

# Microbe-Assisted Remediation of Crude-Oil-Contaminated Environment

Subjects: Microbiology

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Crude oil contamination is an emerging environmental concern on a global scale due to inadvertent oil leakage. Oil spillage and discharges frequently happen as a consequence of explosion incidents during oilfield drilling; leakage from oil and gas pipelines and reservoirs, fuel tankers, and well waxing; and during overhauls of refineries and petrochemical manufacturing equipment. Crude oil and its products damage water and air quality and also reduce soil fertility. Crude oil can harm plants by clogging soil pores, reducing soil aeration and water permeability, which can have both ecological and toxicological effects and disrupt the soil's natural structure. Aside from these environmental concerns, one of the most serious consequences of such anthropogenic emissions is current global climate change. Crude oil contamination not only impacts different ecosystems but also influences global socio-economic status.

Keywords: crude oil contamination ; bioremediation ; bacteria ; biodegradation ; metabolic engineering

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## 1. Introduction

With all the shortcomings of other techniques of remediation, the complete or partially effective removal of crude oil is still a challenge <sup>[1][2][3]</sup>. The microbial degradation of complex and diverse crude oil molecules requires specific enzymes for a specific class of compounds <sup>[4][5]</sup>. Depending on the bacterial species and community, the resulting catabolites are either utilised by the bacteria themselves or released into the environment for further degradation by the other bacteria in the community <sup>[6]</sup>. The sole determining factor critical to the degradation is the survival of the bacteria in a medium with high crude oil contamination <sup>[7]</sup>. Due to their capacity to degrade a variety of crude oil components, bacteria are considered to be the most effective degraders of crude oil <sup>[8][9][10]</sup>. According to the research by El-Liethy et al. <sup>[11]</sup> the rate of crude oil degradation by the bacterial strain *Enterobacter hormaechei* was 0.6% in a minimal medium, but it accelerated to 70.7% when the medium was biostimulated with peptone. A bioremediation experiment carried out with biosurfactant-producing *Pseudomonas aeruginosa* showed a maximum of 68.3% degradation of *n*-hexadecane in 60 days <sup>[12]</sup>. Alkanes with different molecular chains may be degraded by the *Acinetobacter* strain DSM17874 using degrading genes, such as *alkB* and *AlmA* <sup>[13]</sup>. *Pseudomonas nitroreducens* efficiently degraded 70% of paraffinic contaminants in a short period of 10 days <sup>[14]</sup>. However, a consortium can attain better degradation of crude oil by the availability of a broader range of enzymes acting upon crude oil compounds, as well as the co-existence of multiple bacteria that assist in the formation of higher metabolic networks, which can interact with the persistent compounds and carry out the process of biodegradation <sup>[10][15]</sup>. A microbial consortium consisting of *Actinotalea ferrariae*, *Arthrobacter ginsengisoli*, *Dietzia cinnamea*, *Dietzia papillomatosis*, and *Pseudomonas songnensis* showed a degradation of crude oil ranging from 73.6 to 69.3% in soil with 1–10% crude oil contamination <sup>[16]</sup>.

Bacteria, such as *Acidobacter*, *Alteromonas*, *Arthrobacter*, *Burkholderia*, *Dietzia*, *Enterobacter*, *Kocuria*, *Marinobacter*, *Mycobacterium*, *Pandoraea*, *Pseudomonas*, *Staphylococcus*, *Streptobacillus*, *Streptococcus*, and *Rhodococcus*, play crucial roles in the degradation of crude oil <sup>[17][18]</sup>. *Acinetobacter*, *Aquabacterium* along, *Brevundimonas*, and *Pseudomonas* predominated in crude-oil-contaminated soil by bioremediating 49.3% (1756 mg/kg) of HMW crude oil components with food waste composting <sup>[19]</sup>. Similarly, *Exiguobacterium aurantiacum* and *Burkholderia cepacian* could degrade 52.93% and 51.37% of crude oil (1%) in 15 days <sup>[20]</sup>, respectively. Similarly, obligate hydrocarbonoclastic bacteria, such as *Nesiotobacter*, *Nitratireductor*, *Acinetobacter*, *Marinobacter*, *Pseudoalteromonas*, and *Sphingomonas*, have very low abundances in uncontaminated soil <sup>[21]</sup>, otherwise rising with an increase in the contamination <sup>[22][23][24]</sup>. *Sphingomonas* strain 4c can break down 87.2% (100 mg/L) of the fluorene in 7 days <sup>[25]</sup>. *Nesiotobacter exalbescens* COD22 isolated from a marine environment can degrade 92% of hydrocarbon contaminants under high-pressure conditions. However, at a normal pressure rate, the degradation was found to be slowed down by a rate of 87.5% <sup>[26]</sup>. The symbiotic association of *Talaromyces* species and *Acinetobacter baumannii* can potentially degrade a complete alkane

fraction and about 80% of the total petroleum hydrocarbon contamination within 14 days in culture conditions at a pH of 9 [27]. The bacterial functions that allow them to thrive in such extremely polluted conditions are their unique genetic features and catabolic attributes [28]. A consortium consisting of *Aeromonas hydrophila*, *Alcaligenes xylosoxidans*, *Gordonia* sp., *Pseudomonas fluorescens*, *Pseudomonas putida*, *Rhodococcus equi*, *S. maltophilia*, and *Xanthomonas* sp. was able to potentially degrade 89% of diesel oil from heavily contaminated field soil [49].

Crude oil has been an integral part of the marine environment for millions of years, and microbes use it as an energy source. PAHs are ubiquitous in the marine environment and enter through chronic or acute pollution events, such as oil spills. The fate of PAHs after entering the marine environment is determined by several factors, such as sinking, sedimentation, resuspension, volatilisation, photodegradation, and biodegradation [29]. Over 175 prokaryotic genera in 7 phyla of bacteria and archaea and a similar number of fungal genera have been identified that can use hydrocarbons as their sole carbon source [30]. Some commonly reported genera for oil degradation are *Alcanivorax*, *Cycloclasticus* [31], *Fundibacter* [32], and *Oleispira* [33]. *Cycloclasticus* can use pyrene, naphthalene, 2-methylnaphthalene, 2,6-dimethylnaphthalene, biphenyl, fluorene, acenaphthalene, dibenzofuran, phenanthrene, or anthracene as a sole carbon source [34]. A recent study in Mediterranean waters revealed that the microbial communities in pristine, unpolluted locations responded very differently than those inhabiting chronically polluted sites. The latter showed a faster degradation response when accidental oil pollution occurred [30]. Mixed bacterial species of *Pseudomonas* sp. and *Bacillus* sp. degrade 80.64% of crude oil and 76.30% of alkanes in oily wastewater.

## **2. Regulatory Factors Involve in Microbial Degradation of Crude Oil**

Microbes are primarily responsible for the biodegradation of crude oil in the contaminated environment; however, several biotic and abiotic factors also influence the efficacy of bioremediation approaches, such as soil oxygen content, soil type, pH, temperature, nutrient availability, water content, and the concentration of crude oil and its bioavailability in the existing environment [35].

The bioavailability of crude oil components in a soil environment is an essential factor in regulating biodegradation rate [36]. Though the increase in the concentration of crude oil negatively affects the degradation rate, the degradation of aromatic compounds and linear hydrocarbons occurs discretely [37]. After the rapid initial rate of crude oil degradation, the residue becomes partially diffused to the surrounding solid surfaces, reducing its bioavailability, phosphorus and nitrogen contents and degradation rate at later stages [38]. The improper disposal of crude oil with high concentrations of volatile hydrocarbons and sludge (15–20%) can harm the microbial population and hinder the biodegradation of the oil [39][40]. Temperature has multiple effects in the bioremediation process, as it influences the rate at which the crude oil is degraded by the microorganisms, affecting the composition of the whole soil microbiome [41]. The chemical and physical properties of crude oil are also affected by temperature variations. The soil–water partition coefficient decreases with increasing temperature in soil with high moisture content, and contaminant dissolution may occur [42]. Though the ability of microorganisms to degrade crude oil is negatively affected in high-pH environments [43], the pH of the soil plays a major role in the degradation of crude oil by the bacteria. It has been observed that pH 7 is optimum for the significant degradation of PAHs and other *n*-alkanes (C-7 to C-25). However, bacterial strains such as *Bacillus subtilis* BL-27 showed a wide tolerance to pH values ranging from 4 to 10 during the degradation of crude oil [44].

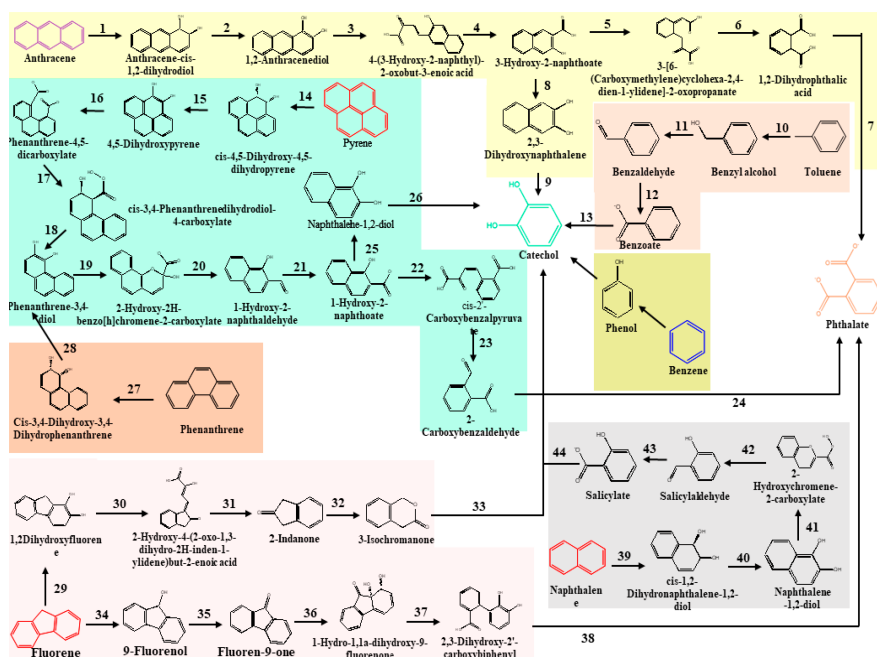
Dubinsky et al. [45] observed that during unmitigated flow, *n*-alkane and cycloalkane availability were higher, which probably helped alkane-degrading bacteria flourish. According to metagenomic and meta-transcriptomic analyses carried out on a subset of samples obtained during uncontrolled flow, alkane degradation was shown to be the predominant hydrocarbon-degrading pathway expressed when *Oceanospirillaceae* and *Pseudomonas* were prevalent in the community. Numerous investigations demonstrated that oxygen loss causes biodegradation activities in soils and marine sediments to decrease drastically [46][47][48]. There are studies that suggest the estimated oxygen requirements for aerobic hydrocarbon degradation. As 3kg O<sub>2</sub> is required for every kg of petroleum contaminants, 8.6 moles of oxygen are required for every mole of diesel to be degraded [49]. Even though a variety of microbial communities typically contribute to the *in-situ* breakdown of alkane mixtures under varying soil circumstances, many soils show a similar pattern in the types of microorganisms that respond to alkane disturbance [50]. For appropriate bioremediation experiments, factors such as expense, duration, human resource, and targets must all be considered.

## **3. Classical Metabolic Pathways Involved in Degradation of Total Petroleum Hydrocarbons**

Hydrocarbonoclastic bacterial species degrade PAHs due to their adaptability, vigour, diversity, and capacity to generate less hazardous metabolic intermediates. Microbes use the energy from the breakdown of hydrocarbons ranging from

simple alkanes to PAHs [51]. Oxygen is necessary for the ring hydroxylation, ring cleavage, and final electron uptake in the aerobic breakdown of PAHs by bacteria. However, reductive processes are the basis of anaerobic PAH consumption [52]. High-molecular-weight PAHs are harder to break down because their fused aromatic structures make them very thermodynamically stable, water-repellent, and less bioavailable. Pyrene mineralisation can take place either at the C-1 and C-2 positions or at the C-4 and C-5 positions of the aromatic ring due to the action of the dioxygenase. Following that, the ring aromatisation of the dihydrodiols and the ring cleavage dioxygenase led to the synthesis of phenanthrene dicarboxylate, which is then decarboxylated to generate carboxylate. The phenanthrene dicarboxylate is eventually converted into phenanthrene carboxylate. At this stage, a cis-3,4-dihydroxyphenanthrene-4-carboxylate is formed by a deoxygenation process. This cis-3,4-dihydroxyphenanthrene-4-carboxylate then rearomatizes to form dihydroxyphenanthrene, which is then metabolised to yield hydroxynaphthoate. In addition, the pyrene breakdown occurs via the phthalate pathway as well [53][54]. The phthalate reaction forms several intermediates, including 3,4-dihydroxybenzoate and carboxyhydroxymuconate semialdehyde, both of which eventually feed into the Krebs cycle via the multiple reaction steps. Phenanthrene is a three-ringed compound whose degradation has been reported in both Gram-positive and Gram-negative microbes [55][56]. Most of the intermediates of phenanthrene degradation are shared with the pyrene degradation pathway.

Naphthalene, toluene, and anthracene have a comparatively low molecular weight and constitute most petroleum hydrocarbons [57]. Ring hydroxylating genes, such as *nagAc*, *phnAc*, *nahAc*, *nidA*, and *pdoA*, are target genes for naphthalene breakdown. In the first stage, an oxidoreductase inserts two oxygen atoms into the naphthalene rings to generate dihydronaphthalene. The following step involves the dehydrogenase enzyme, followed by ring cleavage and oxidation [58]. The dioxygenase facilitates the cleavage of catechol and proceeds via various pathways, ultimately leading to the formation of succinyl-CoA and entering the TCA cycle [59]. **Figure 1** represents the classical pathways for the aerobic biodegradation of various hydrocarbons, such as pyrene, anthracene, phenanthrene, and naphthalene (reconstructed from the KEGG database). Microorganisms synthesised several catabolic enzymes, such as naphthalene 1,2-dioxygenase, oxidoreductases, pyrene dioxygenase, aldehyde dehydrogenase, and many others, in every step of the degradation process. The enzymes involved in the different steps of the degradation process are also given in **Figure 1**.

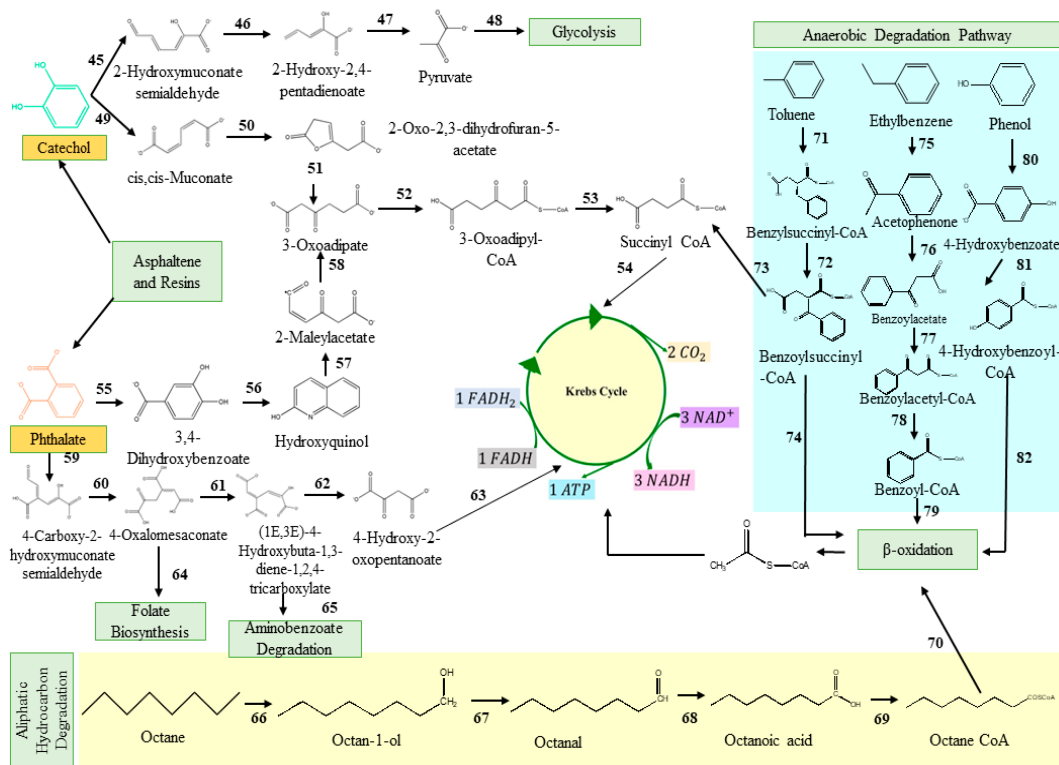


**Figure 1.** Pathways for degradation of anthracene, pyrene, phenanthrene, fluorene, toluene, naphthalene, and benzene until the formation of intermediate product catechol and phthalate. The enzymes involved in the reactions are 1. naphthalene 1,2-dioxygenase; 2. cis-1,2-dihydro-1,2-dihydroxynaphthalene dehydrogenase; 3. anthracene-1,2-diol 1,2-dioxygenase; 4. Dioxygenase; 5. 3-hydroxy-2-naphthoate 2,3-dioxygenase; 6. Oxidoreductases; 7. Oxidoreductases; 8. 3-hydroxy-2-naphthoate 2,3-dioxygenase; 9. chlorobenzene dihydrodiol dehydrogenase; 10. toluene monooxygenase system protein A; 11. aryl-alcohol dehydrogenase; 12. benzaldehyde dehydrogenase; 13. dihydroxycyclohexadiene carboxylate dehydrogenase; 14. pyrene dioxygenase; 15. dibenzothiophene dihydrodiol dehydrogenase; 16. 4,5-dihydroxypyrene dioxygenase; 17. phenanthrene-4,5-dicarboxylate decarboxylase; 18. phenanthrene-4-carboxylate dioxygenase; 19. extradiol dioxygenase; 20. hydratase-aldolase; 21. aldehyde dehydrogenase; 22. 1-hydroxy-2-naphthoate dioxygenase; 23. 4-(2-carboxyphenyl)-2-oxobut-3-enoate aldolase; 24. 2-formylbenzoate dehydrogenase; 25. salicylate hydroxylase; 26. Nah C; 27. PAH dioxygenase; 28. cis-3,4-dihydrophenanthrene-3,4-diol dehydrogenase; 29. Hydroxylases; 30. 1,2-dihydroxyfluorene 1,1a-dioxygenase; 31. Alcohol dehydrogenase; 32. 2-indanone

monooxygenase; 33. 34. naphthalene 1,2-dioxygenase; 35. fluorene-9-ol dehydrogenase; 36. dibenzofuran dioxygenase; 37. 1,1a-dihydroxy-1-hydro-9-fluorenone dehydrogenase; 38. 2-hydroxy-6-oxo-6-(2'-carboxyphenyl)-hexa-2,4-dienoate hydrolase; 39. Nah Ac; 40. Nah B; 41. Nah C; 42. Nah E; 43. Nah F; 44. Salicylate hydroxylase.

The solubility of alkanes in water is low and decreases with increasing molecular weight [60]. In the case of *n*-alkanes with two or more carbon atoms, aerobic degradation often begins with the oxidation of a terminal methyl group to produce a primary alcohol, which is then oxidised to the corresponding aldehyde and transformed into a fatty acid. The conjugation of fatty acids to CoA is followed by  $\beta$ -oxidation to create acetyl-CoA [61]. Some examples of regulators studied from different families are *LuxR/MalT*, *AraC/XylS*, etc. [62][63]. Asphaltenes and resins are aromatic compounds with long chains of alkyl groups that dissolve in solvents, such as *n*-heptane and *n*-pentane [64]. These components of crude petroleum are very recalcitrant in nature, and the exact intermediates formed during the degradation process are still under study [65].

The initial activation of hydrocarbons is essential for anaerobic biodegradation, and the overall enzymatic reactions involve the addition of fumarate, catalysed by a glycol radical enzyme to yield aromatic-substituted succinates; the methylation of unsubstituted aromatics followed by the hydroxylation of an alkyl substituent via a dehydrogenase; and direct carboxylation, which may represent previous reactions. After being produced via pathways that begin with these activation reactions and end in ring saturation,  $\beta$ -oxidation, and ring cleavage reactions, with core metabolites, such as benzoyl-CoA, eventually being integrated into the biomass or completely oxidised [36][66]. **Figure 2** represents the aerobic and anaerobic degradation steps of asphaltenes, resins, and other aromatic components of crude oil (reconstructed from the KEGG database).



**Figure 2.** Terminal degradation steps of PAHs following benzoate degradation pathway via catechol and phthalate intermediates. Aliphatic hydrocarbon degradation pathway and anaerobic degradation pathway were also given in separate colour, with both ultimately joining the common carbon metabolism pathway. The numbers representing specific enzymes viz 45. catechol 2,3-dioxygenase 46. 2-hydroxymuconate-semialdehyde hydrolase 47. 4-hydroxy 2-oxovalerate aldolase 48. Common carbon pathway's enzymes 49. catechol 1,2-dioxygenase 50. muconolactone D-isomerase; 51. 3-oxoadipate enol-lactonase; 52. 3-oxoadipate CoA-transferase; 53. acetyl-CoA acyltransferase; 54. Succinyl CoA Synthetase; 55. Monooxygenase; 56. 4-hydroxybenzoate 1-hydroxylase; 57. hydroxyquinol 1,2-dioxygenase; 58. maleylacetate reductase; 59. protocatechuate 4,5-dioxygenase; 60. 2-pyrone-4,6-dicarboxylate lactonase; 61. 4-oxalomesaconate tautomerase; 62. 4-hydroxy-4-methyl-2-oxoglutarate aldolase; 63. Citrate synthetase; 64. Folate synthesis enzymes; 65. Aminobenzoate degradation enzymes; 66. Alkane monooxygenase; 67. Alcohol dehydrogenase; 68. Aldehyde dehydrogenase; 69. Synthetase 70. CoA synthetase; 71–82. Unidentified.

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