a nanoscale level, with favorable characteristics in spintronics and magnetoresistance.

Molybdenum disulfide ( $\text{MoS}_2$ )

Molybdenum disulfide properties

Molybdenum disulfide synthesis

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transition metal dichalcogenides (TMDs)

## 1. Introduction

Molybdenum disulfide ( $\text{MoS}_2$ ) is an inorganic compound of the transition metal dichalcogenides (TMDs) series, that has one atom of Molybdenum and two atoms of Sulfur. Dichalcogenides are chemical compounds consisting of a transition metal, like Molybdenum, and a chalcogen (element of group 16 in the periodic table) like sulfur (S) <sup>[1]</sup>. The physical, chemical, and electronic properties of this compound grabbed the attention of many researchers and were found promising materials to substitute previously used semiconductor and/or graphene devices. As the world is moving towards miniaturization, researchers were searching for a material to substitute semiconductor devices that seemed to reach an end when entering the nanoscale era <sup>[2]</sup>. While semiconductor devices based on Silicon were facing quantum and tunneling effects on a nanoscale level,  $\text{MoS}_2$  showed favorable and promising electronic and quantum characteristics when going from bulk to two-dimensional (2D) structure <sup>[3]</sup>.

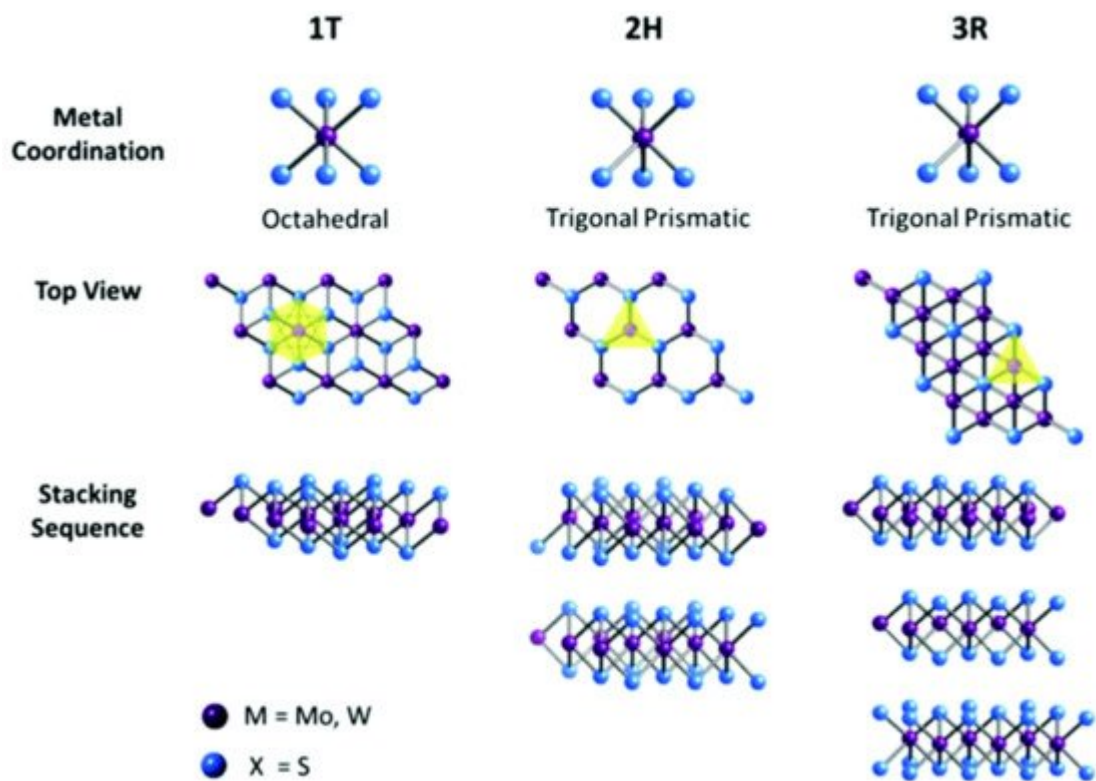
$\text{MoS}_2$  seems to solve many problems facing previous devices <sup>[4][5][6][7][8]</sup>, it has a large band gap (~1.8 eV) which changes from an indirect gap to a direct one in thin structures. This would permit downscaling electronic devices, rather than graphene which nearly has a zero-band gap <sup>[9][10]</sup>. It does not have surface dangling bonds and has high mobility even with high- $\kappa$  dielectric materials. It is ideal for thin-film transistors, and its fabrication is simple which means large production yield and low cost <sup>[11][12]</sup>. The covalent bonds between Molybdenum and Sulfur and the Van der Waals bonds between its layers make it optimal for gas sensing. One of the major problems with Silicon devices was that related to the metal-semiconductor interface <sup>[13]</sup>.  $\text{MoS}_2$  has less contact resistance and high performance. In other words,  $\text{MoS}_2$  has potency to be used in 1 nm gate transistors with excellent on/off switching characteristics and high efficiency <sup>[14]</sup>.

Silicon transistors fabrication faced some problems on the bulk scale that were overcome by new structures such as multi-gate transistors, but when going down to the nanoscale, the fabrication process seems to reach an end. The metal interconnection lines between transistors have high density and are very narrow, which cause an increase in resistance and capacitance between interconnect lines and high delays. Tunneling problems are more obvious with thin gate oxides and when trying to substitute silicon dioxide with other high- $\kappa$  dielectric materials, more serious problems like thermal instability, channel mobility degradation, incompatibility with the interface are aroused. All these issues lead to performance degradation, and high cost with a small yield. The high lithography resolution needed for small half-pitch (HP)  $\sim 20$  nm, is not easily achieved and needs high techniques and tools in lithography [15][16]. On the other side, MoS<sub>2</sub> showed easier and simpler ways of synthesis and device fabrication [17]. It is easily prepared by spreading the molybdenum metal and sulfur and letting them self-propagate under a high temperature [18]. A MoS<sub>2</sub> field effect transistor was fabricated in [19] by immersing it in an electrolyte, the device showed lower contact resistance and a better gate control.

MoS<sub>2</sub> has a wide range of applications in different fields. Like Silicon and graphene, it has applications in biosensing, and optical sensors but the most important ones are those related to bio-applications like DNA, cancer, and Corona Virus detection [20][21][22][23]. While Silicon and graphene were still under study in their compatibility with human bodies, a study, in [24], showed that MoS<sub>2</sub> may be very effective in curing cancer and Alzheimer's disease. It also showed that the compound has no biological interaction which makes it safe for injection in human bodies. Another study in [25] proposed a biocompatible device made of MoS<sub>2</sub> to restore some visual malfunctioning. The compound applications are not restricted to electronics, but it can serve as a lubricant [26] and is used in hydrogen evolution reactions. It is a suitable material for batteries' electrodes [27]. Indeed, MoS<sub>2</sub> properties and structure made it promising for several electronic, sensing, microwave, and terahertz applications.

## 2. Structure and Properties

MoS<sub>2</sub> structures differ from 3D, 2D, one-dimensional (1D), or dot structures. Its characteristics and applications also change from one dimension to another, they can be semiconducting, metallic or superconducting. It exists in several layers and shapes. Its bulk (3D) structure can be tri-agonal (T), hexagonal (H), and Rhombohedral (R), where 2H MoS<sub>2</sub> means 2-layer hexagonal shape MoS<sub>2</sub>. The three main structures are 1T, 2H and 3R, where the 1T phase coordinates in an octahedral structure, 2H and 3R in a trigonal prismatic structure [28][29] as shown in [Figure 1](#). The lattice constants for each structure are shown in [Table 1](#) [30]. The 1T structure is known to be metallic while the other two are semiconducting. The monolayer of hexagonal MoS<sub>2</sub> is also semiconducting. Both 2H and 3R are used as dry lubricants. Due to the nonlinear optical properties of 3R phase it is used in nonlinear optical mass sensing in quantum measurements and biomedicine [31]. As an example, for gas sensors, the different phase materials of MoS<sub>2</sub> can be interesting in obtaining high sensitivity and rapid desorption [32].



**Figure 1.** Different coordination and stacking sequences of the three MoS<sub>2</sub> structures 1T, 2H and 3R. Reproduced from [29]. Published by The Royal Society of Chemistry.

**Table 1.** Comparison between different MoS<sub>2</sub> structures.

	1T	2H	3R
Structure Coordination	Octahedral	Trigonal Prismatic	Trigonal Prismatic
Lattice parameters	a = 5.60 Å, c = 5.99 Å and an edge sharing octahedral [30]	a = 3.15 Å, c = 12.30 Å [30]	a = 3.17 Å, c = 18.38 Å [30]
property	paramagnetic and metallic	Semiconducting	
Electrical conductivity	10 <sup>5</sup> times higher than 2H phase	Low (~0.1 S/m)	

	1T	2H	3R
Structure Coordination	Octahedral	Trigonal Prismatic	Trigonal Prismatic
Absorption peaks <sup>2</sup>	No peaks at 604 nm and 667 nm	Showed peaks at 604 nm and 667 nm	
Common applications	Intercalation in chemistry	Dry lubricants	Dry <sup>[34]</sup> lubricants and non-linear optical devices <sup>[35]</sup>

photoluminescence properties are summarized in [Table 2](#) <sup>[36]</sup>.

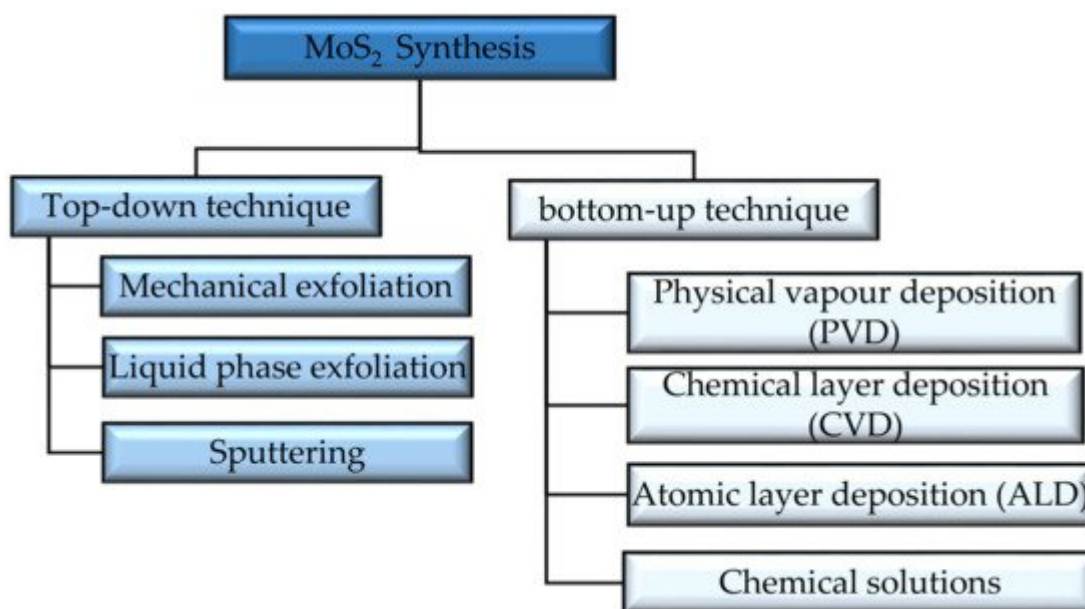
**Table 2.** Comparison between Bulk and monolayer MoS<sub>2</sub>.

	Bulk	Monolayer
Bandgap	Indirect (~1.2 eV)	Direct (1.8 eV)
Binding energy	0.1 eV	1.1 eV
Photoluminescence intensities	between 10 <sup>-5</sup> and 10 <sup>-6</sup>	10 <sup>4</sup> times higher than that of bulk, up to and up to 4 × 10 <sup>-3</sup>

MoS<sub>2</sub> exists in different 2D structure like nanosheets, and nanoribbons or 1D structures as nanowires and nanotubes, or 0D structure as quantum dots and nanoplatelets. The thickness of 2D nanoribbons was found to be of 1 to 3 layers of MoS<sub>2</sub>, while the thickness of 1D nanowires (NW) can have lengths from 14 to 30 nm and a width of 0.6 nm approximately <sup>[37]</sup>. The structure of 1D nanoplatelets and their properties were investigated in <sup>[38]</sup>. The nanoplatelets are 12–30 nm with one-unit cell width. They have very high catalytic activity for hydrodesulphurization. The quantum dots range from 2 to 10 nm in size. They have higher band gap than nanosheets, and stronger bonds between Mo atoms than monolayers. The change in band gap of MoS<sub>2</sub> from one dimension to another, changes the photoluminescence characteristics and thus has different optical properties according to its dimension. Additionally, monolayers or other low dimension forms are also easy to be implemented in optical nanostructures to enhance the photoluminescence intensities and emission rates through light-matter interactions <sup>[39][40][41]</sup>. This is a strong motivation for MoS<sub>2</sub> to be included in optical applications <sup>[42]</sup>.

### 3. Synthesis

There are different techniques used to obtain material layers [Figure 2](#), and each one of them results in different quantities, shapes, and sizes. Mainly the approaches used in synthesizing TMDs nanostructures, are the top-down approach and the bottom-up approach [\[43\]\[44\]](#). The first approach depends on etching crystal planes from a substrate that has the crystals laid over it, while in the second approach, the crystals are stacked over the substrate. Exfoliation is one of the top-bottom techniques for obtaining MoS<sub>2</sub> layers. The weak Van der Waal forces between layers of TMDs paved the way in front of different exfoliation synthesizing techniques [\[45\]](#). Mechanical exfoliation is done using a sticky tape which is rubbed out and shifted on a substrate having MoS<sub>2</sub> flakes over it. The method gives low yield and is good for lab use. Exfoliation can also be done in the liquid phase by adding a chemical compound and stirring, bubbling, or grinding. This method is simple and cheap but has low quality. The low yield in liquid exfoliation was avoided using carbon aerogel composites in [\[46\]](#). The synthesizing is fast and completed in 30 minutes. It also avoids pyrophoric materials that are typically used in liquid exfoliation and increases the electrical conductivity and porosity of MoS<sub>2</sub>. Sonication is one of the techniques that showed simple synthesis when used with liquid exfoliation [\[47\]\[48\]](#). It disposes of the use of hazardous materials used in liquid exfoliation. It is based on ultrasonic waves emitted from a probe in shape of bubbles that peels MoS<sub>2</sub> layers when they burst. The challenge in using sonication assisted techniques is that it produces relatively small area MoS<sub>2</sub> nanosheets that limits its use in practical applications. Generally Top down techniques are said to have low controllability, and scalability and high cost [\[43\]](#). The yield is increased to >90% when using ultrasound sonication with supercritical carbon in [\[49\]](#) with an intercalating solvent N-methyl-2-pyrrolidone (NMP). The method is fast, simple and scalable. Sputtering is used to prepare layers of MoS<sub>2</sub> to be used as lubricants, the layers have a low friction coefficient, but these frictional properties can be changed under humidity, especially for thin films of MoS<sub>2</sub>.



**Figure 2.** Different MoS<sub>2</sub> synthesis techniques [\[27\]](#).

Physical layer deposition (PVD) is one of the bottom-up techniques that includes ion implantation like molecular beam epitaxy (MBE) [\[50\]](#). The method can be applied only to thin layers of MoS<sub>2</sub> and the resulting grain sizes are variable [\[51\]](#). Chemical vapor deposition (CVD) is applied to thin and thick layers, where Mo is laid over a substrate

and Sulfur vapor passes over it. This method has good quality, but low yield. The atomic layer deposition (ALD) method is used to fabricate thick and thin films. The method is considered efficient and the layers have fewer impurities that can be used in different applications, including electronics and sensors. MoS<sub>2</sub> layers can be synthesized with the help of chemical solutions, using hydrothermal and solvothermal reactions wherein both cases Mo and S react in an aqueous solution above the boiling point and in a nonaqueous solution at high temperature, respectively. The size and shape of the layers can be controlled where we can get powder and thin films of MoS<sub>2</sub> by this method. It is considered cheap and scalable [52].

In [53], MoS<sub>2</sub> is synthesized using a liquid organic precursor on an insulating substrate using CVD. The used method is more reproducible and is used to obtain larger areas of MoS<sub>2</sub> layers than those obtained with methods using powder Molybdenum oxide and sulfur powder. Another method in [54] used thermal evaporation and ALD, where it used metalloporphyrin as a promoter layer. The method allowed to manipulate the carrier density and conductivity of MoS<sub>2</sub> according to the thickness of the metalloporphyrin layer used. It is used to produce MoS<sub>2</sub> nanosheets on a large scale. In [18], MoS<sub>2</sub> is synthesized using self-propagating, under high temperature, where Mo nanopowders and elementary Sulfur are used. The mixture is put into cylinders and then under pressure. The main resulting structure is 2H MoS<sub>2</sub>, but there are other phases like rhombohedral MoS<sub>2</sub> and Mo<sub>2</sub>S<sub>3</sub>. Thermal sulfidation is another method like CVD that uses Sulfur gas as a precursor. It uses a Mo [55] or Mo-oxide [56] film deposited on a substrate, where evaporated sulfur passes over it under certain temperature. The method is known to reduce the effect of gas flows that occurs in CVD, and results in self-aligned patterns of MoS<sub>2</sub>. The sulfidation of two different oxides of Mo: MoO<sub>3</sub>, and MoO<sub>2</sub> discussed in [56], showed more stable MoS<sub>2</sub> monolayer films, produced from MoO<sub>2</sub>. The films were integrated with bottom-gate transistors and they showed on/off ratio of 10<sup>3</sup>–10<sup>4</sup> and electron mobility of 10<sup>−4</sup> cm<sup>2</sup>/V·s. The PL spectrum of the synthesized monolayers has an exciton peak at 1.89 eV.

Another approach to avoid the drawbacks of exfoliation and intercalation or liquid exfoliation like low electrical performance (low mobility of 0.3-0.4 square centimeters per volt per second and low on/off ratios ~10–100) is using electrochemical intercalation [57]. The method involves quaternary ammonium molecules into 2D crystals, with mild sonication and exfoliation techniques. The technique gives high performance MoS<sub>2</sub> nanosheets with 10 square centimeters per volt per second mobility, and on/off ratios of 10<sup>6</sup>. Table 3 summarizes some synthesis techniques that are already known until now.

**Table 3.** Summary of synthesis techniques.

Technique	Charecteristics of the Obtained MoS <sub>2</sub> Sheets	Publisher	Year	Reference
Using aerogel	A rapid synthesis technique for TMDs- carbon aerogel composites to be used in supercapacitor. The preparation time is 30 min which is approximately 2% of typical aerogel techniques that take 24 h	Nature	2017	[46]

Technique	Charecteristics of the Obtained MoS <sub>2</sub> Sheets	Publisher	Year	Reference
Liquid assisted Sonication	Studied the PL, Raman analysis resulting from bath and probe sonication synthesis	Elsevier	2020	[47]
Liquid exfoliation & ultrasonic cavitation	Obtain less defective and high concentration nanosheets in a short time	Nature	2014	[48]
Exfoliation of & Ultrasound Sonication in Supercritical CO <sub>2</sub>	Exfoliation efficiency > 90%	Springer	2019	[49]
Exfoliation & sonication	Mobility = 10 cm <sup>2</sup> /V, on/off ratios = 10 <sup>6</sup>	Nature	2018	[57]
ALD & thermal evaporation	Enhanced electrical conductivity	Nature	2016	[54]
CVD & liquid precursor	The method uses water to remove impurities like carbon and sulphur. It ensures full coverage of MoS <sub>2</sub> for the substrate	Nature	2017	[53]
ALD & thermal evaporation	The carrier density and conductivity of MoS <sub>2</sub> are adjusted according to the thickness of the metalloporphyrin layer used.  Produce large scale MoS <sub>2</sub> nanosheets	Nature	2016	[54]
Self propagation of Mo powder and elementary Sulfur	Resulted MoS <sub>2</sub> nanosheets have thermal stability up to 400 °C.	IEEE	2013	[18]



Technique	Charecteristics of the Obtained MoS <sub>2</sub> Sheets	Publisher	Year	Reference
CVD with Sulfur as a precursor  (Sulfidation)	on/off current ratio of 10 <sup>5</sup> , a mobility of 0.12 cm <sup>2</sup> /V·s  (mean mobility value of 0.07 cm <sup>2</sup> /V·s	Royal society of chemistry	2014	[55]
Sulfidation	The on/off ratio of 10 <sup>3</sup> –10 <sup>4</sup> and electron mobility of 10 <sup>−4</sup> cm <sup>2</sup> /V·s	Elsevier	2020	[56]
Intercalation & exfoliation	To be used in high-performance thin-film transistors, on/off ratio of 10 <sup>6</sup> and electron mobility of 10 cm <sup>2</sup> /V·s	Springer Nature	2018	[57]

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#### 4. Applications

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MoS<sub>2</sub> has unique structural, and chemical properties allowed this metal dichalcogenide to have a wide range of applications in different fields. Its electronic properties allowed it to enter the nanoelectronics and sensor application field and in turn the medical field. These special electronic properties, together with its biocompatibility,

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Application	Categeory	Description	Publisher	Year	Reference
Electronics	Analogue	Developing a 2D MoS <sub>2</sub> field-effect transistors (FETs) to be used as operational amplifier	Nature	2020	[58]
Electronics	CMOS	Enhanced back gate MoS <sub>2</sub> FETs using Sulfur treatment, gate length from 500 to 80 nm, contact resistance = 1.3 kΩ	IEEE	2020	[59]

MOS<sub>2</sub> transistor with molybdenum source/drain electrodes. IEEE Electron Device Lett. 2015,



Application	Categeory	Description	Publisher	Year	Reference
Electronics	nonvolatile memoryNVM	Hystresses and capacitance of MoS <sub>2</sub> NVM with 9nm and 15 nm blocking oxide layer of SiO <sub>2</sub> is investigated. The device has largest hystresses window. And can store both electrons and holes	IEEE	2016	[60]
Electronics	Power electronics	Ultra low power transistor of SS ~4.5 mm/decade and steep on/off characteristics	Nature	2020	[61]
Electronics	optoelectronics	Developing a highly-efficient and fast photodetector using amorphous silicon and MoS <sub>2</sub>	Nature	2013	[62]
Electronics	optoelectronics	Efficient photodetector with photocurrent gain of 1.6, specific detectivity of $4.32 \times 10^8$ Jones and quantum efficiency $\sim 1.0 \times 10^{10}$ at 365 nm	Nature	2020	[63]
Electronics	optoelectronics	A nanoscale photodetector, energy efficient (consumes from 1 to 1000 nanojoules), and has a small fingerprint of ( $\sim 1 \mu\text{m} \times 5 \mu\text{m}$ ).	Nature	2020	[64]
Electronics	Image sensors	A MoS2-organic heterostructure image sensor similar to human vision system using simple design. The output has less noise and without redundant input data	Nature	2020	[65]

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Application	Categeory	Description	Publisher	Year	Reference
Medical	Cancer cure	Using MoS <sub>2</sub> /GO nanocomposites to cure lung cancer in mice	Nature	2018	[66]
Medical	Cancer cure	Using defect engineering (Sulfur defects) in MoS <sub>2</sub> quantum dots to kill cancer cells. The Sulfur defects increased oxidation strees and was able to inhibit cancer growth cells	Nature	2019	[67]
Medical	Cancer deteccion	A real-time MoS <sub>2</sub> FET sensor to detect H <sub>2</sub> O <sub>2</sub> in biological cells	Nature	2019	[68]
Medical	Cancer deteccion	MoS <sub>2</sub> nanosheet fluorescent biosensor for detecting prostate antigen. A 20 µg/mL of MoS2 nanosheets quenched up to 97% of the dye	Springer	2014	[23]
Medical	Cancer deteccion	Detecting breast cancer based on PL property of MoS <sub>2</sub> . A redshift of 16 nm takes place for miRNA21c (a cancer biomarker)	Nature	2020	[21]
Medical	DNA detection	Sensing DNA nucleobases using MoS <sub>2</sub> nanopores. Molar absorption of MoS <sub>2</sub> nanopore of 0.65 nm thick, and lengths 2 nm, 3 nm, 5 nm.	Nature	2019	[69]
Medical	DNA detection	Detecting DNA based on dye quenching property of MoS <sub>2</sub> . The device has linear	Royal Societly of chemistry	2015	[70]

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Application	Categeory	Description	Publisher	Year	Reference
		range (0 to 50 nM), and a detection limit = 500 pM			, M. en.
Medical	Amino acid detection	Using MoS <sub>2</sub> nanopores and machine learning to detect ionic current and residence time of 20 different amino acids with accuracy range 72.45% to 99.6%	Nature	2018	[71]h, K.P.; atum ; Lu, I.- ; ennae.
Medical	Microfluidic immunosensor	Use of MoS <sub>2</sub> nanosheets on microfluidic electrodes to detect Salmonella typhimurium (S.typhimurium) sensitivity = 1.79 kΩ/(CFU/mL.cm <sup>2</sup> ) detection limit = 1.56 CFU/mL detection range = 10–10 <sup>7</sup> CFU/mL.	Elsevier	2017	[72]%; Tay, n Strong Adv. etal
Medical	Antibacterial materials	Using nanohole enriched MoS <sub>2</sub> to destroy bacteria. MoS <sub>2</sub> concentration used is 4 µg/mL. It has affinity response with biofilms of 14.71 nM, 1.3-fold > 11.44 nM obtained for pristine MoS <sub>2</sub> Verified in vitro and viro	Nature	2021	[73]thods of ed anoeng.
Energy	Solar cells	Using MoS <sub>2</sub> as a buffer in solar cells to enhance efficiency and stability	IEEE	2016	[74]cation 20, 21, g, H.J.;

Jeong, S.Y.; et al. Extremely Efficient Liquid Exfoliation and Dispersion of Layered Materials by Unusual Acoustic Cavitation. Sci. Rep. 2015, 4, 5133.

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Application	Categeory	Description	Publisher	Year	Reference
Energy	Solar cells	Using MoS <sub>2</sub> as a hole transport layer in solar cells a peak at 404 cm <sup>-1</sup> , at 200 °C, and two more peaks at at 380 and 404 cm <sup>-1</sup> , at 300 °C,	Wiley	2019	[75]
Energy	Solar cells	MoS <sub>2</sub> as a transport layer in perovskite solar cells PCE (η) = 3.9% compared to other cells of (η = 3.1%). High stability for 800-hour shelf life (ΔPCE/PCE = -17%) when compared to other cells of (ΔPCE/PCE = -45%).	IEEE	2015	[76]
Energy	Solar cells	Using MoS <sub>2</sub> /Si heterojunction in solar cells to enhance conversion efficiency from 1.1% to 4.6%.	Elsevier	2018	[77]
Energy	Solar cells	Enhancing organometallic-halide perovskite solar using MoS <sub>2</sub> as a buffer Cells (PCE) = 14.9%, and maintaing 93.1% of its PCE after 1 hour	Nature	2020	[78]
Energy	lithium-ion batteries	Using MoS <sub>2</sub> as as anode material for lithium-ion batteries.It has capacity of 1103.6 mAh/g and maintains a	Elsevier	2019	[79]

State and Integrated Circuit Technology (ICSICT), Hangzhou, China, 25–28 October 2016; IEEE: Hangzhou, China, 2016; pp. 489–491.

Application	Categeory	Description	Publisher	Year	Reference
		reversible capacity of 786.4 mAh/g after 50 cycles at 0.1 A/g			[80]
THz applications	THz wave reflector	Switchable THz wave reflector made of MoS <sub>2</sub> , SiO <sub>2</sub> , and gold layers. reflection phase ranges from 0 to 2π linear phase shift according to the geometric dimension	IEEE	2018	[80]
THz applications	THz modulator	Terahertz modulator of multilayer-MoS <sub>2</sub> on Silicon (MOS). Moudulation efficiency of annealed MOS is higher that that of graphene over Silicon and graphene metamaterials	Nature	2016	[81]
Other	Hydrogen detection	MoS <sub>2</sub> -PVP (Polyvinyl pyridine) nanocomposite with ZnO to detect hydrogen. A 5 mg/mL sensor has 8 times better ensing than pristine ZnO	IEEE	2021	[82]
Other	Hydrogen generation	coordination polymer based on [Mo <sub>3</sub> S <sub>13</sub> ] <sup>2-</sup> clusters from amorphous MoS <sub>2</sub> to be used in hydrogen evolution reactions.	Nature	2016	[83]
Other	Hydrogen generation	A edge-terminated and interlayer-expanded MoS <sub>2</sub> catalyst fabricated using a microwave heating strategy. The proposed structure has the highest hydrogen evolution activity and best stability	Nature	2015	[84]

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Application	Categeory	Description	Publisher	Year	Reference
Other	Oil separation from water	MoS <sub>2</sub> Sponge to absorb oil from water. Simple fabrication and easy scaling up.	Nature	2016	[85]

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