3-Phenylcoumarins as a Privileged Scaffold in Medicinal Chemistry

Subjects: Others | Chemistry, Medicinal Contributor: Maria João Matos

3-Phenylcoumarins are a family of heterocyclic molecules that are widely used in both organic and medicinal chemistry. 3-Phenylcoumarins have been used by several research groups in the search for new chemical entities with potential in the discovery of new therapeutic solutions for several diseases. The versatility and chemical properties of this scaffold have been attracting the attention of researchers all over the world.

3-phenylcoumarins

1. Introduction

The versatility and chemical properties of 3-Phenylcoumarins have been attracting the attention of researchers all over the world. Different molecular spots that may be modified, with different reactivities, allow for a huge number of different derivatives with different properties. This scaffold can be considered an isostere of the isoflavone in which the carbonyl group is translated from position 4 to position 2 on the pyran ring (**Figure 1**). Isoflavones are produced almost exclusively by the members of the bean family Fabaceae (Leguminosae). It can also be considered a coumarin-resveratrol hybrid. Resveratrol (3,5,4-trihydroxy-*trans*-stilbene) is a stilbenoid, a type of natural phenol, and a phytoalexin produced by several plants (**Figure 1**). The stilbenoids share most of their biosynthetic pathway with chalcones.



Figure 1. Coumarin, *trans*-resveratrol, 3-phenylcoumarin and isoflavone chemical structures.

Coumarins, the basic structure of 3-phenylcoumarins, are a group of substances of natural or synthetic origin that are highly studied and have a great variety of pharmacological interests. In addition, a connection of 3-phenylcoumarins can also be established (although it is more from a structural or steric point of view) with steroid hormones, especially with estrogens, due to the aromatization of the A ring. For these reasons, 3-phenylcoumarins (**Figure 1**) are considered a privileged scaffold in medicinal chemistry.

2. Presence of 3-Phenylcoumarins in Nature

The naturally-occurring 3-phenylcoumarins that have been published in the past decade are listed in **Table 1**. Mucodianin A was isolated from the vine stems of *Mucuna birdwoodiana* ^[1]. The 3-(4-ethynylphenyl)-4-formylcoumarin has been isolated from a methanol extract of the red ants of ChangBai Mountain, *Tetramorium* sp. ^[2]. Pterosonin F was isolated from the heartwood of *Pterocarpus soyauxii* ^[3]. Sphenostylisin A was isolated from the root bark of *Sphenostylis marginata* using a bioactivity-guided isolation approach. It is worth highlighting that this compound is a potent NF- κ B (nuclear factor kappa-light-chain-enhancer of activated B cells) inhibitor, displaying different physiological functions. This compound is also overexpressed in some cancer cells ^[4]. 2',4'-Dinitro-3-phenylcoumarin was isolated from *Rhizophora mucronata* ^[5]. Selaginolide A was found in *Selaginella rolandi-principis* ^[6]. Glycycoumarin and licorylcoumarin (**Figure 2**) are two 3-phenylcoumarins previously isolated from *Glycyrrhiza uralensis* or *glabra* (Licorice), on which there are several studies in the past decade ^{[7]B}.



Figure 2. Chemical structures of glycycoumarin and licoarylcoumarin.

Table 1. Naturally-occurring 3-phenylcoumarins identified in the past decade.



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 IC_{50} of 660 nM in the presence of bovine serum albumin (BSA) ^[27]. Finally, QSAR predictive models for antioxidant activity of new coumarin derivatives have also been described as an interesting tool in drug discovery ^[28].



Figure 5. 3-Phenylcoumarins in oxidation.

3.4. Cardiovascular Diseases

A series of 6-halo-3-hydroxyphenylcoumarins has been evaluated for their vasorelaxation activity in intact rat aorta rings pre-contracted with phenylephrine, as well as for their inhibitory effects on platelet aggregation induced by thrombin in washed human platelets (**Figure 6**). These compounds proved to relax the vascular smooth muscle in a concentration-dependent manner. Compound **31** presents an IC₅₀ of 36.6 μ M against a concretion induced by phenylephrine. Some of the compounds showed a platelet antiaggregatory activity that was up to 30 times higher than that shown by *trans*-resveratrol, used as control, i.e., compound **32** (IC₅₀ = 6.41 μ M) ^[29]. The niacin receptor 1 (GPR109a) is a receptor that inhibits lipolytic and atherogenic activity and induces vasodilatation. A series of coumarin-dihydroquinazolinone conjugates has been evaluated for its agonist potential, displaying, in compound **33**, robust agonist action to GPR109a with an EC₅₀ < 11 nM. Further, the efficacy of the active compound has been corroborated by in vivo assays, showing the animals reduced body weight in a diet-induced obese mice model. Compound **33** proved to reduce leptin in blood plasma and total serum cholesterol ^[30].



Figure 6. 3-Phenylcoumarins in cardiovascular diseases.

4. Other Interests: Fluorescent Probes

The 3-phenylcoumarins are a privileged scaffold not only for their interesting and varied specific pharmacological activities, but also for other physicochemical properties and, in particular, for their potential as fluorescence probes. Due to the interesting and large number of studies that exist on this in recent years, they deserve an independent review. In the current overview, very succinct examples have been selected which are considered the most significant. Fluorescent biosensors have been developed to enable imaging and monitoring of a variety of metabolites and cellular events. This can be done by direct visualization and analysis; however, 3-phenylcoumarins also offer enormous possibilities in biorthogonal fluorogenic reactions, which allow not only the visualization of a fixed situation but also, in many cases, to follow the transformations throughout complex metabolic processes [31] ^[32]. These compounds can be useful as fluorescent biosensors for the detection of hydroxyl radicals, and therefore can be potentially applied in the diagnosis of oxidative stress in the human body [33], metal cations such as Fe³⁺ [34] or anions such as carbonate [35]. 3-Phenylcoumarins can be used to analyze and study different biochemical compounds such as flavin [36][37] or anatomical/physiological states, such as the state of neuronal myelination [38], histamine released by mast cells ^[39], access to mitochondria ^[40] and others, which can be identified very directly with some metabolic problem, which in turn can be linked to a disease or disorder. On occasion, this may allow or facilitate the study of new pharmacological agents in oxidation/reduction processes [41], but also in more specific processes such as estrogen receptors and breast cancer [42], bioinorganic anticancer compounds [43], MAOs [44], etc. For these functions, they can also be associated with other groups or molecules, as there are examples with fluorenes or xanthenes $\frac{[45]}{4}$, tetrazines $\frac{[46]}{4}$, rhodamine $\frac{[47]}{4}$ or with different polymers $\frac{[48]}{4}$.