

On-Site Carwash Wastewater Treatment

Subjects: [Water Resources](#)

Contributor: Wen-Hui Kuan , Ching-Yao Hu , Li-Wei Ke , Jung-Ming Wu

The main pollutants in car wash wastewater are detergents, dirt, oil, and grease. Untreated wastewater released into rainwater sewer systems or other water bodies may pollute the water and generate excessive bubble foams, which negatively affects urban appearance. Car washes are divided into mechanical car washes and manual or self-service car washes. In general, car washes have a small operation and scale, occupy limited land, and cannot afford wastewater treatment costs. Therefore, most car washes are not equipped with wastewater treatment facilities. Consequently, the discharge of wastewater from car washes negatively affects the water quality in the surrounding environment and results in wasteful use of water resources.

carwash

wastewater

1. Introduction

The rapid growth in human population has resulted in increased car use, which has increased the demand for car wash services and thereby generated large amounts of car wash wastewater. In metropolitan areas, the foam in the wastewater produced during car washing overflows and spoils the appearance of the city. However, compared with industrial wastewater it is relatively easy to treat car wash wastewater and improve its water quality.

Car wash wastewater generally contains suspended particles that originate from the dirt on vehicles, the oil on vehicle exteriors, the oil and grease generated from car wax, and the anionic surfactants caused by detergent use [1][2]. This wastewater has a high COD.

A reasonable amount of water for washing a car is 100~200 L. Several studies have collected data on car wash wastewater for unique vehicles. The car wash water consumption for heavy vehicles and waste container washing vehicles was recorded as approximately 350~900 L and 5000 L, respectively [3][4].

2. Car Wash Wastewater Treatment Technique

2.1. Electrocoagulation (EC)

Figure 1 depicts the general mechanism of the electrocoagulation process. EC uses metal hydroxides produced by electrolysis to remove pollutants in wastewater. During the electrolysis reaction, a sacrificial anode undergoes an oxidation reaction to release metal ions, while the cathode undergoes a reduction reaction to reduce the metal ions to metal and generate hydrogen. Commonly used metal anodes include aluminum and iron. The EC process has a

turbidity removal rate of approximately 90% [5][6][7]. When coupled with adsorption treatment or electro-oxidation treatment, the turbidity removal rate of the EC process can be increased. Moreover, the EC process has a COD removal rate of approximately 80%. When combined with other treatments, the COD removal rate of the EC process can be increased (Table 1).

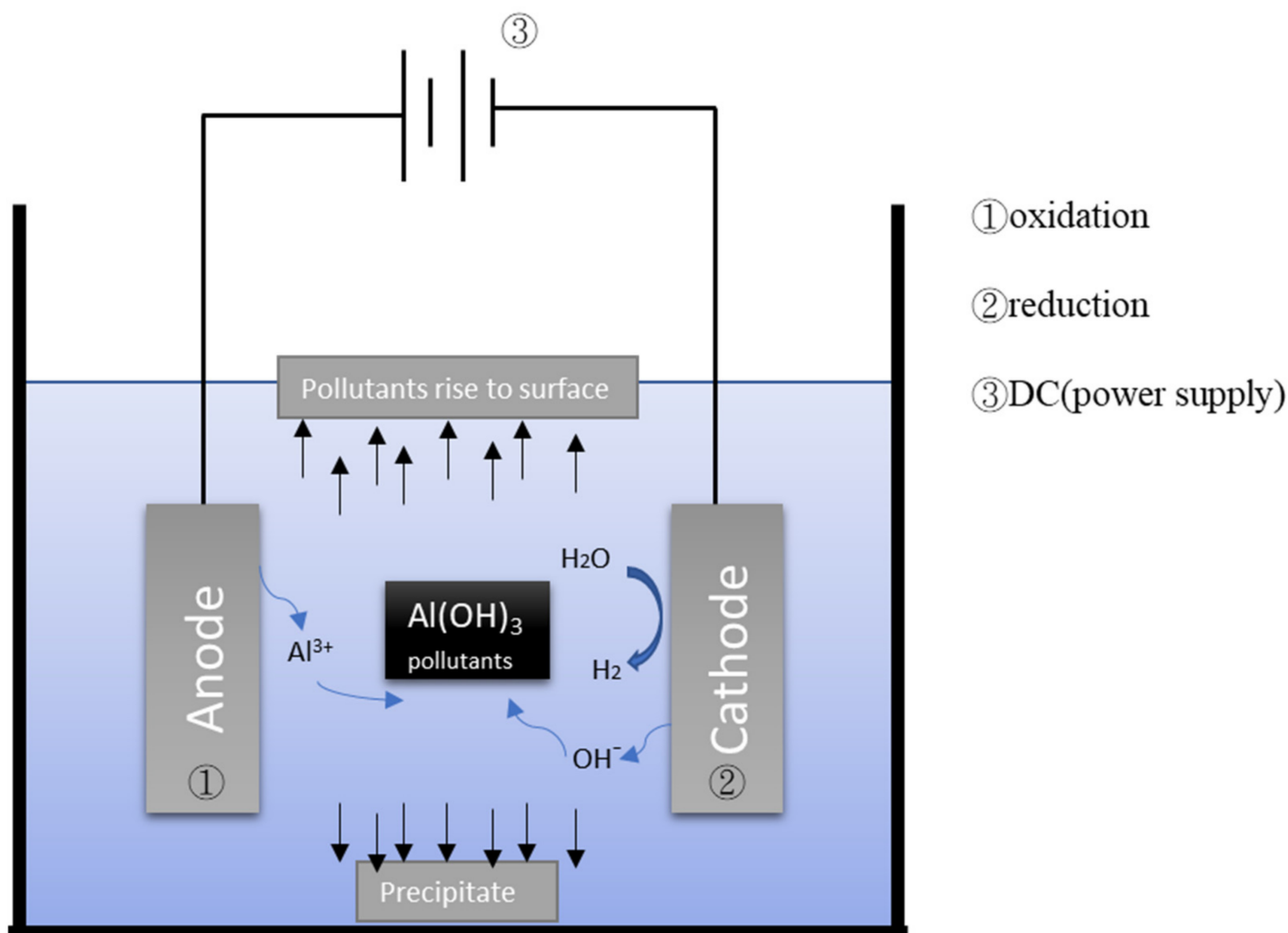


Figure 1. Schematic illustration of electrocoagulation/flotation.

Table 1. Removal rate of various water qualities by EC method.

Country	Area	Ref.	Technique	SS (mg/L)	Turbidity (NTU)	COD (mg/L)	O&G (mg/L)	AS (mg/L)
Mexico	Toluca	[5]	EC + AD	—	92–98%	78–94%	—	—
Mexico	Toluca	[6]	EC + EO	—	98–98.4%	76–96%	92–100%	81–92%
Italy	Genoa	[8]	EC + EO	—	—	75–97%	—	—
Iran	Tehran	[9]	EC	—	85.5%	80.8%	—	—

Country	Area	Ref.	Technique	SS (mg/L)	Turbidity (NTU)	COD (mg/L)	O&G (mg/L)	AS (mg/L)
Iran	Tehran	[10]	EC	—	—	88%	—	—
Iran	Ahvaz	[11]	EC	—	—	90%	—	—
USA	Texas	[12]	EC	—	—	79%	—	—
Egypt	Shatby	[7]	EC	—	~87%	~85%	—	—
Turkey	Istanbul	[13]	EC	—	—	88%	82%	99%
Turkey	Tekirdag	[14]	EC	—	99%	76%	—	—
China	Zhenjiang	[15]	EC + Ultrasound	—	96%	69%	—	—

addition and air bubble flotation to separate pollutants in carwash wastewater. The SS and turbidity removal rates of the FF process are approximately 85% and 90%, respectively [16][17]. When coupled with other treatments, the SS and turbidity removal rates of this process can reach as high as 96% [18][19]. The FF process has a COD removal rate of approximately 70~80%, which can be increased when this process is coupled with other treatments. Thus, the FF process exhibits a turbidity and COD removal performance comparable to that of the EC process (Table 2).

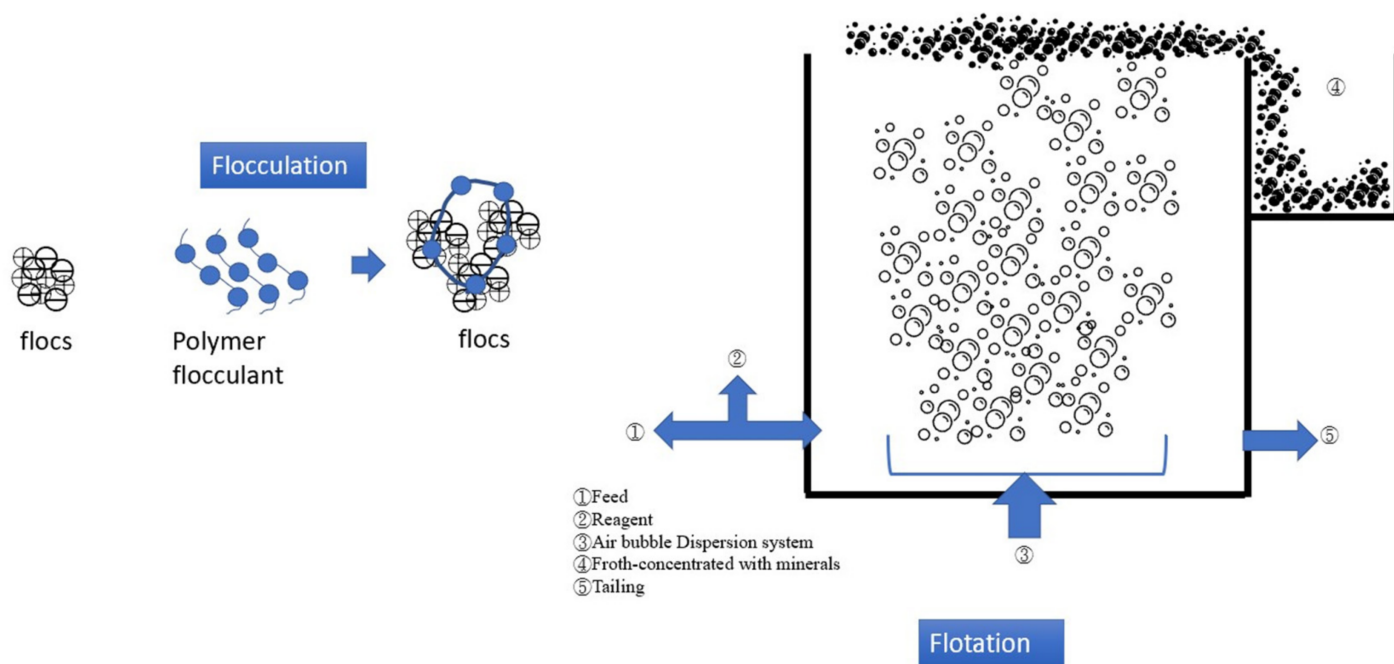


Figure 2. Schematic illustration of flocculation–flotation.

Table 2. Removal rate of various water qualities by flocculation–flotation method.

Country	Area	Ref.	Technique	SS (mg/L)	Turbidity (NTU)	COD (mg/L)	O&G (mg/L)	AS (mg/L)
Brazil	Porto Alegre	[17]	FF + O	83–99%	89–95% 93–98%	39–85% 81–99%	—	78–89% 81–99%
Brazil	Porto Alegre	[16]	FF	—	91–96%	—	—	40%
Brazil	Sao Paulo	[18]	FF + SF	—	87–91%	—	—	—
Brazil	Porto Alegre	[20]	FF	89%	93%	11%	—	—
Brazil	Porto Alegre	[19]	FF + SC	91–93%	91–96%	63–76%	—	—
Pakistan	Hyderabad, Sindh	[21]	DAF + F	—	97%	—	[22] 99%	—

has especially been used in many fields, for example mineral processing [23], removing surfactants [24], suspension filtration [25], and more. **Figure 3** illustrates the mechanism of filtration

Note: sand filtration (SF), ozonation (O), sand filtration and chlorination (SC), Filtration (F), Dissolved Air Flotation

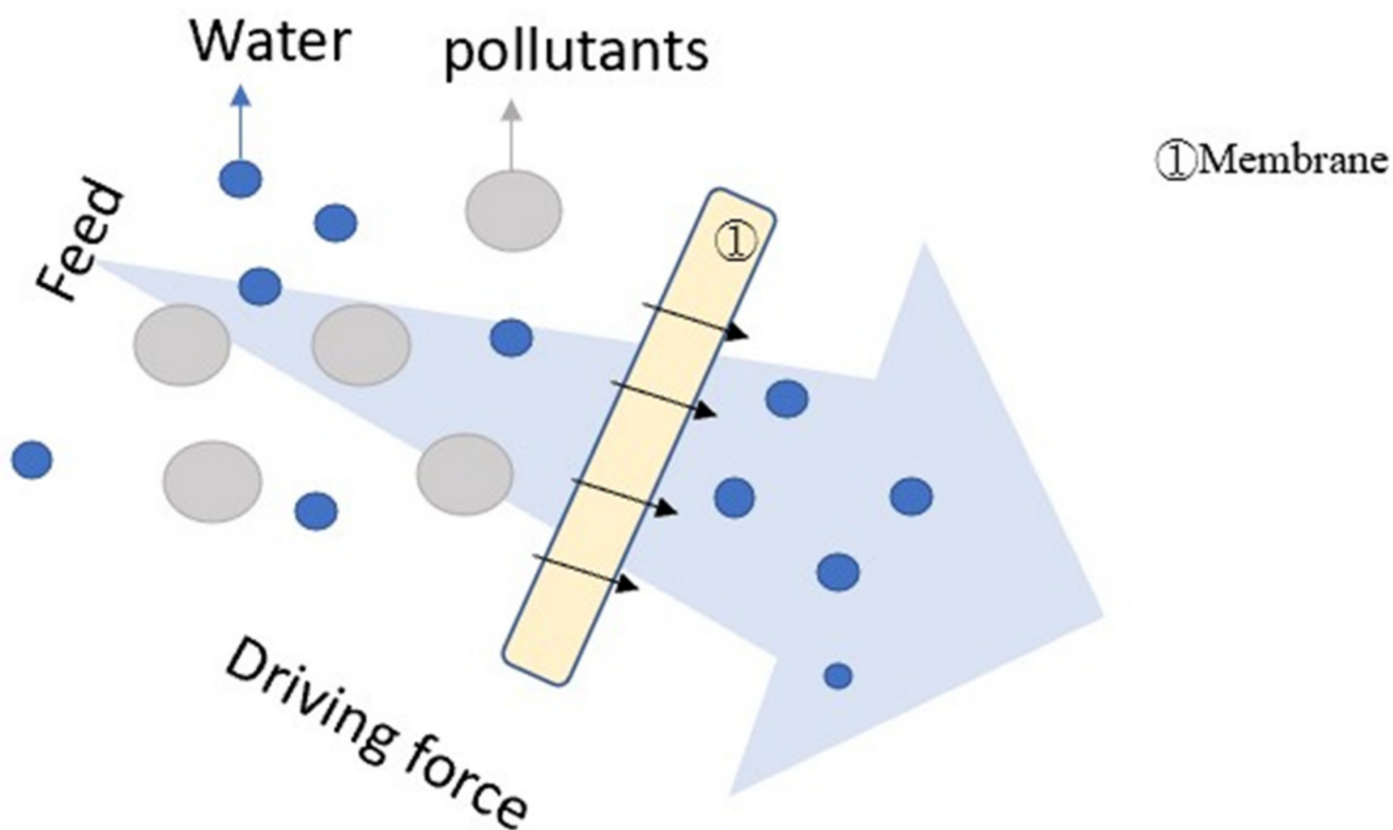


Figure 3. Illustration of filtration.

When the filter element has sufficient selectivity, the flocculation–filtration process can achieve SS and turbidity removal rates over 99%, as presented in [26][27]. However, the filtrate flux of flocculation–ultrafiltration and flocculation–nanofiltration are only approximately 50 and 10 LMH (L/m²-h), respectively. To provide wastewater

treatment for the medium-scale car wash factory discussed in [28], an ultrafiltration plant with a size of approximately 100 m² would be required. Such a plant would occupy a large space, and would thus be unsuitable for highly developed urban areas. Despite being able to remove partial COD, the general COD removal rate of the flocculation–filtration process is approximately 60% (Table 2).

The coagulation–filtration process has turbidity and COD removal rates of approximately 90% and 60%, respectively (Table 3).

Table 3. Removal rate of various water qualities by filtration.

Country	Area	Ref.	Technique	SS (mg/L)	Turbidity (NTU)	COD (mg/L)	O&G (mg/L)	AS (mg/L)
Belgium	Leuven	[28]	UF + NF	—	—	60–95%	—	88–95%
Sweden	—	[29]	UF	—	—	60%	—	—
Turkey	Istanbul	[13]	EC + NF	99%	—	88%	90%	91%
Malaysia	Johor, Skudai	[30]	UF + NF	—	—	55–92%	—	—
Brazil	Belo Horizonte	[31]	MF + UF	—	96.2–99.3%	81–85%	—	—
Turkey	Istanbul	[32]	UF + NF	—	—	Negligible–97%	—	—
Indonesia	Semarang	[27]	UF	—	100%	91%	83%	—
Japan	Tokyo	[33]	F + UF	—	75%	50–90%	—	—
China	Shanghai	[34]	C + UF	—	85%	80%	—	—
Vietnam	Hanoi	[35]	MBR + F	—	—	90%	88%	—
Australia	Melbourne	[36]	UF + RO	100%	99.9%	96%	—	—
Pakistan	Peshawar	[26]	SED + F	80%	99%	—	49.2%	—
India	Aligarh	[37]	SF	89.2%	—	83.5%	—	—
India	Trichy	[38]	UF	—	82%	47–60%	—	—

2.4. Coagulation–Flocculation (CF)

Figure 4 depicts the processes of coagulation–flocculation (CF), which is a two-stage reaction system. In the first stage, a coagulant (e.g., aluminum sulfate (Al₂(SO₄)₃), ferric chloride (FeCl₃), or ferric sulfate (Fe₂(SO₄)₃)) is added to the wastewater to neutralize the surface charge (ZPC) of the particles, thereby eliminating the electrostatic repulsion between the particles. The flocculant (i.e., polymer) is then added to the wastewater to aggregate the near-neutral electrostatic particles and form flocs for easier pollutant removal. Generally speaking, the turbidity removal rate of CF with car wash wastewater is good, generally over 90%; however, the removal rates of COD, O&G, and AS are not as good [4][39]. In addition, CF needs to add a suitable flocculant, which can easily cause cost increases and

secondary pollution. **Table 4** lists the effects of using CF and its combinations on car wash wastewater treatment as found in the literature.

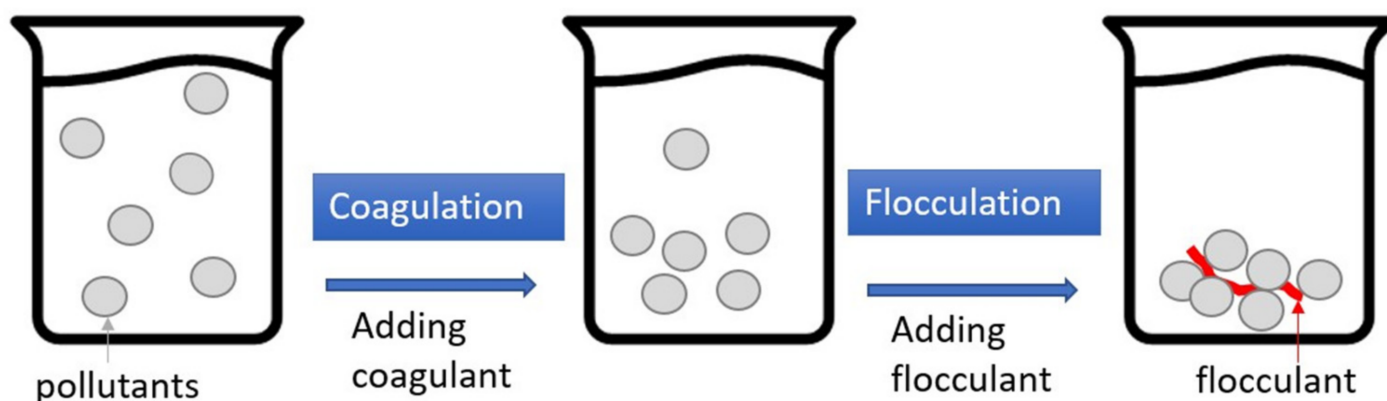


Figure 4. Schematic illustration of coagulation–flocculation (CF).

Table 4. Removal rate of various water qualities by coagulation–flocculation.

Country	Area	Ref.	Technique	SS (mg/L)	Turbidity (NTU)	COD (mg/L)	O&G (mg/L)	AS (mg/L)
Italy	Brescia	[4]	CF	—	98%	74%	—	—
Iran	Zahedan	[40]	C	37%	—	44%	—	76%
Egypt	Elminia	[41]	CF + SF + O + SF	—	100%	88%	—	—
Malaysia	Parit Raja, Johor	[39]	CF	—	97%	35%	—	—
Malaysia	Parit Raja	[42]	C	—	94%	60%	—	—
Malaysia	Taman University	[43]	C	—	90%	60%	—	—
China	Shenyang	[44]	C + UF	—	94%	—	>40%	—
China	Shanghai	[45]	C + M	—	70%	—	—	—
India	Bangalore	[46]	CF + F	—	—	80–90%	92–93%	—
Pakistan	Abbottabad	[47]	C + H ₂ O ₂	—	97%	93%	96%	—

2.5. Bio-Treatment

Figure 5 illustrates the mechanism of bio-treatment. Note: Chemical coagulation (C), membrane filtration (M). Aerobic microorganisms in the wastewater degrade the organics into H₂O and CO₂, while the dead biomass of microorganisms forms a sludge in the wastewater. **Table 5** lists the effects of biological treatment combined with other technologies on car wash wastewater treatment. For biological treatment followed by filtration treatment, the removal rate of turbidity, COD, and AS can reach more than 95% [3][48][49][50].

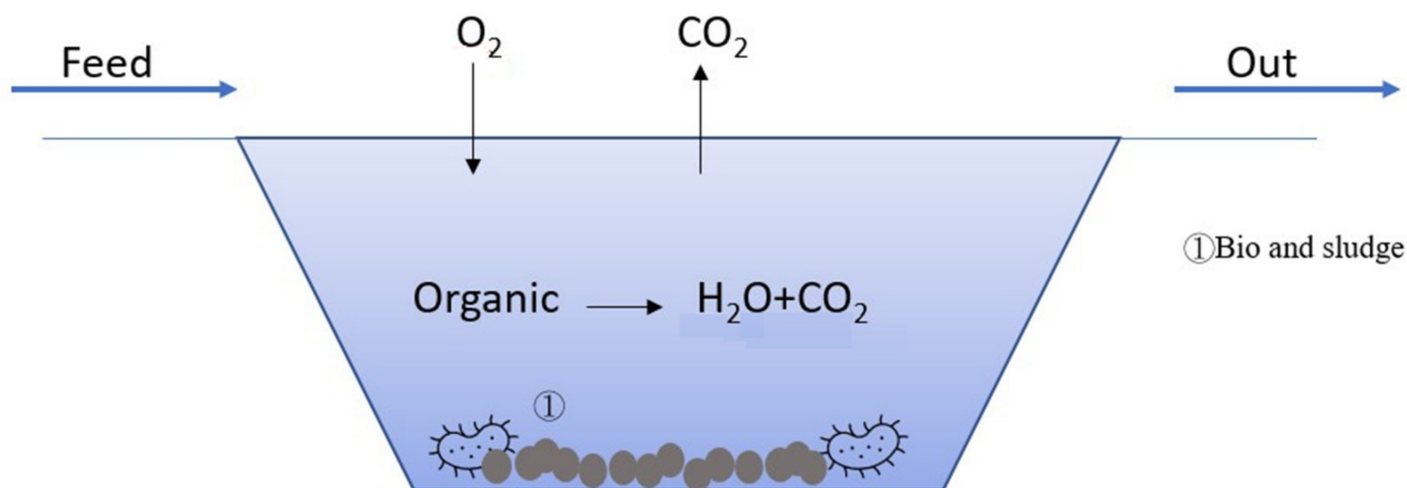


Figure 5. Illustration of bio-treatment.

Table 5. Removal rate of various water qualities by bio-treatment.

Country	Area	Ref.	Technique	SS (mg/L)	Turbidity (NTU)	COD (mg/L)	O&G (mg/L)	AS (mg/L)
Brazil	Sao Paulo	[3]	RBC + F	—	72~97%	56~94%	—	—
USA	New Jersey	[48]	four bioretention mesocosms	84~95%	—	—	—	89~96%
Taiwan	Hsinchu	[51]	Bio + M	95.7%	—	70.2%	—	—
Australia	Geelong	[49]	C + MBR	99.8%	99.6%	—	—	—
Australia	Melbourne	[50]	enhanced MBR (eMBR)	—	99.9%	99.8%	5.9~6.7 LMH	—

2.6. Other Methods

Rotating Biological Contactor (RBC).

A few other single treatment methods, such as the Photo-Fenton's process [52], adsorption [53], electro-oxidation [54], etc., are listed in the table below (Table 6). Except for electro-oxidation, these single-unit processing technologies have a removal rate of less than 90%.

Table 6. Removal rate of various water qualities by other single unit treatment techniques.

Country	Area	Ref.	Technique	SS (mg/L)	Turbidity (NTU)	COD (mg/L)	O&G (mg/L)	AS (mg/L)
Egypt	South of Egypt	[52]	Photo-Fenton's process	—	—	82~93.4%	—	—
Syria	Aleppo	[53]	AD	—	—	81.6%	86.8%	88.3%
Brazil	Natal	[54]	EO	—	—	96%	—	83~96%

References

1. Torkashvand, J.; Farzadkia, M.; Younesi, S.; Gholami, M. A systematic review on membrane technology for carwash wastewater treatment: Efficiency and limitations. *Desalin. Water Treat.* 2021, 210, 81–90.
2. Talebzadeh, F.; Valeo, C.; Gupta, R.; Constabel, C.P. Exploring the Potential in LID Technologies for Remediating Heavy Metals in Carwash Wastewater. *Sustainability* 2021, 13, 8727.
3. Subtil, E.L.; Rodrigues, R.; Hespanhol, I.; Mierzwa, J.C. Water reuse potential at heavy-duty vehicles washing facilities—The mass balance approach for conservative contaminants. *J. Clean. Prod.* 2017, 166, 1226–1234.
4. Vaccari, M.; Gialdini, F.; Collivignarelli, C. Study of the reuse of treated wastewater on waste container washing vehicles. *Waste Manag.* 2013, 33, 262–267.
5. Rubi-Juarez, H.; Barrera-Diaz, C.; Uena-Nunez, F. Adsorption-assisted electrocoagulation of real car wash wastewater with equilibrium and kinetic studies. *Pollut. Res.* 2017, 36, 175–184.
6. Rubi-Juarez, H.; Barrera-Diaz, C.; Linares-Hernandez, I.; Fall, C.; Bilyeu, B. A combined electrocoagulation-electrooxidation process for carwash wastewater reclamation. *Int. J. Electrochem. Sci.* 2015, 10, 6754–6767.
7. El-Ashtoukhy, E.S.Z.; Amin, N.K.; Fouad, Y.O. Treatment of real wastewater produced from Mobil car wash station using electrocoagulation technique. *Environ. Monit. Assess.* 2015, 187, 628–638.
8. Panizza, M.; Cerisola, G. Applicability of electrochemical methods to carwash wastewaters for reuse. Part 2: Electrocoagulation and anodic oxidation integrated process. *J. Electroanal. Chem.* 2010, 638, 236–240.
9. Mirshahghassemi, S.; Aminzadeh, B.; Torabian, A.; Afshinnia, K. Optimizing electrocoagulation and electro-Fenton process for treating car wash wastewater. *Environ. Health Eng. Manag. J.* 2017, 4, 37–43.
10. Mohammadi, M.J.; Salari, J.; Takdastan, A.; Farhadi, M.; Javanmardi, P.; Yari, A.R.; Dobaradaran, S.; Almasi, H.; Rahimi, S. Removal of turbidity and organic matter from car wash wastewater by electrocoagulation process. *Desalin. Water Treat.* 2017, 68, 122–128.
11. Mohammadi, M.J.; Takdastan, A.; Jorfi, S.; Neisi, A.; Farhadi, M.; Yari, A.R.; Dobaradaran, S.; Khaniabadi, Y.O. Electrocoagulation process to Chemical and Biological Oxygen Demand treatment from carwash grey water in Ahvaz megacity, Iran. *Data Brief* 2017, 11, 634–639.
12. Gomes, A.J.; Das, K.K.; Jame, S.A.; Cocke, D.L. Treatment of truck wash water using electrocoagulation. *Desalin. Water Treat.* 2016, 57, 25991–26002.
13. Gonder, Z.B.; Balcioglu, G.; Vergili, I.; Kaya, Y. An integrated electrocoagulation-nanofiltration process for carwash wastewater reuse. *Chemosphere* 2020, 253, 126713.

14. Kara, S. Treatment of transport container washing wastewater by electrocoagulation. *Environ. Prog. Sustain. Energy* 2013, 32, 249–256.
15. Chu, J.Y.; Li, Y.R.; Li, N.; Huang, W.H. Treatment of Car-washing Wastewater by Electrocoagulation-Ultrasound Technique for Reuse. *Adv. Mater. Res.* 2012, 433–440, 227–232.
16. Zaneti, R.; Etchepare, R.; Rubio, J. More environmentally friendly vehicle washes: Water reclamation. *J. Clean. Prod.* 2012, 37, 115–124.
17. Etchepare, R.; Zaneti, R.; Azevedo, A.; Rubio, J. Application of flocculation–flotation followed by ozonation in vehicle wash wastewater treatment/disinfection and water reclamation. *Desalin. Water Treat.* 2015, 56, 1728–1736.
18. Zaneti, R.; Etchepare, R.; Rubio, J. Car wash wastewater reclamation. Full-scale application and upcoming features. *Resour. Conserv. Recycl.* 2011, 55, 953–959.
19. Zaneti, R.N.; Etchepare, R.; Rubio, J. Car wash wastewater treatment and water reuse—A case study. *Water Sci. Technol.* 2013, 67, 82–88.
20. Rubio, J.; Zaneti, R.N. Treatment of washrack wastewater with water recycling by advanced flocculation–column flotation. *Desalin. Water Treat.* 2009, 8, 146–153.
21. Bhatti, S.; Siddiqui, Z.; Memon, S.; Kandhir, I.; Memon, M.A.; Mahesar, A.W. Analysis and treatment wash off water from vehicular service station in Hyderabad. *Sindh Univ. Res. J. SURJ (Sci. Ser.)* 2017, 49, 473–478.
22. Zapién Serrano, L.Z.; Ortiz Lara, N.O.; Ríos Vera, R.R.; Cholico-González, D. Removal of Fe(III), Cd(II), and Zn(II) as Hydroxides by Precipitation–Flotation System. *Sustainability* 2021, 13, 11913.
23. Park, J.H.; Han, Y.S.; Ji, S.W. Investigation of Mineral-Processing Wastewater Recycling Processes: A Pilot Study. *Sustainability* 2018, 10, 3069.
24. Siddig, O.; Al-Afnan, S.; Elkatatny, S.; Bahgat, M. Novel Cake Washer for Removing Oil-Based Calcium Carbonate Filter Cake in Horizontal Wells. *Sustainability* 2020, 12, 3427.
25. Mohamed, A.; Basfar, S.; Elkatatny, S.; Al-Majed, A. Prevention of Barite Sag in Oil-Based Drilling Fluids Using a Mixture of Barite and Ilmenite as Weighting Material. *Sustainability* 2019, 11, 5617.
26. Syed, N.H.; Ahmad, J.; Khan, N.A.; Khan, N.; Shafiq, M.A. A low-cost wastewater treatment unit for reducing the usage of fresh water at car wash stations in Pakistan. *Pak. J. Sci. Ind. Res. A Phys. Sci.* 2019, 62A, 57–66.
27. Istirokhatun, T.; Destianti, P.; Hargianintya, A.; Oktiawan, W.; Susanto, H. Treatment of car wash wastewater by UF membranes. In *Proceedings of the International Conference of Chemical and Material Engineering*, Kyoto, Japan, 29 December 2015; p. 060025.

28. Boussu, K.; Kindts, C.; Vandecasteele, C.; van der Bruggen, B. Applicability of nanofiltration in the carwash industry. *Sep. Purif. Technol.* 2007, 54, 139–146.
29. Jönsson, C.; Jönsson, A.S. The influence of degreasing agents used at car washes on the performance of ultrafiltration membranes. *Desalination* 1995, 100, 115–123.
30. Lau, W.J.; Ismail, A.F.; Firdaus, S. Car wash industry in Malaysia: Treatment of car wash effluent using ultrafiltration and nanofiltration membranes. *Sep. Purif. Technol.* 2013, 104, 26–31.
31. Pinto, A.C.S.; de Barros Grossi, L.; de Melo, R.A.C.; de Assis, T.M.; Ribeiro, V.M.; Amaral, M.C.S.; de Souza Figueiredo, K.C. Carwash wastewater treatment by micro and ultrafiltration membranes: Effects of geometry, pore size, pressure difference and feed flow rate in transport properties. *J. Water Process Eng.* 2017, 17, 143–148.
32. Uçar, D. Membrane processes for the reuse of car washing wastewater. *J. Water Reuse Desalin.* 2018, 8, 169–175.
33. Hamada, T.; Miyazaki, Y. Reuse of carwash water with a cellulose acetate ultrafiltration membrane aided by flocculation and activated carbon treatments. *Desalination* 2004, 169, 257–267.
34. Tan, X.; Tang, L. Application of enhanced coagulation aided by UF membrane for car wash wastewater treatment. In *Proceedings of the 2008 2nd International Conference on Bioinformatics and Biomedical Engineering, Shanghai, China, 16–18 May 2008*; pp. 3653–3656.
35. Do, K.U.; Kim, J.H.; Chu, X.Q. Sludge characteristics and performance of a membrane bioreactor for treating oily wastewater from a car wash service station. *Desalin. Water Treat.* 2018, 120, 166–172.
36. Moazzem, S.; Wills, J.; Fan, L.; Roddick, F.; Jegatheesan, V. Performance of ceramic ultrafiltration and reverse osmosis membranes in treating car wash wastewater for reuse. *Environ. Sci. Pollut. Res.* 2018, 25, 8654–8668.
37. Alam, J.; Farooqi, I.H. Management of grey water of an automobile workshop—A case study. In *Proceedings of the International Workshop on Civil Engineering and Architecture, Istanbul, Turkey, 8–9 August 2014*; pp. 133–138.
38. Kiran, S.A.; Arthanareeswaran, G.; Thuyavan, Y.L.; Ismail, A.F. Influence of bentonite in polymer membranes for effective treatment of car wash effluent to protect the ecosystem. *Ecotoxicol. Environmen. Saf.* 2015, 121, 186–192.
39. Al-Gheethi, A.A.; Mohamed, R.M.S.R.; Rahman, M.A.A.; Johari, M.R.; Kassim, A.H.M. Treatment of wastewater from car washes using natural coagulation and filtration system. *IOP Conf. Ser. Mater. Sci. Eng.* 2016, 136, 012046.
40. Bazrafshan, E.; Mostafapoor, F.K.; Soori, M.M.; Mahvi, A.H. Application of combined chemical coagulation and electrocoagulation process to carwash wastewater treatment. *Fresenius Environ.*

- Bull. 2012, 21, 2694–2701.
41. Abdelmoez, W.; Barakat, N.A.M.; Moaz, A. Treatment of wastewater contaminated with detergents and mineral oils using effective and scalable technology. *Water Sci. Technol.* 2013, 68, 974–981.
 42. Radin Mohamed, R.M.S.; Abdul Rahman, N.; Mohd Kassim, A.H. *Moringa Oleifera* and *Strychnos Potatorum* seeds as natural coagulant compared with synthetic common coagulants in treating car wash wastewater: Case study 1. *Asian J. Appl. Sci.* 2014, 2, 693–700.
 43. Mohamed, R.M.S.R.; Saphira, R.M.; Kutty, A.I.; Mariam, N.; Kassim, M.; Hashim, A. Efficiency of using commercial and natural coagulants in treating car wash wastewater treatment. *Aust. J. Basic Appl. Sci.* 2014, 8, 227–234.
 44. Zhang, J.K.; Yang, Y.B.; Wang, H.Y.; Dong, Z.B. CFU combined process for the treatment of oily car washing wastewater. *Appl. Mech. Mater.* 2013, 253–255, 999–1004.
 45. Tang, L.; Tan, X.J.; Cui, F.Y.; Zhou, Q.; Yin, J. Reuse of carwash wastewater with hollow fiber membrane aided by enhanced coagulation and activated carbon treatments. *Water Sci. Technol.* 2007, 56, 111–118.
 46. Asha, M.N.; Chandan, K.S.; Harish, H.P.; NikhileswarReddy, S.; Sharath, K.S.; Liza, G.M. Recycling of waste water collected from automobile service station. *Procedia. Environ. Sci.* 2016, 35, 289–297.
 47. Bhatti, Z.A.; Mahmood, Q.; Raja, I.A.; Malik, A.H.; Khan, M.S.; Wu, D. Chemical oxidation of carwash industry wastewater as an effort to decrease water pollution. *Phys. Chem. Earth Parts A/B/C* 2011, 36, 465–469.
 48. Bakacs, M.E.; Yergeau, S.E.; Obropt, C.C.; ASCE, P.E.M. Assessment of car wash runoff treatment using bioretention mesocosms. *J. Environ. Eng.* 2013, 139, 1132–1136.
 49. Boluarte, I.A.R.; Andersen, M.; Pramanik, B.K.; Chang, C.Y.; Bagshaw, S.; Farago, L.; Jegatheesan, V.; Shu, L. Reuse of car wash wastewater by chemical coagulation and membrane bioreactor treatment processes. *Int. Biodeterior. Biodegrad.* 2016, 113, 44–48.
 50. Moazzem, S.; Ravishankar, H.; Fan, L.; Roddick, F.; Jegatheesan, V. Application of enhanced membrane bioreactor (eMBR) for the reuse of carwash wastewater. *J. Environ. Manag.* 2020, 254, 109780.
 51. Hsu, S.K.; Chen, C.H.; Chang, W.K. Reclamation of car washing wastewater by a hybrid system combining bio-carriers and non-woven membranes filtration. *Desalin. Water Treat.* 2011, 34, 349–353.
 52. Tony, M.A.; Bedri, Z. Experimental design of photo-Fenton reactions for the treatment of car wash wastewater effluents by response surface methodological analysis. *Adv. Environ. Chem.* 2014,

2014, 958134.

53. Baddor, I.M.; Farhoud, N.; Abdel-Magid, I.M.; Alshami, S.; Hassan Ahmad, F.; Olabi, E.A. Study of car wash wastewater treatment by adsorption. In Proceedings of the International Conference of Engineering, Information Technology, and Science, Kuala Lumpur, Malaysia, 1 May 2014; pp. 2–22.
54. Ganiyu, S.O.; dos Santos, E.V.; de Araújo Costa, E.C.T.; Martínez-Huitle, C.A. Electrochemical advanced oxidation processes (EAOPs) as alternative treatment techniques for carwash wastewater reclamation. *Chemosphere* 2018, 211, 998–1006.

Retrieved from <https://www.encyclopedia.pub/entry/history/show/58598>