TiO2-NPs: Wastewater Treatment and Ago-Environment

Subjects: Materials Science, Biomaterials | Environmental Sciences | Agriculture, Dairy & Animal Science Contributor: Zahra Zahra , Zunaira Habib , Sujin Chung , Mohsin Ali Badshah

The tremendous increase in the production and consumption of titanium dioxide (TiO_2) nanoparticles (NPs) in numerous industrial products and applications has augmented the need to understand their role in wastewater treatment technologies. The use of TiO_2 NPs as the representative of photocatalytic technology for industrial wastewater treatment is coming to the horizon. As the use of industrial wastewater to feed agriculture land has been a common practice across the globe and the sewage sludge generated from wastewater treatment plants is also used as fertilizer in agricultural soils. Therefore, it is necessary to be aware of possible exposure pathways of these NPs, especially in the perspective of wastewater treatment and their impacts on the agro-environment.

TiO2 NPs applications wastewater treatment agro-environment

1. Introduction

TiO₂ NPs are one of the most extensively used NPs in different sectors [1]. For example, TiO₂ NPs are widely used in the agriculture sector for different purposes such as nano-pesticides and nano-fertilizers to introduce sustainable agricultural practices ^[2]. The availability of these nano-based agrochemicals in the market is expected to rise in near future 3. Similarly, the use of TiO₂ NPs has also gained the utmost importance in other fields, and eventually from different sources, the inevitable release of these NPs into the environment is obvious either through a direct or indirect route. For example, in 2008, the first evidence of TiO₂ NPs leaching (3.5×10^7 NPs per L) into the aquatic environment from facade paints was reported $[\underline{4}]$. In 2011, TiO₂ NPs were first detected in effluents of wastewater treatment plants, which were discharged into freshwater bodies where these NPs can cause unknown ecological risks [5]. TiO₂ NPs have also been observed to detach from some textiles and paints due to washing or weathering and to run into wastewater treatment plants [6][7] and especially in sewage sludge reaching the approximate concentration of 2 g·kg⁻¹ [8]. Sewage sludge is commonly employed as soil fertilizer in agriculture at the rate of approximately 3 tons per hectare (on a dry weight basis) annually [9][10][11] and become an ultimate source of TiO₂ NPs dissemination in agricultural soils. However, the overall concentration of these NPs in the environment through direct exposure route will be much higher than the indirect release. Interestingly, in both soil and water medium, TiO₂ NPs can be used for purification purposes due to their unique characteristics of photocatalysis in the presence of ultraviolet (UV) light [12][13]. Figure 1 below illustrates the brief overview of TiO₂ NPs applications, their role in wastewater treatment and their impacts on agro-environment.

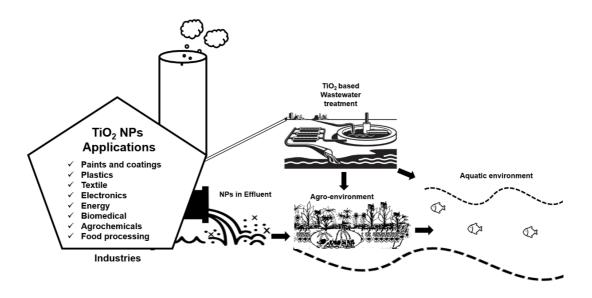


Figure 1. Illustration of the wide range of TiO_2 NPs applications from industries, their release into the wastewater, and their possible exposure routes towards the agro-environment.

2. Wastewater Treatments

With the onset of industrialization, there has been a steady increase in the types and amount of pollutants released in the environment. These environmental problems have garnered much attention on the global scale, especially water scarcity. Global water scarcity is a temporal and graphical mismatch between freshwater resources and the world's water demand. The increasing world population and urban industrialization have made water scarcity more alarming as shown in <u>Figure 2</u>, predicting the gap between supply (4200 billion m³) and demand (6900 billion m³) of freshwater in 2030. A major proportion of this water is used for the agriculture sector and then for the industrial sector.

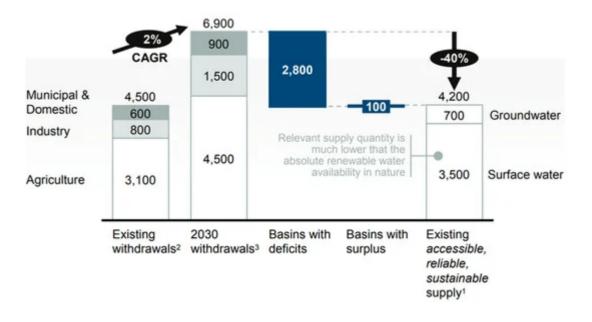


Figure 2. Comparison of current and future water demand, Reproduced with permission from ^[14], published by McKinsey & Company, New York, NY, USA, 2009.

With a growing world population, an ever-increasing demand for food production and potable water is questionable. The agriculture sector requires a surplus amount of water for irrigation. To avoid water scarcity issues, the reuse of wastewater is tremendously increasing across the planet. Reusing wastewater is a sustainable strategy to manage natural water resources ^[15]. However, the use of untreated wastewater for irrigation is a usual practice in developing countries causing serious threats to the ecosystem as well as human health. Specifically, carcinogenic pollutants pose a solemn threat to agricultural land, irrigated with industrial effluent without any treatment ^[16].

The whole world stands as a witness to unintended repercussions caused by rapid industrialization. The wastewater generated from industrial sectors has pronounced effects on humans as well as landmass fertility. Some industrial estates have operational wastewater treatment plants but unfortunately, they cannot handle a large proportion of industrial effluent. To meet the international standards of wastewater discharge, suitable technologies are required for wastewater treatment before discharging to streams. It could help to reduce the burden on freshwater resources by reusing treated water in various industrial processes. Due to the widely used application of nanotechnology, challenges, and opportunities of using engineered nanomaterials (ENMs) in wastewater treatment is a matter of endless concern. Based on the wastewater standards, a technique using TiO₂ NPs for resilient pollutants in the context of wastewater treatment has become popular in recent years. Up to date, TiO₂ NPs have drawn attention over other photocatalysts in every field of life. Over the last few decades, TiO₂ NPs with high photocatalytic efficacy has been tested to reduce the pollution load from various industrial units. The conventional wastewater treatment methods mostly come up with high costs as well as lower efficiencies. However, the advantages of the use of TiO₂ NPs (non-toxic, inexpensive, stable, and reusable NPs) appeared as a promising strategy to save the environment from pollution.

3. Impacts of TiO₂ NPs in the Agro-Environment

In agro-environment, soil is the main and complex matrix in which analyzing the fate of TiO_2 NPs is a challenging task. Furthermore, the impacts of TiO_2 NPs are difficult to measure in the soil due to the high geogenic background of Ti (\approx 0.6% of the terrestrial crust). Up until now, modeling studies had helped to estimate the approximate amount of TiO₂ NPs that is accumulating in the environment. According to recent forecasts, TiO₂ NPs sludge treated soils (with 45,000 tons) were observed to be the largest sink for NPs release among different environmental compartments ^[4]. The crop plants served as an entry route for NPs' uptake into the food chain. Presently, there are limited data available about these NPs interactions within the soil matrix. As nanotechnology is emerging in the field of agriculture sector in terms of growing global food production, nutritional contents, quality, food safety, and security ^[17]. Besides all these aspects, there are several other applications of NPs in agro-environments such as food processing and production, nano-fertilizer, nano-pesticides, etc. but the important concern arises here is the fate of these NPs.

Scientists have investigated the effects of TiO₂ NPs on the soil–plant continuum and have observed diverse impacts based on different characteristics of NPs, plant species, experimental conditions, and exposure period. For example, Figure 3, shows the TiO₂ NPs effects on plants with respect to different stages, concentration range, and exposure time. In a recent study, experiments were conducted on growth-promoting rhizobacteria (PGPR) inoculation with and without TiO₂ NPs in peat soil under the three stress situations. TiO₂ NPs were reported to enhance the performance of growth-promoting rhizobacteria which further promotes the solubilization of insoluble phosphates ^[18]. A grassland soil was treated with TiO₂ NPs at the rate of 0, 500, 1000, and 2000 mg kg⁻¹ of soil. These NPs were observed to negatively affect the soil bacterial communities after 60 days of exposure ^[19]. TiO₂ NPs effects on several bacterial taxa were also studied using incubated soil microcosms having concentrations range of TiO₂ NPs 0, 0.5, 1.0, and 2.0 mg g⁻¹ soil. Of the identified taxa that exist in all samples, 9 taxa were found to be positively correlated with TiO₂ NPs effects were investigated at concentrations ranging from 0.05 to 500 mg kg⁻¹ dry soil on different bacterial communities. The abundance of ammonia-oxidizing archaea was reported to decrease by 40% in response to TiO₂ NPs whereas *Nitrospira* was not affected at all. Furthermore, the abundance of ammonia-oxidizing bacteria and *Nitrobacter* were also reported to reduce due to TiO₂ NPs treatments ^[21].

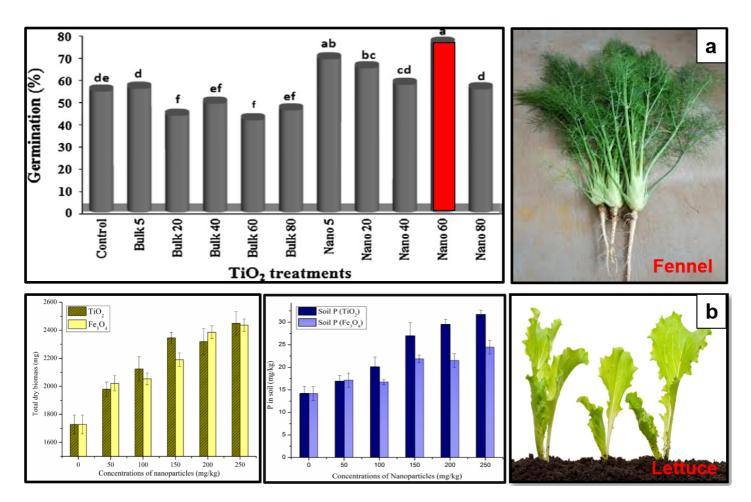


Figure 3. Effects of TiO_2 NPs on plants with respect to different stages, concentration range, and exposure time. (a) represents the effects of TiO_2 NPs on germination % of fennel seeds after short term exposure in a petri dish, the lowercase letters show the level of significance such as 'a' represent significant increase in germination percentage at Nano 60 treatment compared to control group. Adapted with permission from ^[22], published by ELSEVIER, 2013, (**b**) shows the effects of these NPs on lettuce plants after long term exposure of 90 days in soil, Adapted with permission from ^[24], published by American Chemical Society, 2015.

TiO₂ NPs (0, 5, 20, 40, 60, and 80 mg/kg) were used to study phytotoxicity and stimulatory impacts on fennel after 14 days of exposure. The mean germination percentage was increased by 76% at 60 mg L⁻¹, while the mean germination time was decreased by 31% at 40 mg L⁻¹ [22]. Similarly, in another study, plant shoot-root length was increased by 49% and 62%, respectively at 100 mg kg⁻¹ of NPs treatment in lettuce after 14 days exposure in soil medium ^[23]. Another study was performed using TiO₂ NPs treatments (0, 50–250 mg kg⁻¹) in soil medium for a period of 90 days. The total dry biomass was observed to increase 1.4-fold and phyto-available phosphorus (P) in soil by 2.2-fold, respectively ^[24]. <u>Table 1</u> enlists the recent studies conducted for the investigation of TiO₂ NPs effects on different plants.

Experimental Conditions	Plants	Impacts of TiO ₂	Ref.
TiO ₂ NPs Size: 20–30 nm Treatments: 0, 50, 100 and 200 mg L ⁻¹) in the growth medium of cocopite and perlite. Period: 60 days	Moldavian balm	Plants cultivated in salt stress conditions were observed to have improved physical traits and increased antioxidant enzyme activity in response to TiO ₂ NPs treatment compared to control.	[25]
TiO ₂ NPs Size: 50 and 68 nm Treatments: 100 mg nTiO ₂ /kg on 10 mg kg ⁻¹ of Cd-spiked soils Period: 14 days	Cowpea	No change in chlorophylls occurred. In leaves and roots, both ascorbate peroxidase and catalase activities were improved by NPs. TiO ₂ NPs have the potential for soil nano- remediation and could be an environmentally friendly option to tolerate soil Cd toxicity in cowpea plants.	[26]
TiO ₂ NPs Size: 30 nm Treatments: 0, 30, 50 and 100 mg kg ^{-1} Period: 60 days	Size: 30 nm Freatments: 0, 30, 50 M Wheat M Und 100 mg kg ⁻¹ M Wheat M Cu (296%), Al (171%), and Mg (187%) contents in shoots at 50 mg kg ⁻¹ TiO ₂ NPs treatment which shows improved grain quality and crop growth		[27]
TiO ₂ NPs Treatments: 0, 5, 10, 15, and 20 mg L ⁻¹ (foliar spray) Medium: Soil Period: 55 days	Rice (Oryza sativa)	The foliar spray of TiO_2 NPs reduced the soil bioavailable Cd by 10, 14, 28, and 32% in response to 5, 10, 20, and 30 mg/L NPs treatments compared to their control values. These NPs also significantly decreased the Cd concentration in the shoot as well.	[<u>28]</u>

Table 1. TiO₂ NPs applications since 2010 on different plants and their impacts.

Experimental Conditions	Plants	Impacts of TiO ₂	
TiO ₂ NPs Size: <40 nm Treatments: 0, 50, and 100/mg kg ⁻¹ Medium: Soil Period: 40 days	Wheat (Triticum aestivum)	Shoots and root lengths of wheat plants increased by16% and 4%, respectively. Phosphorus in shoots and roots was increased by 23.4% and 17.9% at 50/mg kg ⁻¹ of soil compare to control.	[<u>29</u>]
TiO ₂ NPs Size: <40 nm Treatments: 0, 25, 50, 150, 250, 500, 750 and 1000 mg L^{-1} Medium: Soil	ts: 0, 25, 50, Wheat (<i>Triticum</i> 500, 750 <i>aestivum</i>) mg L ⁻¹ Wheat (<i>Triticum</i> <i>kg</i> ⁻¹ , plant growth, biomass. Phosphorus content along with other tested parameters did not shown any improvement in the testing soils		[<u>30</u>]
ann 500 mn kn =		No effect of phytotoxicity was observed in plant growth, chlorophyll content, and biomass.	[<u>31</u>]
TiO ₂ NPs Treatments: 0–750 mg kg ⁻¹ Medium: Soil Period: 90 days	Rice (Oryza sativa)	Phosphorus concentration was increased in roots by 2.6-fold, shoots 2.4-fold, and grains 1.3-fold upon 750 mg kg ⁻¹ of NPs treatment. Metabolomics study revealed that levels of amino acids, glycerol content, and palmitic acid were also improved in grains.	
TiO ₂ NPs Treatments: 0, 100, 150, 200, 400, 600, and 1000 mg L ^{-1} Medium: Hydroponics Period: 7 days	Treatments: p_{1} (p_{1} (p_{2} p_{3}), 100, 150, 200, 400, Barley (<i>Hordeum vulgare</i> p_{3}), 100, 150, 200, 400, Barley (<i>Hordeum vulgare</i> p_{3}), 100, 150, 200, 400, Barley (<i>Hordeum vulgare</i> p_{3}), 100, 150, 200, 400, Barley (<i>Hordeum vulgare</i> p_{3}), 100, 150, 200, 400, Barley (<i>Hordeum vulgare</i> p_{3}), 100, 150, 200, 400, Barley (<i>Hordeum vulgare</i> p_{3}), 100, 150, 200, 400, Barley (<i>Hordeum vulgare</i> p_{3}), 100, 150, 200, 400, Barley (<i>Hordeum vulgare</i> p_{3}), 100, 150, 200, 400, Barley (<i>Hordeum vulgare</i> p_{3}), 100, 100, 100, 100, 100, 100, 100, 10		[33]
Ineatments: 0–100Wheat (Triticumincreasedmg kg ⁻¹ aestivum)32.3% of		NPs treatment at the rate of 20, 40, and 60 mg kg ⁻¹ increased plant growth and phosphorus uptake. 32.3% of chlorophyll content increased at 60 mg kg ⁻¹ while 11.1% decrease at 100 mg kg ⁻¹ .	[<u>34</u>]
$TiO_2 NPs$ Arabidopsis thaliana (L.)Size: >20 nmTreatments:0, 100, 250, 500 and1000 mg L ⁻¹ Medium: SoilPeriod: 5 weeks		Plant biomass and chlorophyll content decreased as the NPs treatment increase. Higher concentrations of NPs improved root growth. NPs treatments from 100 to 1000 μg mL ⁻¹ affect vitamin E content in plants. Decrease in plant biomass by 3-fold in response to 500 and 1000 mg/ml NPs treatment, whereas, at	

Experimental Conditions	Plants	Impacts of TiO ₂	
		100 mg/mL, the biomass decreases to half relative to control.	
TiO ₂ NPs Treatments: 250 and 500 μg/mL	Cabbage, Cucumber, Onion	The germination of cabbage significantly increased. In cucumber and onion, significant root elongation was observed.	[<u>36</u>]
TiO ₂ NPs P25: 29 \pm 9 nm, E171: 92 \pm 31nm, Non-nanomaterial TiO ₂ : 145 \pm 46 nm Treatments: 1, 10, 100, 1000 mg kg ⁻¹ Period: 12 weeks	Wheat, Red clover	TiO ₂ NPs showed restricted mobility from soil to leachate. No significant translocation of Ti was observed in both plant species, while average Ti content increased from 4 to 8 mg kg ⁻¹ at the highest treatments.	[<u>37]</u>
TiO ₂ NPs Size: 22 and 25 nm Period: 6 weeks	Soya bean	Plant growth significantly decreased which corresponds to the reduced carbon content in leaves.	[<u>38</u>]
TiO ₂ NPs Treatments: 0, 10, 20, 40 and 80 mg L^{-1} Medium: Petri dish Period: 10 days	Alyssum homolocarpum, Salvia mirzayanii, Carum copticum, Sinapis alba, and Nigella sativa	TiO ₂ NPs affected the germination and seedling vigor of 5 medicinal plants. Appropriate concentration levels had improved the germination as well as the vigor index of the subjected plant.	[<u>39]</u>
TiO ₂ NPs Treatments: 0, 10, 20, 30, and 40 mg mL ⁻¹ Parsley		Significant increase in seedlings germination percentage, germination rate index, shoot-root length, fresh biomass, vigor index, and chlorophyll content. 30 mg mL ⁻¹ was observed to be the optimum concentration of NPs. Increased germination percentage (92.46%) was observed at 40 mg mL ⁻¹ treatment, relative to the lowest one (44.97%) at control.	
TiO ₂ NPs Treatments: 0, 0.01%, 0.02%, and 0.03% Medium: Soil Period: 14 days	Wheat (Triticum aestivum)	Under the water-stressed conditions, the plant's length, biomass, and seed number along with the other tested traits like gluten and starch content were increased at 0.02% of NPs treatment.	[<u>41]</u>
TiO ₂ NPs Size: 14–655 nm	Wheat (Triticum aestivum)	NPs treatment improved root length. NPs above 140 nm diameter are not accumulated in wheat roots. NPs above 36 nm threshold diameter, can be accumulated (at concentration 109 mg Ti/kg dry weight) in wheat root parenchyma cells but are	[<u>42</u>]

Experimental Conditions	Plants	Impacts of TiO ₂	Ref.	
		unable to translocate to the shoot. Enhanced wheat root elongation was observed when exposed to 14 and 22 nm TiO_2 NPs.		-
TiO ₂ NPs Size: 5 nm Treatments: 0.25% NPs Medium: Hoagland nutritive fluid Period: 35 days	Arabidopsis thaliana	Improved photosynthesis and growth in plants were reported. Generally, the absorption of light in chloroplast and light-harvesting complex II was supposed to be stimulated by TiO ₂ NPs; thus, enhancing the transformation of light energy to electronic energy, the evolution of oxygen, and water photolysis.	[<u>43</u>]	_
TiO ₂ NPs (43%) with sucrose coating Size: >5 nm	Arabidopsis thaliana	Results revealed that small NPs entered plant cells and got accumulated in distinct subcellular locations.	[<u>44]</u>	_
TiO ₂ NPs Size: <100 nm Treatments: 0, 5, 10 and 20 mg L ⁻¹ Period: 20 days	Zea mays L. [<u>48]</u>	TiO ₂ NPs treatment significantly reduced the shoot, root biomass, and chlorophyll contents of leaves in a dose-dependent manner. Whereas positive effects were reported on the N, P, K, Zn Mn, and Cu contents except for Fe.	[<u>45</u>]	negati
TiO ₂ NPs Size: <100 nm Treatments: 15, 30, 60, 120 and 240 mg L^{-1} Period: at different time intervals up to a maximum of 82 days	Vicia faba	2 TiO ₂ NPs were reported to induce variations in a meiotic activity which results in an increased number of chromosomal abnormalities in the plant's reproductive parts.	[<u>46]</u>	mg kg enotype activation ts in co ondition toleran
[49] TiO ₂ NPs Size: <100 nm (tetragonal crystals), <10 nm (spherical shape) Treatments: 50 mg L ⁻¹ Period: 3 days	Vicia faba	² Based on the characteristics of size and shape, TiO_2 NPs can induce different levels of toxicity in terms of seed vigor index, aberration index and oxidative stress in plants.	[<u>47</u>]	uld be ar Ve further effects of onmenta

4. Conclusions and Future Perspectives

Based on the recent developmental facts related to TiO_2 NPs in wastewater treatment technologies and the agroenvironment, the use of TiO_2 NPs will further increase for promising applications in the near future. In wastewater treatment technologies, downstream separation of these NPs after photocatalytic degradation is still a matter of concern which can be minimized by using TiO_2 in photocatalytic reactors either in slurry form or immobilized on a solid substrate. Immobilization might result in loss of potential active sites which could be minimized by adding NMs into the polymeric substrate. The polymer can provide firm anchoring to TiO_2 NPs, however, there is still a chance of NPs leaching into the treated water and reaching the agricultural soils via irrigation. Since the agriculture sector is the backbone of the economy in most countries, studies based on crop improvement using TiO_2 NPs could help to overcome the burden of nutrient deficit in soils providing better crop yield. Apart from the potential benefits of TiO₂ NPs there are also some limitations that we could not ignore. At this stage, we could not claim with surety that the use of NPs is fully safe for human health and the environment or if it is harmful. Risks associated with chronic exposure of these NPs, interaction with flora and fauna, and their possible bioaccumulation effects have not been fully considered yet. The other limitations include the lack of information about a safe range of NPs' concentration, scalability of research and development for prototypes, industrial production, and public concern about health and safety issues. Detailed investigations are necessarily required to resolve these concerns and provide conclusive statements. We need to optimize the useful concentration levels of TiO₂ NPs for various applications and limit their usage for environmental safety.

References

- 1. Hou, J.; Wang, L.; Wang, C.; Zhang, S.; Liu, H.; Li, S.; Wang, X.; Toxicity and mechanisms of action of titanium dioxide nanoparticles in living organisms. *J. Environ. Sci.* **2019**, *75*, 40–53.
- 2. Kalpana, S.R.; Rashmi, H.B.; Rao, N.H.; Nanotechnology Patents as R&D Indicators for Disease Management Strategies in Agriculture. *J. Intellect. Prop. Rights* **2010**, *15*, 197–205.
- 3. Gogos, A.; Knauer, K.; Bucheli, T.D.; Nanomaterials in plant protection and fertilization: Current state, foreseen applications, and research priorities. *J. Agric. Food Chem.* **2012**, *60*, 9781–9792.
- Kaegi, R.; Ulrich, A.; Sinnet, B.; Vonbank, R.; Wichser, A.; Zuleeg, S.; Simmler, H.; Brunner, S.; Vonmont, H.; Burkhardt, M.; et al.et al. Synthetic TiO2 nanoparticle emission from exterior facades into the aquatic environment.. *Environ. Pollut.* **2008**, *156*, 233–239.
- Westerhoff, P.; Song, G.; Hristovski, K.; Kiser, M.A.; Occurrence and removal of titanium at full scale wastewater treatment plants: Implications for TiO2 nanomaterials. *J. Environ. Monit.* 2011, 13, 1195–1203.
- Mackevica, A.; Foss Hansen, S.; Release of nanomaterials from solid nanocomposites and consumer exposure assessment—A forward-looking review. *Nanotoxicology* 2016, 10, 641–653.
- Windler, L.; Lorenz, C.; von Goetz, N.; Hungerbühler, K.; Amberg, M.; Heuberger, M.; Nowack, B.; Release of Titanium Dioxide from Textiles during Washing. *Environ. Sci. Technol.* 2012, 46, 8181– 8188.
- Sun, T.Y.; Bornhöft, N.A.; Hungerbühler, K.; Nowack, B.; Dynamic Probabilistic Modeling of Environmental Emissions of Engineered Nanomaterials. *Environ. Sci. Technol.* 2016, 50, 4701– 4711.
- 9. Gottschalk, F.; Sun, T.; Nowack, B.; Environmental concentrations of engineered nanomaterials: Review of modeling and analytical studies. *Environ. Pollut.* **2013**, *181*, 287–300.

- 10. Kim, B.; Murayama, M.; Colman, B.P.; Hochella, M.F.; Characterization and environmental implications of nano- and larger TiO2 particles in sewage sludge, and soils amended with sewage sludge. *J. Environ. Monit.* **2012**, *14*, 1129.
- 11. Sharma, B.; Sarkar, A.; Singh, P.; Singh, R.P.; Agricultural utilization of biosolids: A review on potential effects on soil and plant grown. *Waste Manag.* **2017**, *64*, 117–132.
- 12. Wu, M.; Deng, J.; Li, J.; Li, Y.; Li, J.; Xu, H.; Simultaneous biological-photocatalytic treatment with strain CDS-8 and TiO2 for chlorothalonil removal from liquid and soil. *J. Hazard. Mater.* **2016**, *320*, 612–619.
- Zimbone, M.; Cacciato, G.; Boutinguiza, M.; Privitera, V.; Grimaldi, M.G.; Laser irradiation in water for the novel, scalable synthesis of black TiOx photocatalyst for environmental remediation. *Beilstein J. Nanotechnol.* 2017, 8, 196–202.
- Addams, L.; Boccaletti, G.; Kerlin, M.; Stuchtey, M. Charting Our Water Future: Economic Frameworks to Inform Decision-Making: 2030 Water Resources Group; McKinsey & Company: New York, NY, USA, 2009
- 15. Analouei, R.; Taheriyoun, M.; Safavi, H.R.; Risk assessment of an industrial wastewater treatment and reclamation plant using the bow-tie method . *Environ. Monit. Assess.* **2020**, *192*, 1–16.
- 16. Cao, H.; Chen, J.; Zhang, J.; Zhang, H.; Qiao, L.; Men, Y.; Heavy metals in rice and garden vegetables and their potential health risks to inhabitants in the vicinity of an industrial zone in Jiangsu, China. *J. Environ. Sci.* **2010**, *22*, 1792–1799.
- 17. Singh Sekhon, B.; Nanotechnology in agri-food production: An overview. *Nanotechnol. Sci. Appl.* **2014**, *7*, 31–53.
- 18. Timmusk, S.; Seisenbaeva, G.; Behers, L.; Titania (TiO2) nanoparticles enhance the performance of growth-promoting rhizobacteria . *2018* **Sci. Rep.**, *8*, 1–13.
- 19. Ge, Y.; Schimel, J.P.; Holden, P.A.; Evidence for negative effects of TiO2 and ZnO nanoparticles on soil bacterial communities. *Environ. Sci. Technol.* **2011**, *45*, 1659–1664.
- 20. Ge, Y.; Schimel, J.P.; Holdena, P.A.; Identification of soil bacteria susceptible to TiO2 and ZnO nanoparticles. *Appl. Environ. Microbiol.* **2012**, *78*, 6749–6758.
- Simonin, M.; Martins, J.M.F.; Le Roux, X.; Uzu, G.; Calas, A.; Richaume, A.; Toxicity of TiO2 nanoparticles on soil nitrification at environmentally relevant concentrations: Lack of classical dose–response relationships. *Nanotoxicology* **2017**, *11*, 247–255.
- 22. Feizi, H.; Kamali, M.; Jafari, L.; Rezvani Moghaddam, P.; Phytotoxicity and stimulatory impacts of nanosized and bulk titanium dioxide on fennel (Foeniculum vulgare Mill). *Chemosphere* **2013**, *91*, 506–511.

- Zahra, Z.; Arshad, M.; Rafique, R.; Mahmood, A.; Habib, A.; Qazi, I.A.; Khan, S.A.; Metallic Nanoparticle (TiO2 and Fe3O4) Application Modifies Rhizosphere Phosphorus Availability and Uptake by Lactuca sativa. *J. Agric. Food Chem.* **2015**, *63*, 6876–6882.
- 24. Gohari, G.; Mohammadi, A.; Akbari, A.; Panahirad, S.; Dadpour, M.R.; Fotopoulos, V.; Kimura, S.; Titanium dioxide nanoparticles (TiO2 NPs) promote growth and ameliorate salinity stress effects on essential oil profile and biochemical attributes of Dracocephalum moldavica. *Sci. Rep.* **2020**, *10*, 1–14.
- Ogunkunle, C.O.; Gambari, H.; Agbaje, F.; Okoro, H.K.; Asogwa, N.T.; Vishwakarma, V.; Fatoba, P.O.; Effect of Low-Dose Nano Titanium Dioxide Intervention on Cd Uptake and Stress Enzymes Activity in Cd-Stressed Cowpea [Vigna unguiculata (L.) Walp] Plants. *Bull. Environ. Contam. Toxicol.* 2020, *104*, 619–626.
- Ullah, S.; Adeel, M.; Zain, M.; Rizwan, M.; Irshad, M.K.; Jilani, G.; Hameed, A.; Khan, A.; Arshad, M.; Raza, A.; et al.et al. Physiological and biochemical response of wheat (Triticum aestivum) to TiO2 nanoparticles in phosphorous amended soil: A full life cycle study. *J. Environ. Manag.* 2020, 263, 110365.
- 27. Rizwan, M.; Ali, S.; ur Rehman, M.Z.; Malik, S.; Adrees, M.; Qayyum, M.F.; Alamri, S.A.; Alyemeni, M.N.; Ahmad, P.; Effect of foliar applications of silicon and titanium dioxide nanoparticles on growth, oxidative stress, and cadmium accumulation by rice (Oryza sativa). *Acta Physiol. Plant.* 2019, *41*, 1–12.
- 28. Zahra, Z.; Maqbool, T.; Arshad, M.; Badshah, M.A.; Choi, H.K.; Hur, J.; Changes in fluorescent dissolved organic matter and their association with phytoavailable phosphorus in soil amended with TiO2 nanoparticles. *Chemosphere* **2019**, *227*, 17–25.
- Zahra, Z.; Ali, M.A.; Parveen, A.; Kim, E.; Khokhar, M.F.; Baig, S.; Hina, K.; Choi, H.-K.; Arshad, M.; Exposure–Response of Wheat Cultivars to TiO2 Nanoparticles in Contrasted Soils . *Soil Sediment Contam. Int. J.* 2019, *28*, 184–199.
- 30. Larue, C.; Baratange, C.; Vantelon, D.; Khodja, H.; Surblé, S.; Elger, A.; Carrière, M.; Influence of soil type on TiO2 nanoparticle fate in an agro-ecosystem. *Sci. Total Environ.* **2018**, *630*, 609–617.
- Zahra, Z.; Waseem, N.; Zahra, R.; Lee, H.; Badshah, M.A.; Mehmood, A.; Choi, H.-K.; Arshad, M.; Growth and Metabolic Responses of Rice (Oryza sativa L.) Cultivated in Phosphorus-Deficient Soil Amended with TiO2 Nanoparticles. *J. Agric. Food Chem.* **2017**, *65*, 5598–5606.
- 32. Kořenková, L.; Šebesta, M.; Urík, M.; Kolenčík, M.; Kratošová, G.; Bujdoš, M.; Vávra, I.; Dobročka, E.; Physiological response of culture media-grown barley (Hordeum vulgare L.) to titanium oxide nanoparticles.. *Acta Agric. Scand. Sect. B Soil Plant Sci.* **2017**, 67, 285–291.
- 33. Rafique, R.; Zahra, Z.; Virk, N.; Shahid, M.; Pinelli, E.; Park, T.J.; Kallerhoff, J.; Arshad, M.; Dosedependent physiological responses of Triticum aestivum L. to soil applied TiO2 nanoparticles:

Alterations in chlorophyll content, H2O2 production, and genotoxicity . *Agric. Ecosyst. Environ.* **2018**, *255*, 95–101.

- Szymańska, R.; Kołodziej, K.; Ślesak, I.; Zimak-Piekarczyk, P.; Orzechowska, A.; Gabruk, M.;
 Zadło, A.; Habina, I.; Knap, W.; Burda, K.; et al.et al. Titanium dioxide nanoparticles (100–1000 mg/l) can affect vitamin E response in Arabidopsis thaliana.. *Environ. Pollut.* 2016, *213*, 957–965.
- 35. Andersen, C.P.; King, G.; Plocher, M.; Storm, M.; Pokhrel, L.R.; Johnson, M.G.; Rygiewicz, P.T.; Germination and early plant development of ten plant species exposed to titanium dioxide and cerium oxide nanoparticles. *Environ. Toxicol. Chem.* **2016**, *35*, 2223–2229.
- Gogos, A.; Moll, J.; Klingenfuss, F.; Heijden, M.; Irin, F.; Green, M.J.; Zenobi, R.; Bucheli, T.D.; Vertical transport and plant uptake of nanoparticles in a soil mesocosm experiment. *J. Nanobiotechnol.* **2016**, *14*, 40.
- 37. Burke, D.; Pietrasiak, N.; Situ, S.; Abenojar, E.; Porche, M.; Kraj, P.; Lakliang, Y.; Samia, A.; Iron Oxide and titanium dioxide nanoparticle effects on plant performance and root associated microbes. *Int. J. Mol. Sci.* **2015**, *16*, 23630–23650.
- Hatami, M.; Ghorbanpour, M.; Salehiarjomand, H.; Nano-anatase TiO2 modulates the germination behavior and seedling vigority of some commercially important medicinal and aromatic plants. *J. Biol. Environ. Sci.* **2014**, *8*, 53–59.
- 39. Dehkourdi, E.H.; Mosavi, M.; Effect of anatase nanoparticles (TiO2) on parsley seed germination (petroselinum crispum) in vitro. *Biol. Trace Elem. Res.* **2013**, *155*, 283–286.
- 40. Jaberzadeh, A.; Moaveni, P.; Tohidi Moghadam, H.R.; Zahedi, H.; Influence of bulk and nanoparticles titanium foliar application on some agronomic traits, seed gluten and starch contents of wheat subjected to water deficit stress. *Not. Bot. Horti Agrobot. Cluj Napoca* **2013**, *41*, 201–207.
- 41. Larue, C.; Veronesi, G.; Flank, A.M.; Surble, S.; Herlin-Boime, N.; Carrière, M.; Comparative uptake and impact of TiO2 nanoparticles in wheat and rapeseed. *J. Toxicol. Environ. Health Part A Curr. Issues* **2012**, *75*, 722–734.
- 42. Ze, Y.; Liu, C.; Wang, L.; Hong, M.; Hong, F.; The Regulation of TiO2 Nanoparticles on the Expression of Light-Harvesting Complex II and Photosynthesis of Chloroplasts of Arabidopsis thaliana. *Biol. Trace Elem. Res.* **2011**, *143*, 1131–1141.
- Kurepa, J.; Paunesku, T.; Vogt, S.; Arora, H.; Rabatic, B.M.; Lu, J.; Wanzer, M.B.; Woloschak, G.E.; Smalle, J.A.; Uptake and distribution of ultrasmall anatase TiO2 alizarin reds nanoconjugates in Arabidopsis thaliana. *Nano Lett.* **2010**, *10*, 2296–2302.
- 44. Dağhan, H.; Effects of TiO2 nanoparticles on maize (Zea mays L.) growth, chlorophyll content and nutrient uptake. *Appl. Ecol. Environ. Res.* **2018**, *16*, 6873–6883.

- 45. Kushwah, K.S.; Patel, S.; Effect of Titanium Dioxide Nanoparticles (TiO2 NPs) on Faba bean (Vicia faba L.) and Induced Asynaptic Mutation: A Meiotic Study. *J. Plant Growth Regul.* **2019**, ., 1–12.
- 46. Ruffini Castiglione, M.; Giorgetti, L.; Bellani, L.; Muccifora, S.; Bottega, S.; Spanò, C.; Root responses to different types of TiO2 nanoparticles and bulk counterpart in plant model system Vicia faba L.. *Environ. Exp. Bot.* **2016**, *130*, 11–21.
- 47. Jahan, S.; Alias, Y.B.; Bakar, A.F.B.A.; Yusoff, I.; Bin Toxicity evaluation of ZnO and TiO2 nanomaterials in hydroponic red bean (Vigna angularis) plant: Physiology, biochemistry and kinetic transport. *J. Environ. Sci.* **2018**, *72*, 140–152.
- 48. Mohammadi, R.; Maali-Amiri, R.; Abbasi, A.; Effect of TiO2 Nanoparticles on Chickpea Response to Cold Stress. *Biol. Trace Elem. Res.* **2013**, *152*, 403–410.

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