

Functional Coatings

Subjects: **Others**

Contributor: Radu Claudiu Fierascu , Irina Fierascu , Irina Elena Chican

Coating materials have gained researchers' interest, finding applications in different areas such as antimicrobial coatings for biomedical applications, coatings for increasing the shelf-life of commercial products, or coatings for the conservation of cultural heritage artifacts.

functional coatings

natural materials

antimicrobial

1. Introduction

In recent decades, coating materials have gained researchers' interest, and the registered progress has led to the evaluation of their potential use in different applications, including some of great importance, such as the development of antimicrobial coatings for biomedical applications, coatings applied for increasing the shelf-life of commercial products, or coatings used for the conservation of cultural heritage artifacts. In its widest sense, a coating represents a thin layer of material that, when deposited or applied on the surface of a support material, improves the surface properties of the substrates and creates a protective barrier against detrimental external factors ^[1]. Considering their envisaged properties and final applications, different types of coatings were developed, from monolayer to nanostructured and/or nanometric-scale multilayer coatings ^[2].

The permanent search for new types of coating materials, accompanied by the shift in consumer's habits towards "bio-products" led to an increasing interest regarding the application of functional coatings based on natural products, especially in industries such as food or cosmetic. In the food industry, as an example regarding the potential of natural alternatives, the use of wax coatings on citrus (as a method to protect them against microbial spoilage) was known and applied in ancient China ^[3]; nowadays, the development of new materials and production technologies has new valences—to obtain enhanced biodegradability and eco-friendly properties ^[4].

For other applications, such as development of antifouling coatings, the interest is to replace classical used materials (such as tributyltin or zinc pyrithione), which can have negative effects on non-target organisms, with some natural-based alternatives ^[5].

Widely applied, natural and synthetic polymers are commonly used as coating layers, and their enrichment with nanoparticles, inorganic and organic materials increases the variety of available coatings, with the developed hybrid materials overcoming deficiencies by combining the advantages of each component ^{[6][7]}.

2. Coating Materials for Different Industries

2.1. Food Industry

Recently, different materials were developed, in order to be used in the food industry to prolong the shelf-life of food products, presented as packings or even coatings used as layers on commercialized merchandise. The interest in using natural compounds started in controlling the postharvest disease of fruits and vegetables [8]. Usually, these coatings are based on natural materials, classified, according to their origin, as of animal origin (casein, gelatin, collagen, etc.), and of vegetable origin (proteins, polysaccharides, waxes, etc.) [9]. In contrast to the harmful chemicals, natural functional coatings represent a non-toxic, biocompatible and biodegradable alternative. This type of coatings can decrease the respiration level and migration of water for perishable products [10], and, serve as good delivery vectors for antimicrobial substances, nutraceuticals and flavors [11].

For obtaining coatings as edible films for the food industry, the methods are based on two different routes, wet and dry processes, each of them having advantages and drawbacks, related not only to the raw materials, but also the envisaged application. They are based on solvent casting and extrusion processes, solubility of the materials being a main parameter for the first method and gelatinization for the second one [12].

The solvent casting method involves a three steps process: solubilization of raw materials in a suitable solvent, casting of the obtained solution, respectively drying of casted solution under proper conditions; each step parameter influences the intermediate and final products, the main advantage being low-cost film manufacturing without a specialized equipment. However, due to multiple operation conditions, this method present several bottlenecks: the proper choice of the solvent—solvents must be non-toxic, as there it remains a possibility that the solvent can impurify the polymer and to negatively affect the active substances characteristics; the proper choice of the molds—for food industry, the mechanical characteristics of the obtained films must permit molding on different types of products; extra time for drying into the molds—slows down and makes the process more expensive; proper temperatures for drying—the process must be optimized for proper temperatures in order not to damage the films and final products, which often degrades at higher temperatures [13][14].

Extrusion methods are in use at industrial scales for obtaining polymeric coating films, but they involve the use of specialized equipment and supplementary materials, such as plasticizers and stabilizers. Through thermomechanical processes such as extrusion, injection, kneading, and casting, different films can be obtained [15], a proper choice of the added plasticizer influencing the physicochemical properties of the final material, decreasing hydrogen bonding between polymers, and increasing intermolecular spacing. For example, hydrophilic plasticizers such as protein hydrolysates can enhance water vapor permeability levels of the vegetal based films, having a double role: plasticizer and active ingredient [16].

Moreover, for extrusion methods, a major challenge is maintaining a balance in choosing the suitable equipment operation parameters and proper plasticizers and stabilizers, this balance being a key point for obtaining coating materials with specific physico-chemical characteristics (homogeneity, shear rate, shear stress, and residence time control). An example of choosing proper conditions is the study of Sun et al. [17]. The authors developed starch/polyhydroxyalkanoate films through extrusion method using as cross-linking agent, citric acid, adipic acid,

boric acid, and borax, at a blending and compounding screw speed of 60 rpm, and film blowing screw speed of 30 rpm. The obtained films proved to have the best properties (high tensile strength (9 MPa) and elongation at break over than 60%) for the above presented operations conditions, only when the citric acid and adipic acid were applied.

The extrusion method for obtaining coating materials is used at industrial levels with high performance and low costs [18]. The disadvantage of this method is mainly related to the processing temperature when the raw material is based on more than two components, as all the components must have similar melting points.

Coating materials are an interesting domain and the concept of active packaging gained new valences, offering the possibility of interaction among the food product, the packaging, and the environment, new smart packaging systems based on natural products being developed in order to improve the quality of food products, thus prolonging their shelf-life [19][20][21][22][23].

The deposition methods are dependent on the nature of food that should be coated, the support surface being a main parameter for a successful coating [4]. Additionally, surface tension, density, and viscosity of the coating material influence the selection of applying methods and equipment used. Dipping and spraying technologies are usually used as application methods of coating materials for the fresh products, ensuring uniformity across a rough and complex shape. In the case of spraying method, several shortcomings are mainly related to the working parameters of the equipment, which must be corroborated with the coating's physico-chemical properties [24]. For a low-density coating material, fluidized-bed processing is the method of choice, and in confectionery sector, panning method is used to apply thick layers to hard materials.

2.2. Cosmetic Industry

For coating systems used in cosmetic formulations and pharmaceutical industry, the approach is different than in the above discussed sections dedicated to medical devices or food industry. The subject is very vast, and itself can be a topic for a review paper, but some interesting examples of coating materials and active substances will be discussed. The development of new coating delivery systems in cosmetic formulations and pharmaceutical industry, nowadays follow concerns regarding the use of green technologies and non-toxic environmental materials. Developing cosmetic formulations is a significant challenge as there can appear diverse bottlenecks, the final target efficacy (sun screen protection, antiaging, antiwrinkle, etc.) being dependent on the vehicle formulation, choice of emulsifiers, solvents and emollients.

Different types of nanoengineered coating systems (e.g., liposomes, niosomes, transferosomes, lipid nanoparticles, core-shell materials), polymeric microparticles, nanoparticles, inorganic materials, etc., have been successfully used in cosmetic formulations, improving the penetration of active substances into the skin (**Figure 1**).

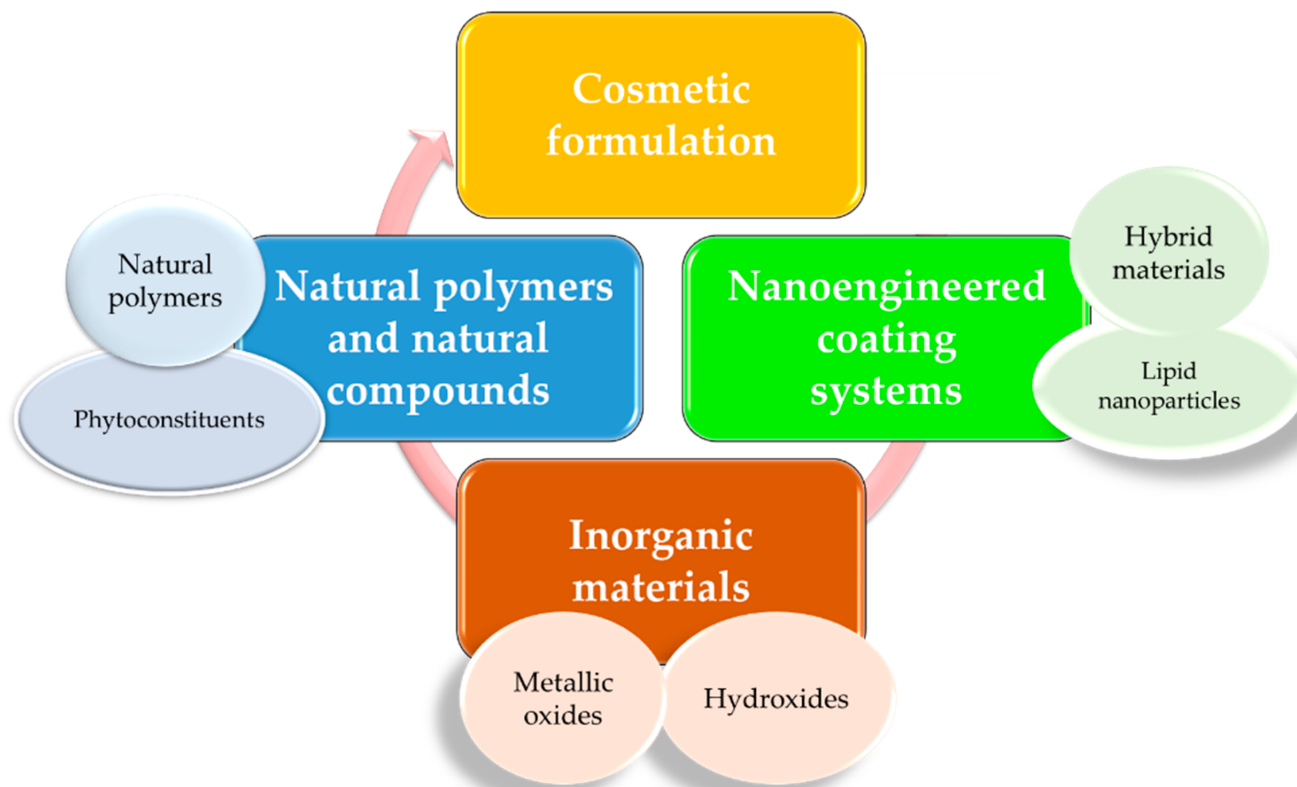


Figure 1. Potential coating materials for cosmetic formulation.

For cosmetic formulations, the requirements are different than for a topical pharmaceutical preparation, many factors influencing the properties of the final products. The coating material of the active substance must be compatible with the oil-water emulsion, water-oil emulsion or double emulsion in the case of cosmetic formulations [25].

2.3. Coatings for Miscellaneous Applications

Different types of pollutants are produced nowadays in large amounts due to a rapid increase in the world population and widespread industrialization, so the development of new materials for environmental remediation represents a goal of outmost importance [26]. Advanced oxidation processes (AOP) are by far the most applied methods for removing hazardous compounds from water. Catalytic ozonation represents an AOP process with high efficiency in the application on effluents with a low and average flow rate and pollutants concentration [27]. The optimization of this process, in order to achieve better performances is very difficult and requires multiparametric monitoring of oxidation reaction, the catalyst playing a main role. In the last decade, numerous materials and technologies were proposed for a better and complete oxidation processes. In porous structures, nanoparticles are formed inside pores as a result of a multistep, costly and difficult process without having a very good control over their chemical composition, dimensions, and shape [28].

An interesting alternative is represented by the immobilization of catalyst in the form of thin-films coatings using different substrates (glass, quartz, polymers, textile materials, etc.), according to their uses, having the advantage

of being cost effective and easily recycled after washing [29][30]. Thin film morphology containing phytosynthesized metallic or metal oxides nanoparticles gained more attention in the last decades, being suitable for catalytic processes, since this approach includes advantages of using nanoparticles and avoid the diffusivity problems (in fixed bed systems) or daunting separation step (in slurry approach). Another innovative and eco-friendly aspect of this type of materials consists in the preparation of novel thin-film catalysts using plant extracts, without harmful substances [31][32]. Additionally, phyto-mediated metallic or metal oxide nanoparticles can act as catalysts themselves (being coated by the phytoconstituents), or be included in nanocomposite materials for the catalytic reduction of organic dyes [30][33][34], or can be coated by natural minerals, such as perlite [35].

For other environmental applications, coating materials based on natural product can be successfully applied. This is the case of sensors developed using silver nanoparticles coated with natural polymers, such as xylan, which are able to detect Hg^{2+} at a 4nM detection limit [36] or natural minerals such as Cs_2SnI_6 perovskite used for coating ZnO nanorods for solar cells [37]. Coatings based on rhamnolipid biosurfactant, gum karaya and xanthan gum are able to inhibit the agglomeration of nanoparticles, such as zero valent iron nanoparticles, in order to be used for organic and inorganic bioremediation [38][39][40][41]. For soil remediation applications, coating layer has a double role: protect the inorganic core against agglomeration and percolate into the spaces between soil particles in order to travel longer distances before being trapped by the soil matrix [42].

Environmental factors affect not only the human health, being able to damage important cultural heritage objects which are kept in outdoor conditions. The use of protective coatings gained importance in heritage conservation, and coatings based on natural products replace successfully chemical hazardous substances [43]. From the category of inorganic compounds, calcium hydroxide can be used as a coating material, due to its property to be transformed under natural atmospheric conditions into a hard coating of insoluble CaCO_3 . Moreover, when $\text{Ca}(\text{OH})_2$ is applied in its nanoparticle state, the deposition and penetration of prepared suspensions through external porous of the surface are enhanced. Daniele and Taglieri reported the efficiency of consolidation with this type of coating more than 75% for stone samples [44], while Lanzón et al. reported that nanoparticles with dimensions in the range 200 to 600 nm can penetrate through conventional pores of mechanically weak materials (lime mortars), this type of coating being used for the consolidation of deteriorated walls of the Roman Theatre of Cartagena [45]. Despite the disadvantages (160 repetitions) presented by Slízková et al. [46], the use of alcohols as dispersion media can overcome the drawbacks of a large number of applications of the coating material [45].

As a very interesting aspect regarding the potential use of natural resources for the development of next-generation materials for highly specialized scientific, medical or precision equipment, the biomachining process can be mentioned. It represents the metal processing using lithotrophic (usually extremophile) bacteria, as an intermediate step, before applying the final functional coatings. An example on this topic is the work of Díaz-Tena et al., presenting the application of *Acidithiobacillus ferrooxidans* for biomachining oxygen-free copper, resulting in a finished and/or engraved material surface, without the shortcomings of traditional processing methods [47].

For wooden cultural heritage objects, the appropriate materials for coating layers are natural products, such as proteins, lipids, polysaccharides, or terpenoids, having a waterproofing role and, occasionally, as ingredients of

binding media [48], while for iron artefacts, adding natural compounds (tannins) into commercial resins can inhibit their corrosion [49]. For conservation studies, natural polymer coatings are preferred instead of resins, for indoor objects, due to their water-solubility, thus avoiding the use of harmful solvents, necessary for the application and removal of commonly used commercial protective coatings. Giuliani et al. used a coating based on chitosan, which acted as a reservoir for the inhibitors benzotriazole and mercaptobenzothiazole, for bronze artefacts, contributing to the formation of a barrier layer, thus improving the protective properties of the treatment [50].

Conservation methods based on the chemical strategies of using natural coatings can be adapted for paper artefacts, too. Jia et al. obtained an antibacterial and antifungal composite material based on ZnO/cellulose nanocrystals, and the chemical and mechanical properties of coated papers after dry heat and UV accelerated aging were measured, in order to demonstrate the efficiency of the treatment [51], the cellulose layer having a good compatibility and affinity with paper cellulose fibers. The treated papers had higher thermal and UV stability and exhibited an inferior loss of strength. For papers coated with chitosan nanoparticles, the protection is due to the formation of chitosan protective layer when the nanoparticles combined with H₂O and H⁺, the deacidification of the paper being produced (a pH increase from 5 to 7 being recorded), also increasing tensile strength and folding endurance [52].

References

1. Ong, G.; Kasi, R.; Subramaniam, R. A review on plant extracts as natural additives in coating applications. *Prog. Org. Coat.* 2021, 151, 106091.
2. Rodríguez-Barrero, S.; Fernández-Larrinoa, J.; Azkona, I.; De Lacalle, L.N.L.; Polvorosa, R. Enhanced performance of nanostructured coatings for drilling by droplet elimination. *Mater. Manuf. Process.* 2016, 31, 593–602.
3. Kumar, N. Polysaccharide-based component and their relevance in edible film/coating: A review. *Nutr. Food Sci.* 2019, 49, 793–823.
4. Suhag, R.; Kumar, N.; Petkoska, A.T.; Upadhyay, A. Film formation and deposition methods of edible coating on food products: A review. *Food Res. Int.* 2020, 136, 109582.
5. Chen, L.; Duan, Y.; Cui, M.; Huang, R.; Su, R.; Qi, W.; He, Z. Biomimetic surface coatings for marine antifouling: Natural antifoulants, synthetic polymers and surface microtopography. *Sci. Total Environ.* 2021, 766, 144469.
6. Zentel, R. Polymer coated semiconducting nanoparticles for hybrid materials. *Inorganics* 2020, 8, 20.
7. Hussain, A.K.; Sudin, I.; Basheer, U.M.; Yusop, M.Z.M. A review on graphene-based polymer composite coatings for the corrosion protection of metals. *Corros. Rev.* 2019, 37, 343–363.

8. Fierascu, R.; Fierascu, I.; Baroi, A.; Ortan, A. Selected aspects related to medicinal and aromatic plants as alternative sources of bioactive compounds. *Int. J. Mol. Sci.* 2021, 22, 1521.
9. Mokrejs, P.; Langmaier, F.; Janacova, D.; Mladek, M.; Kolomaznik, K.; Vasek, V. Thermal study and solubility tests of films based on amaranth flour starch–protein hydrolysate. *J. Therm. Anal. Calorim.* 2009, 98, 299–307.
10. Beikzadeh, S.; Khezerlou, A.; Jafari, S.M.; Pilevar, Z.; Mortazavian, A.M. Seed mucilages as the functional ingredients for biodegradable films and edible coatings in the food industry. *Adv. Colloid Interface Sci.* 2020, 280, 102164.
11. Wongphan, P.; Harnkarnsujarit, N. Characterization of starch, agar and maltodextrin blends for controlled dissolution of edible films. *Int. J. Biol. Macromol.* 2020, 156, 80–93.
12. Sharma, R.; Ghoshal, G. Emerging trends in food packaging. *Nutr. Food Sci.* 2018, 48, 764–779.
13. Velaga, S.P.; Nikjoo, D.; Vuddanda, P.R. Experimental studies and modeling of the drying kinetics of multicomponent polymer films. *AAPS PharmSciTech* 2018, 19, 425–435.
14. Rodríguez, G.M.; Sibaja, J.C.; Espitia, P.J.P.; Otoni, C.G. Antioxidant active packaging based on papaya edible films incorporated with *Moringa oleifera* and ascorbic acid for food preservation. *Food Hydrocoll.* 2020, 103, 105630.
15. Avramescu, S.M.; Butean, C.; Popa, C.V.; Ortan, A.; Moraru, I.; Temocico, G. Edible and functionalized films/coatings—Performances and perspectives. *Coatings* 2020, 10, 687.
16. Zhang, C.; Wang, Z.; Li, Y.; Yang, Y.; Ju, X.; He, R. The preparation and physiochemical characterization of rapeseed protein hydrolysate-chitosan composite films. *Food Chem.* 2019, 272, 694–701.
17. Sun, S.; Liu, P.; Ji, N.; Hou, H.; Dong, H. Effects of various cross-linking agents on the physicochemical properties of starch/PHA composite films produced by extrusion blowing. *Food Hydrocoll.* 2018, 77, 964–975.
18. Anukiruthika, T.; Sethupathy, P.; Wilson, A.; Kashampur, K.; Moses, J.A.; Anandharamakrishnan, C. Multilayer packaging: Advances in preparation techniques and emerging food applications. *Compr. Rev. Food Sci. Food Saf.* 2020, 19, 1156–1186.
19. Aragüez, L.; Colombo, A.; Borneo, R.; Aguirre, A. Active packaging from triticale flour films for prolonging storage life of cherry tomato. *Food Packag. Shelf Life* 2020, 25, 100520.
20. Ceballos, R.L.; Ochoa-Yepes, O.; Goyanes, S.; Bernal, C.; Famá, L. Effect of yerba mate extract on the performance of starch films obtained by extrusion and compression molding as active and smart packaging. *Carbohydr. Polym.* 2020, 244, 116495.
21. Cheng, Y.; Wang, W.; Zhang, R.; Zhai, X.; Hou, H. Effect of gelatin bloom values on the physicochemical properties of starch/gelatin–beeswax composite films fabricated by extrusion

- blowing. *Food Hydrocoll.* 2021, 113, 106466.
22. El-Sayed, S.M.; El-Sayed, H.S.; Ibrahim, O.A.; Youssef, A.M. Rational design of chitosan/guar gum/zinc oxide bionanocomposites based on Roselle calyx extract for Ras cheese coating. *Carbohydr. Polym.* 2020, 239, 116234.
 23. Vedove, T.M.; Maniglia, B.C.; Tadini, C.C. Production of sustainable smart packaging based on cassava starch and anthocyanin by an extrusion process. *J. Food Eng.* 2021, 289, 110274.
 24. Peretto, G.; Du, W.-X.; Avena-Bustillos, R.J.; Berrios, J.D.J.; Sambo, P.; McHugh, T.H. Electrostatic and conventional spraying of alginate-based edible coating with natural antimicrobials for preserving fresh strawberry quality. *Food Bioprocess Technol.* 2017, 10, 165–174.
 25. Costa, R.; Santos, L. Delivery systems for cosmetics—From manufacturing to the skin of natural antioxidants. *Powder Technol.* 2017, 322, 402–416.
 26. Pedanekar, R.; Shaikh, S.; Rajpure, K. Thin film photocatalysis for environmental remediation: A status review. *Curr. Appl. Phys.* 2020, 20, 931–952.
 27. Rekhate, C.V.; Srivastava, J. Recent advances in ozone-based advanced oxidation processes for treatment of wastewater- A review. *Chem. Eng. J. Adv.* 2020, 3, 100031.
 28. Vebber, M.C.; Aguzzoli, C.; Beltrami, L.V.R.; Fetter, G.; Crespo, J.D.S.; Giovanela, M. Self-assembled thin films of PAA/PAH/TiO₂ for the photooxidation of ibuprofen. Part II: Characterization, sensitization, kinetics and reutilization. *Chem. Eng. J.* 2019, 361, 1487–1496.
 29. Chang, C.; Wei, Y.; Kuo, W. Free-standing CuS-ZnS decorated carbon nanotube films as immobilized photocatalysts for hydrogen production. *Int. J. Hydrog. Energy* 2018, 44, 30553–30562.
 30. Nasrollahzadeh, M.; Sajjadi, M.; Dasmeh, H.R.; Sajadi, S.M. Green synthesis of the Cu/sodium borosilicate nanocomposite and investigation of its catalytic activity. *J. Alloys. Compd.* 2018, 763, 1024–1034.
 31. Baruah, R.; Yadav, A.; Das, A.M. *Livistona jekinsiana* fabricated ZnO nanoparticles and their detrimental effect towards anthropogenic organic pollutants and human pathogenic bacteria. *Spectrochim. Acta Mol. Biomol. Spectrosc.* 2021, 251, 119459.
 32. Diaz-Urbe, C.; Vallejo, W.; Romero, E.; Villareal, M.; Padilla, M.; Hazbun, N.; Muñoz-Acevedo, A.; Schott, E.; Zarate, X. TiO₂ thin films sensitization with natural dyes extracted from *Bactris guineensis* for photocatalytic applications: Experimental and DFT study. *J. Saudi Chem. Soc.* 2020, 24, 407–416.
 33. Bordbar, M.; Negahdar, N.; Nasrollahzadeh, M. *Melissa Officinalis* L. leaf extract assisted green synthesis of CuO/ZnO nanocomposite for the reduction of 4-nitrophenol and Rhodamine B. *Sep.*

- Purif. Technol. 2018, 191, 295–300.
34. Maham, M.; Nasrollahzadeh, M.; Sajadi, S.M.; Nekoei, M. Biosynthesis of Ag/reduced graphene oxide/Fe₃O₄ using *Lotus garcinii* leaf extract and its application as a recyclable nanocatalyst for the reduction of 4-nitrophenol and organic dyes. *J. Colloid Interface Sci.* 2017, 497, 33–42.
 35. Maryami, M.; Nasrollahzadeh, M.; Mehdipour, E.; Sajadi, S.M. Green synthesis of the Pd/perlite nanocomposite using *Euphorbia neriifolia* L. leaf extract and evaluation of its catalytic activity. *Sep. Purif. Technol.* 2017, 184, 298–307.
 36. Luo, Y.; Shen, S.; Luo, J.; Wang, X.; Sun, R. Green synthesis of silver nanoparticles in xylan solution via Tollens reaction and their detection for Hg₂⁺. *Nanoscale* 2015, 7, 690–700.
 37. Qiu, X.; Jiang, Y.; Zhang, H.; Qiu, Z.; Yuan, S.; Wang, P.; Cao, B. Lead-free mesoscopic Cs₂SnI₆ perovskite solar cells using different nanostructured ZnO nanorods as electron transport layers. *Phys. Status Solidi RRL Rapid Res. Lett.* 2016, 10, 587–591.
 38. Zhan, J.; Zheng, T.; Piringier, G.; Day, C.; McPherson, G.L.; Lu, Y.; Papadopoulos, K.; John, V.T. Transport characteristics of nanoscale functional zerovalent iron/silica composites for in situ remediation of trichloroethylene. *Environ. Sci. Technol.* 2008, 42, 8871–8876.
 39. Comba, S.; Sethi, R. Stabilization of highly concentrated suspensions of iron nanoparticles using shear-thinning gels of xanthan gum. *Water Res.* 2009, 43, 3717–3726.
 40. Basnet, M.; Ghoshal, S.; Tufenkji, N. Rhamnolipid biosurfactant and soy protein act as effective stabilizers in the aggregation and transport of palladium-doped zerovalent iron nanoparticles in saturated porous media. *Environ. Sci. Technol.* 2013, 47, 13355–13364.
 41. Vinod, V.T.P.; Waclawek, S.; Senan, C.; Kupčík, J.; Pešková, K.; Černík, M.; Somashekarappa, H.M. Gum karaya (*Sterculia urens*) stabilized zero-valent iron nanoparticles: Characterization and applications for the removal of chromium and volatile organic pollutants from water. *RSC Adv.* 2017, 7, 13997–14009.
 42. Li, Y.; Zhao, H.-P.; Zhu, L. Remediation of soil contaminated with organic compounds by nanoscale zero-valent iron: A review. *Sci. Total Environ.* 2020, 760, 143413.
 43. García-Vera, V.E.; Tenza-Abril, A.J.; Solak, A.M.; Lanzón, M. Calcium hydroxide nanoparticles coatings applied on cultural heritage materials: Their influence on physical characteristics of earthen plasters. *Appl. Surf. Sci.* 2020, 504, 144195.
 44. Daniele, V.; Taglieri, G. Nanolime suspensions applied on natural lithotypes: The influence of concentration and residual water on carbonation process and on treatment effectiveness. *J. Cult. Herit.* 2010, 11, 102–106.
 45. Lanzón, M.; Madrid, J.A.; Martínez-Arredondo, A.; Mónaco, S. Use of diluted Ca(OH)₂ suspensions and their transformation into nanostructured CaCO₃ coatings: A case study in

- strengthening heritage materials (stucco, adobe and stone). *Appl. Surf. Sci.* 2017, 424, 20–27.
46. Slízková, Z.; Drdácký, M.; Viani, A. Consolidation of weak lime mortars by means of saturated solution of calcium hydroxide or barium hydroxide. *J. Cult. Herit.* 2015, 16, 452–460.
47. Díaz-Tena, E.; Rodríguez-Ezquerro, A.; Marcaide, L.L.D.L.; Bustinduy, L.G.; Sáenz, A.E. A sustainable process for material removal on pure copper by use of extremophile bacteria. *J. Clean. Prod.* 2014, 84, 752–760.
48. Invernizzi, C.; Fiocco, G.; Iwanicka, M.; Targowski, P.; Piccirillo, A.; Vagnini, M.; Licchelli, M.; Malagodi, M.; Bersani, D. Surface and interface treatments on wooden artefacts: Potentialities and limits of a non-invasive multi-technique study. *Coatings* 2020, 11, 29.
49. Jaén, J.A.; De Obaldía, J.; Rodriguez, M.V. Application of Mössbauer spectroscopy to the study of tannins inhibition of iron and steel corrosion. *Hyperfine Interact.* 2011, 202, 25–38.
50. Giuliani, C.; Pascucci, M.; Riccucci, C.; Messina, E.; de Luna, M.S.; Lavorgna, M.; Ingo, G.M.; Di Carlo, G. Chitosan-based coatings for corrosion protection of copper-based alloys: A promising more sustainable approach for cultural heritage applications. *Progr. Org. Coat.* 2018, 122, 138–146.
51. Jia, M.; Zhang, X.; Weng, J.; Zhang, J.; Zhang, M. Protective coating of paper works: ZnO/cellulose nanocrystal composites and analytical characterization. *J. Cult. Herit.* 2019, 38, 64–74.
52. Jia, Z.; Yang, C.; Zhao, F.; Chao, X.; Li, Y.; Xing, H. One-step reinforcement and deacidification of paper documents: Application of Lewis base—Chitosan nanoparticle coatings and analytical characterization. *Coatings* 2020, 10, 1226.
-

Retrieved from <https://encyclopedia.pub/entry/history/show/66522>