

Development of Thermoelectric Half-Heusler Alloys

Subjects: **Chemistry, Applied**

Contributor: Gerda Rogl , Peter Franz Rogl

Half-Heusler alloys are among the most promising thermoelectric materials.

Half Heusler alloys, physical properties, Figure of merit

1. Introduction

In this chapter, all data from references [1][2][3][4][5][6][7][8][9][10][11][12][13][14][15][16][17][18][19][20][21][22][23][24][25][26][27][28][29][30][31][32][33][34][35][36][37][38][39][40][41][42][43][44][45][46][47][48][49][50][51][52][53][54][55][56][57][58][59][60][61][62][63][64][65][66][67][68][69][70][71][72][73][74][75][76][77][78][79][80][81][82][83][84][85][86][87][88][89][90][91][92][93][94][95][96][97][98][99][100][101][102][103][104][105][106][107][108][109][110][111][112][113][114][115][116][117][118][119][120][121][122][123][124][125][126][127][128][129][130][131][132][133][134][135][136][137][138][139][140][141][142][143][144][145][146][147][148][149][150][151][152][153][154][155][156][157][158][159][160][161][162][163][164][165][166][167][168][169][170][171][172][173][174][175][176][177][178][179][180][181][182][183][184][185][186][187][188][189][190][191][192][193][194][195][196][197][198][199][200][201][202][203][204][205][206][207][208] were used for figures and discussions. The references [19][20][21][22][23][24][25][26][27][28][29][30][31][32][33][34][35][36][37][38][39][40][41][42][43][44][45][46][47][48][49][50][51][52][53][54][55][56][57][58][59][60][61][62][63][64][65][66][67][68][69][70][71][72][73][74][75][76][77][78][79][80][81][82][83][84][85][86][87][88][89][90][91][92][93][94][95][96][97][98][99][100][101][102][103][104][105][106][107][108][109][110][111][112][113][114][115][116][117][118][119][120][121][122][123][124][125][126][127][128][129][130][131][132][133][134][135][136][137][138][139][140][141][142][143][144][145][146][147][148][149][150][151][152][153][154][155][156][157][158][159][160][161][162][163][164][165][166][167][168][169][170][171][172][173][174][175][176][177][178][179][180][181][182][183][184][185][186][187][188][189][190][191][192][193][194][195][196][197][198][199][200][201][202][203][204][205][206][207][208] are numbered by the publication year, within one year, in alphabetical order.

Looking at the highest ZT s published during the last 25 years, one can see in **Figure 1** that, as already mentioned in the introduction, considerably more n-type half-Heusler alloys were investigated than p-type. It is also clearly visible that the trend of high peak ZT s increases almost linearly from 2004 to 2018. There are exceptions for the n-type, such as (i) the work of Sakurada et al. [33], who published already in 2005 outstanding high values ($ZT = 1.25$ to 1.5) for $\text{Ti}_{0.5}\text{Zr}_{0.25}\text{Hf}_{0.25}\text{NiSn}$ doped with small amounts of antimony, or (ii) from Shen et al. [8][23] and Kawasaki et al. [27], who reached almost $ZT = 1$ at 800 K in 2001 and 2004 for $\text{ZrNiSn}_{0.99}\text{Sb}_{0.01}$. It is also worth mentioning that Yu et al. [140] reached $ZT = 1.6$ at 1200 K for his p-type HH series $(\text{Nb}_{1-x}\text{Ta}_x)_{0.8}\text{Ti}_{0.2}\text{FeSb}$. Furthermore, it seems that after a boom during 2015–2020, the number of half-Heusler publications has dropped significantly.

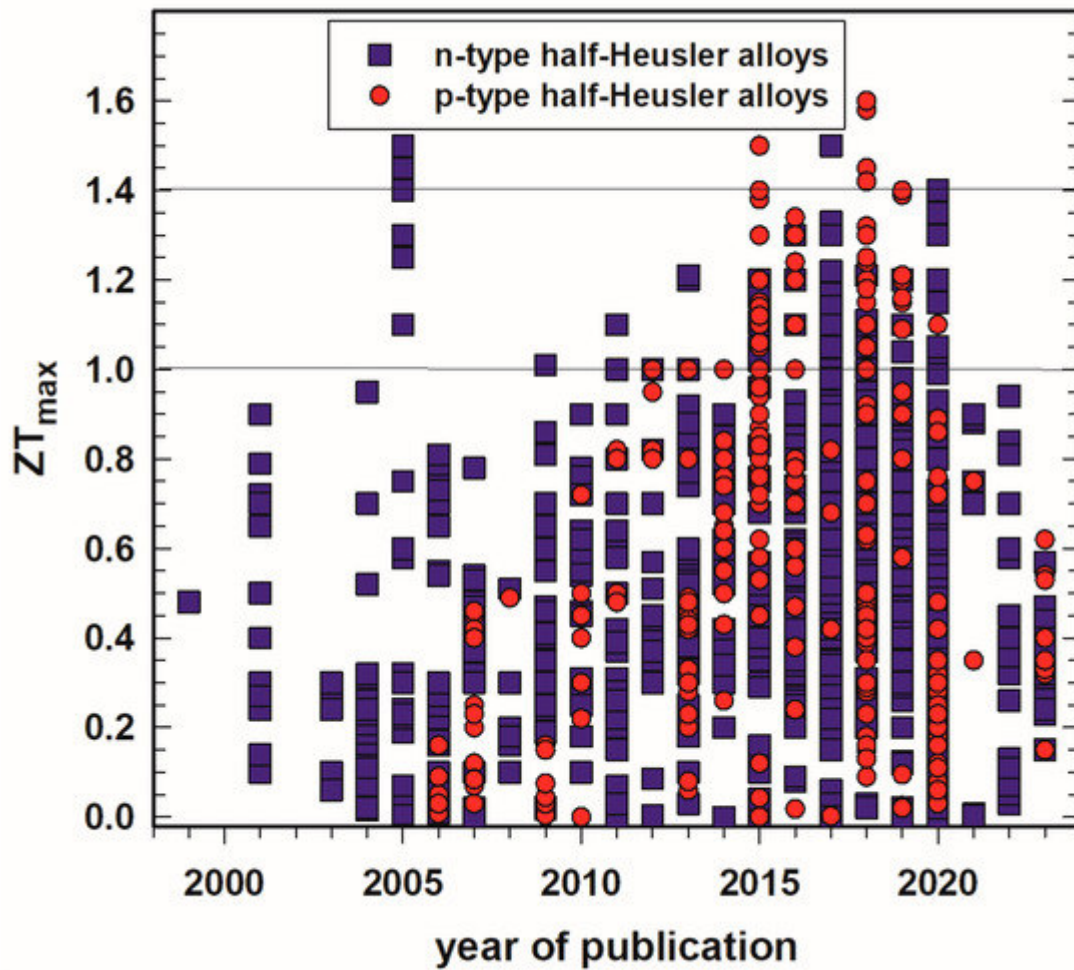


Figure 1. Peak ZT , $ZT_{\max}$, of p- and n-type half-Heusler alloys vs. year of publication.

Figure 2 displays the highest ZT s as a function of temperature. It is necessary to point out that in some cases $ZT_{\max}$ is not really the peak ZT of the respective sample but the ZT at the highest measured temperature. This is especially the case for publications earlier than 2005. Besides that, in most cases $ZT_{\max}$ is published for temperatures between 700 K and 1000 K; the highest temperatures for $ZT_{\max}$ for p- as well as n-type are around 1200 K.

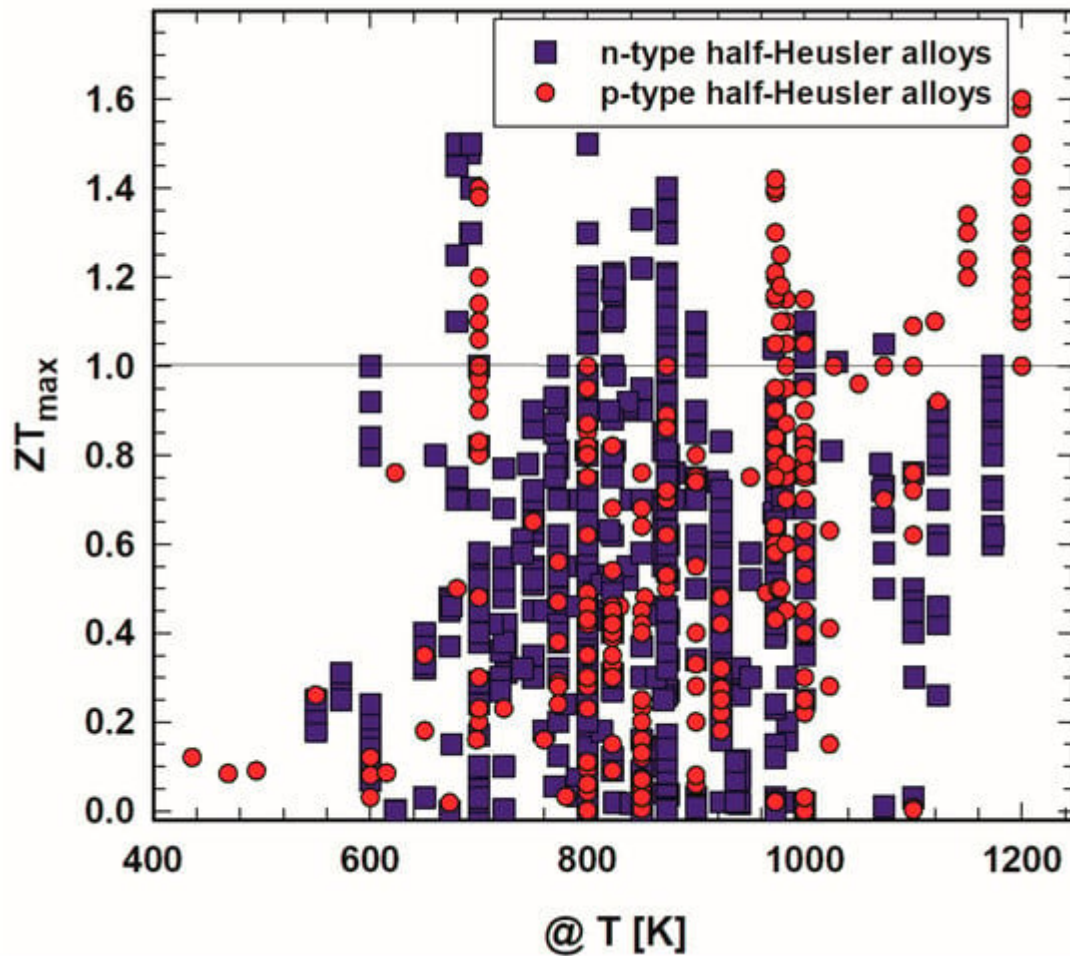


Figure 2. Peak ZT , $ZT_{\max}$, of p- and n-type half-Heusler alloys vs. temperature, T .

Figure 3 and **Figure 4** depict the dependence of the power factor on the thermal conductivity at 300 K and 800 K, respectively. **Figure 3** shows that at 300 K, almost all power factor values for the n-type are within the range of 4 mW/mK² for thermal conductivities up to 250 mW/cmK, whereas the p-type displays much higher power factors, reaching values higher than 10 mW/mK² for thermal conductivities up to about 150 mW/cmK, e.g., for the Nb_{1-x}Ti_xFeSb-series of He et al. [121] or for Nb_{0.95}Hf_{0.05}FeSb and Nb_{0.95}Zr_{0.05}FeSb of Ren and colleagues [150].

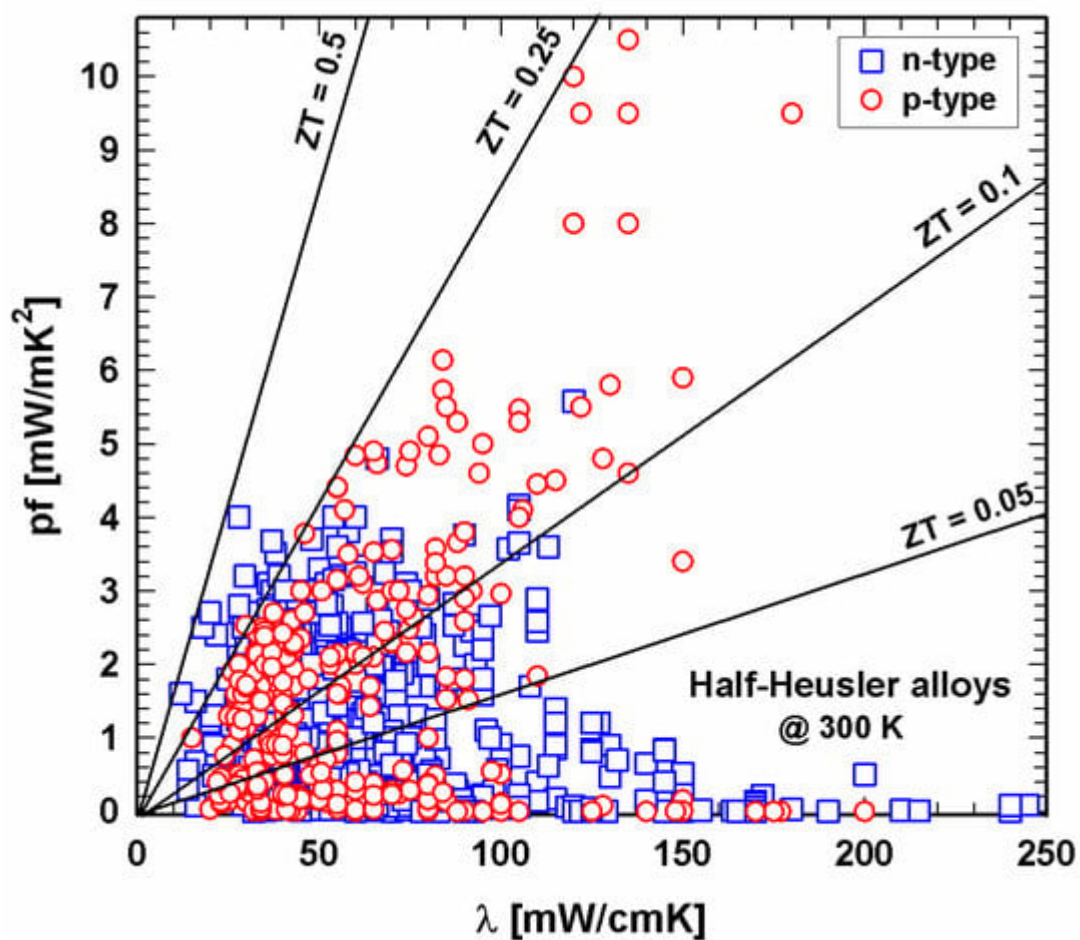


Figure 3. P- and n-type half-Heusler alloys: power factor, pf , vs. thermal conductivity, λ , at 300 K.

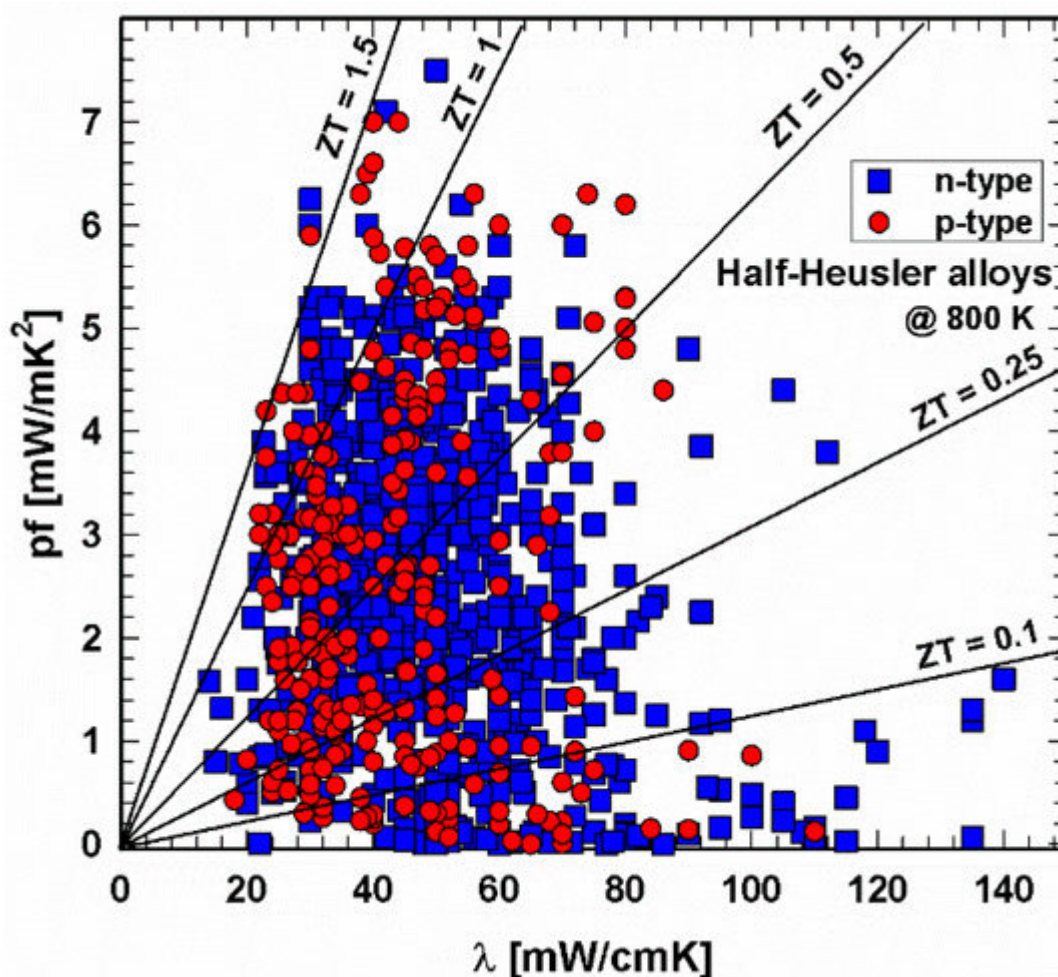


Figure 4. P- and n-type half-Heusler alloys: power factor, pf , vs. thermal conductivity, λ , at 800 K.

The situation has changed comparing this figure now with the one for 800 K (**Figure 4**), and as for p- and n-type, all data are mainly evenly distributed within a power factor–thermal conductivity window of almost 0–5.5 mW/mK² by 20–80 mW/cmK.

Lines for the resulting ZT values are drawn in both **Figure 3** and **Figure 4**, revealing that at 300 K, no compound exceeds $ZT = 0.5$. Whilst many of the p-type compounds range between $ZT = 0.1$ and 0.25 or have values as low as almost 0, one can find the values for the n-type all over, the very low values due to high thermal conductivities. At 800 K, both p- and n-type exceed the $ZT = 1.5$ line. The majority of data can be found for thermal conductivities between 20 and 80 mW/cmK, representing ZT s from almost 0 to $ZT = 1.5$. This figure also illustrates that p-type HH materials generally exhibit lower thermal conductivities than the n-type.

Figure 5 and **Figure 6** depict the relation between electrical resistivity and ZT at 300 K and 800 K, respectively. It shows that, especially in the 300 K temperature region, individual specimens have very high resistivities, resulting, as a consequence, in very low ZT s. Generally, the electrical resistivity decreases with increasing temperature. **Figure 6** displays lower resistivity values. Still, here one can find ZT s in the range of 10^{-3} or even lower. The insert,

a cutout, in each ZT -resistivity figure, gives a better overview, indicating that, especially for 800 K, most data can be found between 500 and 2500 $\mu\Omega\text{cm}$. In this region are also the highest ZT s.

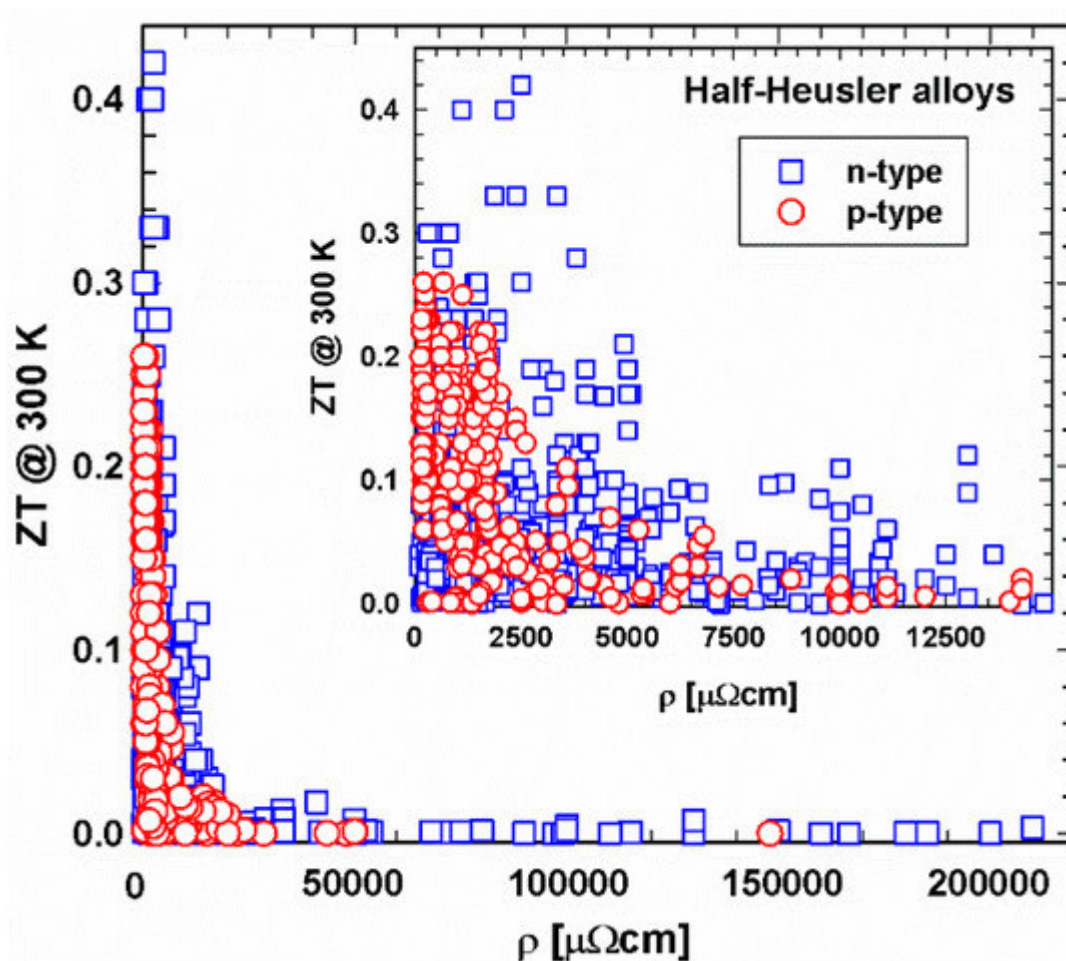


Figure 5. ZT at 300 K vs. electrical resistivity, ρ , for p- and n-type half-Heusler alloys. Insert: cutout for $\rho < 15000 \mu\Omega\text{cm}$.

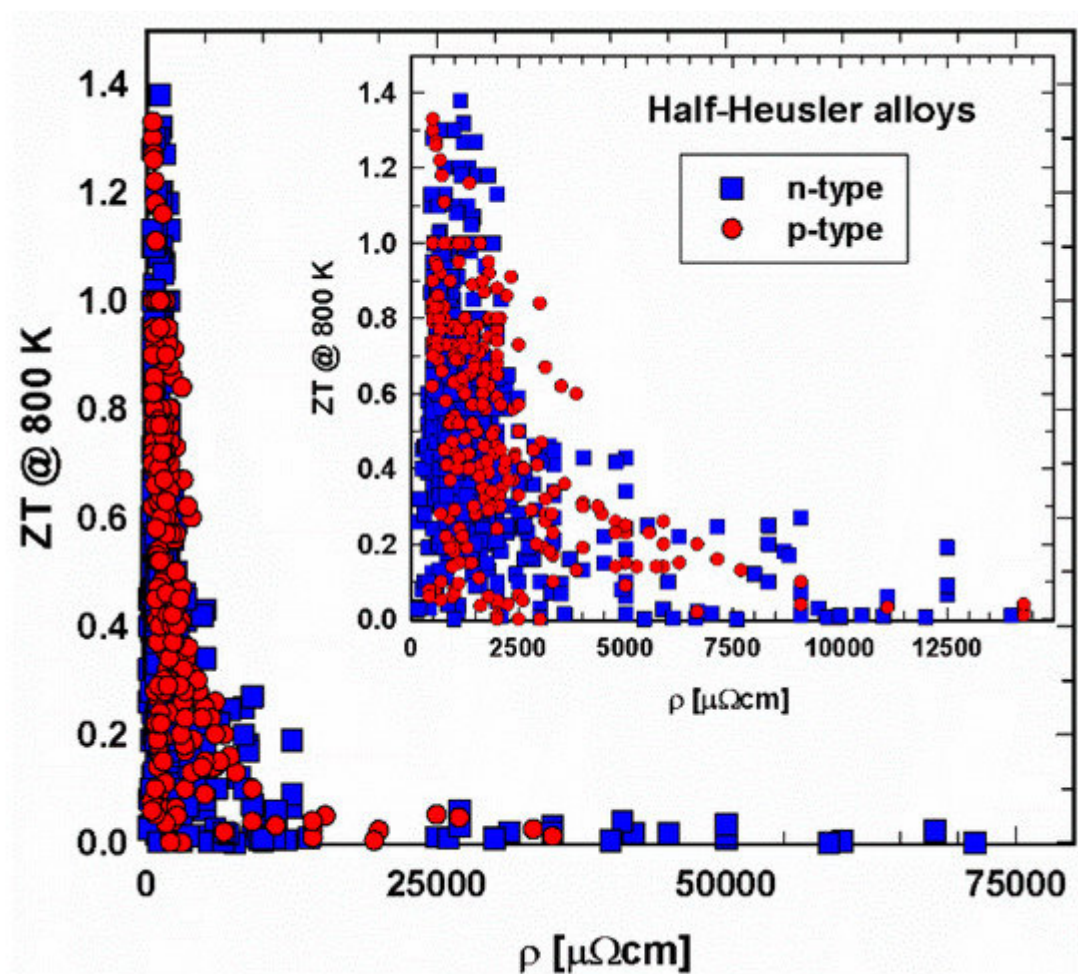


Figure 6. ZT at 800 K vs. electrical resistivity, ρ , for p- and n-type half-Heusler alloys. Insert: cutout for $\rho < 12000 \mu\Omega\text{cm}$.

Figure 7 and **Figure 8** present the dependence of ZT on the Seebeck coefficient. Especially at 300 K, the area of Seebeck values is much wider for the n- than for the p-type. Generally, it seems that an absolute Seebeck value of $200 \pm 20 \mu\text{V/cm}$ leads to high ZT s, a great finding for the production of new half-Heuslers.

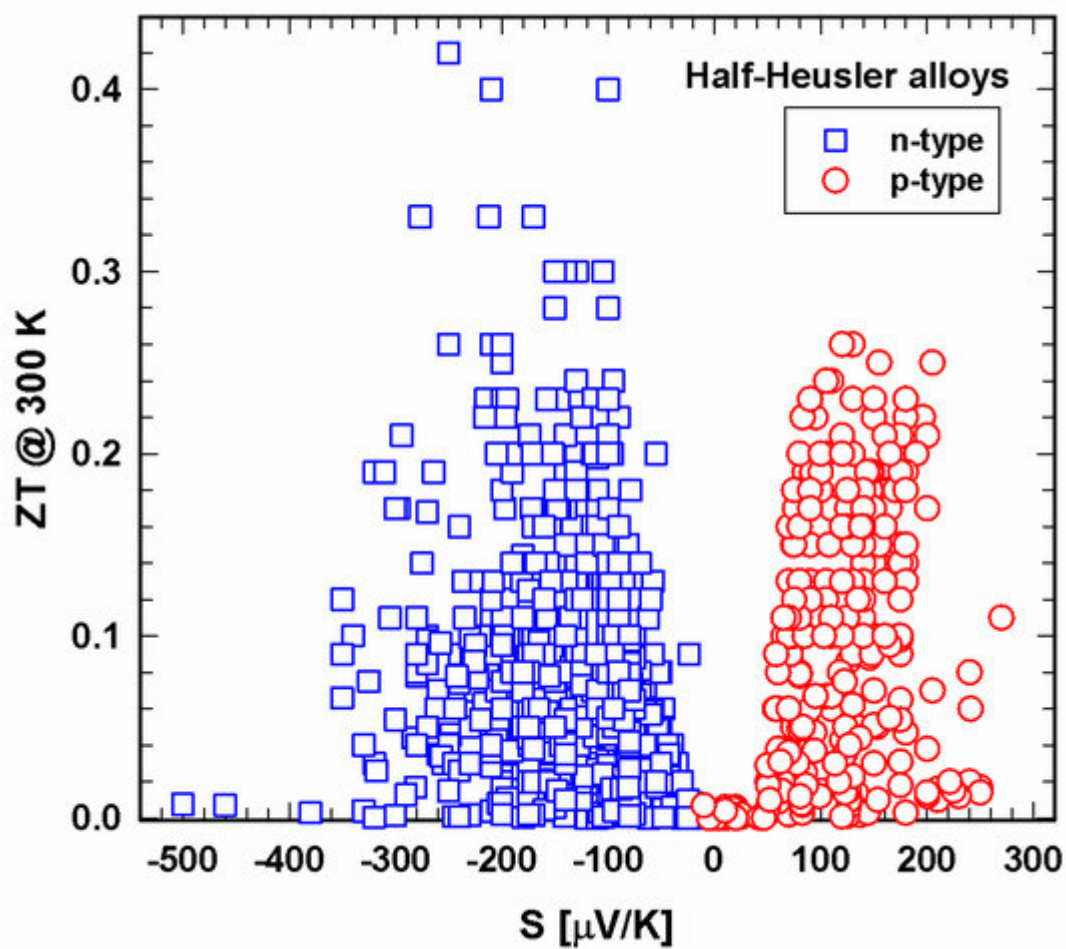


Figure 7. ZT at 300 K vs. Seebeck coefficient, S , for p- and n-type half-Heusler alloys.

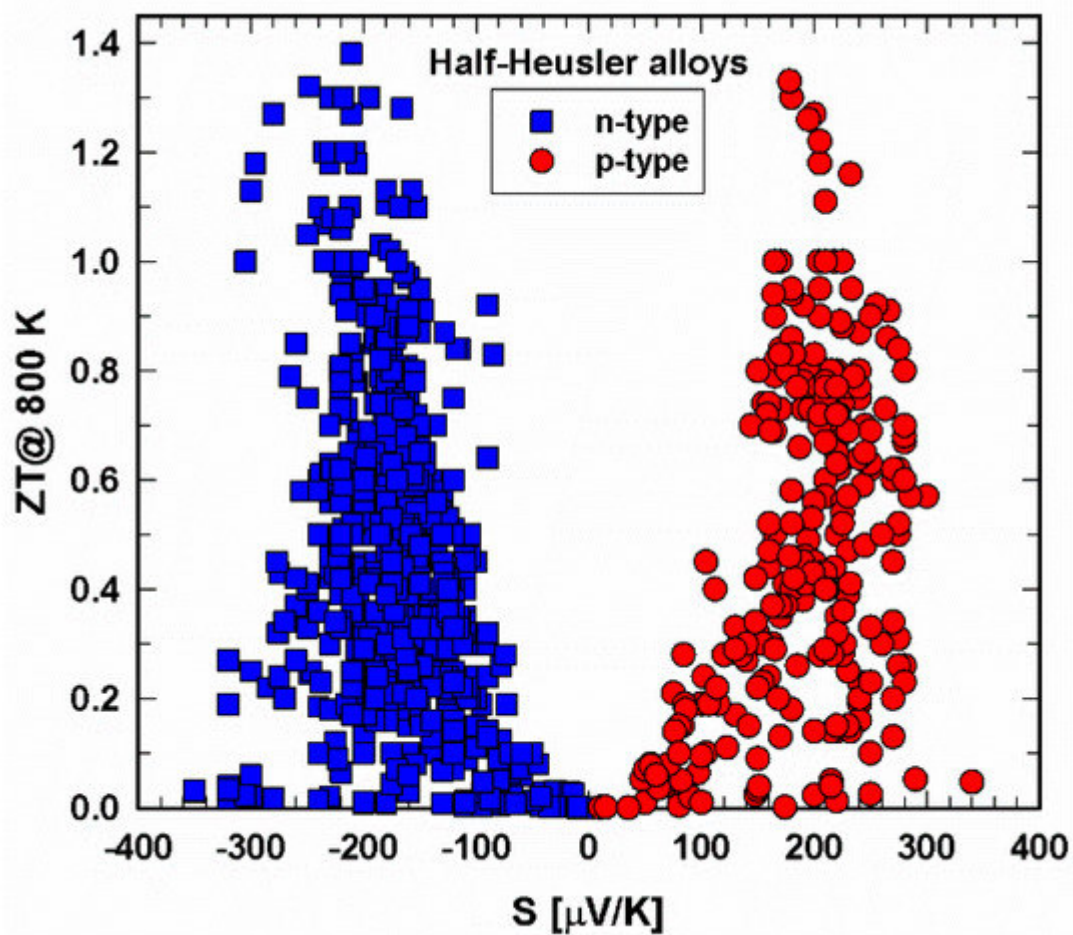


Figure 8. ZT at 800 K vs. Seebeck coefficient, S , for p- and n-type half-Heusler alloys.

Whereas for the high-temperature regime (**Figure 9**), the data ZT vs. λ are more or less evenly mixed at 300 K (insert in **Figure 9**), most p-type half-Heuslers have thermal conductivities between 20 and 70 mW/cmK. Of course, in both regions, ZT s are high for low thermal conductivities. Comparing the total and the lattice thermal conductivities in **Figure 9** and **Figure 10**, the distribution of the data looks very similar, indicating that the electron part of the thermal conductivity is rather low, and, as a consequence, the phonon part is in the range of the total thermal conductivity.

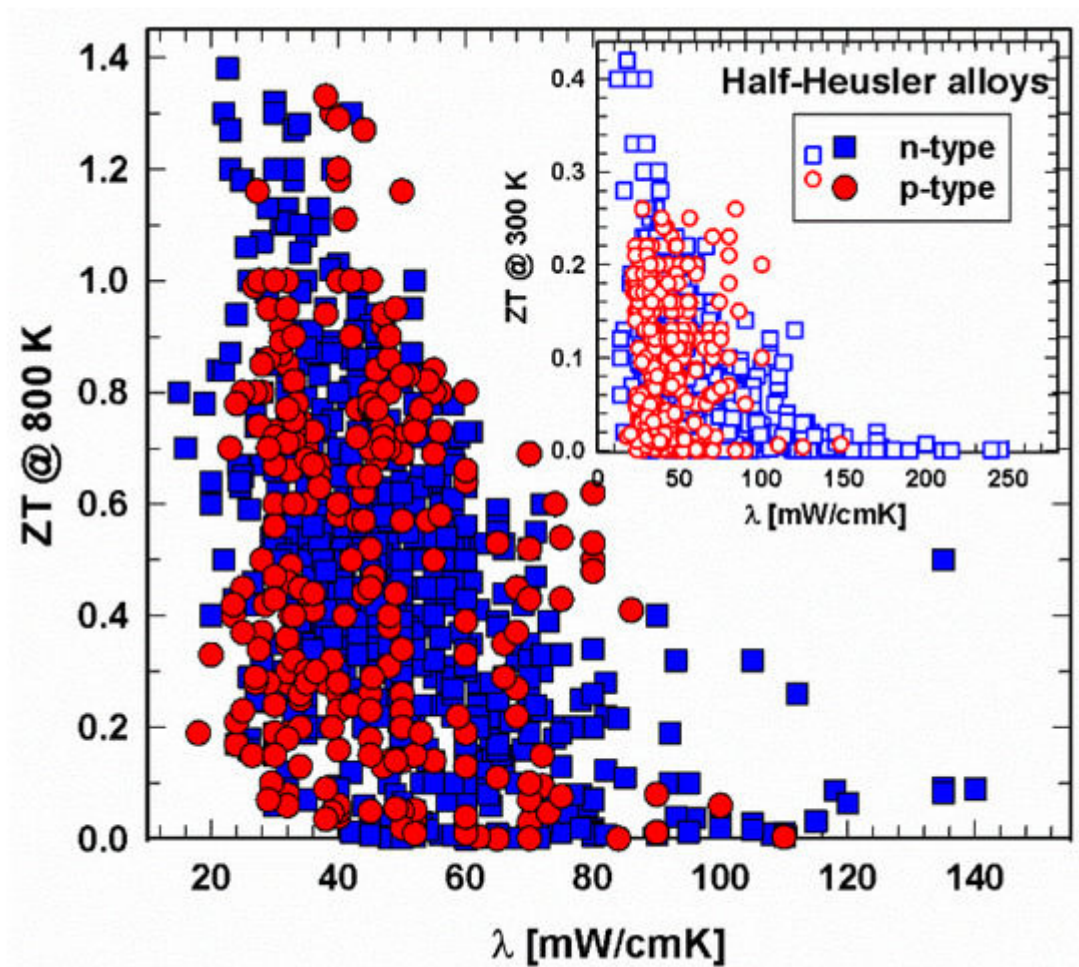


Figure 9. ZT at 800 K vs. thermal conductivity, λ , for p- and n-type half-Heusler alloys. Insert: ZT at 300 K vs. thermal conductivity, λ .

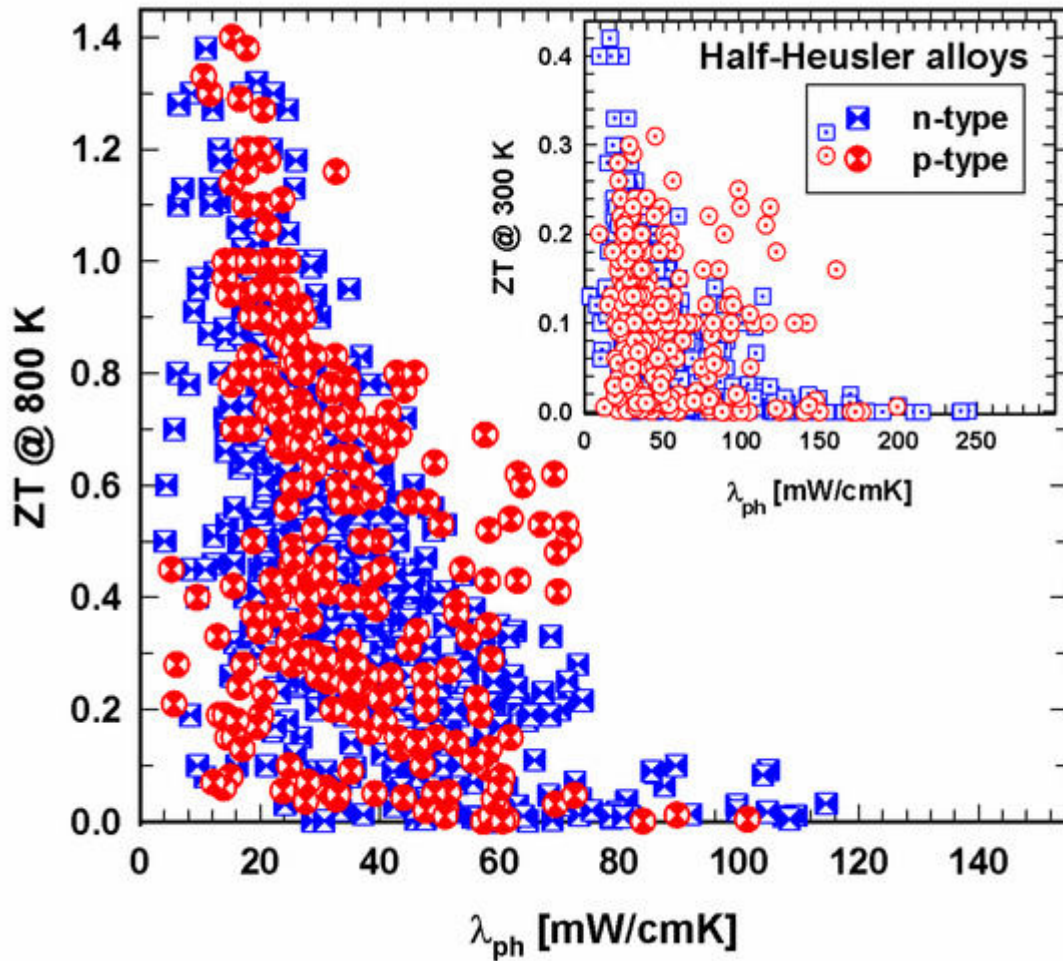


Figure 10. ZT at 800 K vs. lattice thermal conductivity, λ_{ph} , for p- and n-type half-Heusler alloys. Insert: ZT at 300 K vs. lattice thermal conductivity, λ_{ph} .

2. N-Type Half-Heusler Alloys

2.1. (Ti,Zr,Hf)NiSn-System

Comparing the ZT values of TiNiSn, ZrNiSn and HfNiSn, as shown in **Figure 11** and **Figure 12**, one can immediately see that ZrNiSn is the most investigated alloy of these three, whereas HfNiSn is not so popular. It is evident from **Figure 11** and **Figure 12** that the highest ZT s at 300 K as well as at 800 K in this system are for (Zr,Hf)NiSn and (Ti,Zr,Hf)NiSn and even more so when Sb-doped at the Sn site.

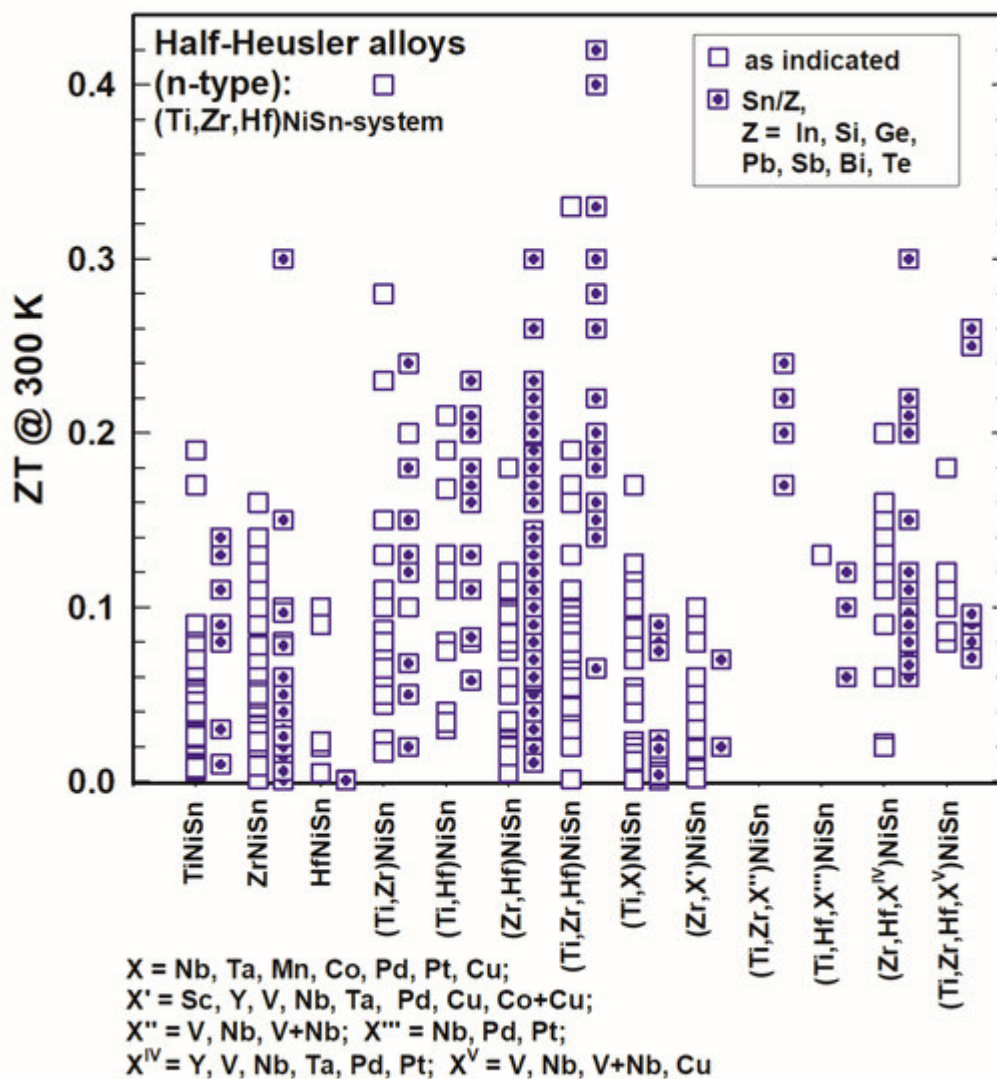


Figure 11. ZT at 300 K for the HH n-type (Ti,Zr,Hf)NiSn-system.

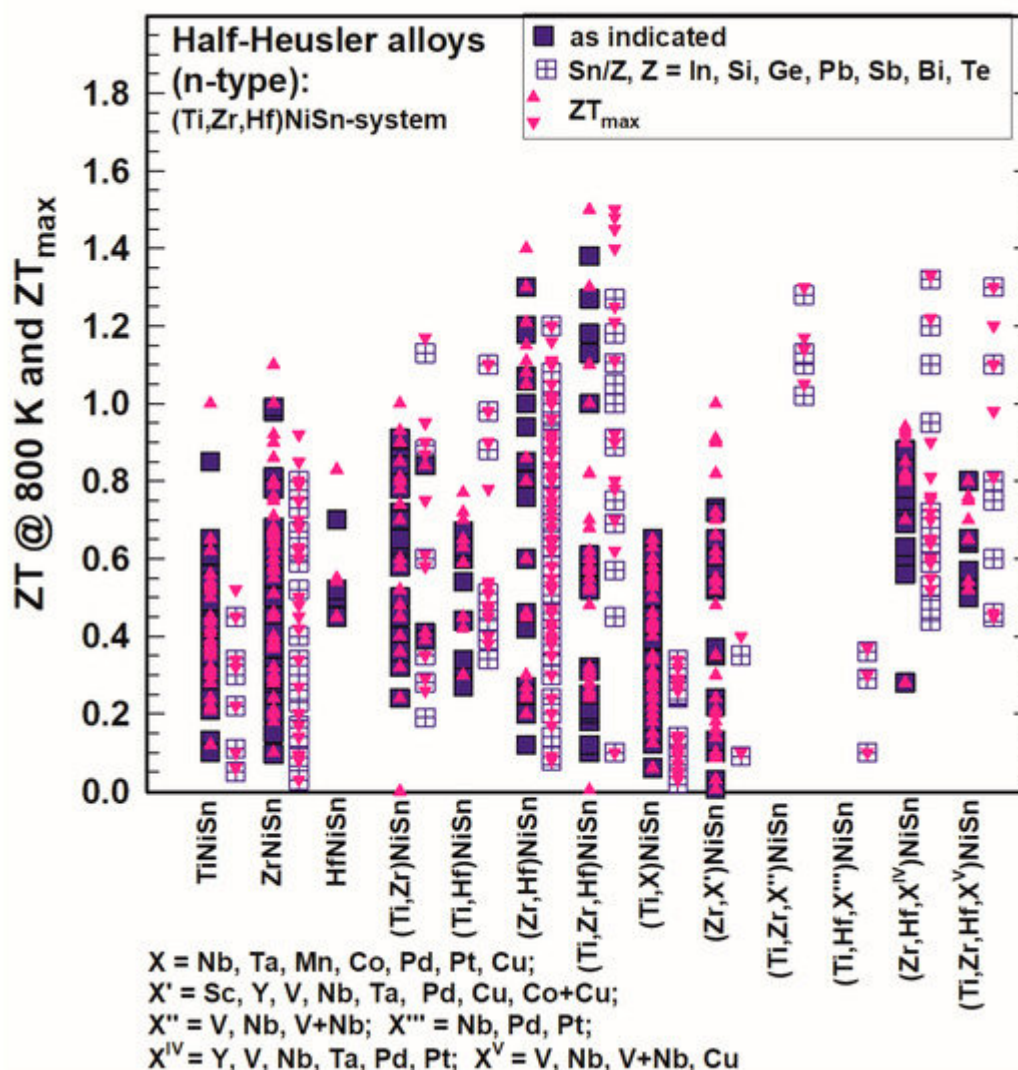


Figure 12. ZT at 800 K and ZT_{max} for the HH n-type (Ti,Zr,Hf)NiSn-system.

Gürth et al. [119] published $ZT = 1$ at 823 K for TiNiSn and for ZrNiSn, Misra et al. [96] $ZT = 0.9$ at 773 K for ZrNiSn and Liu et al. [109] $ZT = 0.8$ at 1000 K for HfNiSn. In comparison to these values, those for compounds with Ti and Zr, Ti and Hf, Zr and Hf or with Ti and Zr and Hf reach much higher values, such as for $Zr_{0.7}Hf_{0.3}NiSn$ $ZT = 1.2$ at 873 K (Chauhan et al. [144]), for $Ti_{0.5}Zr_{0.25}Hf_{0.25}NiSn$ $ZT = 1.3$ at 693 K (Sakurada and Shutoh [33]; Shutoh and Sakurada [37]) or even $ZT = 1.5$ at 800 K (Rogl et al. [136]).

Doping in the (Ti,Zr,Hf)NiSn-system with X (X = Sc, Y, V, Nb, V + Nb, Ta, Mn, Co, Cu, Co + Cu, Pd, Pt, Pb) was not really successful as generally, all these alloys did not have higher ZTs; however, from Figure 11 as well as from Figure 12 it is obvious that with an additional substitution of Sn in many cases ZT is enhanced. Usually, Sn is substituted by a very small amount of Sb or Ge, but also, in rarer cases, In, Si, Pb, Bi or Te was used. The highest ZT values were reached for $Ti_{0.5}Zr_{0.25}Hf_{0.25}NiSn_{0.998}Sb_{0.002}$ at 700 K ($ZT = 1.5$, Shutoh and Sakurada [37]) and $Ti_{0.5}Zr_{0.25}Hf_{0.25}NiSn_{0.99}Sb_{0.01}$ at 825 K ($ZT = 1.45$, Rogl et al. [136]). Besides that, there are several compounds with $ZT \sim 1.3$, such as $(Hf_{0.6}Zr_{0.4})_{0.99}V_{0.01}NiSn_{0.995}Sb_{0.005}$ at 900 K (Chen et al. [130]) or $Ti_{0.49}Zr_{0.49}V_{0.02}NiSn_{0.98}Sb_{0.02}$ at 823 K (Rogl et al. [136]) (see also Figure 12).

Various groups used additions, mainly nanoparticles, to reduce the thermal conductivity and to enhance ZT . Schwall and Balke [70] incorporated Nb into $\text{Zr}_{0.5}\text{Hf}_{0.5}\text{NiSn}$ but presented data only below room temperature. Akram et al. [98] admixed La with ZrNiSn : the enhancement in ZT was within the error bar of ZT (from 0.52 to 0.54), which also applies to Visconti et al. [139], who used HfO_2 for $\text{Ti}_{0.5}\text{Zr}_{0.5}\text{NiSn}_{0.994}\text{Sb}_{0.006}$ (ZT rises from 0.9 to 1). A better result was achieved by Chauhan et al. [143] by mixing ZnO with ZrNiSn (highest $ZT = 1$). Nanoparticles of ZrO_2 were added to ZrNiSn (Huang et al. [25]) as well as to $\text{Zr}_{0.5}\text{Hf}_{0.5}\text{Ni}_{0.8}\text{Pd}_{0.2}\text{Sn}_{0.99}\text{Sb}_{0.01}$ and $\text{Zr}_{0.25}\text{Hf}_{0.6}\text{Ti}_{0.15}\text{NiSn}_{0.995}\text{Sb}_{0.005}$ (Chen et al. [39] and [117], respectively). For ZrNiSn , the “enhancement” of ZT was within the error bar; however, it was much better for the other two compounds with $ZT_{\text{max}} = 0.74$ and even $ZT_{\text{max}} = 1.3$ for the Ti-doped half-Heusler alloy.

2.2. (Ti,Zr,Hf)CoSb-System, (V,Nb,Ta)FeSb-System, 19-Electron System and Other N-Type Half-Heusler Alloys

In Figure 13 and Figure 14, the ZT s are shown for the (Ti,Zr,Hf)CoSb-system, the (V,Nb,Ta)FeSb-system, the 19-electron system and of all those half-Heuslers, which do not fit into any of the so-far mentioned categories. For all these systems, only a few half-Heuslers with Sb/Sn substitutions were published; generally, ZT s are much lower in comparison with the (Ti,Zr,Hf)NiSn-system for 300 K as well as for 800 K.

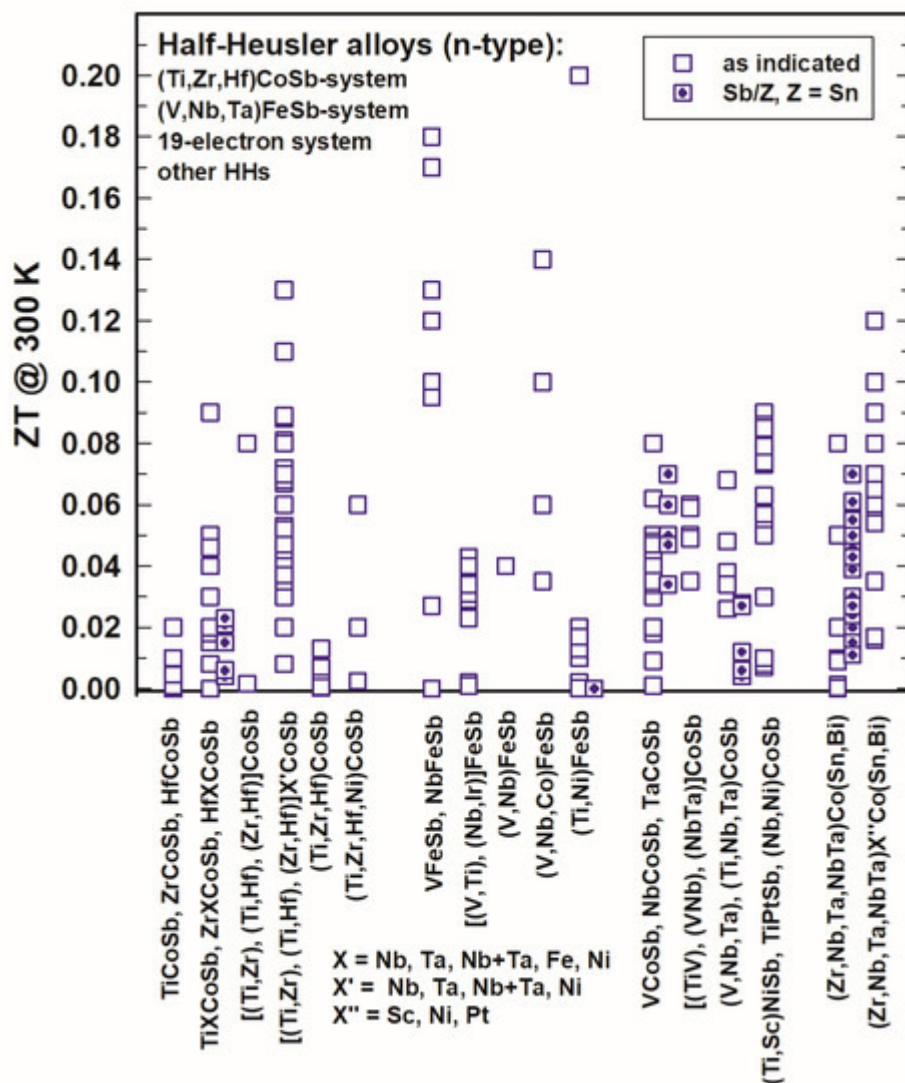


Figure 13. ZT at 300 K for the HH n-type (Ti,Zr,Hf)CoSb-system, (V,NbTa)FeSb-system, 19-electron system and other half-Heuslers.

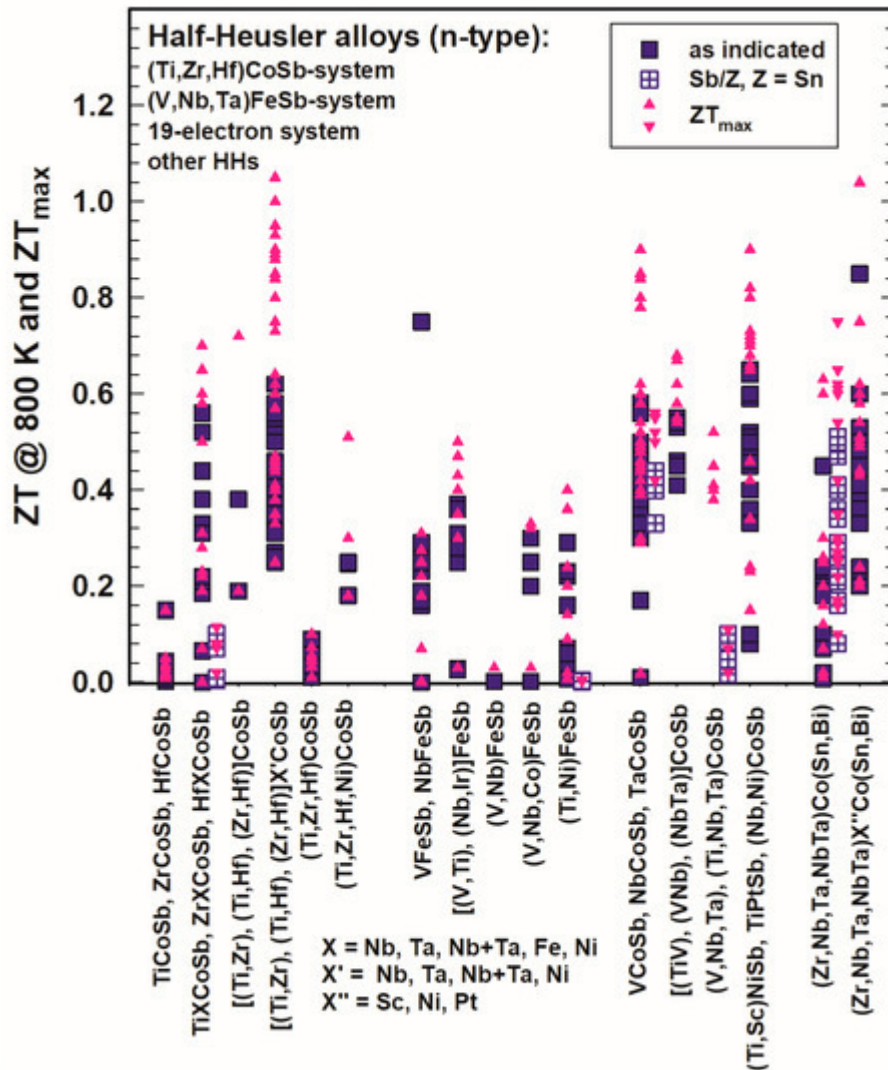
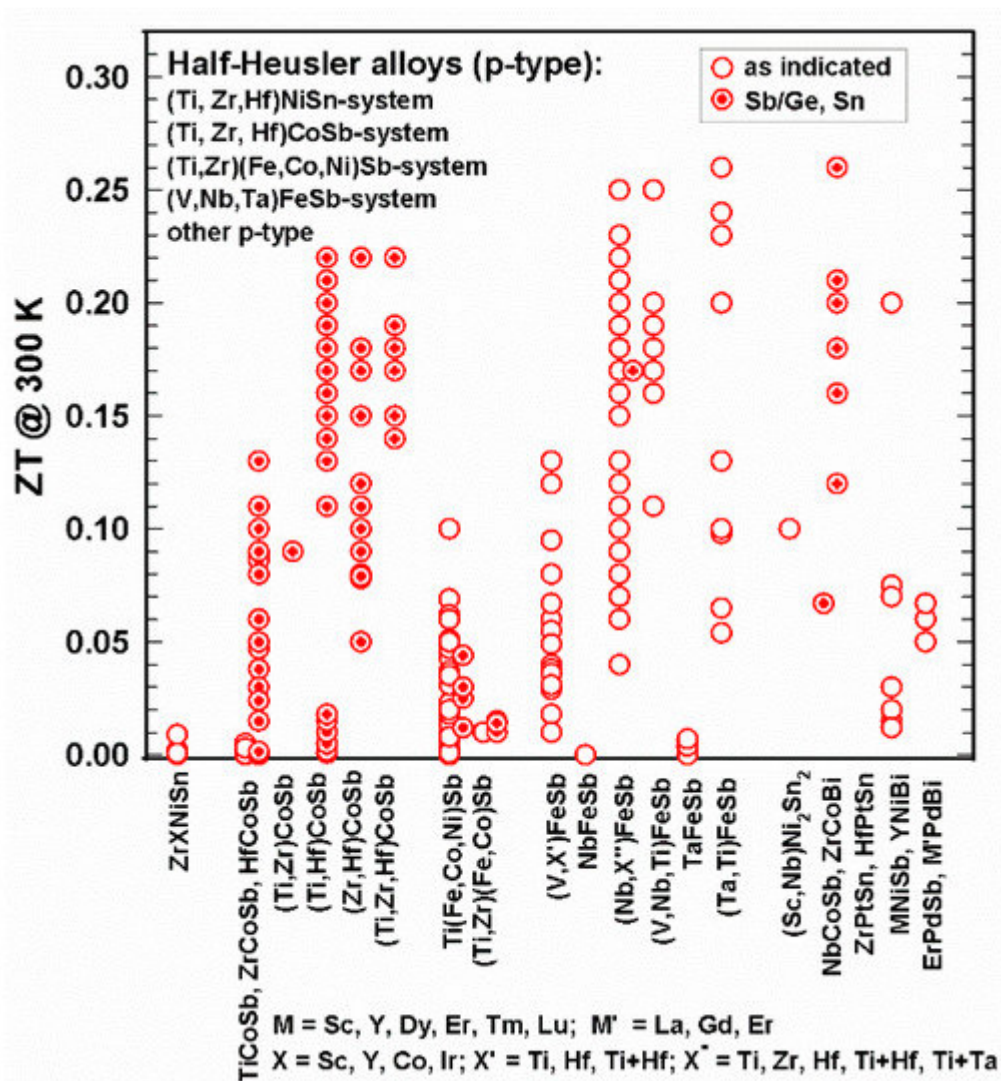


Figure 14. ZT at 800 K for the HH n-type (Ti,Zr,Hf)CoSb-system, (V,NbTa)FeSb-system, 19-electron system and other half-Heuslers.

At 800 K (**Figure 14**), almost all ZT s are lower than 0.7 and also ZT_{max} rarely exceeds 1. He et al. [131] reached $ZT > 1$ for $Zr_{0.5}Hf_{0.5}Co_{0.9}Ni_{0.1}Sb$ at 1073 K, Liu et al. [148] acquired ZT s of almost 1 and around 1 for the series $(Zr_{1-x}Hf_x)_{0.88}Nb_{0.12}CoSb$. Xia et al. [106][155] investigated $Nb_{1-x}CoSb$ and $Nb_{0.8}Co_{1-x}Ni_xSb$ with $ZT = 0.8$ – 0.9 . Zhu et al. [177] obtained $ZT = 0.85$ at 800 K with a peak $ZT = 1.04$ at 972 K for $ZrCo_{0.9}Ni_{0.1}Bi_{0.85}Sb_{0.15}$.

3. P-Type Half-Heusler Alloys

In some cases, measurements were only performed below room temperature. **Figure 15** and **Figure 16** represent the ZT s for p-type half-Heusler alloys at 300 K and 800 K as well as the peak ZT s, respectively.

Figure 15. ZT at 300 K for the p-type HH alloys.

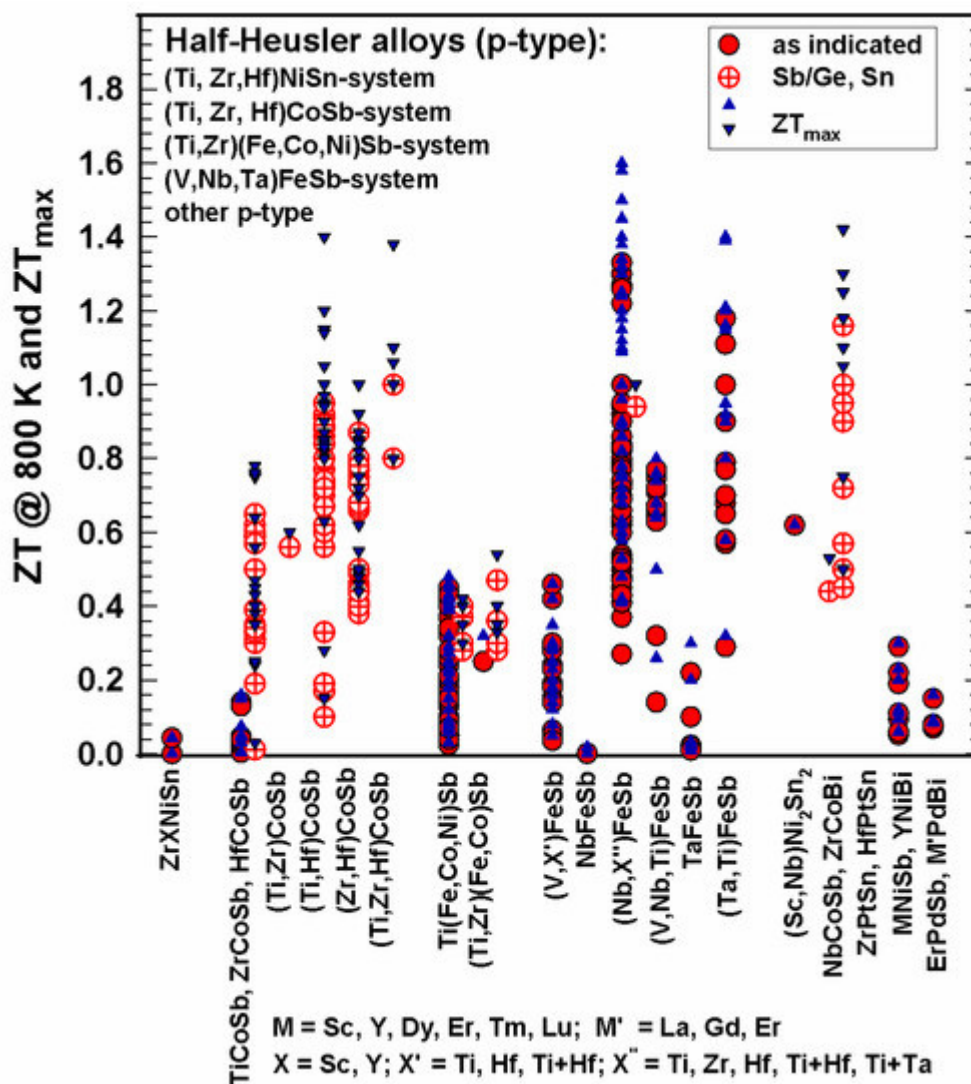


Figure 16. ZT at 800 K and ZT_{max} for the p-type HH alloys.

It is obvious that many more n-type than p-type alloys were investigated. As depicted in **Figure 15**, at 300 K, all ZTs are below 0.3; at 800 K (**Figure 16**), most ZTs are lower than 1; however, several half-Heuslers in the (NbX'')FeSb series with X'' = Ti, Zr, Hf, Zr and Hf, as well as Ti + Ta, reach ZT = 1.4 at 800 K; the values for ZT_{max} in some cases reach or even exceed 1.4 and go up to 1.6.

It can be seen in **Figure 15** and **Figure 16** that only data of two p-type compounds are available for the (Ti,Zr,Hf)NiSn-system, namely (Zr,Sc)NiSn [112] and (Zr,Y)NiSn [104] with rather low ZTs.

In the CoSb-based system, almost all compounds are substituted at the Sb-site by Sn, except for TiCoSb, which was also substituted by Ge [58]. Rausch et al. published in two separate works [110][111] rather high ZTs of the (Ti,Zr,Hf)CoSb_{1-x}Sn_x series: the highest ZT_{max} = 1.4 for Ti_{0.12}Zr_{0.44}Hf_{0.44}CoSb_{0.8}Sn_{0.2} at 963 K [111].

In the FeSb-based system, Yu et al. [140] achieved outstanding high peak ZT values (ZT = 1.2–1.6) for the series Ti_{0.2}(Nb_{1-x}Ta_x)_{0.8}FeSb; Zhu et al. gained ZT~1.4 at 1200 K for Hf_{0.12}Nb_{0.88}FeSb [158] and for Ti_{0.16}Ta_{0.84}FeSb [178].

Whereas many p-type half-Heuslers have $ZT < 1$, the ZrCoBi-based half-Heuslers showed excellent thermoelectric performance, even more so for substituting at the Bi-site with $ZT = 1.42$ for $\text{ZrCoBi}_{0.65}\text{Sb}_{0.15}\text{Sn}_{0.20}$ at 973 K as reported by Zhu et al. [159].

Also, for p-type half-Heusler alloys, some research groups tried to enhance ZT via additions. Kimura et al. [61] added Co and Ir to ZrNiSn and converted this way the n-type to a p-type half-Heusler alloy; however, ZT s were rather low, i.e., below 0.3. Mallick et al. [149] investigated $(\text{Hf}_{0.7}\text{Ti}_{0.3}\text{CoSb}_{0.8}\text{Sn}_{0.2})_{1-x}(\text{Cu}_{1.98}\text{Ni}_{0.04}\text{Te}_{0.97}\text{Se}_{0.03})_x$; for $x = 0.25$ they could enhance $ZT = 0.15$ (without the chalcogenide) to $ZT = 0.63$ at 1023 K, which corresponds an enhancement of 320%. Hsu et al. [93] were successful in producing nanostructured HfO_2 during the ball-milling and spark-plasma-sintering process of $\text{Zr}_{0.5}\text{Hf}_{0.5}\text{CoSb}_{0.8}\text{Sn}_{0.2}$ and reached as highest $ZT = 0.75$. Xie et al. [183] enhanced ZT of $\text{Ti}(\text{Co,Fe})\text{Sb}$ with the addition of 1 atom percent nano inclusions of InSb by 450%; however, $ZT = 0.33$ is still very low.

4. Miscellaneous

On 8-electron half-Heusler alloys: Several authors have screened 8-electron half-Heusler alloys in DFT throughput calculations, identifying a series of compounds of interest in thermoelectrics. Vikram et al., in one publication [209], presented the TE properties of 21 half-Heusler alloys with a sum of 8 valence electrons (8-electron half-Heuslers); 6 of them showed ZT s between 0.8 and 1, such as LiSnAl , NaAlSn or LiAlGe . In another paper [210], he calculated the thermoelectric properties of a large number of ternary compounds of the type: group IA or IB element—group IIA or IIB element—group V element. From these results, he highlighted even ZT s such as $ZT = 1.37$ for RbPBa and $ZT = 1.56$ for AgPMg . Hoat et al. [211] also made theoretical investigations of the 8-electron half-Heusler alloy RbYSn , but they published only the power factors for the p- and n-types. So far, however, no experimental confirmation exists for the ZT s estimated from these DFT calculations.

On “double half-Heusler alloys”, etc.: Some authors have coined the term “double half-Heusler alloys”, “triple half-Heusler alloys”, etc. These terms are not without problems. Firstly, double a half is full, but the authors certainly did not want to discuss full-Heusler alloys. Secondly, a formula such as $\text{Ti}_2\text{FeCoSnSb}$ from a structural chemical point of view indicates a fully ordered compound, where Fe,Co and Sn,Sb atoms fully occupy separate two-fold sites in a crystallographic subgroup of the space group of the half-Heusler phase. As such a subgroup cannot be found among the HH face-centered cubic space group type, the structure, therefore, needs to be transferred into a suitable body-centered tetragonal space group (cubic setting a,a,a needs to be transferred to a tetragonal cell: $a/2--\sqrt{2}a/2--\sqrt{2}a$) in order to find sites of two-fold multiplicity. Only if authors can prove the full atom order will the formula $\text{Ti}_2\text{FeCoSnSb}$ be proper. Similar arguments hold for “triple half-Heusler alloys”, etc.

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