### Bioactive Flavonoid Hybrids Linked by 1,2,3-Triazole

#### Subjects: Chemistry, Medicinal

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Considering the biological activities of both flavonoids and 1,2,3-triazole ring, as well as the metabolic stability associated to this heterocyclic ring, hybridization of flavonoids with a 1,2,3-triazole ring has been widely reported over the last years. The most common way to obtain these hybrids is through the copper (I) catalyzed azide-alkyne cycloaddition (CuAAC), also known as Click Chemistry reaction. It was highlighted the flavonoid hybrids linked by the1,2,3-triazole ring obtained since 2017, including chalcones, flavones, flavanones and flavonols, among others, with antitumor, antimicrobial, antidiabetic, neuroprotective, anti-inflammatory, antioxidant, and antifouling activities.

flavonoids Click Chemistry 1,2,3-triazole biological activities

#### 1. Introduction

The 1,2,3-triazole ring in Medicinal Chemistry has been gaining increased attention over the past years. The ability of this heterocycle moiety to participate in hydrogen bonding and dipole interactions provides an improvement of solubility and binding to biomolecular targets. Triazole ring is stable towards hydrolysis, oxidative, and reductive conditions and enzymatic degradation <sup>[1]</sup>. Moreover, triazole can be used as a linker and as a bioisostere of different functional groups for the synthesis of new active compounds <sup>[2][3]</sup>. Besides that, therapeutic properties have been attributed to 1,2,3-triazole hybrids, including antibacterial <sup>[4]</sup>, anticancer <sup>[5][6]</sup>, anti-inflammatory <sup>[7]</sup> activities, among others <sup>[3]</sup>.

The most common reaction to synthesize the 1,2,3-triazole ring is the Copper (I) catalyzed azide-alkyne cycloaddition (CuAAC). CuAAC, commonly referred to as Click Chemistry, was first introduced by Sharpless and Meldal groups in 2002 and involves the reaction of a terminal alkyne and an aliphatic azide using copper as a catalyst to give 1,4-disubstituted-triazoles <sup>[8][9]</sup>. Huisgen 1,3-dipolar cycloaddition (**Figure 1A**) presented several limitations, including long reaction time, high temperature, and the formation of a mixture of 1,4 and 1,5-disubstitution products. In contrast, CuAAC (**Figure 1B**) allows selective, high-yield, faster and mild reaction conditions, with the formation of few side products <sup>[10][11]</sup>.

Moreover being widely used in drug discovery <sup>[12][13]</sup>, CuAAC has been applied in other areas, including bioimaging <sup>[14][15]</sup>, biological <sup>[16]</sup>, and biomedical<sup>[17]</sup> applications, synthesis of polymers <sup>[18][19]</sup> and dendrimers <sup>[20][21]</sup>, and preparation of chemosensors<sup>[22]</sup>, among others.

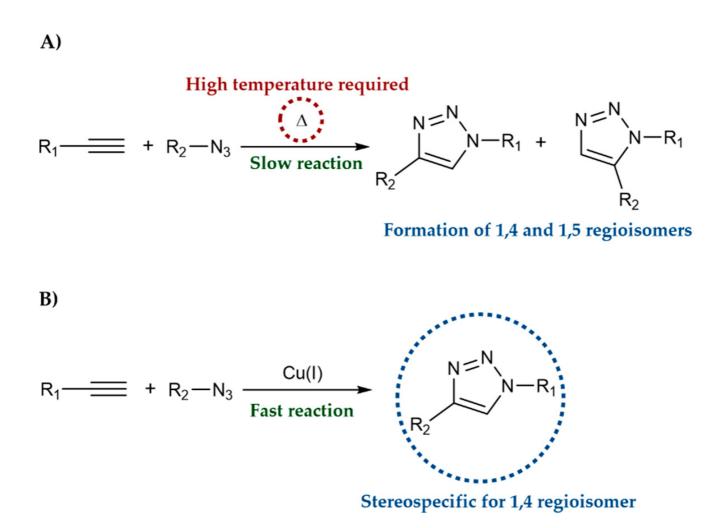


Figure 1. A) Huisgen 1,3-dipolar cycloaddition. B) Copper (I) catalyzed azide-alkyne cycloaddition (CuAAC).

Flavonoids are well-known for their wide range of pharmacological activities, such as anti-inflammatory, cardioprotective, antimicrobial, anticancer, and neuroprotective <sup>[23][24][25][26][27][28][29][30][31][32]</sup>. Beyond their potential in Medicinal Chemistry, flavonoids have been reported for their antioxidant, antiaging, anti-browning, anticorrosive, and antifouling activities which have highlighted the potential of flavonoids for several industrial applications (food, cosmetic, marine industries) <sup>[33][34][35][36][37][38]</sup>. Hybridization of flavonoids with other pharmacophoric moieties has become a strategy in Medicinal Chemistry for the development of compounds with higher potency by combining two or more pharmacophores into a single entity <sup>[39][40]</sup>. The incorporation of 1,2,3-triazole ring in flavonoid hybridization has become a strategy due to its bioactivity, metabolic stability, and convenient synthesis <sup>[41]</sup>.

It is to highlight the biological activities of flavonoid hybrids linked by the 1,2,3-triazole ring synthesized in the last five years, emphasizing the mechanism of action and SAR studies. Most of the flavonoid hybrids were obtained by CuAAC using copper sulphate and sodium ascorbate as catalysts, with slight modifications in the solvent system, since the *in situ* generation of Cu(I) salts from copper sulphate through reduction by sodium ascorbate is more convenient and more commonly used than Cu(I) salts, which needs the presence of a base as stabilizer and to assist in the ionization of the terminal acetylene [42]. Nevertheless, the utilization of copper iodide (CuI) as a

catalyst is also widely reported. Some research papers report the use of other catalysts, such as bromotris(triphenylphosphine)copper(I). Prior to the CuAAC step, a terminal alkyne or an azide is introduced in the flavonoid scaffold. Interestingly, for the synthesis of chalcones, most of the research papers reported the propargylation on the building blocks acetophenone or benzaldehyde (**Figure 2A**), followed by Claisen–Schmidt condensation, whereas for other flavonoids the propargylation generally occurred on the flavonoid scaffold (**Figure 2B**).



**Figure 2**. General steps for the synthesis of flavonoid hybrids conjugated with 1,2,3-triazole. A) Synthesis of chalcone-1,2,3-triazole hybrids. B) Synthesis of flavone/flavonol/flavanone/isoflavone-1,2,3-triazole hybrids.

The biological activities of the different subclasses of flavonoid hybrids, synthesized since 2017 are following presented, beginning with antitumor activity for which higher number of compounds were reported.

# **2. 1,2,3-Triazole linked Flavonoid Hybrids with Antitumor Activity**

In this chapter, more than 300 synthetic 1,2,3-triazole linked flavonoid hybrids with antitumor activity, including chalcones, flavanones, flavones, flavonols and isoflavones, have been reported since 2017. The conjugation of flavonoids by CuAAC giving rise to bioactive flavonoid dimers was also described. As shown in the **Figure 3**, chalcones was the most common flavonoid subclass.

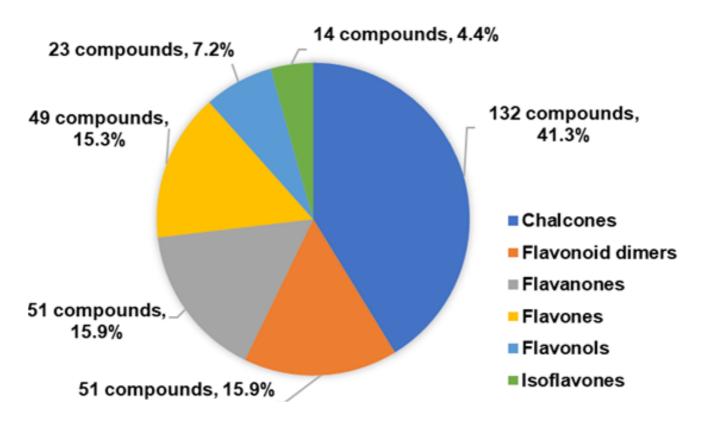


Figure 3. Distribution of the flavonoid hybrids with antitumor activity according to flavonoid subclasses.

## **3. 1,2,3-Triazole-Linked Flavonoid Hybrids with Antimicrobial Activity**

In this chapter, the synthesis and antimicrobial activities of more than 200 flavonoid hybrids linked by the 1,2,3triazole ring, reported since 2017, are described. Most of these compounds correspond to chalcone derivatives, as shown in **Figure 4.** Nevertheless, other classes of flavonoids, such as flavanones, flavones, and isoflavones, were also reported. Moreover, most of the described flavonoid hybrids showed antibacterial and antifungal activity. However, some reports about flavonoids with antiparasitic and antiviral activity can also be found in literature. It should be noted that several compounds simultaneously exhibited antibacterial, antifungal, and antiparasitic activities.

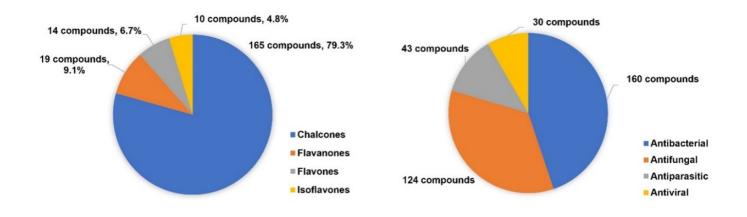


Figure 4. Distribution of the flavonoid hybrids according to the reported antimicrobial activity.

### 4. 1,2,3-Triazole-Linked Flavonoid Hybrids with Other Activities

Besides antitumor and antimicrobial activities, a total of 163 flavonoid hybrids conjugated by the 1,2,3-triazole ring with other activities, including neuroprotective, anti-inflammatory, antioxidant, antidiabetic, and antifouling, were also reported since 2017 and recorded. Moreover, some of these compounds have been shown to be modulators of other enzymes, namely carbonic anhydrase, serine proteases and glutatione S-transferase. The distribution of the different classes of flavonoids is highlighted in **Figure 5**.

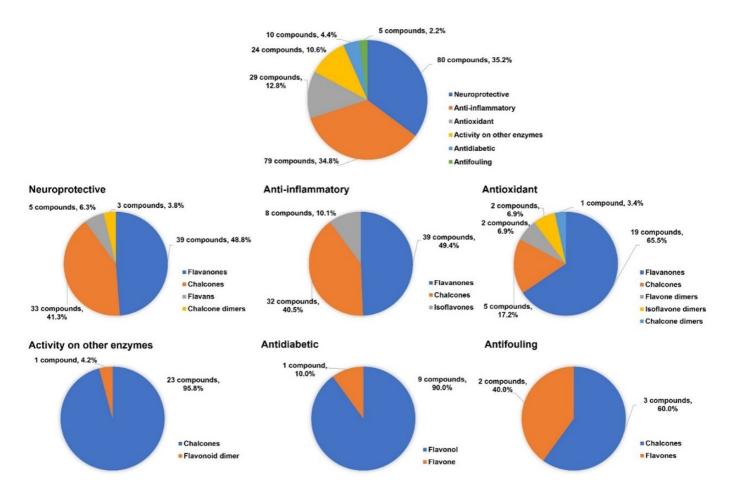


Figure 5. Other activities attributed to 1,2,3-triazole linked flavonoid hybrids.

### 5. Conclusion

Many compounds (691) belonging to the 1,2,3-triazole linked flavonoid hybrids reported since 2017, were highlighted, namely concerning their synthesis and biological activities. Concerning the synthesis of the triazole ring, it was found that for most, the Click Chemistry step was accomplished at room temperature using copper sulphate as a catalyst and sodium ascorbate as a reducing agent, with slight modifications on the solvent system,

generally giving rise to the desired compounds in high yields. Nevertheless, the use of conventional heating or MW irradiation was also reported, with the advantage of the decrease of reaction time. Instead of copper sulphate, the use of other copper catalysts such as copper iodide or bromotris(triphenylphosphine)copper(I) was also described with overall good results. Interestingly, some reports focused the Click Chemistry step using cellulose supported copper iodide nanoparticles as a reusable catalyst without loss of efficiency.

Among biological activities, antitumor and antimicrobial activities have been the most learned, mainly for chalcone derivatives. Some of derivatives of flavonoids with these activities revealed to be more potent than parent compounds or even commercially available drugs, using *in vitro* assays.

Considering the flavonoids with antimicrobial activity, several hybrids showed comparable or even higher antibacterial activity to reference drug ciprofloxacin, currently used for the treatment of microbial diseases. Moreover, some hybrids revealed promising antiplasmodial activity against multidrug resistant strains. These data suggest the potential of some flavonoid-hybrids linked by 1,2,3-triazole ring to circumvent microbiological diseases, especially those with multidrug resistance.

In addition to the antitumor and antimicrobial activity, several hybrids of flavonoids exhibited quite promising antiinflammatory and antidiabetic activities. Interestingly, some of these compounds also showed suitable pharmacokinetic and toxicity characteristics in in silico ADMET ones, reinforcing that the flavonoid hybridization using 1,2,3-triazole as linker may allow not only to improve the potency, but also the pharmacokinetic profile of flavonoids.

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