Synthesis of Variolins, Meridianins, and Meriolins

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Marine natural products are a source of essential significance due to a plethora of highly diverse biological properties. The naturally occurring (aza)indole alkaloids variolin B (1), meridianins (2), and their synthetic hybrids meriolins (3) exhibit potent kinase inhibitory activities.

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1. Syntheses of Variolins

1.1. First Total Synthesis by Morris and Anderson

The first total synthesis of variolin B (**1**) was achieved by Morris and Anderson in 2001 ^[1]. Later in 2005, they published the full details of their synthetic strategy together with the synthesis of the synthetic analog desoxyvariolin B ^[2]. They recognized the C2-symmetry of intermediate **8**, which is cyclized to the pyridopyrrolopyrimidine in the following key step. After halogen lithium exchange in the methylthiopyrimidine **4**, the reaction with diethyl carbonate (**5**) gave the symmetric ketone **6**. The reaction with the lithiated pyridine **7**, followed by the key step tandem deoxygenation and cyclization in the presence of triethylsilane and TFA led to the variolin core structure **9**. The introduction of the amino groups was achieved by oxidizing the dimethylthiol **9** with *m*-chloroperbenzoic acid (mCPBA) to the corresponding disulfoxide, which was reacted with *p*-methoybenzylamine (PMB amine) (**10**) to give the bisprotected amine **11**. Demethylation of **11** and removal of the PMB protecting groups gave the trifluoroacetate salt of the title compound, which was neutralized with concentrated ammonia to give variolin B (**1**) in an eight-step synthesis and an overall yield of **11%** (**Figure 1**) ^[1].



Figure 1. First total synthesis of variolin B (1) by Morris and Anderson ^[1].

1.2. Synthesis by Molina and Fresneda

The next synthetic approach was conducted by Molina and Fresneda, who published their syntheses of **1** in 2002 ^[3] and a modified synthetic route together with the synthesis of an analog in 2003 ^[4]. This approach starts with the synthesis of the 7-azaindole **16**. Aldehyde **13** was condensed with azidoacetate **14** and the resulting vinyl azide **15** cyclized to azaindole **16** via a nitrene insertion. After *N*-protection with 2-(trimethylsilyl)ethoxymethyl (SEM), the chloride key intermediate **19** was synthesized in a two-step procedure (**Figure 2**).



Figure 2. Synthesis of key intermediate azaindole **19**. Reaction conditions for a: first: 1.4 equivs NaH, DMF, rt, 45 min. Then: 1.4 equivs SEM-Cl, rt, 12 h ^[3].

Next, two different approaches are reported (**Figure 3**). Aldehyde **19** was similarly condensed as aldehyde **13** to give vinyl azide **20**. After *N*-SEM-deprotection, a Staudinger reaction with triphenylphosphane led to iminophosphorane **21** in a one-pot reaction. Reaction with benzyl isocyanate (**22**) in the key aza-Wittig reaction gave a non-isolable carbodiimide that subsequently cyclized to the desired pyridopyrrolopyrimidine moiety **23**.



Figure 3. Two approaches to the synthesis of the tricyclic pyridopyrrolopyrimidine structures, 23 and 27 [3].

Molina and Fresneda developed a second approach to obtain the tricyclic variolin core without the ester group at C-7. After the *N*-SEM-deprotection of **19**, a nitroaldol condensation with nitromethane led to the formation of **25**. Treatment with lithium aluminum hydride gave the corresponding 2-(2-aminoethyl)-7-azaindole, which was sequentially converted to the urea derivative **26** with benzyl isocyanide (**22**) without isolation. The **26** was dehydrated to the carbodiimide, which subsequently cyclized to the dihydropyrimidine **27** using the Appel reagent (CCl₄/PPh₃/NEt₃). Applying both synthetic approaches, an oxygen substituent is placed at C-4 and a nitrogen

substituent at C-9. The next step was to introduce the 2-aminopyrimidine ring at C-5, consequently leading both approaches to the acylated intermediate **31**. The reaction of **23** with phosphorus oxychloride and *N*,*N*-dimethylacetamide (DMA) (**28**) allowed the direct introduction of an acetyl group at C-5. Ester hydrolysis led to the carbonic acid **30**, and the thermal treatment forced the formation of intermediate **31** by decarboxylation. The route starting from **27** began with the introduction of a bromine substituent at C-5 and the reaction of bromine **32** with *n*-tributyltin(1-ethoxyvinyl)stannane (**33**) in the presence of dichlorobis (triphenylphosphine)-palladium(II) introduced to the acetyl group at C-5. Oxidation with 2,3-dichloro-5,6-dicyano-1,4-benzoquinone (DDQ) gave the intermediate **31** (**Figure 4**).



Figure 4. Introduction of an acetyl group at C-5 ^[3].

The 2-aminopyrimidine substituent was synthesized using a protocol developed by Bredereck (**Figure 5**) ^[5]. Enaminone **36** was synthesized from **31** with *N*,*N*'-dimethylformamide di-*tert*-butylacetal (**35**) in DMF. Condensation with guanidine hydrochloride (**37**) led to ring closure and formed the desired 2-aminopyrimidine **38**.



Figure 5. Synthesis of the 2-aminopyrimidine ring to give access to variolin B (1) [3].

1.3. Variolin B Approach by Alvarez

In 2003, Alvarez published the synthesis of variolin B (1) and the synthetic analog desoxyvariolin B $^{[6][7][8]}$. Starting from 4-methoxy-7-azaindole (40), a lithium carboxylate was used as an *N*-protecting group as well as an *ortho*directing substituent to form a 2-lithio-7-azaindole with a protocol by Katritzky $^{[9]}$. Reaction with 2-(1,3dioxoisindolin-2-yl)acetaldehyde (41) gave the alcohol 42 that was protected with dihydropyran. *N*-deprotection of 43 by hydrazinolysis gave the aminoacetal 44. Ring closure was achieved by the reaction with *N*-tosylcarbonimidic dichloride (45) and diisopropylethylamine (DIPEA) giving 46 in a diasteriometric mixture in a ratio of 1:1. Removal of the *O*-tetrahydropyran (THP) protecting group and elimination of the resulting hydroxy group by the formation of its mesylate and treatment with triethylamine afforded the pyridopyrrolopyrimidine scaffold (48). Regioselective iodination with *N*-iodosuccinimide (NIS) gave the key intermediate 49 (Figure 6).



Figure 6. Synthesis of the key intermediate iodide 49^[6].

A Stille reaction of **49** and 2-acetylamino-4-trimethylstannylpyrimidine (**50**) in the presence of tris(dibenzylideneacetone)dipalladium(0) afforded **51**. The *O*-demethylation and *N*-acetyl-deprotection were achieved by the treatment of **51** with hydrobromic acid, and after reductive photolysis with hydrazine as a reducing agent and 1,4-dimethoxybenzene as an electron source, the tosyl group was cleaved to give variolin B (**1**) in a 10-step synthesis with an overall yield of 1% (**Figure 7**).



Figure 7. Synthesis of variolin B (1) via Stille coupling as a key reaction step ^[6].

1.4. Synthesis of Variolin B by Burgos and Vaquero

The 2008 approach by Burgos and Vaquero to synthesize variolin B (1) followed the strategy to design the highly functionalized trihalo-substituted pyridopyrrolopyrimidine core **55** and introduce the substituents via palladiummediated cross-coupling reactions ^{[10][11]}. The functionalized 7-azaindole **53** was synthesized from 7-azaindole in six single steps ^[12]. The **53** was reacted with *N*-tosylmethyl dichloroformimide (**54**) under phase-transfer conditions in the two-phase system LiOH (aq., 30%)/CH₂Cl₂ (1:1) with tetrabutylammonium chloride to give the trihalosubstituted compound **55**. The C-9 amino substituent was introduced by a palladium-mediated C-N bond formation, using lithium bis(trimethylsilyl)amide (LiHMDS) and triphenylsilylamine as an ammonia source. The reaction required the use of the ligand [1,1'-biphenyl]-2-yldi-*tert*-butylphosphane (JohnPhos). After *N*-acetyl-protection, **56** was obtained (**Figure 8**).



Figure 8. Synthesis of the trihalo core and introduction of the C-9 amino substituent ^[10].

Next, in a debromination-iodination process, tris(trimethylsilyl)silane (TTMSS) and azobisisobutyronitrile (AIBN) and subsequently NIS were used to exchange the bromo compound **56** to the more reactive iodo derivative **57**. In a palladium-catalyzed cross-coupling reaction with the pyrimidyl stannyl reagent **58**, the C-C bond at C-5 was formed and the deprotection of both amino groups led to **59**. Then, in a palladium-promoted C-O coupling microwave (MW) reaction with sodium *tert*-butoxide, the *tert*-butyl group was introduced at C-4 to give the *tert*-butyl ether **60**, and in a final step, the *tert*-butyl moiety was cleaved to give variolin B (**1**) (**Figure 9**). Starting from **53**, variolin B was synthesized in seven steps with an overall yield of 5% ^{[10][11]}.



Figure 9. Palladium-mediated synthesis of variolin B (1) ^[10].

2. Syntheses of Meridianins

2.1. First Total Synthesis of Meridianins D and G by Jiang and Yang

In the early 2000s, Jiang and Yang published a straightforward synthesis of meridianins D and G. Starting from the corresponding indolyl boronic acid, **61** with 4-chloropyrimidine **62a** is the key reaction in this meridianin synthesis to furnish protected meridians **63** (**Figure 10**). After *N*-tosyl-deprotection of compounds **63** with sodium hydroxide, meridianin G (**2g**) is obtained in an overall yield of 63%, and meridianin D (**2d**) in an overall yield of 40% in this two-step synthesis ^[13].



Figure 10. First synthesis of meridianins D and G ^[13].

2.2. Synthesis of Meridianins by Fresneda and Molina

Shortly after the publication of the first meridianin syntheses, Fresneda and Molina developed a facile two-step synthesis of meridianins, starting from *N*-tosyl-3-acetylindoles **64**. The reaction of **64** with dimethylformamide dimethylacetal (DMF-DMA) gave the enaminone **66**. The key step was the formation of the 2-aminopyrimidine ring by condensation of **66** with guanidine hydrochloride (**37**). Molina and Fresneda described the synthesis of meridianin D (**2d**, 65% overall yield) as well as the first total synthesis of meridianin C (**2c**, 59% overall yield) and *O*-benzyl-protected derivative **2h**. After *O*-deprotection and dehalogenation of **2h** with hydrogen and palladium on carbon, meridianin A was synthesized for the first time (**2a**, 31% overall yield) or respectively by treating **2h** with the milder deprotecting agent; no dehalogenation occurred to give the first total synthesis of meridianin E (**2e**, 24% overall yield) (**Figure 11**) ^{[14][15]}.



Figure 11. Synthesis of meridianins A, C, D and E by Molina and Fresneda [14].

2.3. Meridianin Synthesis by Müller via Carbonylative Alkynylation

In 2005, Karpov et al. published a concise synthesis of meridianins C, D, and G. *Tert*-butoxycarbonyl(Boc)protected indoles (67) reacted in a palladium-catalyzed three-component carbonylative alkynylation with TMSprotected acetylene (TMSA) (68) to the TMS-alkynones 69. Subsequent cyclocondensation with guanidine (37) concluded the meridianin synthesis as both the TMS- and the Boc-group are cleaved under the chosen reaction condition. Meridianin G was obtained with an overall yield of 45%, and meridianin C and D could be isolated with 50% overall yield (Figure 12).



Figure 12. Synthesis of meridianins C, D, and G by Karpov et al. via carbonylative alkynylation [16].

2.4. Meridianin Synthesis by Penoni via Indolozation of Nitrosoarenes

Efficiently, Penoni approached the synthesis of meridianins C and G. In a one-pot process, the corresponding nitrosobenzene **70** was reacted with 2-amino-4-ethynylpyrimidine (**71**) to give the natural products **2c** (28%) and **2g** (41%) (**Figure 13**) ^{[<u>17]</u>.}



Figure 13. Indolization of nitrosoarenes for the synthesis of meridianins C and G by Penoni [17].

2.5. Synthesis of Meridianins via One-Pot Masuda Borylation-Suzuki Coupling Sequence by Müller

In 2011 and 2022, Müller and coworkers published a different synthetic strategy addressing meridianins. In a palladium-catalyzed Masuda borylation-Suzuki coupling (MBSC), one-pot procedure meridianins C, D, F and G, as well as the meridianin A precursor *O*-methyl meridianin A (**2i**), could be synthesized. 3-lodo-*N*-protected indoles **67** react with pinacolyl borane (HBpin) (**72**) and without the isolation of the resulting boronic acid ester, the subsequent Suzuki coupling with 2-aminopyridine (**62a**) leads to the formation of *O*-methyl meridianin A (**2i**) and meridianins D (**2d**) and G (**2g**). The Boc-protecting group is cleaved under the Suzuki conditions. In contrast to this, *N*-tosyl-protected indoles **73** require an additional deprotecting step that can be implemented in the one-pot process. Treatment with potassium hydroxide leads to meridianins C (**2c**), F (**2f**), and G (**2g**) (**Figure 14**).



Figure 14. Synthesis of meridianins by Müller via MBSC sequence [18][19].

When *O*-methyl meridianin A (2i) was melted with pyridinium hydrochloride, the natural product 2a could be isolated in 85% (Figure 15) ^[18].



Figure 15. Demethylation of 2i furnished meridianin A.

2.6. Synthesis of Meridianin F by Grainger

Grainger and coworkers worked on the regioselective dibromination of methyl indole-3-carboxylate and its application in the synthesis of indole building blocks. In this context, the synthesis of meridianin F was performed. Dibrominated indole **74** was reacted with *N*,*O*-dimethylhydroxylamine (**75**) to form the corresponding Weinreb amide **76**. Treatment with lithium(trimethylsilyl)acetylide (**77**) led to the formation of alkynone **78**. In the aftermath, meridianin F (**2f**) was furnished by cyclocondensation with guanidine (**37**) according to the aforementioned protocol by Müller in an overall yield of 37% (**Figure 16**) ^{[16][20]}.



Figure 16. Synthesis of meridianin F by Grainger ^[20].

2.7. Domino Amino-Palladation Reaction for the Synthesis of Meridianins C and G by Morris

Morris and coworkers came up with a modified Cacchi protocol ^[21] to synthesize meridianins from readily available monocyclic precursors in a catalytic domino amino-palladation reaction ^[22]. The four-step synthesis starts with a Sonogashira coupling of 2-iodoaniline **79** and TMSA (**68**) to give 2-alkynyl anilines **80** followed by *N*-mesylation to give the activated sulfonamide **82**. The reaction of **82** with the *N*-Boc-protected 4-iodo-2-aminopyrimidine **83** in a Cacchi-type protocol led to the formation of the protected meridianin precursors **84**. The global deprotection was achieved in a one-pot acid/base process and furnished meridianins C (**2c**, 31% overall yield) and G (**2g**, 45% overall yield) in four steps (**Figure 17**).



Figure 17. Synthesis of meridianins C and G via palladium-catalyzed domino reaction by Morris ^[22].

3. Syntheses of Meriolins

3.1. First Synthesis of Meriolin 1 by Molina and Fresneda

With their protocol for the synthesis of meridianins in hand, Molina and Fresneda were able to extend their strategy to 7-azaindoles, leading to the first synthesis of meriolin 1 (**3a**). 7-Azaindole (**85a**) was treated with acetyl chloride (**86**) in the presence of tin (IV) tetrachloride, which afforded 3-acetyl-7-azaindole (**87**). After *N*-tosyl-protection, the enaminone **90** was furnished after the reaction of **89** with DMF-DMA (**65**) similarly to the meridianin synthesis. Cyclocondensation with guanidine (**37**) led to the 2-aminopyrimidine formation and meriolin 1 (**3a**) was obtained in the 16% overall yield (**Figure 18**) ^[14].



Figure 18. Synthesis of meriolin 1 by Molina and Fresneda $\frac{14}{14}$.

3.2. Synthesis of Meriolin Derivatives by Joseph and Meijer

Joseph, Meijer, and coworkers used the strategy by Molina and Fresneda for the synthesis of meriolin 1 and in addition to that were able to generate a large substance library of novel meriolin derivatives. Starting from substituted 7-azaindoles **85**, acylation in 3-position was achieved by treatment with acetic anhydride (**91**) and trifluoroacetic acid. *N*-protection with benzenesulfonyl chloride (**91**) afforded the derivatives **94** (**Figure 19**) ^{[23][24]}.



Figure 19. Preparation of 3-acetyl-*N*-protected intermediates 94 [23].

In the case of **85g**, an alternative pathway was chosen to prevent *O*-demethylation under acidic conditions. After its iodination, the resulting **95** was first treated with benzenesulfonyl chloride (**93**) to give the *N*-protected intermediate **96** that was reacted with **33** in a palladium-mediated Stille cross-coupling reaction to form **94f**. Treatment of the 4-methoxy derivative **92b** with dimethyl sulfate (**97**) gave the *N*-methylated intermediate **94 g** (**Figure 20**).



Figure 20. Alternative pathways to access 94f and 94g ^[23].

The functionalized 7-azaindoles **94** were then transformed to the corresponding enaminones **97** according to the Molina and Fresneda protocol, and after cyclocondensation with guanidine (**37**), meriolins 3–7 (**3c**, **3e–g**, **3b**') and 9–11 (**3h**, **3b**, **3d**) were obtained. Meriolin 7 (**3b**') was isolated as a side product in the synthesis of meriolin 10 (**3b**), where a nucleophilic substitution of the chlorine substituent took place. Treatment of **97b** with 2-methyl-2-thiopseudourea sulfate (**98**) led to the formation of the 2-methylthiopyrimidine-substituted meriolin 12 (**3i**). The meriolins 3–7 and 9–12 were isolated in overall yields ranging from 12 to 37% starting from the corresponding 7-azaindole **85** (**Figure 21**).



Figure 21. Synthesis of meriolins 3–7 and 9–12 by Joseph and Meijer ^[23].

The 4-methoxy-substituted meriolins **3c**, **3h**, and **3j** could be transformed to the corresponding 4-hydroxysubstituted meriolins 2 (**3k**, 26% overall yield), 8 (**3l**, 31% overall yield), and 13 (**3m**, 22% overall yield) by *O*demethylation with hydrobromic acid in acetic acid (**Figure 22**) ^{[23][24]}.



Figure 22. O-demethylation of meriolins 3, 9, and 14 gives 4-hydroxy-substituted meriolins 2, 8, and 13 ^[23].

3.3. Meriolin Syntheses by Müller via Carbonylative Alkynylation

The Müller approach to meridianins via carbonylative alkynylation and subsequent pyrimidine synthesis could be transferred to the synthesis of meriolins. A small library of potential kinase inhibitors has been synthesized for screenings, among them meriolin derivatives **3a** and **3n**. Therefore, 3-iodo-*N*-Boc-7-azaindole (**99a**) was transformed to the alkynones **101** in a Sonogashira coupling with TMSA (**68**) or 1-hexyne (**100**). Alkynones **101** were then cyclized with guanidine (**37**), either in a mixture of *tert*-butanol and acetonitrile or in DMF, to the meriolin derivatives **3a** (37% overall yield) and **3n** (51% overall yield) (**Figure 23**) ^{[16][25]}.



Figure 23. Synthesis of meriolins 3a and 3n via carbonylative alkynylation and subsequent pyrimidine synthesis by Müller [16][25].

3.4. Three-Component Glyoxylation Decarbonylative Alkynylation Synthesis of Alkynones by Müller

Another approach by Merkul et al. to address meriolins was performed via a three-component glyoxylation alkynylation reaction, leading to *N*-benzylated and *N*-methylated meriolins **30** and **3p**. Starting from 7-azaindoles,

85 in the first step reaction with oxalyl chloride (**102**) furnished the indole-3-glyoxyl chlorides. These reactive synthetic equivalents of acid chlorides were directly transformed to alkynones **104** in a decarbonylative Sonogashira coupling with 1-hexyne (**100**) or phenylacetylene (**103**). The cyclocondensation reaction with guanidine (**37**) went similarly as in the previously described strategy, which gave meriolins **30** and **3p** in 51 and 52% overall yield (**Figure 24**) ^[26].



Figure 24. Synthesis of meriolins via three-component glyoxylation decarbonylative alkynylation by Merkul et al. [26]

3.5. Synthesis of Meriolins with a Suzuki Coupling as a Key Reaction by Huang

To investigate their kinase inhibitory effects, Huang and coworkers established a synthetic route to derivatize meriolins via a nucleophilic substitution on the pyrimidine moiety, as well as by functionalizing the N-1 and C-2 position on the azaindole moiety. 7-Azaindole (**85a**) was *N*-protected by treatment with benzenesulfonyl chloride (**93**) before it was selectively brominated in the C-3 position. Brominated and *N*-protected **106** was then transformed to the boronic acid ester **108** in a $Pd(dppf)_2Cl_2$ -catalyzed Miyaura borylation with bis(pinacolato)diboron (**107**). Suzuki coupling with 2,4-dichloropyrimidine (**109**) in the presence of $Pd(PPh_3)_4$ gave **110** in superior regioselectivity. The chlorine substituent on the pyrimidine ring could then be substituted by various amines **111** in a nucleophilic aromatic substitution. Two equivalents of amine were used, as one equivalent was consumed by the concurrent cleavage of the benzenesulfonyl group. This furnished 15 meriolin derivatives (**3q-ae**) with overall yields ranging from 48 to 58% (**Figure 25**) ^[27].



Figure 25. Meriolin synthesis by Huang via nucleophilic substitution on the pyrimidine moiety [27].

For the installment of solubilizing amino side chains, derivative **3ae** was treated with methanesulfonyl chloride (**112**), and the mesylate **3af** was obtained. After treatment with different amines **113**, the meriolin derivatives **3ag-ai** have been isolated in overall yields of 76–86% (starting from **3ae**) (**Figure 26**) ^[27].



Figure 26. Introduction of solubilizing side chains gave the meriolin derivatives **3ag-ai** ^[27].

To assess the role of the NH group of the 7-azaindole unit in CDK1 binding, *N*-functionalization was anticipated. Compound **3ab** was treated with potassium *tert*-butoxide before it reacted with different electrophiles **114** to give the derivatives **3aj-am** (**Figure 27**) ^[27].



Figure 27. *N*-functionalization of compound **3ab** with different electrophiles gave meriolins **3aj-am** ^[27].

Lastly, compound **110** was treated with lithiumdiisopropylamine (LDA) and methyl iodide (**115**) to introduce a methyl group in the C-2 position. After the reaction with amine **117**, an additional deprotection step was added, since the C-2 methyl group caused the *N*-benzenesulfonyl group to be stable under hot aminolysis conditions. This furnished meriolin **3an** in 6% overall yield starting from **110** (**Figure 28**) ^[27].



Figure 28. Synthesis of meriolin 3an [27].

3.6. Meriolin Synthesis via the Masuda borylation-Suzuki Coupling Sequence by Müller

The Müller group could show the versatility of the MBSC sequence by transferring their meridianin protocol to the synthesis of meriolins and other biaryl systems ^{[18][19][28][29]}. In a one-pot-process, 7-azaindoles **99** were transformed to the corresponding pinacolyl boronic acid esters in a palladium-mediated Masuda borylation. In the sense of a sequentially catalyzed reaction, a subsequent Suzuki coupling with arylhalide **62** follows (**Figure 29**). Under Suzuki conditions, the Boc group is concomitantly cleaved, leading to meriolin 1 (**3a**), meriolins **3ao-au**, and the *N*-benzylated meriolins **3av** and **3aw**, with yields ranging from 37 to 96% (**Table 1**). If the reaction sequence is started with *N*-tosylated azaindoles **99b** a subsequent deprotection step with a hydroxide base is required. This step can be included in the one-pot process, which led to meriolins **3a** and **3ax-bi** with overall yields ranging from 40 to 91% starting from 7-azaindoles **99b** (**Figure 29**) ^{[18][28]}.



Figure 29. MBSC sequence for the synthesis of meriolins 3a and 3ao-bi by Müller [18][28].

Entry	Azaindole 99	R ¹	(hetero)aryl R ² 62	Meriolins 3 (yield)
1	99a	Н	4-pyrimidin-2-amine (62b)	3a , meriolin 1 (63%)
2	99a	Н	6-pyrazin-2-amine (62c)	3ao (53%)
3	99a	Н	5-pyrimidin-2-amine (62d)	3ap (66%)
4	99a	Н	2-pyrimidin-4-amine (62e)	3aq (37%)
5	99a	Н	6-pyridin-2-amine (62f)	3ar (81%)
6	99a	Н	4-pyridin-2-amine (62g)	3as (64%)
7	99a	Н	2-aniline (62h)	3at (74%)
8	99a	Н	4-phenol (62i)	3au (57%)
9	99c	Bn	4-pyrimidin-2-amine (62b)	3av (96%)
10	99c	Bn	4-pyridin-2-amine (62g)	3aw (93%)
11	99b	Н	4-pyrimidin-2-amine (62b)	3a, meriolin 1 (81%)
12	99b	Н	5-pyridin-2-amine (62j)	3ax (91%)
13	99b	Н	4-pyridin-2-amine (62g)	3ay (75%)
14	99b	Н	N-benzyl-5-pyridin-2-amine (62k)	3az (77%)

Table 1. Introduced heterocycles R^2 62 and the corresponding yields of the synthesized meriolins 3

15	99b	Н	4-(2-methoxypyrimidine) (621)	3ba (40%)	
16	99b	Н	4-pyridin-2,6-diamine (62m)	3bb (67%)	
17	99b	Н	5-pyrimidin-2-amine (62d)	3bc (47%)	
18	99b	Н	4-(2-methylthiopyrimidine) (62n)	3bd (83%)	
19	99b	Н	4-(6-methoxypyrimidin-2-amine) (620)	3be (62%)	
20	99b	Н	4-pyrimidin-2,6-diamine (62p)	3bf (53%)	
21	99b	Н	N-benzyl-4-pyridin-2-amine (62q)	3bg (83%)	
22	99b	Н	2-pyrimidin-4-amine (62e)	3bh (75%)	
23 ^a	99b	Н	5-isoquinolin (62r)	3bi (82%)	orri

Morris and coworkers tried to adapt their meridianin protocol to the synthesis of meriolins starting from iodinated aminopyridines **119** to give 3-alkynylated 2-amino pyridines **120**. In contrast to anilines (vide supra), it was not possible to prepare the monomesylated aminopyridines directly, which required treatment with trifluoroacetyl anhydride (TFAA) (**121**) first to furnish trifluoroacetamides **122**. Reaction with mesyl chloride (**112**) led to the desired intermediate **123** that could be converted in the optimized domino reaction with *N*-Boc-4-iodopyrimidine-2-amine (**83**) and subsequent acid/base deprotection protocol to give meriolin **1** (**3a**) in 34% overall yield, as well as the 5-bromo meriolin **3bj** in 31% overall yield (**Figure 30**) ^[22].



Figure 30. Synthesis of meriolins 3a and 3bj via domino amino-palladation reaction by Morris [22].

3.8. Metal-Free CH-Activation of a Pyrimidine and an Indolylboronic Ester by Singh

In 2016, Singh presented a metal-free CH-activation approach toward the synthesis of meriolin 1. The group reported cross-coupling between diazines and related electron-deficient heteroarenes with organoboron species. Treatment of *N*-Boc-protected boronic acid ester **124** with 2-aminopyrimidine (**125**) and potassium persulfate in an acetone-water mixture led to the formation of meriolin 1 in 35% yield (**Figure 31**). The proposed mechanism includes the formation of a sulfate anion radical that activates the boronic acid ester and generates an azaindolyl radical. The radical reacts with the in situ-formed pyrimidinyl salt to form a radical cation. After it undergoes single electron transfer, the protonated form of the desired product is obtained ^[30].



Figure 31. Metal-free synthesis of meriolin 1 via CH-activation by Singh ^[30].

3.9. Functionalization of Meriolins via Suzuki Coupling or Nucleophilic Substitution Reactions by Singh and Malik

A different approach by Singh in cooperation with Malik was more pragmatic to synthesize a large library of meriolins to establish structure-activity-relationships and determine their potency against CDKs. It was elaborated that functionalization in the C-5 position and N-1 position on the 7-azaindole unit, as well as on the pyrimidine ring, should be accomplished. Starting from 5-bromo-7-azaindole (**85j**) in a Suzuki coupling with several boronic acids, **126** led to functionalized 7-azaindoles **127** (**Figure 32**). After iodination and protection with Boc-anhydride (**128**) to

give 3-iodo-7-azaindoles **129**, a Masuda borylation with pinacolyl borane (**72**) and subsequent Suzuki coupling with 4-chloropyrimidine-2-amine (**62**) gave the meriolin derivatives **3bk-cc** in 25–46% overall yield (**Figure 32**, **Table 2**) [<u>31</u>].



Figure 32. Functionalization of the C-5 position led to meriolins 3bk-cc [31].

 Table 2. Boronic acids 126 used for the functionalization in C-5 position and the corresponding yields of meriolins

 3bk-cc.

Entry	Boronic Acid R ¹ -B(OH) ₂ (126)	Meriolins 3 (yield)
1	(4-(trifluoromethyl)phenyl)boronic acid (126a)	3bk (63%)
2	(4-fluorophenyl)boronic acid (126b)	3bl (65%)
3	(4-chlorophenyl)boronic acid (126c)	3bm (62%)
4	(4-(trifluoromethoxy)phenyl)boronic acid (126d)	3bn (60%)
5	(4-methoxyphenyl)boronic acid (126e)	3bo (59%)
6	(4-(methylthio)phenyl)boronic acid (126f)	3bp (54%)
7	(3-fluorophenyl)boronic acid (126g)	3bq (55%)
8	<i>m</i> -tolylboronic acid (126h)	3br (50%)
9	(3-(trifluoromethyl)phenyl)boronic acid (126i)	3bs (58%)

10	(2-(methylthio)phenyl)boronic acid (126j)	3bt (59%)
11	(2-ethylphenyl)boronic acid (126k)	3bu (48%)
12	naphthalen-1-ylboronic acid (126I)	3bv (60%)
13	(2-methoxynaphthalen-1-yl)boronic acid (126m)	3bw (55%)
14	furan-3-ylboronic acid (126n)	3bx (48%)
15	thiophen-3-ylboronic acid (1260)	3by (44%)
16	pyridin-3-ylboronic acid (126p)	3bz (45%)
17	benzo[<i>b</i>]thiophen-2-ylboronic acid (126q)	3ca (40%)
18	benzofuran-2-ylboronic acid (126r)	3cb (39%)
19	(5-methoxy-1 <i>H</i> -indol-2-yl)boronic acid (126s)	3cc (45%)

To derivatize the pyrmutine mig, rounated and *N*-protected *I*-azamole **12** was transformed to the corresponding pinacolyl boronic ester and reacted with **130** in a Suzuki coupling to give compound **3cd**. The thiomethyl group was oxidized to give the sulfone **3ce** using *m*-CPBA. Nucleophilic substitution by several primary **131** or secondary amines **132** furnished meriolins **3bk-cc** in 29–39% overall yield (**Figure 33**, **Table 3**). To vary the substituents in the N-1 position, at first the synthesis of meriolin 1 was approached using a Masuda borylation and subsequent Suzuki coupling. After treatment with sodium hydride, reaction with different sulfonyl chlorides **133** gave meriolins **3cn-ct** in 39–47% overall yield (**Figure 33**, **Table 4**).



Figure 33. Functionalization of the pyrimidine ring and the N-1 position gave meriolins 3cf-ct [31].

 Table 3. Primary 131 and secondary amines 132 used for the functionalization of the pyrimidine ring and the corresponding yields of meriolins 3cf-cm.

Entry	Amine R ² -NH ₂ (131) or R ² NHR ³ (132)	Meriolins 3 (yield)
1	2-phenylethan-1-amine (131a)	3cf (65%)
2	2-(4-methoxyphenyl)ethan-1-amine (131b)	3cg (60%)
3	2-(3,4-dimethoxyphenyl)ethan-1-amine (131c)	3ch (68%)
4	2-(1 <i>H</i> -indol-3-yl)ethan-1-amine (131d)	3ci (50%)
5	pyrrolidine (132a)	3cj (52%)
6	piperidine (132b)	3ck (52%)
7	morpholine (132c)	3cl (60%)
8	1-methylpiperazine (132d)	3cm (54%)

 Table 4. Sulfonyl chlorides 133 for the functionalization in N-1 position and the corresponding yields of meriolins

 3cn-ct.

Entry	Sulfonyl Chloride R ⁴ -SO ₂ Cl (133)	Meriolins 3 (yield)
1	4-fluorobenzenesulfonyl chloride (133a)	3cn (70%)
2	4-bromobenzenesulfonyl chloride (133b)	3co (70%)

3	4-(trifluoromethyl)benzenesulfonyl chloride (133c)	3cp (77%)
4	4-(trifluoromethoxy)benzenesulfonyl chloride (133d)	3cq (80%)
5	4-acetamidobenzenesulfonyl chloride (133e)	3cr (71%)
6	2,3-dihydrobenzo[b][1,4]dioxine-6-sulfonyl chloride (133f)	3cs (79%)
7	1-methyl-1 <i>H</i> -imidazole-5-sulfonyl chloride (133g)	3ct (65%)

3.10. Meriolin Synthesis via Friedel Crafts Acylation by Grädler

Grädler and coauthors started their approach on meriolins from 5-bromo-7-azaindole (**85**j) with a Friedel Crafts acylation using aluminium chloride and acid chloride **134**. The intermediate **135** was reacted in a cyclocondensation with guanidine carbonate (**136**), which furnished meriolin **3cu** in 40% overall yield. The bromine atom in C-5 position was then employed for further derivatization. Suzuki coupling with Boc-protected pinacolyl boronic acid ester **137** and subsequent Boc-deprotection with hydrochloric acid gave meriolin **3cv** in 41% yield (**Figure 34**). The overall yield after three steps is 17% ^[25]. Using this method, as well as the carbonylative alkynylation by Müller ^[16], several derivatives have been synthesized and tested for their PDK1 inhibitory properties ^[25].



Figure 34. Preparation of meriolin 3cv via Friedel Crafts acylation by Grädler ^[25].

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