

Limosilactobacillus reuteri Fermented Brown Rice

Subjects: Biology

Contributor: Deog-Hwan Oh

Oxidative stress has been postulated to play a role in several diseases, including cardiovascular diseases, diabetes, and stress-related disorders (anxiety/depression). Presently, natural plant-derived phytochemicals are an important tool in reducing metabolomic disorders or for avoiding the side effects of current medicinal therapies. Brown Rice (*Oryza sativa* L.) is an important part of Asian diets reported as a rich source of bioactive phytonutrients.

Keywords: brown rice ; fermentation ; antioxidants ; oxidative stress ; untargeted metabolomics ; UHPLC-QTOF/MS ; health benefits

1. Introduction

Oxidative stress is a condition that is caused by an imbalance between antioxidants and free radicals of living organisms. This imbalance occurs due to the excessive production of reactive oxygen species (ROS) or antioxidant deficiency that leads to the damage of aerobic organisms as well as chronic inflammation; referred to as oxidative stress ^[1]. Lower ROS concentration is important for normal cellular signalling, while excess ROS can cause oxidative damage to DNA, lipids, proteins, and is associated with several chronic diseases ^{[2][3]}. The current definition of oxidative stress includes metabolic stress-related pathways that participate in both cellular and extracellular metabolic events. The biology of oxidative stress is extremely complex, with multiple mechanisms at work ^[4]. Regardless of the mechanism, oxidative stress causes the onset of many diseases including cardiovascular diseases, diabetes, and anxiety or depression which are considered a major public health issue worldwide. As a result, consuming antioxidants to prevent oxidative stress is becoming important for health. Moreover, an increase in the health-consciousness of consumers has increased the demand for nutritional and disease-preventing functional foods, probiotics, prebiotics, and postbiotics. Numerous studies have focused on probiotics, specifically *Lactobacilli* strains, that have the potential to act as antioxidants to protect the host from oxidative stress ^[5]. Some *Lactobacilli* strains have been found to quench oxygen free radicals using a chemical antioxidant method.

Many studies have been reported that phytochemicals (e.g., polyphenols and phenolic acids) derived from natural plants have the potential to target oxidative stress and inflammatory pathways ^{[6][7]}. Rice is a staple food (in many countries) that belongs to the grass family (*Oryza sativa*). The total worldwide production of rice was about 769,657,791 tonnes in an area of 167,249,103 ha. Epidemiological studies have shown that the low incidence of chronic diseases in rice-consuming regions can be correlated with rice antioxidants ^{[8][9]}. The antioxidant activity and phytochemical content of brown rice (BR) have been recorded in several studies. Components such as γ -oryzanol, phenolic acids, gamma-aminobutyric acid (GABA), flavonoids, and γ -tocotrienol contribute to the health-promoting properties of brown rice ^[10].

Evidence supports the effect of solid-state fermentation (SSF) techniques using lactic acid bacteria (LABs) and fungal strains on antioxidant levels and bioactive properties in a variety of substrates, including barley ^[11], pearl millet ^[12], and rice ^[2]. Many researchers, food scientists, and industrialists use the SSF process to enhance the nutritional quality of food and food products. Biological methods are environmentally friendly, relatively safe, and rely on the use of appropriate and specific microorganisms ^[13]. Our study aimed to provide knowledge to quantify the quality of these phytochemical antioxidants in whole brown rice to meet the needs of food producers and consumers of rice: (1) to analyze the antioxidant properties of differently fermented brown (FBR) rice over raw brown rice (BR); (2) detection of bioactive compounds in raw BR and different LABs fermented brown rice using ultra-high-performance liquid chromatography quadrupole time-of-flight mass spectrometry (UHPLC-QTOF/MS), and (3) detection of cellular antioxidant activity of the best LAB fermenting bacterial strain (*L. reuteri* FBR).

2. Untargeted Metabolomics Using UHPLC Q-TOF-MS/MS in Brown Rice Samples

UHPLC Q-TOF-MS/MS detection is considered a gold standard technique for the precise detection and quantification of a wide variety of components. Therefore, in this study, we have used this detection technique for the identification of

phenolic compounds in brown rice.

2.1. Phenolic Compounds

In the present research, the phenolic compositions of BR treated with different fermentation bacteria were selected and positively or tentatively identified by UHPLCQ-TOF-MS/MS. Phenolic identification and characterization were achieved by comparing our results with mass spectral literature evidence and cross-referencing it with other available spectral databases, such as Metlin and Metabolomics Workbench. A total of 15 phenolic compounds were tentatively found from our soluble extracts of raw BR, *L. reuteri* FBR, *L. fermentum* FBR, and *L. plantarum* FBR respectively, as shown in **Table 1**. In the ethanol extract, we identified compounds 1 to 14 at different adduct charges $[M - H]^-$ and $[M + H]^+$ which are identified by comparing with mass spectral libraries, XCMS online (Metlin), and Metabolomics Workbench. Heat map analysis was used for clustering phenolic compounds based on their concentrations (**Figure 1**) where the colour scheme from blue to red shows concentration in decreasing order.

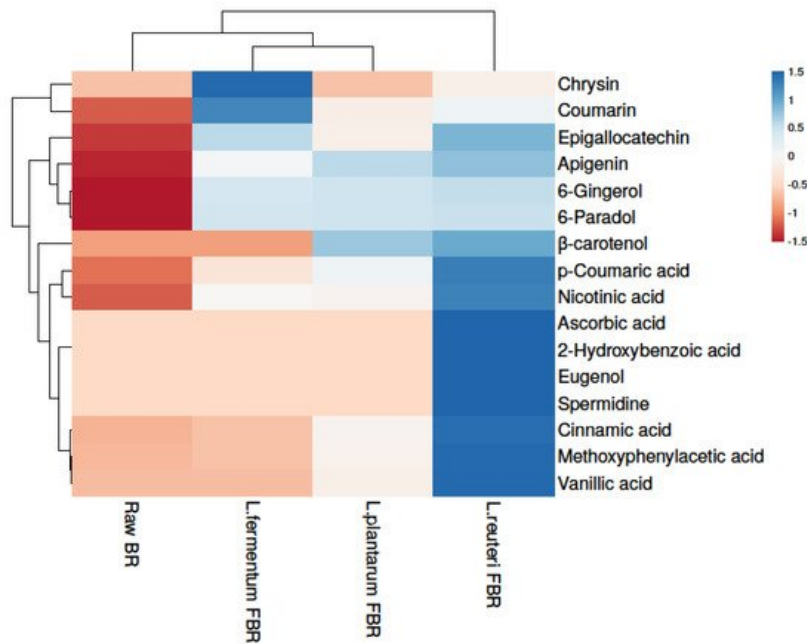


Figure 1. Heat map showing levels of phenolic compounds in raw and LABs fermented BR samples.

Table 1. Phenolic compounds detected in raw and LABs fermented BR.

S. No	Sample Name	Retention Time (min)	Peak Area	Adduct/Charge	Precursor Mass	Found at Mass	Molecular Formula	Tentative Phenolic Compound
1	Raw BR	Nd	Nd	$[M - H]^-$	Nd	Nd		
	<i>L. plantarum</i> FBR	45.56	4.53×10^5	$[M - H]^-$	353.268	353.187		
	<i>L. fermentum</i> FBR	Nd	Nd	$[M - H]^-$	Nd	Nd	$C_{25}H_{36}O$	Beta-carotenol
	<i>L. reuteri</i> FBR	45.45	5.19×10^5	$[M - H]^-$	353.268	353.284		
2	Raw BR	Nd	Nd	$[M + H]^+$	Nd	Nd		
	<i>L. plantarum</i> FBR	Nd	Nd	$[M + H]^+$	Nd	Nd		
	<i>L. fermentum</i> FBR	Nd	Nd	$[M + H]^+$	Nd	Nd	$C_{10}H_{12}O_2$	Eugenol
	<i>L. reuteri</i> FBR	20.81	2.24×10^5	$[M + H]^+$	179.107	179.1067		

S. No	Sample Name	Retention Time (min)	Peak Area	Adduct/Charge	Precursor Mass	Found at Mass	Molecular Formula	Tentative Phenolic Compound
3	Raw BR	33.80	2.20 × 10 ⁵	[M – H]–	293.177	293.1761	C ₁₇ H ₂₆ O ₄	6-Gingerol
	<i>L. plantarum</i> FBR	33.80	2.08 × 10 ⁶	[M – H]–	293.177	293.176		
	<i>L. fermentum</i> FBR	33.81	2.03 × 10 ⁶	[M – H]–	293.177	293.1761		
	<i>L. reuteri</i> FBR	33.80	2.13 × 10 ⁶	[M – H]–	293.177	293.1762		
4	Raw BR	Nd	Nd	[M + H]+	Nd	Nd	C ₁₅ H ₁₀ O ₄	Chrysin
	<i>L. plantarum</i> FBR	Nd	Nd	[M + H]+	Nd	Nd		
	<i>L. fermentum</i> FBR	14.78	4.09 × 10 ⁵	[M + H]+	253.052	253.0524		
	<i>L. reuteri</i> FBR	14.81	1.01 × 10 ⁵	[M + H]+	253.052	253.0527		
5	Raw BR	Nd	Nd	[M + H]+	Nd	Nd	C ₁₆ H ₈ N ₂ O ₅	Apigenin
	<i>L. plantarum</i> FBR	14.79	4.68 × 10 ⁵	[M + H]+	269.047	269.0457		
	<i>L. fermentum</i> FBR	14.78	3.38 × 10 ⁵	[M + H]+	269.047	269.0457		
	<i>L. reuteri</i> FBR	14.78	5.15 × 10 ⁵	[M + H]+	269.047	269.0458		
6	Raw BR	Nd	Nd	[M + H]+	Nd	Nd	C ₉ H ₆ O ₂	Coumarin
	<i>L. plantarum</i> FBR	1.92	1.24 × 10 ⁵	[M + H]+	147.044	147.0444		
	<i>L. fermentum</i> FBR	1.87	2.94 × 10 ⁵	[M + H]+	147.044	147.0447		
	<i>L. reuteri</i> FBR	1.90	1.55 × 10 ⁵	[M + H]+	147.044	147.0445		
7	Raw BR	Nd	Nd	[M + H]+	Nd	Nd	C ₁₅ H ₁₄ O ₇	Epigallocatechin
	<i>L. plantarum</i> FBR	12.26	7.44 × 10 ⁵	[M + H]+	305.071	305.067		
	<i>L. fermentum</i> FBR	12.28	1.20 × 10 ⁶	[M + H]+	305.071	305.067		
	<i>L. reuteri</i> FBR	12.28	1.38 × 10 ⁶	[M + H]+	305.071	305.067		
8	Raw BR	Nd	Nd	[M + H]+	Nd	Nd	C ₇ H ₁₉ N ₃	Spermidine
	<i>L. plantarum</i> FBR	Nd	Nd	[M + H]+	Nd	Nd		
	<i>L. fermentum</i> FBR	Nd	Nd	[M + H]+	Nd	Nd		
	<i>L. reuteri</i> FBR	0.96	8.75 × 10 ⁵	[M + H]+	188.176	188.1761		

S. No	Sample Name	Retention Time (min)	Peak Area	Adduct/Charge	Precursor Mass	Found at Mass	Molecular Formula	Tentative Phenolic Compound
9	Raw BR	38.06	3.21 × 10 ⁴	[M – H]–	277.182	277.1812	C ₁₇ H ₂₆ O ₃	6-Paradol
	<i>L. plantarum</i> FBR	38.06	4.38 × 10 ⁵	[M – H]–	277.182	277.1812		
	<i>L. fermentum</i> FBR	38.08	4.33 × 10 ⁵	[M – H]–	277.182	277.1813		
	<i>L. reuteri</i> FBR	38.06	4.42 × 10 ⁵	[M – H]–	277.182	277.1812		
10	Raw BR	Nd	Nd	[M – H]–	Nd	Nd	C ₉ H ₈ O ₂	Cinnamic acid
	<i>L. plantarum</i> FBR	4.01	3.26 × 10 ⁵	[M – H]–	147.046	147.0455		
	<i>L. fermentum</i> FBR	4.03	4.46 × 10 ⁴	[M – H]–	147.046	147.0456		
	<i>L. reuteri</i> FBR	3.98	1.09 × 10 ⁶	[M – H]–	147.046	147.0454		
11	Raw BR	Nd	Nd	[M + NH ₄]+	Nd	Nd	C ₉ H ₈ O ₃	p-Coumaric acid
	<i>L. plantarum</i> FBR	1.86	1.06 × 10 ⁶	[M + NH ₄]+	182.081	182.0813		
	<i>L. fermentum</i> FBR	1.92	6.90 × 10 ⁵	[M + NH ₄]+	182.081	182.0813		
	<i>L. reuteri</i> FBR	1.87	2.13 × 10 ⁶	[M + NH ₄]+	182.081	182.0812		
12	Raw BR	Nd	Nd	[M – H]–	Nd	Nd	C ₉ H ₁₀ O ₃	Methoxyphenylacetic acid
	<i>L. plantarum</i> FBR	15.28	5.20 × 10 ⁶	[M – H]–	165.057	165.0558		
	<i>L. fermentum</i> FBR	15.29	4.09 × 10 ⁵	[M – H]–	165.057	165.056		
	<i>L. reuteri</i> FBR	15.27	1.92 × 10 ⁷	[M – H]–	165.057	165.0557		
13	Raw BR	Nd	Nd	[M – H]–	Nd	Nd	C ₇ H ₆ O ₃	Sesamol/2-Hydroxybenzoic acid
	<i>L. plantarum</i> FBR	Nd	Nd	[M – H]–	Nd	Nd		
	<i>L. fermentum</i> FBR	Nd	Nd	[M – H]–	Nd	Nd		
	<i>L. reuteri</i> FBR	19.63	3.24 × 10 ⁵	[M – H]–	137.025	137.0249		
14	Raw BR	Nd	Nd	[M – H]–	Nd	Nd	C ₈ H ₈ O	Vanillic acid
	<i>L. plantarum</i> FBR	15.28	3.18 × 10 ⁵	[M – H]–	119.051	119.0504		
	<i>L. fermentum</i> FBR	Nd	Nd	[M – H]–	Nd	Nd		
	<i>L. reuteri</i> FBR	15.27	1.27 × 10 ⁶	[M – H]–	119.051	119.0504		

S. No	Sample Name	Retention Time (min)	Peak Area	Adduct/Charge	Precursor Mass	Found at Mass	Molecular Formula	Tentative Phenolic Compound
15	Raw BR	Nd	Nd	Nd	Nd	Nd	C ₆ H ₈ O ₆	Ascorbic acid (Vitamin C)
	<i>L. plantarum</i> FBR	Nd	Nd	Nd	Nd	Nd		
	<i>L. fermentum</i> FBR	Nd	Nd	Nd	Nd	Nd		
	<i>L. reuteri</i> FBR	1.02	3.05 × 10 ³	[M + H] ⁺	209.009	209.0107		

Nd—not detected, BR—brown rice, and FBR—fermented brown rice.

Results showed that the highest phenolic compounds were detected in the *L. reuteri* FBR sample. Because phenolic compounds are not readily available, they typically occur in cereals in esterified linkages to the cereal wall matrix [14]. Fermentation is considered to be a possible strategy to release insoluble or bound phenolic compounds and thus leading to improve the poor bioavailability of grain phenolics. Comparing different fermenting bacteria in the present study we found that *L. reuteri* fermentation releases most of the phenolic compounds compared with other bacterial strains and thus improves the bioavailability and bioaccessibility of cereal grains such as brown rice phenolics [15]. Many phenolic compounds detected in the current study such as p-coumaric acid [16], ascorbic acid [17], cinnamic acid [18], and vanillic acid [19] are already reported in the literature for their strong antioxidant capacities.

2.2. Levels of Amino Acid in Brown Rice

In the growth and development of organisms, amino acids play an important role and can also improve the taste of food. In our present study, a total of 18 amino acids were detected in raw and differently fermented BR samples (**Figure 2** and **Table 2**) which shows statistically significant differences from each other after comparing their levels. Raw BR contained the least number of amino acids, which may be due to more bound molecules with the parent, whereas fermentation leads to an increase in amino acid content. The levels of amino acids were detected highest in the *L. reuteri* FBR sample which might be strain-specific as fermentation microorganisms produce enzymes that lead to the formation of several metabolites and bioactive compounds from the food matrix [20]. In the ethanol extract, we found levels of some essential amino acids (tryptophan, lysine, methionine, and histidine), as well as certain conditionally essential amino acids (arginine, ornithine, serine, and glutamine), increased drastically after fermentation (**Figure 2** and **Table 2**). The identification was done by comparing with mass spectral libraries, XCMS online (Metlin) and Metabolomics Workbench. In amino acids, *L. reuteri* FBR also shows the highest number of amino acids content as observed in phenolic compounds.

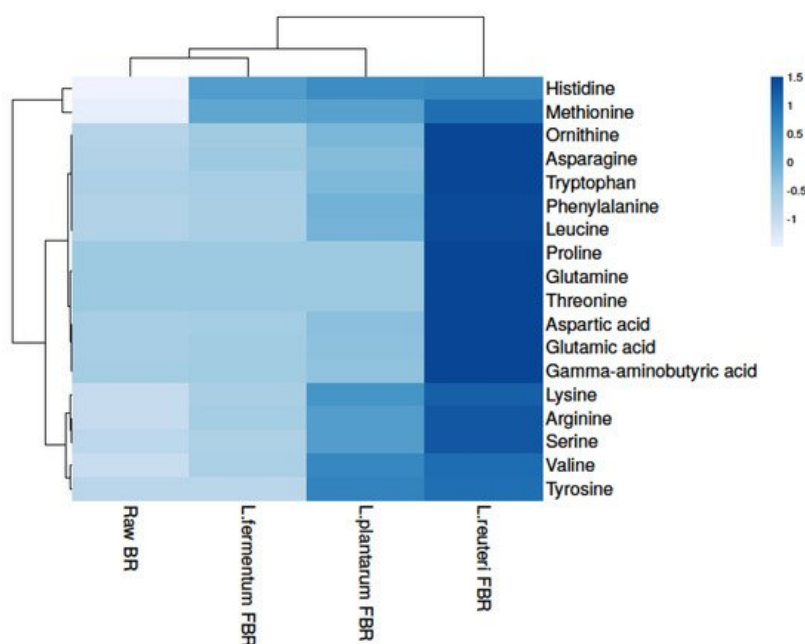


Figure 2. Heat map showing levels of amino acids in raw and different fermented BR samples.

Table 2. Amino acids detected in raw and LABs fermented brown rice.

S. No	Sample Name	Retention Time (Min)	Peak Area	Adduct/ Charge	Precursor Mass	Found at Mass	Formula Finder Result	Amino Acid
1	Raw BR	1.00	1.52 × 10 ³	[M + H] ⁺	156.077	156.077	C ₆ H ₉ N ₃ O ₂	Histidine
	<i>L. plantarum</i> FBR	1.00	7.31 × 10 ⁵	[M + H] ⁺	156.077	156.077		
	<i>L. fermentum</i> FBR	1.02	6.33 × 10 ⁵	[M + H] ⁺	156.077	156.0771		
	<i>L. reuteri</i> FBR	1.02	7.45 × 10 ⁵	[M + H] ⁺	156.077	156.0771		
2	Raw BR	ND	ND	[M – H] –	ND	ND	C ₆ H ₁₄ N ₂ O ₂	Lysine
	<i>L. plantarum</i> FBR	1.02	9.16 × 10 ⁵	[M – H] –	145.099	145.0982		
	<i>L. fermentum</i> FBR	1.02	2.17 × 10 ⁵	[M – H] –	145.099	145.0984		
	<i>L. reuteri</i> FBR	1.02	1.40 × 10 ⁶	[M – H] –	145.099	145.0983		
3	Raw BR	ND	ND	[M + H] ⁺	ND	ND	C ₅ H ₁₁ NO ₃ S	Methionine
	<i>L. plantarum</i> FBR	1.17	5.66 × 10 ⁵	[M + H] ⁺	166.053	166.0536		
	<i>L. fermentum</i> FBR	1.17	5.33 × 10 ⁵	[M + H] ⁺	166.053	166.0537		
	<i>L. reuteri</i> FBR	1.18	8.41 × 10 ⁵	[M + H] ⁺	166.053	166.0536		
4	Raw BR	1.17	2.56 × 10 ²	[M – H] –	146.047	146.0457	C ₅ H ₉ NO ₄	Glutamic acid
	<i>L. plantarum</i> FBR	1.47	3.18 × 10 ⁵	[M – H] –	146.047	146.046		
	<i>L. fermentum</i> FBR	1.47	4.58 × 10 ⁴	[M – H] –	146.047	146.046		
	<i>L. reuteri</i> FBR	1.47	2.65 × 10 ⁶	[M – H] –	146.047	146.0458		
5	Raw BR	ND	ND	[M + H] –	ND	ND	C ₄ H ₉ NO ₂	Gamma-aminobutyric acid
	<i>L. plantarum</i> FBR	1.16	1.38 × 10 ⁵	[M – H] –	102.057	102.056		
	<i>L. fermentum</i> FBR	1.16	2.76 × 10 ⁴	[M – H] –	102.057	102.0563		
	<i>L. reuteri</i> FBR	1.16	1.28 × 10 ⁶	[M – H] –	102.057	102.0561		
6	Raw BR	ND	ND	[M + H] ⁺	ND	ND	C ₆ H ₁₄ N ₄ O ₂	Arginine
	<i>L. plantarum</i> FBR	1.11	2.81 × 10 ⁶	[M + H] ⁺	175.118	175.1183		
	<i>L. fermentum</i> FBR	1.14	9.09 × 10 ⁵	[M + H] ⁺	175.118	175.1194		
	<i>L. reuteri</i> FBR	1.11	4.99 × 10 ⁶	[M + H] ⁺	175.118	175.1184		

S. No	Sample Name	Retention Time (Min)	Peak Area	Adduct/ Charge	Precursor Mass	Found at Mass	Formula Finder Result	Amino Acid
7	Raw BR	ND	ND	$[M - H]^-$	ND	ND	$C_5H_{11}NO_2$	Valine
	<i>L. plantarum</i> FBR	1.49	6.15×10^5	$[M - H]^-$	116.073	116.0717		
	<i>L. fermentum</i> FBR	1.50	1.37×10^5	$[M - H]^-$	116.073	116.0718		
	<i>L. reuteri</i> FBR	1.45	7.61×10^5	$[M - H]^-$	116.073	116.0719		
8	Raw BR	1.14	1.19×10^3	$[M - H]^-$	132.031	132.0307	$C_4H_7NO_4$	Aspartic acid
	<i>L. plantarum</i> FBR	1.15	7.26×10^4	$[M - H]^-$	132.031	132.0303		
	<i>L. fermentum</i> FBR	1.15	8.76×10^3	$[M - H]^-$	132.031	132.0305		
	<i>L. reuteri</i> FBR	1.12	4.89×10^5	$[M - H]^-$	132.031	132.0302		
9	Raw BR	ND	ND	$[M - H]^-$	ND	ND	$C_9H_{11}NO_2$	Phenylalanine
	<i>L. plantarum</i> FBR	4.01	2.26×10^6	$[M - H]^-$	164.072	164.0718		
	<i>L. fermentum</i> FBR	4.03	3.50×10^5	$[M - H]^-$	164.072	164.072		
	<i>L. reuteri</i> FBR	3.98	7.21×10^6	$[M - H]^-$	164.072	164.0718		
10	Raw BR	ND	ND	$[M - H]^-$	ND	ND	$C_5H_{12}N_2O_2$	Ornithine
	<i>L. plantarum</i> FBR	1.01	1.53×10^5	$[M - H]^-$	131.084	131.0827		
	<i>L. fermentum</i> FBR	1.01	6.67×10^4	$[M - H]^-$	131.084	131.0828		
	<i>L. reuteri</i> FBR	1.01	5.35×10^5	$[M - H]^-$	131.084	131.0827		
11	Raw BR	1.12	3.31×10^2	$[M - H]^-$	104.036	104.0353	$C_3H_7NO_3$	Serine
	<i>L. plantarum</i> FBR	1.12	3.06×10^5	$[M - H]^-$	104.036	104.0353		
	<i>L. fermentum</i> FBR	1.13	4.70×10^4	$[M - H]^-$	104.036	104.0356		
	<i>L. reuteri</i> FBR	1.12	5.73×10^5	$[M - H]^-$	104.036	104.0353		
12	Raw BR	ND	ND	$[M - H]^-$	ND	ND	$C_6H_{13}NO_2$	Leucine
	<i>L. plantarum</i> FBR	2.46	1.95×10^6	$[M - H]^-$	130.088	130.0874		
	<i>L. fermentum</i> FBR	2.48	3.24×10^5	$[M - H]^-$	130.088	130.0875		
	<i>L. reuteri</i> FBR	2.41	6.63×10^6	$[M - H]^-$	130.088	130.0875		

S. No	Sample Name	Retention Time (Min)	Peak Area	Adduct/ Charge	Precursor Mass	Found at Mass	Formula Finder Result	Amino Acid
13	Raw BR	ND	ND	$[M - H]^-$	ND	ND	$C_5H_{10}N_2O_3$	Glutamine
	<i>L. plantarum</i> FBR	ND	ND	$[M - H]^-$	ND	ND		
	<i>L. fermentum</i> FBR	ND	ND	$[M - H]^-$	ND	ND		
	<i>L. reuteri</i> FBR	1.10	1.62×10^4	$[M - H]^-$	145.063	145.0619		
14	Raw BR	ND	ND	$[M - H]^-$	ND	ND	$C_9H_{11}NO_3$	Tyrosine
	<i>L. plantarum</i> FBR	1.98	3.38×10^5	$[M - H]^-$	180.068	180.0667		
	<i>L. fermentum</i> FBR	ND	ND	$[M - H]^-$	ND	ND		
	<i>L. reuteri</i> FBR	1.89	4.08×10^5	$[M - H]^-$	180.068	180.0667		
15	Raw BR	ND	ND	$[M - H]^-$	ND	ND	$C_4H_9NO_3$	Threonine
	<i>L. plantarum</i> FBR	ND	ND	$[M - H]^-$	ND	ND		
	<i>L. fermentum</i> FBR	ND	ND	$[M - H]^-$	ND	ND		
	<i>L. reuteri</i> FBR	1.14	1.14×10^5	$[M - H]^-$	118.052	118.051		
16	Raw BR	ND	ND	$[M - H]^-$	ND	ND	$C_4H_8N_2O_3$	Asparagine
	<i>L. plantarum</i> FBR	1.10	6.8×10^4	$[M - H]^-$	131.047	131.0462		
	<i>L. fermentum</i> FBR	1.13	3.41×10^4	$[M - H]^-$	131.047	131.0464		
	<i>L. reuteri</i> FBR	1.13	3.01×10^5	$[M - H]^-$	131.047	131.0461		
17	Raw BR	ND	ND	$[M - H]^-$	ND	ND	$C_{11}H_{12}N_2O_2$	Tryptophan
	<i>L. plantarum</i> FBR	7.64	6.19×10^5	$[M - H]^-$	203.084	203.0829		
	<i>L. fermentum</i> FBR	7.66	7.77×10^4	$[M - H]^-$	203.084	203.0832		
	<i>L. reuteri</i> FBR	7.61	2.78×10^6	$[M - H]^-$	203.084	203.0829		
18	Raw BR	ND	ND	$[M + H]^+$	ND	ND	$C_5H_9NO_2$	Proline
	<i>L. plantarum</i> FBR	ND	ND	$[M + H]^+$	ND	ND		
	<i>L. fermentum</i> FBR	ND	ND	$[M + H]^+$	ND	ND		
	<i>L. reuteri</i> FBR	0.86	7.90×10^5	$[M + H]^+$	116.07	116.0704		

ND—not detected, BR—brown rice, and FBR—fermented brown rice.

2.3. Level of Fatty Acid in Brown Rice Samples

In particular, fermentation has been proposed as a tool for enhancing foods' nutritional values, both in terms of enhanced bioavailability of bioactive components as well as the production of health-promoting end-products. Due to their proven benefit, in the last decade, short-chain fatty acids (SCFAs) have emerged as some of the most researched compounds. In the present study, 13 fatty acids were detected in raw and different LABs fermented BR samples (**Table 3**) and fatty acid levels were found to be significantly different in all samples. The results show that the highest levels of fatty acids were found in *L. reuteri* FBR. Heat map analysis was used for separating fatty acids based on the different concentrations, represented in different shades of green (dark to light) in decreasing order (**Figure 3**).

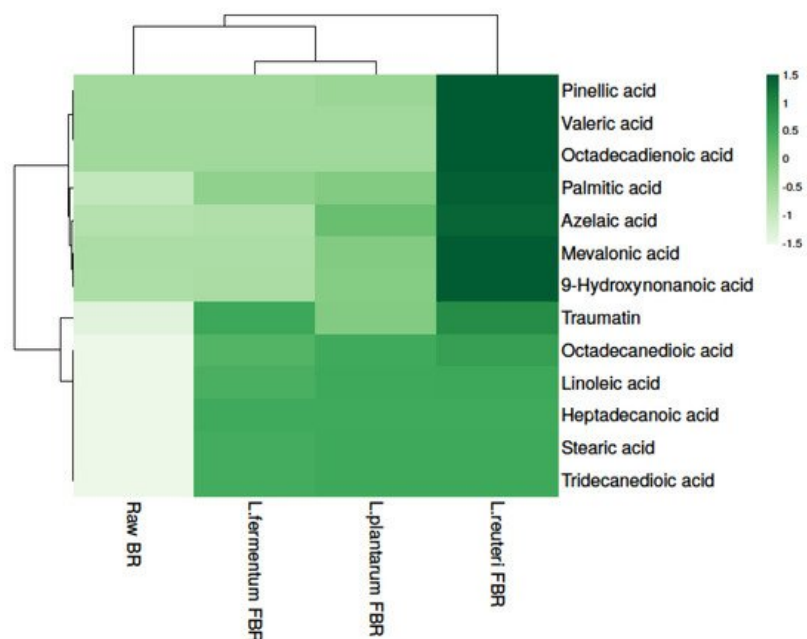


Figure 3. Heat map showing levels of fatty acids in raw and different LABs fermented BR samples.

Table 3. Fatty acids detected in raw and LABs fermented brown rice.

S. No	Sample Name	Retention Time	Peak Area	Adduct/Charge	Precursor Mass	Found at Mass	Formula Finder Result	Fatty Acid
1	Raw BR	39.97	1.91 × 10 ²	[M – H]–	255.234	255.2331	C ₁₆ H ₃₂ O ₂	Palmitic Acid
	<i>L. plantarum</i> FBR	1.94	5.15 × 10 ³	[M – H]–	255.234	255.2331		
	<i>L. fermentum</i> FBR	28.03	4.28 × 10 ³	[M – H]–	255.234	255.2332		
	<i>L. reuteri</i> FBR	23.73	1.53 × 10 ⁴	[M – H]–	255.234	255.2332		
2	Raw BR	ND	ND	[M + H]+	ND	ND	C ₅ H ₁₀ O ₂	Valeric acid
	<i>L. plantarum</i> FBR	ND	ND	[M + H]+	ND	ND		
	<i>L. fermentum</i> FBR	ND	ND	[M + H]+	ND	ND		
	<i>L. reuteri</i> FBR	22.88	1.02 × 10 ³	[M + H]+	185.066	185.0663		
3	Raw BR	46.24	5.29 × 10 ³	[M – H]–	279.234	279.2332	C ₁₈ H ₃₂ O ₂	Linoleic Acid
	<i>L. plantarum</i> FBR	46.26	6.56 × 10 ⁵	[M – H]–	279.234	279.2332		
	<i>L. fermentum</i> FBR	46.24	6.16 × 10 ⁵	[M – H]–	279.234	279.2334		
	<i>L. reuteri</i> FBR	46.25	6.64 × 10 ⁵	[M – H]–	279.234	279.2334		

S. No	Sample Name	Retention Time	Peak Area	Adduct/Charge	Precursor Mass	Found at Mass	Formula Finder Result	Fatty Acid
4	Raw BR	47.28	1.45 × 10 ⁵	[M + H] ⁺	271.264	271.2637	C ₁₇ H ₃₄ O ₂	Heptadecanoic acid
	<i>L. plantarum</i> FBR	47.28	1.44 × 10 ⁶	[M + H] ⁺	271.264	271.2636		
	<i>L. fermentum</i> FBR	47.26	1.44 × 10 ⁶	[M + H] ⁺	271.264	271.2637		
	<i>L. reuteri</i> FBR	47.27	1.45 × 10 ⁶	[M + H] ⁺	271.264	271.2638		
5	Raw BR	27.69	5.28 × 10 ³	[M – H] [–]	283.265	283.2644	C ₁₈ H ₃₆ O ₂	Stearic acid
	<i>L. plantarum</i> FBR	49.10	1.23 × 10 ⁶	[M – H] [–]	283.265	283.2645		
	<i>L. fermentum</i> FBR	49.09	1.20 × 10 ⁶	[M – H] [–]	283.265	283.2644		
	<i>L. reuteri</i> FBR	49.10	1.24 × 10 ⁶	[M – H] [–]	283.265	283.2645		
6	Raw BR	34.57	3.18 × 10 ⁴	[M – H] [–]	243.161	243.1605	C ₁₃ H ₂₄ O ₄	Tridecanedioic acid
	<i>L. plantarum</i> FBR	34.58	2.98 × 10 ⁵	[M – H] [–]	243.161	243.1606		
	<i>L. fermentum</i> FBR	34.58	2.90 × 10 ⁵	[M – H] [–]	243.161	243.1604		
	<i>L. reuteri</i> FBR	34.57	3.01 × 10 ⁵	[M – H] [–]	243.161	243.1605		
7	Raw BR	ND	ND	[M + H] ⁺	ND	ND	C ₁₂ H ₂₀ O ₃	Traumatin
	<i>L. plantarum</i> FBR	32.80	2.64 × 10 ⁵	[M + H] ⁺	213.149	213.1491		
	<i>L. fermentum</i> FBR	32.82	4.26 × 10 ⁵	[M + H] ⁺	213.149	213.1492		
	<i>L. reuteri</i> FBR	32.82	5.20 × 10 ⁵	[M + H] ⁺	213.149	213.1492		
8	Raw BR	ND	ND	[M – H] [–]	ND	ND	C ₁₈ H ₃₂ O ₅	Octadecadienoic acid/Corchorifatty acid F
	<i>L. plantarum</i> FBR	ND	ND	[M – H] [–]	ND	ND		
	<i>L. fermentum</i> FBR	ND	ND	[M – H] [–]	ND	ND		
	<i>L. reuteri</i> FBR	30.87	4.08 × 10 ⁵	[M – H] [–]	327.219	327.2181		
9	Raw BR	ND	ND	[M – H] [–]	ND	ND	C ₆ H ₁₂ O ₄	Mevalonic Acid
	<i>L. plantarum</i> FBR	3.49	7.39 × 10 ⁴	[M – H] [–]	147.067	147.0667		
	<i>L. fermentum</i> FBR	ND	ND	[M – H] [–]	ND	ND		
	<i>L. reuteri</i> FBR	3.47	3.13 × 10 ⁵	[M – H] [–]	147.067	147.0667		

S. No	Sample Name	Retention Time	Peak Area	Adduct/Charge	Precursor Mass	Found at Mass	Formula Finder Result	Fatty Acid
10	Raw BR	22.59	3.99 × 10 ⁴	[M – H]–	187.099	187.0979	C ₉ H ₁₆ O ₄	Azelaic Acid
	<i>L. plantarum</i> FBR	22.50	2.77 × 10 ⁵	[M – H]–	187.099	187.0979		
	<i>L. fermentum</i> FBR	22.61	6.12 × 10 ⁴	[M – H]–	187.099	187.0978		
	<i>L. reuteri</i> FBR	22.50	6.25 × 10 ⁵	[M – H]–	187.099	187.0979		
11	Raw BR	ND	ND	[M – H]–	ND	ND	C ₉ H ₁₈ O ₃	9-Hydroxynonanoic acid
	<i>L. plantarum</i> FBR	23.49	3.10 × 10 ⁴	[M – H]–	173.119	173.1187		
	<i>L. fermentum</i> FBR	23.50	3.49 × 10 ³	[M – H]–	173.119	173.1188		
	<i>L. reuteri</i> FBR	23.48	1.39 × 10 ⁵	[M – H]–	173.119	173.1186		
12	Raw BR	39.06	2.45 × 10 ³	[M – H]–	313.24	313.2389	C ₁₈ H ₃₄ O ₄	Octadecanedioic acid
	<i>L. plantarum</i> FBR	39.07	3.52 × 10 ⁵	[M – H]–	313.24	313.2388		
	<i>L. fermentum</i> FBR	39.06	3.22 × 10 ⁵	[M – H]–	313.24	313.2389		
	<i>L. reuteri</i> FBR	39.06	3.83 × 10 ⁵	[M – H]–	313.24	313.2386		
13	Raw BR	ND	ND	[M – H]–	ND	ND	C ₁₈ H ₃₄ O ₅	Pinellic acid
	<i>L. plantarum</i> FBR	33.48	3.12 × 10 ⁵	[M – H]–	329.234	329.2337		
	<i>L. fermentum</i> FBR	33.48	3.54 × 10 ⁴	[M – H]–	329.234	329.2339		
	<i>L. reuteri</i> FBR	32.82	7.37 × 10 ⁶	[M – H]–	329.234	329.2331		

ND—not detected, BR—brown rice, and FBR—fermented brown rice.

3. Cell Viability Assay and Cellular Antioxidant Activity (CAA)

3.1. Cell Viability Assay

Cytotoxicity is regarded as an important step in determining the suitability and further applications of any food extract. Using the Ez cytox assay kit, the cytotoxic effect of *L. reuteri* FBR extracts at 0.3–10 mg/mL concentrations was investigated using Caco-2 cell lines. **Figure 4** depicts the cell viability results of the extract after 12 h of incubation. It was observed that cell viability was not much decreased after increasing the concentration up to 10 mg/mL. No significant differences were observed in cytotoxicity assay by using 0.3–10 mg/mL concentrations (**Figure 4**). The extract was observed to be non-toxic after 12 h assay as the extract still shows about 97 per cent of cell viability. Our results were found similar to the results presented by Yue et al. [21].

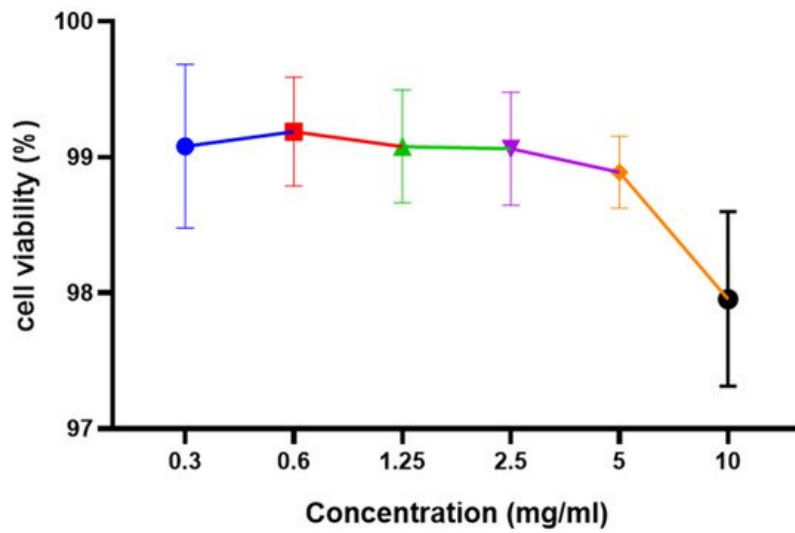


Figure 4. Effect of *L. reuteri* FBR extracts on viability of Caco-2 cells analyzed by Ez cytotoxic assay kit. Cells were treated with an increased concentration of *L. reuteri* FBR extracts for 12 h. Data are represented as means \pm standard deviations ($n = 3$).

3.2. Cellular Antioxidant Activity (CAA)

The effect of pretreatment of Caco-2 cells with *L. reuteri* fermented extract of brown rice on intracellular reactive oxygen species (ROS) was determined using a cell-based assay. The fluorescent probe DCFH-DA is used as an indicator of ROS and oxidative stress in our study. The nonionic and nonpolar DCFH-DA probe diffuses passively into cells before being hydrolyzed by intracellular esterases to form nonfluorescent 2',7'-dichlorofluorescein (DCFH). Later in the presence of ROS, DCFH that is trapped inside cells is oxidized into fluorescent 2',7'-dichlorofluorescein (DCF) [22]. When the cellular antioxidant defence system fails to compensate for ROS production, oxidative stress occurs. This reaction can be slowed down using bioactive compounds, preventing the generation of DCF. Following the uptake of antioxidant compounds can be accomplished on the cell membrane surface or within the cell [23]. We evaluated the effect of our *L. reuteri* FBR extract against oxidative stress in Caco-2 cells. In our study, ABAP was chosen as an intracellular oxidizing agent to simulate oxidative stress in cells. 600 $\mu\text{mol L}^{-1}$ ABAP was chosen as the optimal concentration to induce oxidation. As represented in **Figure 5A**, CAA values in *L. reuteri* FBR extract were observed to be 5.7 times higher than the raw BR sample at a concentration of 1mg/mL. Our results indicate that extracts reduced ROS levels at rest in a dose-dependent manner (**Figure 5B**); CAA values were increased with concentration (0.5 mg/mL to 5 mg/mL) from $49.50 \pm 1.67\%$ to $72.49 \pm 1.23\%$. The strength of inhibition strongly followed a curvilinear pattern as *L. reuteri* FBR extract concentrations increased. A similar effect was previously observed in the study of Grauzdytė et al., where they observed *Phyllanthus phillyreifolius* extracts in HEK-293 cells [24], and the study of Kellett et al. [25] in Caco-2 cells.

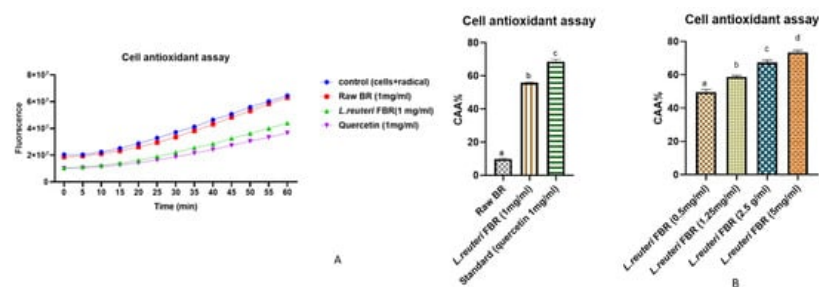


Figure 5. In Caco-2 cells, peroxyl radical-induced oxidation of DCFH to DCF and ROS inhibition by raw BR and *L. reuteri* FBR extract (**A,B**) showing the effect of dose-dependent inhibition of *L. reuteri* FBR extracts (0.5–5 mg/mL). Data were represented as means \pm standard deviations ($n = 3$) with one way ANOVA. The columns with different letters (a–d) show significant differences using Tukey's test at $p < 0.05$.

4. Conclusions

In our study, we discovered that *L. reuteri* FBR had higher antioxidant activity as well as a higher concentration of phenolics and flavonoids among all LABs used for the study. This shows the ability of *L. reuteri* as a promising fermentation strain to increase the bioavailability of cereals or grains in producing health-promoting functional materials. *L. reuteri* fermentation improves phenolic constituents and antioxidant activity of BR, improves food quality, and confers organoleptic characteristics. Furthermore, we discovered that *L. reuteri* FBR enhanced the production of essential amino

acids and fatty acids using untargeted metabolomics. The present study has provided information on bioactive compounds and antioxidant activities as well as the cellular antioxidant capacities of *L. reuteri* FBR. These data are required for the processing of the whole BR and its products for the pharmaceutical and food markets. As a result, new strategies and collaborations among industry, researchers, and relevant agencies are required to publicize whole grain consumption. Additionally, the current research is part of ongoing efforts to increase the added value of brown rice production and use in the prevention of human chronic diseases caused by oxidative stress. Moreover, these findings also make this sample a promising material for the development of health-promoting functional food. Whereas, it is necessary to perform more research into the mechanisms of different types of fermentation (solid and liquid-state) on single/pure phenolic compounds and antioxidant properties. Furthermore, in vivo models should be used to study the bioavailability and absorption of phenolic compounds in the gut.

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